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Source Test and Evaluation Report
Stoic Low-BTU Gasifier

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ENVIRONMENTAL ASSESSMENT:
SOURCE TEST AND EVALUATION REPORT--
STOIC LOW-BTU GASIFIER

by

M.R. Fuchs, M.A. McDevitt, D.S. Lewis, and C.E. Hudak
Radian Corporation
8501 Mo-Pac Blvd.
Austin, Texas 78766

Contract No. 68-02-3137

Project Officer

William J. Rhodes
U. S. Environmental Protection Agency
Air and Energy Engineering Research Laboratory
Research Triangle Park, NC 27711

AIR AND ENERGY ENGINEERING RESEARCH LABORATORY
OFFICE OF RESEARCH AND DEVELOPMENT
U.S. ENVIRONMENTAL PROTECTION AGENCY
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ABSTRACT

Under contract to the Environmental Protection Agency, Radian Corporation performed a Source Test and Evaluation of the Foster Wheeler/Stoic gasifier located at the University of Minnesota-Duluth Campus (UMD). The test, which ran from February 23 to March 6, was designed to collect data pertinent to an environmental assessment of the Foster Wheeler/Stoic two-stage gasification process, including evaluation of the emissions from the combustion of both the product gas and the by-product ESP oil. Results of the test are presented in this report.

Distribution of minor and trace elements in the product, by-product, and discharge streams were calculated from elemental mass flow rates. Evaluation of elemental behavior was also performed by comparing the concentration of an element in a stream to the concentration of the element in the ash mineral fraction of that stream.

Extractable organic material in product gas, by-product ESP oil, and discharge streams was characterized by total organic loading and by gas chromatographic analysis. Priority pollutants were identified and quantitated for major streams. The gas clean-up devices (ESP and cyclone) were evaluated for both efficiency and effects on the gas composition.

Evaluation of test results indicated that combustion of ESP oil and product gas destroyed priority organic pollutants to the extent that less than one percent of the original levels was detected. Also, most of the hydrocarbons (C1-C6) present in the combined product gas were destroyed by combustion.

Although federal New Source Performance Standards (NSPS) do not apply to the Foster Wheeler/Stoic gasifier boilers because of their size, if these standards were applied, SO₂ emissions would fall slightly above or slightly below NSPS levels, depending on the method of calculation. NO_x emissions would be marginally above NSPS, and emission levels of particulates would fall well below NSPS.

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SECTION 1

INTRODUCTION AND SUMMARY

Radian Corporation is under contract to the Environmental Protection Agency (EPA) to perform a comprehensive assessment of low- and medium-Btu coal gasification technology. The program includes the performance of Source Tests and Evaluations to gather data for the assessment of commercially-viable low- and medium-Btu coal gasification systems.

This Source Test and Evaluation Report (STER) presents the findings of a limited environmental assessment of the Foster Wheeler/Stoic two-stage gasifier. The Foster Wheeler/Stoic two-stage gasifier was developed by Stoic Combustion (Pty.) Limited of Johannesburg, South Africa and is marketed by Foster Wheeler Energy Corporation, with licensing rights in North and South America, Europe, Asia, and Australia.

The University of Minnesota, with funding from the Department of Energy's (DOE) "Gasifiers in Industry" Program, constructed and now operates a Foster Wheeler/Stoic two-stage gasifier on the University of Minnesota-Duluth (UMD) campus. The test program was conducted during February and March, 1981 as part of a joint field test effort between Oak Ridge National Laboratories (ORNL) and EPA. Personnel from the Lake Superior Basin Studies Center (LSBSC), a research arm of UMD, assisted during the test, contributing to both the ORNL and EPA programs.

1.1 Test Overview

The source test was designed to collect data pertinent to the environmental assessment of the Foster Wheeler/Stoic two-stage gasifier. Testing was conducted from February 23 to March 6, 1981 under "steady-state"

operating conditions. During the 12-day test period, approximately 280 metric tons of bituminous coal were gasified at an average rate (including standby periods) of 270 g/s (2140 lb/hr). Samples of major process and discharge streams, including gaseous discharge streams from boilers combusting the product gas and by-product oil removed from the product gas stream by an electrostatic precipitator (ESP), were characterized. The selection of streams for testing was made based upon the stated program objectives:

- Performance of an environmental assessment of the Foster Wheeler/Stoic two-stage gasification process including the gaseous discharge streams resulting from combustion of the product gas and by-product ESP oil.
- Calculation of material balances around the plant systems.
- Evaluation of the emissions from combustion of the product gas, as related to the product gas composition.
- Evaluation of the emissions from combustion of the ESP oil, as related to the ESP oil composition.
- Evaluation of the efficiency of product gas clean-up systems (ESP and cyclone) and their effect upon gas composition.

1.2 Results

This section summarizes the results obtained and evaluation made in this report. The areas include:

- process flow rates and material and component balances,
- elemental behavior,
- extractable organics,
- gas composition,
- efficiency of gas clean-up devices,
- water quality, and
- leaching studies.

1.2.1 Material and Component Balances

Material and component balances around the gasifier closed within 3 to 14 percent. Carbon and sulfur displayed the greatest difference between mass out and mass in. The overall material balance and flows normalized to coal flow per ton are presented in Table 1.

The gasifier had a cold gas efficiency* of approximately 80 percent during the test period. However, this value should not be taken as representative of this gasifier's typical performance. This calculated efficiency is not supported by a detailed energy balance since heat balance and efficiency calculations were not within the scope of work.

Material and component balances around the product gas boilers closed within 10 to 50 percent. Hydrogen, which included gravimetrically determined moisture in the combustion gases, exhibited the poorest balance, with an out/in ratio of 1.5. Sulfur also exhibited poor closure around the product gas boilers, with an out/in ratio of 1.3.

Material and component balances around the boiler firing ESP oils indicate closure within 4 to 10 percent for all species except sulfur, which showed an out/in ratio of 1.6.

1.2.2 Elemental Behavior

Distribution

Of the 65 elements for which analytical data allowed an evaluation of elemental behavior, 57 are found in the gasifier ash and 43 of these have 95

*Cold gas efficiency is a measure of the ability of a gasifier to convert potential combustion energy in coal to potential combustion energy in dry, tar-free product gas.

TABLE 1. OVERALL MATERIAL BALANCE OF FOSTER WHEELER/STOIC TWO-STAGE GASIFIER (UMD)

Inlet Streams	Mass Flow Rate* (g/s)	Flow Rate (per ton dry coal)	Flow Rate (per Mg dry coal)
Coal (dry basis)†	270	1.0 ton	1.0 Mg
Moisture in coal#	30	--	--
Ash pan water make-up	7	52 lb	24 kg
Blast air (with steam)	690	71,300 scf	1,870 Nm ³
Purge steam	22	163 lb	74 kg
TOTAL IN	1,019		
Outlet Streams			
Gasifier ash	44	0.162 tons	147 kg
Cyclone dust	0.2	1.48 lb	0.67 kg
ESP oil**	4	0.093 bbl	0.015 m ³
Combined product gas††	1,005	121,000 scf	3,180 Nm ³
TOTAL OUT	1,053.2		

*From Table 10.

†Ultimate analysis: C = 69.88%

H = 4.53%

N = 1.45%

S = 0.52%

Ash = 9.51%

O = 14.11 (by difference)

Btu/lb = 12,012 (2.7E7 J/kg)

#Moisture content taken as 10%.

**HHV of ESP oil is 18,930 Btu/lb (4.4E7 J/kg).

††HHV of combined product gas is 158 Btu/scf (6.3E6 J/Nm³).

percent or more of their mass discharged in the gasifier ash. Five elements (antimony, bromine, chlorine, sulfur, and tin) have significantly more than 25 percent of their total mass in the combined product gas stream and may be considered to exhibit volatile behavior.

While bismuth, lead, and zinc are discharged primarily in the gasifier ash, 30 to 45 percent of the total mass discharged is found in the cyclone dust, ESP oil, and product gas. Tin, mentioned above as having more than 25 percent of its discharged mass in the product gas stream, is actually distributed almost equally across the bottom ash, cyclone dust, and product gas.

Enrichment

Another approach to evaluation of elemental behavior in the gasification process is the use of elemental concentration quotients. A concentration quotient is defined here as the ratio of the concentration of an element in a specific stream to the concentration of that same element in the feed coal.

Elements enter the gasifier in the feed coal and are concentrated in the ash minerals as the carbonaceous materials and many of the volatile components react to form the gaseous products of gasification. If elemental enrichment is due to nothing more than the concentration of elements in the resulting ash mineral fraction, then the change in elemental concentration would be equal to the change in concentration of ash mineral fraction in a stream. However, due to various chemical or physical characteristics, a number of elements exhibit changes in concentration that are different from that of the ash mineral fraction (i.e., their concentration quotients are greater or smaller than the concentration quotients calculated for the ash mineral fraction). If their concentration quotient is greater than that for ash mineral fraction, they are considered to be enriched in the stream with respect to ash mineral fraction. Conversely, if their concentration quotient is smaller, they are considered to be depleted with respect to ash mineral fraction.

Most of the elements are discharged primarily in the gasifier bottom ash. Fourteen elements which were found in the gasifier bottom ash had not been found at detectable levels in the feed coal. These were:

Bismuth	Hafnium	Thallium
Dysprosium	Holmium	Thulium
Erbium	Lutetium	Tungsten
Europium	Tantalum	Ytterbium
Gadolinium	Terbium	

Eleven elements detected in the coal were found to be enriched with respect to ash mineral fraction. These were:

Arsenic	Fluorine	Manganese
Boron	Iron	Neodymium
Calcium	Lanthanum	Rubidium
Cesium	Magnesium	

Four others, cerium, niobium, potassium, and sodium, show concentration quotients only slightly higher than the concentration quotient for the ash mineral fraction, indicating neither enrichment nor depletion. The other elements found in the gasifier bottom ash were depleted with respect to the ash mineral fraction (see Table 2).

Ten elements were found in the cyclone dust which were not found at detectable levels in the coal. These were:

Bismuth	Gadolinium	Terbium
Dysprosium	Hafnium	Thallium
Erbium	Holmium	Tungsten
Europium		

Ten elements detected in the coal were found to be enriched in the cyclone dust with respect to ash mineral fraction. These were:

Antimony	Germanium	Mercury
Arsenic	Iodine	Tin
Fluorine	Lead	Zinc
Gallium		

TABLE 2. CONCENTRATION QUOTIENTS OF ELEMENTS AND ASH MINERALS IN GASIFIER ASH, CYCLONE DUST AND ESP OIL

Element	Concentration Quotient*			
	Gasifier Ash	Cyclone Dust	ESP Oil	
Aluminum	6	2	<0.01	
Antimony	1	95	1	
Arsenic	13	17	0.45	
Barium	6.9	0.7	0.01	
Beryllium	3	2.7	---	
Bismuth	0.5	20	0.2	
Boron	28	4	0.04	
Bromine	1.5	3.5	0.05	
Cadmium	5	3	---	
Calcium	15	3.6	<0.01	
Cerium	9	2.5	<0.01	
Cesium	27	2.3	---	
Chlorine	0.37	3.2	0.01	
Chromium	8	3.8	0.02	
Cobalt	7.5	2.7	0.4	
Copper	6.6	4.6	0.03	
Dysprosium	30	20	---	
Erbium	15	4	---	
Europium	5	5	---	
Fluorine	16	>143	0.6	
Gadolinium	20	10	---	
Gallium	4.6	56	0.01	
Germanium	3.5	95	0.1	
Gold	---	---	---	
Hafnium	10	10	---	
Holmium	20	5	---	
Iodine	1	10	0.15	
Iridium	---	---	---	
Iron	11	3.3	0.05	
Lanthanum	20	2.2	<0.01	
Lead	4	290	0.13	
Lithium	5.3	2.3	<0.01	
Lutetium	1.5	---	---	
Magnesium	16	3.2	<0.01	
Manganese	18	2	0.02	
Mercury	4.3	10.7	2.3	
Molybdenum	5.5	2.6	0.2	
Neodymium	21	2.2	---	
Nickel	6.7	3.5	0.2	
Niobium	9.5	2.5	---	
Osmium	---	---	---	
Palladium	1	1	<0.01	
Phosphorus	---	---	---	
Platinum	8.6	2.8	0.03	
Potassium	8	2.5	<0.01	
Praseodymium	---	---	---	
Rhenium	---	---	---	
Rhodium	15	2.2	<0.01	
Rubidium	---	---	---	
Ruthenium	3	2	---	
Samarium	2	1	<0.01	
Scandium	1.7	7.2	0.8	
Selenium	1	1	<0.01	
Silicon	0.4	2.5	0.25	
Silver	8.3	3.8	0.27	
Sodium	1	0.8	<0.01	
Strontium	0.53	1.9	0.46	
Sulfur	3	---	---	
Tantalum	0.5	2.5	---	
Tellurium	5	4	---	
Terbium	2	60	---	
Thallium	7.3	2	---	
Thorium	1	---	---	
Thulium	1.2	192	0.1	
Tin	4.1	1.8	<0.01	
Titanium	10	20	---	
Tungsten	6.3	2.3	---	
Uranium	6.8	2.5	0.27	
Vanadium	10	---	---	
Ytterbium	3.6	1.1	---	
Yttrium	2.9	121	0.14	
Zinc	6.4	1.8	<0.01	
Zirconium	8.2	2.8	0.02	
Ash Minerals	---	---	---	

*Concentration quotient = $\frac{\mu\text{g element/g stream}}{\mu\text{g element/g coal}}$ or $\frac{\mu\text{g ash minerals/g stream}}{\mu\text{g ash minerals/g coal}}$

**Element not detected in stream.

Thirteen elements were found at levels equal to or only slightly higher than the ash mineral fraction. These were:

Boron	Chromium	Nickel
Bromine	Copper	Potassium
Cadmium	Iron	Selenium
Calcium	Magnesium	Sodium
Chlorine		

The rest of the elements found in the cyclone dust were depleted in respect to ash mineral fraction (see Table 2).

While a number of elements appeared to be enriched with respect to the ash mineral fraction of the ESP oil (which was quite low), only mercury was present at a concentration (mass/mass) exceeding its concentration in the feed coal. The concentration of antimony in the ESP oil was essentially identical with its concentration in the feed coal, and the concentration of selenium was within 20 percent of its concentration in the feed coal. All other elements in the ESP oil were found at concentrations significantly lower than their concentrations in the feed coal.

1.2.3 Extractable Organics

The majority of the extractable organic materials leaving the gasification process are found in the ESP oil and product gas streams and are combusted in the boilers. The largest amount of priority pollutants leaving the gasifier were found in the ESP oil. However, combustion of the ESP oil was effective in destroying the priority pollutant load. Less than one percent of the priority pollutant load in the ESP oils was found in the emissions from combustion of the ESP oil. Though the priority pollutant loading in the product gas was lower than in the ESP oil, combustion was likewise an effective means of destruction. Less than one percent of the priority pollutants identified in the product gas was detected in the combustion emissions.

Few polynuclear aromatics (PNAs) were detected in the product or emissions streams. Notably absent were acenaphthalene, benzo(a)pyrene, and fluoranthene.

1.2.4 Gas Compositions

The majority of the hydrocarbons (up to C₆) present in the combined product gas were destroyed upon combustion in the product gas boiler. The mass flow rate for C₆ hydrocarbons appeared to be higher in the boiler emissions than in the product gas stream. However, the difference was small and, based upon the standard deviation in the analytical results, no absolute difference can be concluded.

There was no change in the gas composition for the top product gases as a result of the gas clean-up, except for the concentration of aerosolized components. Because of process variations at the time the bottom gases were analyzed, no conclusions were drawn regarding the effect of the cyclone on the composition of the bottom gas.

The majority of gas phase organics exits the gasification process in the top gas. The mass flow of C₁-C₆ hydrocarbons in the treated top gas is about two orders of magnitude higher than in the treated bottom gas.

1.2.5 Efficiency of the Gas Clean-up Devices

The collection efficiency of the ESP used to remove aerosolized oils and particulate matter from the top product gas was found to be approximately 98 percent for oils and 72-74 percent for particulate matter. The collection efficiency for the cyclone was not evaluated because of difficulties in determining a consistent gas flow at the sampling point.

1.2.6 Water Quality

Ash pan water losses occur because of evaporation and carryover with the gasifier ash. Ash pan water may enter the environment as residual drainage from the gasifier ash while it is stored prior to disposal. Water quality analyses were performed on the ash pan water to determine its pollution potential if it should be discharged directly into the environment.

Concentrations of barium ($1.5E3 \mu\text{g/L}$) and mercury ($4.2 \mu\text{g/L}$) in the ash pan water exceed permissible levels set in the National Interim Primary Drinking Water Standards ($1.0E3 \mu\text{g/L}$ Ba and $2 \mu\text{g/L}$ Hg). The presence of mercury, however, may be an artifact of a mercury spill in the ash pan during an early period of operation of the gasifier.

The pH of the ash pan water (12.5) is significantly higher than the range specified (6.5-8.5) in the National Secondary Drinking Water Standards (NSDWS). Total dissolved solids, at $1.8E6 \mu\text{g/L}$, exceeds the NSDWS level of $5.0E5 \mu\text{g/L}$.

The reader should remain aware that the comparisons made above are between process wastewater and drinking water standards, and that the impact of the concentrations observed for barium, mercury, pH, and TDS would be diminished by dilution of the discharge stream in receiving waters as well as by other factors specific to a given site.

1.2.7 Leachates

Elemental materials in gasifier ash or cyclone dust may enter the environment as a result of leaching by ground or run-off waters from a disposal site. Leaching studies provide a measure of the potential for release of contaminants to the environment.

According to the results of analyses performed on leachate from gasifier ash and cyclone dust, these streams can be considered as non-hazardous

solid wastes under RCRA criteria. None of the components found in the extracts appeared at levels greater than ten percent of their maximum allowable concentration.

SECTION 2

PLANT DESCRIPTION

The University of Minnesota, with funding from the Department of Energy's "Gasifiers in Industry" Program, constructed and now operates a Foster Wheeler/Stoic two-stage fixed bed gasifier on the University of Minnesota-Duluth (UMD) Campus. The University of Minnesota selected a Foster Wheeler/Stoic two-stage gasifier because it has a higher capacity than many other gasifiers, enabling one gasifier to supply sufficient product gas to satisfy the heating needs of the campus. In addition, by concentrating the by-product oil in the top gas stream, by-product oil recovery may be accomplished with an electrostatic precipitator (ESP) which is approximately half the size of one that would be required for a single-stage gasifier of the same capacity. The by-product oil can be collected, stored, and burned during peak heating requirements or during periods of very low energy consumption when operation of the gasifier is not economical.

The UMD campus heating plant operates three boilers having a total capacity of 45 Mg (100,000 pounds) of steam per hour. Two of the boilers have been retrofitted to fire either low-Btu product gas or fuel oil. The third boiler is capable of firing the by-product oils from the gasification process, as well as fuel oil or natural gas. The gasification facility is annexed to the existing heating plant on the UMD campus and, along with the coal delivery and solid waste disposal areas, is entirely enclosed.

2.1 Gasification Process

2.1.1 Gasification Process Description

The Foster Wheeler/Stoic two-stage gasifier was developed by Stoic Combustion (Pty.) Limited of Johannesburg, South Africa and is marketed by

Foster Wheeler Energy Corporation, with licensing rights in North and South America, Europe, Asia, and Australia. The gasifier is an atmospheric pressure, air-blown, fixed bed gasifier that produces a low-Btu product gas having a lower heating value of approximately $5.6 \text{E}6 \text{ J/Nm}^3$ (150 Btu/scf). The unit has a diameter of approximately 3 m (10 ft.). The added height, in relation to single-stage units, serves to increase gas residence time and improve volatiles distillation properties of the reactor. The design throughput of the gasifier is 2.7 Mg (3 tons) of coal per hour. At this feed rate, the gasifier provides sufficient product gas to produce about 20 Mg (44,000 pounds) of steam per hour. The unit is capable of operating at pressures of 7.5 kPa (30 inches of water), which increases the delivery pressure of the product gas and minimizes problems associated with combustion burner selection and control.

Sized coal is fed to the gasifier through the top of the unit and blast air, saturated with H_2O , enters through the bottom. Product gas is removed from the reactor in two locations. One stream, known as top product gas, is withdrawn through the top of the reactor. Another stream, known as bottom product gas, is removed from above the reaction zone in the lower area of the reactor.

There are three zones of activity in fixed-bed gasification processes: oxidation, reaction, and volatilization. The oxidation zone exists just above the ash bed, is only inches thick, and is the first zone in the reactor to contact the oxygen in the blast air. In the oxidation zone, a portion of the carbon in the coal is combusted, exhausting the oxygen available in the blast air, and producing heat for subsequent gasification reactions. Above the oxidation zone is the reaction zone where the reactor atmosphere becomes reductive. Carbon monoxide and hydrogen are produced primarily by the reaction of carbon in the coal with water provided by the saturated blast air. The volatilization zone is in the upper part of the gasifier. Here, the hot product gases meet the unreacted incoming coal, causing volatilization of coal hydrocarbons and evaporation of moisture.

In two-stage gasification, the removal of a portion of the product gases near the reaction zone serves to regulate the heat available in the volatilization zone in the upper part of the gasifier. As a result, the incoming coal is heated slowly, causing the volatile hydrocarbons to distill without cracking and with minimal polymerization. The majority of the volatile hydrocarbons are carried from the gasifier in the top gas stream, from which they can be condensed and collected as a by-product. The by-product oil is relatively viscous, but can be used as a liquid fuel. An additional feature of two-stage gasification is that collection and removal of the aerosolized oil is more efficient since it is borne in only approximately one-half of the product gas volume (the top gas).

The particulate concentration in the top product gas may be less than particulate concentrations in single-stage units due to the lower space velocity through the bed. Therefore, oils collected from the top product gas are relatively free of particulate matter.

The combustible components in the bottom gas are primarily carbon monoxide and hydrogen. The bottom product gas contains less oil and more particulate matter than the top product gas. Particulates are removed from the bottom gas by a refractory-lined hot cyclone. The collected particulates are removed from the cyclone by a lockhopper system. The cyclone dust is stored and shipped in drums for disposal.

2.1.2 UMD Facility Description

Figure 1 shows a schematic of the UMD gasification facility. The top product gas is passed through an electrostatic precipitator (ESP) where the aerosolized oils and any particulate matter are collected. The oil flows by gravity from the ESP into an underground heated storage tank where it is stored for future use. The ESP oil is similar to No. 6 fuel oil and has a heating value of about $4.3\text{E}7 \text{ J/kg}$ (18,700 Btu/lb). The cleaned top product gas has a lower heating value of about $6.2\text{E}6 \text{ J/Nm}^3$ (170 Btu/scf) and the cleaned bottom gas has a lower heating value of approximately $4.8\text{E}6 \text{ J/Nm}^3$ (130 Btu/scf).

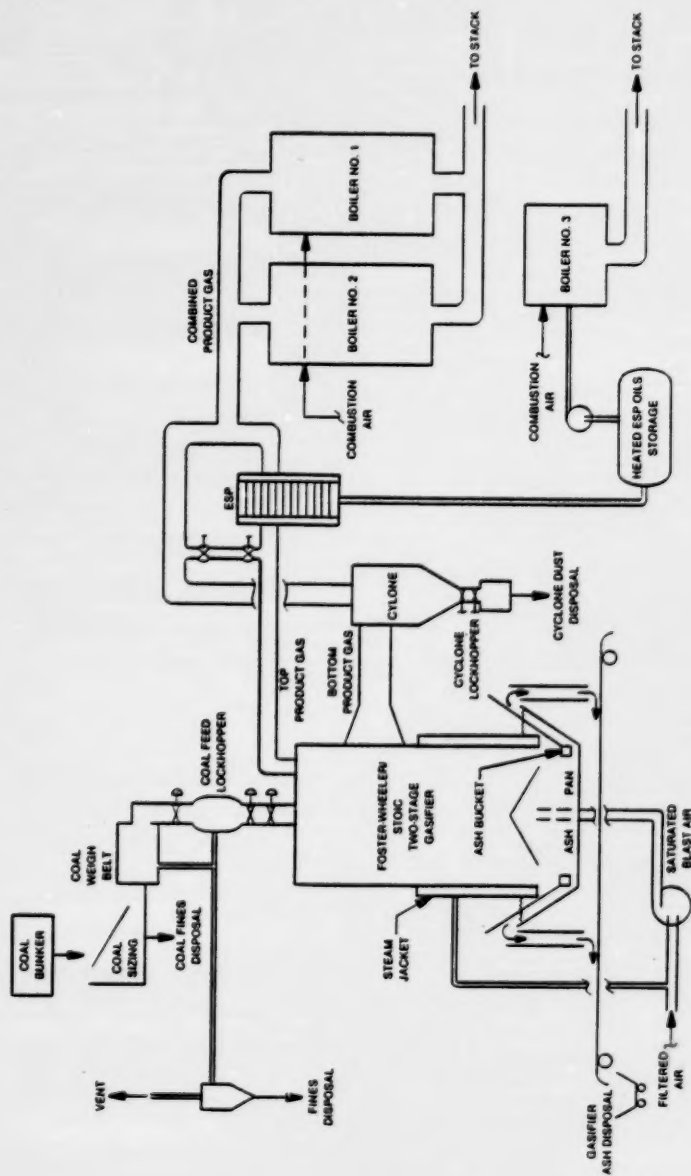


Figure 1. A schematic illustration of the Foster Wheeler/Stoic two-stage gasifier at the University of Minnesota-Duluth campus.

The bituminous feed coal used during the test program was mined from the Pinnacle Seam by Sun Coal Co., Milner, Colorado. Due primarily to the moisture content of the feed coal, the temperature of top product gas was lower than the engineering design value of 120°C (250°F). At this lower temperature, plant officials were concerned that water might condense from the top product gas stream and short out the ESP. A cross-over line was constructed between the bottom product gas line downstream of the cyclone and the top product gas line upstream of the ESP. This permitted the injection of hot, treated bottom gas into the raw top gas stream. Flow in the cross-over line was adjusted manually as necessary to maintain a top product gas temperature of approximately 95°C (200°F). The amount of bottom gas injected into the top gas line upstream of the ESP was not monitored and is, therefore, difficult to assess.

The cleaned top product gas (downstream of the ESP) and the remainder of the cleaned bottom gas (downstream of the cyclone) are combined to form the combined product gas. The temperature of the combined product gas is sufficiently high to maintain all condensable oils as vapors, preventing fouling in downstream gas lines or in the product gas burners. At the UMD facility, the entire production of combined product gas is burned in two boilers, producing steam that is used to heat the campus buildings.

The ash produced by the process is collected in a water-filled ash pan below the gasifier, which also forms a water seal for the gasifier. The depth of the water in the ash pan is a major factor establishing the allowable maximum operating pressure. The ash pan rotates constantly and forces ash into buckets situated at the bottom of the ash pan. The hydraulically operated buckets (two at 180° to each other) lift the ash from the ash pan and drop the ash onto a belt conveyor which transfers the ash to a storage location prior to disposal. Water drains through metal screens in the bottoms of the buckets. The buckets withdraw ash automatically, on a timed sequence.

The system requires make-up water for maintenance of the ash pan water level and for production of steam for the blast air. Water losses from the ash pan are due to evaporation and carryover with gasifier ash.

Coal sized from 1.3 cm to 5 cm (1/2 in. to 2 in.) is delivered to the plant in trucks and transported by conveyor into storage hoppers. From the storage hoppers, the coal is transported to a coal feed bunker which delivers the coal to a screening operation where fines are removed. The coal then passes to a weigh belt which feeds the coal into a lockhopper above the gasifier. The lockhopper reduces emissions to the work area by maintaining a seal on the gasifier during the coal feed operation. The lockhopper has two automatic sliding gate valves below the hopper and one automatic sliding gate valve above the hopper. The valve operation sequence is as follows: the valve above the lockhopper opens to fill the lockhopper with coal and then closes; the lowermost valve below the lockhopper opens followed by the other valve below the lockhopper, thus dropping the coal into the gasifier; both valves below the lockhopper then close and the cycle is repeated. The operating sequence of the two valves below the lockhopper keeps the weight of the coal from affecting the seal between the gasifier and lockhopper. Each dump of the lockhopper feeds about 45 kg (100 lb) of coal to the gasifier.

2.1.3 Plant Operation

The testing described in this source test and evaluation report was conducted February 23 through 27 and March 2 through 6, 1981. Gasifier operation during the week of February 23 was consistent, with an average coal feed rate of 310 g/s (1.25 tons/hr). On the morning of February 25, there was a minor plant disturbance caused by a malfunction in a valve on Boiler No. 1. The gasifier was turned down briefly to allow for repairs on the boiler valve, but was once again operating smoothly within a few hours after the repairs had been completed.

On March 2, the gasifier was shut down due to pressure drop inconsistencies across the bed, which were presumed to have resulted from an increased level of fines in the coal feedstock. Plant personnel made use of this down time to perform routine maintenance on the gasifier (primarily cleaning hardware systems). The gasifier resumed operation about midday on March 4, on coke. Sampling of the process began about midday on March 5 when the gasifier was again operating on coal, and all of the coke had passed through the oxidation zone. During this sampling period, the average coal feed rate was 330 g/s (1.3 tons/hr).

2.2 Heating Plant

2.2.1 Facility Description

The UMD campus heating plant has three boilers available for production of steam used to heat campus buildings. As part of the program to install the gasifier, two of the boilers (Boilers No. 1 and 2) were retrofitted to fire the low-Btu product gas. The retrofit reduced the maximum steam output of the two boilers from 11.3 to 10.9 Mg (25,000 to 24,000 pounds) of steam per hour. The two boilers, manufactured by Keeler, are capable of firing fuel oil in addition to the low-Btu product gas. The combustion emissions of the two boilers are combined and exhausted through a stack approximately 30 meters (100 feet) high.

Boiler No. 3, manufactured by Combustion Engineering, has a capacity of 22.7 Mg (50,000 pounds) of steam per hour and can be fired with ESP oil, fuel oil, or natural gas. The emissions of Boiler No. 3 are exhausted through a stack extending about 3 meters (10 feet) above the roof of the heating plant.

2.2.2 Plant Operation

During the period when combustion emissions from the product gas boilers were sampled (February 24 and 25), the boilers were operating smoothly at about 75% capacity. The combustion emissions of Boiler No. 3 burning ESP

oil were sampled March 2 and 3 when the boiler was operating at about 25% capacity. The combustion emissions from Boiler No. 3 were also sampled March 4 during firing of natural gas.

SECTION 3

SAMPLING

This section describes the joint sampling effort by Radian and Oak Ridge National Laboratories which was conducted February 23 through 27 and March 2 through 6, 1981. The sampling schedule and brief discussions of the technical approach of this project and the techniques used for collection of samples during the field effort are also presented. More detailed descriptions of the sampling techniques are presented in Appendix C.

3.1 EPA/ORNL Joint Effort

Oak Ridge National Laboratories (ORNL) has been conducting a multi-year program to assess environmental and health effects associated with the Foster Wheeler/Stoic gasifier at UMD. Personnel at the Lake Superior Basin Studies Center (LSBSC), a research arm of UMD, have performed environmental on-site sampling and analysis under contract to ORNL.

Discussion between EPA and ORNL prior to the planned test indicated that the scopes of both the ORNL and EPA programs were similar and could be performed more efficiently by conducting a joint field effort. ORNL possesses a full compliment of hardware necessary for sampling the diverse types of gas streams at the gasifier. ORNL agreed to contribute the use of the sampling hardware, which had been designed and fabricated by Radian under a previous ORNL contract. EPA agreed to fund some of the analyses of mutual interest as well as all technical support by Radian during the field effort. LSBSC personnel contributed to both the EPA and ORNL sampling tasks.

The joint effort was developed so that alternate days were used for either ORNL or EPA sampling objectives. In general, the personnel on-site jointly performed basically the same tasks for the sampling efforts during both the ORNL and EPA programs. However, personnel on the ORNL and EPA project teams collected solid and liquid samples separately for their respective programs. Also, on-site analyses (gas phase constituents by gas chromatography and wet chemical analyses) were performed only by personnel associated with the respective ORNL or EPA programs. EPA samples for quantitation of trace organics and trace elements in the two combustion emission streams were collected using the source assessment sampling system (SASS) designed by EPA for Level I assessment (1). ORNL collected samples from these streams using a different sampling system.

3.1.1 Schedule

Alternate days were designated for collection of samples for either the ORNL or EPA program. However, the scope of the alternate days' sampling efforts were similar. The sampling objectives were achieved by collecting samples from the following stream pairs on separate days:

- combined product gas and emissions from product gas boilers,
- cleaned top product gas and raw top product gas,
- cleaned bottom product gas and raw bottom product gas, and
- emissions from Boiler No. 3 firing ESP oil and ESP oil to Boiler No. 3.

The emissions from Boiler No. 3 firing natural gas were sampled on a separate day and were not paired with any other stream.

The schedule of sampling conducted by ORNL and EPA is presented in Table 3. In addition to the sampling indicated in Table 3, grab samples for

TABLE 3. EPA AND ORNL SAMPLING SCHEDULE FOR THE UMD GASIFICATION FACILITY

Stream	Feb. 24	Feb. 25	Feb. 26	Feb. 27	March 2	March 3	March 4	March 5	March 6
Combined Product Gas	E* + 0	E							
Emissions from Product Gas Boilers	0	E							
Cleaned Top Product Gas			0	E					
Raw Top Product Gas			0	E					
Cleaned Bottom Product Gas								E	0
Raw Bottom Product Gas								E	0
Emissions from Boiler No. 3 Firing ESP Oils					0	E			
ESP Oils to Boiler No. 3					E + 0	E	E		
Emissions from Boiler No. 3 Firing Natural Gas									
Coal	E + 0	E	E + 0	E				E	0
Gasifier Ash	E + 0	E	E + 0	E				E	0
Cyclone Dust	E + 0	E	E + 0	E				E	0
ESP Oils to Storage	E + 0	E	E + 0	E				E	0
Ash Pan Water	E + 0	E	E + 0	E				E	0

E = EPA Sampling

0 = ORNL Sampling

*Ammonia and cyanide only

gas phase analyses (fixed gas, sulfur species, and hydrocarbons) were taken by the EPA team concurrently with the ORNL gas sampling on most days.

3.2 Gasification - Gaseous Process Streams Sampling

Samples of five gaseous process streams were collected during the sampling and analysis effort. The combined product gas was sampled to characterize the fuel to Boilers No. 1 and No. 2. The top product gas and bottom product gas were sampled to evaluate the two clean-up devices (electrostatic precipitator and cyclone, respectively) and to determine if the clean-up devices had an effect upon gas composition. The gas streams entering and exiting the clean-up devices were sampled on the same day, but not concurrently. Following are descriptions of the locations and techniques used to collect samples from each stream.

3.2.1 Sampling Locations

Figure 2 is a schematic of the gasifier and combustion facilities showing the sampling locations for all sampled streams. The five gaseous process streams sampled were:

- combined product gas,
- raw top product gas,
- cleaned top product gas,
- raw bottom product gas, and
- cleaned bottom product gas.

Combined Product Gas

The combined product gas is transported to the product gas boilers in 24-inch pipe. The sampling location is in the mid-section of a 3 m vertical run of pipe. On either end of the vertical run of pipe are elbows, mounted perpendicular with respect to each other. One 4-inch gate valve is mounted about midway between the two elbows, perpendicular to the upstream elbow.

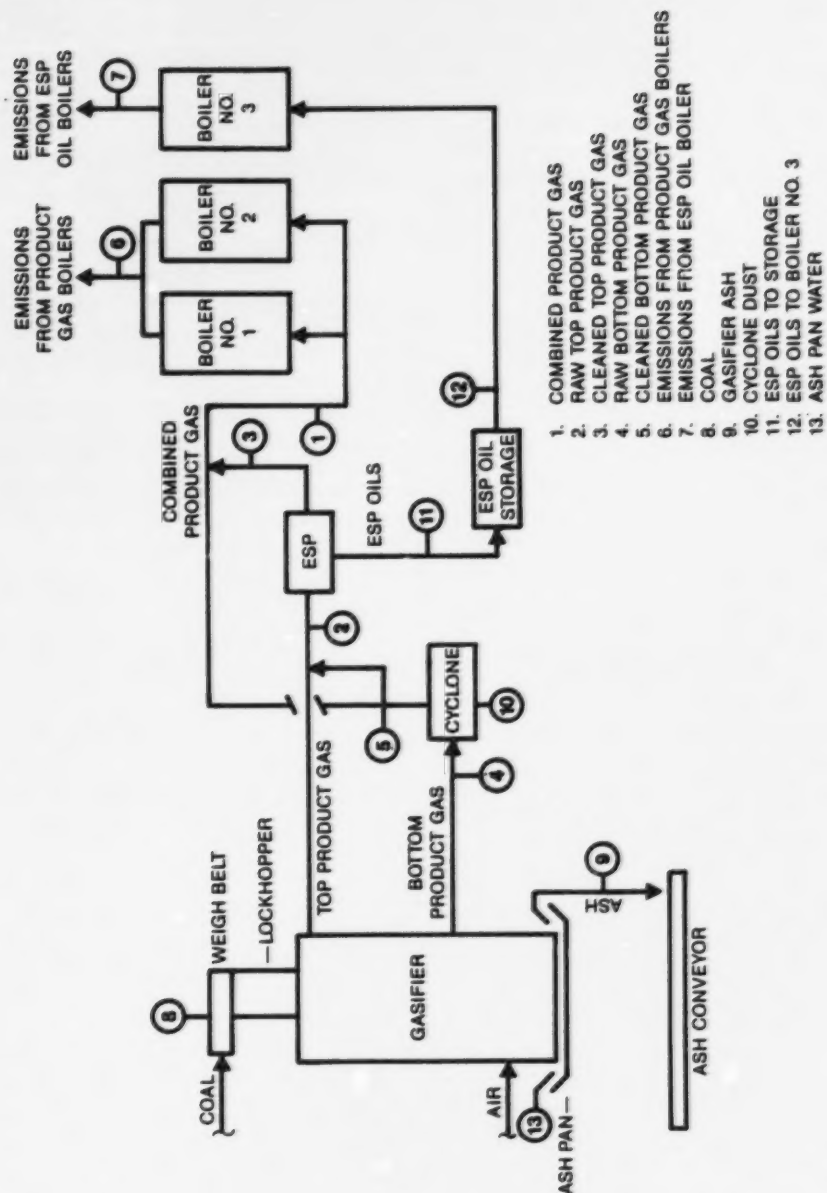


Figure 2. The location of sampling points at the Foster Wheeler/Stoic two-stage gasifier at UMD.

Collection of representative samples of aerosol material in the combined gas may have been hampered by the location of the sampling valve. Particles carried in a gas stream can be concentrated by inertia in the portion of the pipe away from the elbow, thus skewing the particle concentration distribution. Because of the single sampling port and its location, no definition of particle distribution in the stream could be made.

Grab samples for the analysis of major gases, hydrocarbons, and sulfur species by gas chromatography were collected through a 2-inch gate valve mounted near the 4-inch valve.

Raw Top Product Gas

As discussed in Section 2, a portion of the cleaned bottom product gas is combined with the raw top product gas to maintain the temperature of the top gas above the dewpoint. The sampling location for the raw top product gas is about 1 m downstream of the bottom gas injection point. Therefore, the composition of both the raw and cleaned top gas is diluted with cleaned bottom gas. The amount of bottom gas injected into the top gas could not be measured.

One 4-inch gate valve is mounted in a horizontal run of 14-inch pipe about 4 m from an elbow mounted at the gasifier. The injection of the bottom product gas into the raw top product gas line probably disrupts laminar flow patterns at the sampling location. However, because only a single traverse could be performed, no definition of the disruption of the flow could be made.

Grab samples for analysis of major gases, hydrocarbons, and sulfur species by gas chromatography were collected through a 2-inch gate valve mounted near the 4-inch valve.

Cleaned Top Product Gas

The cleaned top product gas exits the electrostatic precipitator (ESP) in a horizontal 16-inch pipe which elbows upward for about 4 m and then elbows once again. The sampling location is about midway between the two elbows. Only one 4-inch sampling valve is mounted in the line, and the valve is not aligned with either of the elbows. Representative samples of aerosols in the cleaned top product gas may not have been collected because of the sampling valve location.

Grab samples for analysis by gas chromatography were collected through a 2-inch gate valve mounted near the 4-inch valve.

Raw Bottom Product Gas

Bottom gas flows from the gasifier to the cyclone in a 2.4 m length of tapered, refractory-lined, rectangular duct. Two 4-inch gate valves are mounted on the duct (one on the side and one on the bottom) about midway between the gasifier and cyclone. Only the valve on the side of the duct could be used for sampling because of physical and process operational restrictions on the bottom valve.

Grab samples for analysis of major gases, hydrocarbons, and sulfur species by gas chromatography were collected through a tee in the gas transport line at the probe exit.

Cleaned Bottom Product Gas

Bottom gas exits the cyclone vertically in a 42-inch refractory-lined pipe. A butterfly flow control valve is located in this section of pipe. Two 4-inch gate valves are mounted horizontally to the pipe, 90° with respect to each other, approximately 30 cm upstream of the butterfly valve. While the butterfly valve does not mechanically or physically interfere with sampling, it probably disrupts flow patterns which may have negatively affected the collection of representative samples.

Only one of the 4-inch valves could be used for probe sampling purposes. The other valve, located at the end of a short horizontal section of pipe which contains a tee to the top product gas, was used to collect grab samples for analysis of major gases, hydrocarbons, and sulfur species by gas chromatography.

3.2.2 Sampling Techniques

Two-stage gasification produces gaseous streams having a wide variation in physical characteristics and requiring different devices and procedures for collection of samples. Typical characteristics of the five sampled process gas streams are presented in Table 4.

All process gas streams are pressurized and contain toxic, flammable gases. Samples from these gas streams must be collected without creating serious health or safety hazards to the personnel in the area. Devices for collection of samples from such streams must be pressure sealed using a packing gland which affixes to a sampling access valve in the process gas line. A sampling probe can then be safely and easily inserted and moved during the sampling process.

Process streams having a high aerosol (oil and/or particulate) loading require specialized sample collection devices and/or techniques. Significant concentrations of aerosolized oils quickly foul conventional particulate collection devices such as filters or cyclones. An ESP, designed and fabricated by Radian under a previous contract with ORNL, was used to remove the aerosolized oils from the process gas stream (2). This device permits the collection of samples that are essentially free of aerosolized oils from gas streams having an oil loading of above 1 g/Nm^3 .

Particulate matter in gas streams which do not contain aerosolized oils may be collected on glass fiber filters. If stream temperatures are below 480°C (900°F), the glass fiber filter may be placed directly in the stream. If stream temperatures are above 480°C , filtration must occur outside of the stream because of the temperature limitations on the glass

TABLE 4. TYPICAL PHYSICAL CHARACTERISTICS OF PROCESS GAS STREAMS

Stream	Temperature		Pressure		Velocity		Aerosolized Oils Loading	Particulate Loading
	°C	(°F)	kPa	(H ₂ O in.)	m/sec	(ft/sec)	g/Nm ³	g/Nm ³
Combined Product Gas	218	(425)	6.0	(+24)	7.0	(23)	2.0E4	1.9E5
Raw Top Product Gas	93	(199)	6.0	(+24)	10.0	(33)	7.3E6	1.2E5
Cleaned Top Product Gas	88	(190)	6.0	(+24)	8.2	(27)	1.5E5	3.1E4
Raw Bottom Product Gas	514	(957)	7.2	(+29)	3.0	(10)	ND	5.7E5
Cleaned Bottom Product Gas	418	(784)	7.2	(+29)	2.7	(8.9)	ND	6.7E5

aEb = a x 10^b.

ND - Not detected.

fiber filter. Out-of-stream filters are usually held at just below 480°C in a temperature-controlled oven to minimize condensation of organic vapors that might contaminate and/or clog the filter.

Three basic hardware systems were used to collect samples from the five process streams. The systems and the streams on which they were used during the sampling and analysis effort are listed in Tables 5 and 6, and illustrated in Appendix C. Table 5 presents the hardware systems used to collect aerosol material, both oils and particulates, from the sample gases. Vapor collection systems followed the aerosol removal systems to provide samples for quantitation of vaporous parameters. Table 6 presents the collection systems for vaporous parameters.

Grab samples for quantitation of major gases, hydrocarbons, and sulfur species were collected in cleaned, silanized glass bombs. The sample gas was first filtered using a Teflon® filter. Moisture was then removed from the sample gas using a Perma-Pure Drier® which uses an osmotic membrane that is permeable to water vapor. The glass bomb was purged a minimum of three sample volumes and then pressurized with sample gas using a diaphragm pump with Teflon®-coated internals contacting the sample gas. If condensation appeared within the sample bomb, the sample was discarded and another sample was collected.

3.3 Combustion - Gaseous Emissions Sampling

Two combustion emissions streams were sampled during the test program. Product gas was combusted in Boilers No. 1 and No. 2, and ESP oil and natural gas were combusted in Boiler No. 3. These streams were characterized to develop an emissions data base and to evaluate potential environmental and health effects of the emissions from the combustion of gasification products. Following are descriptions of the sampling locations and techniques used to collect samples from the streams.

TABLE 5. SAMPLING SYSTEMS FOR PRODUCT GAS STREAMS

Sampling System	Combined Product Gas		Raw Top Product Gas		Cleaned Top Product Gas		Raw Bottom Product Gas		Cleaned Bottom Product Gas	
	Gas	Product	Gas	Product	Gas	Product	Gas	Product	Gas	Product
Heat-Traced Probe										
• packing gland system										
• Type-S pitot										
• Type K thermocouple	X		X		X		X		X	
In-Stack Filtration										
• glass fiber thimble	X				X					
Electrostatic Precipitator (ESP) followed by glass fiber thimble										
				X						
Out-of-Stack Filtration										
• glass fiber thimble in heated oven							X		X	

TABLE 6. COLLECTION SYSTEMS FOR VAPOROUS PARAMETERS

Parameter	Collection System
Condensable Organics	Water-cooled condenser
Gas Phase Organics	XAD-2 resin
Trace Elements*	2-Smith-Greenburg impingers containing 500 mL of 0.2 M $(\text{NH}_4)_2\text{S}_2\text{O}_8$ + 0.2 M AgNO_3
NH_3	2-Smith-Greenburg impingers containing 200 mL of 5% H_2SO_4
HCN	2-Smith-Greenburg impingers containing 200 mL of 10% NaOH

*Gas phase mercury was collected using a gold amalgamation technique.

3.3.1 Sampling Locations

Combustion Emissions from Product Gas Boilers

The combustion emissions from Boilers No. 1 and No. 2 are combined in a rectangular duct which then is exhausted through a circular stack 30 m in height. The sampling location is about 7 m downstream of the point where these two streams converge. Five 3-inch ports are mounted equidistant along one side of the duct. Seven-point velocity traverses with a type-S pitot consistent with EPA Methods (3) were performed through each port; however, all other sampling was performed at a single point having the average flue gas velocity.

Combustion Emissions from Boiler No. 3

The combustion emissions from Boiler No. 3 are exhausted through a 1.4 m diameter metal duct which extends about 3 m above the roof of the building. Six-point velocity traverses were performed through two 4-inch pipe ports, mounted 90° with respect to each other, located about 1 m above the roof of the building. Velocity traverses and collection of samples were performed consistent with EPA Methods (3).

3.3.2 Sampling Techniques

The sampling methodology used to collect samples of the combustion emissions was considerably different from the methodology used to collect samples from the process streams. In gasification process streams, the presence of oils and pressurized gases having toxic and flammable components dictates the sampling protocol. In combustion emissions, trace organic and inorganic components are diluted with combustion air and require a methodology which concentrates these trace components.

Trace organic and trace element samples were collected from the combustion emission streams using the High Volume Source Assessment Sampling System (SASS) (1). The SASS draws sample gas at a rate of 0.1 Nm³/min.,

or 4 SCF/min. (about eight times greater than conventional sampling systems), allowing the collection of a larger mass of sample material in a shorter period of time. The sample nozzle size was selected to give an isokinetic sampling rate initially; however, the rate was not varied to respond to variations in stream flow. The sample gas is drawn through a heat-traced probe, and particulate is removed by a heated cyclone followed by a glass fiber filter. A water-cooled condenser followed by XAD-2 resin collects condensable and gaseous organics. Downstream of the condenser and resin an impinger train containing 0.2 M $(\text{NH}_4)_2 \text{S}_2\text{O}_8$ + 0.2 M AgNO_3 is used for collection of trace elements.

Sulfur oxides and particulate matter were quantitated in the combustion emissions using combined EPA Methods 5 and 6. The sample gas was drawn isokinetically through a heated probe and the particulate was collected on a heated glass fiber filter. Sulfur oxides were collected in 3% hydrogen peroxide impingers downstream of the filter. Nitrogen oxides were quantitated in the combustion emissions using EPA Method 7. Grab samples for analysis of major gases, hydrocarbons, and sulfur species were collected through ports at each location that was not occupied by other sampling hardware.

3.4 Solids Sampling

Three solids streams were sampled during the test period. The three streams are:

- coal,
- gasifier ash, and
- cyclone dust.

Coal is stored in a series of hoppers inside the heating plant and is conveyed from the hoppers to sizing screens located above the coal feed lockhopper. From the screens, the coal is conveyed to a weigh hopper. Coal samples were collected as the coal fell from the conveyor belt into the weigh hopper. The samples were stored in 1-L polyethylene bottles.

Two hydraulically operated buckets remove gasifier ash from the ash pan and dump the ash onto belt conveyors which transport it to a storage area. The ash dewaterers through screens which constitute the bottoms of the buckets. Gasifier ash samples were collected as the ash fell from the buckets onto the conveyor belt. Samples of gasifier ash were stored in 500-mL glass bottles having Teflon®-lined caps.

Dust removed from the bottom product gas by the cyclone is collected in a lockhopper. The dust is emptied periodically from the lockhopper and augered to drums and stored for disposal. Samples of the cyclone dust were collected from the auger and were stored in 500-mL glass jars with Teflon®-lined caps.

3.5 Liquids Sampling

The three liquid streams listed below were sampled during the testing period:

- ESP oil to storage,
- ESP oil to Boiler No. 3, and
- ash pan water.

Oil removed from the top product gas by the electrostatic precipitator (ESP) flows by gravity in heat-traced, insulated lines from the ESP to a heated, insulated intermediate storage tank. Samples of the ESP oil (500 mL) were collected through a valve in the transport line between the ESP and storage tank and were stored in glass bottles having Teflon®-lined caps.

During combustion of ESP oil in Boiler No. 3, ESP oil from an underground heated storage tank was pumped to the boiler. Samples of the ESP oil to Boiler No. 3 were collected through a valve mounted in the transfer line between the pump and boiler. Samples of the ESP oil to Boiler No. 3 (500 mL) were stored in glass bottles having Teflon®-lined caps.

Ash pan water samples were collected directly from the ash pan. Samples for analysis of extractable organics were collected and stored in 1-L glass bottles with Teflon®-lined caps; samples for trace elements analysis were collected and stored in 1-L polyethylene bottles. Samples for analysis of these water quality parameters performed on site were collected in 1-L glass bottles.

SECTION 4

ANALYTICAL METHODOLOGIES

Analytical methodologies are detailed in Appendix D and in the IERL-RTP Procedures Manual: Level I Environmental Assessment (1). The following section briefly describes the individual techniques and provides an overview of the analytical program. Figure 3 gives the analytical matrix for the types of samples collected and the analyses performed. Results of the analyses are presented in Appendix B.

4.1 Organic Screening Analyses

The organic screening analyses were performed according to the procedures described in Section 6 of Appendix D. The use of diethyl ether as an extraction solvent following the methylene chloride extraction was an addition to the cited methods. This step was performed to increase the extraction efficiency for highly polar organic compounds.

Organic screening analyses included the determination of total chromatographable organics (TCO) using a gas chromatographic technique and the gravimetric portion (GRAV) by a gravimetric method. The TCO determination quantitates organics with a boiling point range of 100°C to 300°C; the GRAV determination quantitates organic compounds having boiling points greater than 300°C. Previous work (4) indicates that the overlap of compounds determined by the TCO and GRAV analyses may not be significant.

Those samples with a combined TCO and GRAV value of greater than 15 mg were subjected to liquid chromatographic (LC) fractionation. These samples included gasifier ash, ESP oil to Boiler No. 3, and combined gas (condensable and gas phase organics). The LC fractionation was performed according

	Gas Analytes												Liquids Analytes				Solids Analytes					
	Major Components ^a	Hydrocarbons ^b	Sulfur Species ^c	BTX Organics ^d	Aerosol Organics	Condensable Organics	Gas Phase Organics	MI ₃	HCH	NO _x	SO ₂	Trace Elements ^e	Oil Loading	Particulate Loading	Water Quality Parameters ^f	Extractable Organics	Organics	Trace Elements ^g	Proximate/Ultime	Extractable Organics	Trace Elements ^h	NMA Extraction Procedure
Combined Product Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Combustion Emissions from Product Gas Boilers	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Cleaned Top Product Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Raw Top Product Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Cleaned Bottom Product Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Raw Bottom Product Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Combustion Emissions from Boiler No. 3 Firing ESP Oil	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Combustion Emissions from Boiler No. 3 Firing Natural Gas	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
ESP Oils to Boiler No. 3	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Coal	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Gasifier Ash	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Cyclone Dust	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
ESP Oils	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	
Ash Pan Water	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	

^aMajor components - CO, CO₂, H₂, O₂, N₂, CH₄
^bHydrocarbons - CH₄, C₂H₆, C₃H₈, C₄, C₅, C₆
^cSulfur species - H₂S, COS, CS₂, SO₂, methyl mercaptan, ethyl mercaptan, thiophene
^dHTX - benzene, toluene, xylene
^eTrace elements - SEM, ICPMS, AAS
^fWater quality parameters - PO₄, Cl⁻, F⁻, S²⁻, SO₃²⁻, CN⁻, SCN⁻, SO₄²⁻, NH₃, COD, BOD, TOC, TSS, TDS, pH, conductivity, phenol

Figure 3. An analytical matrix for samples collected at Foster Wheeler/
Stoic two-stage gasifier (UMD).

to the procedure given in Section 6.8 of Appendix D. TCO and GRAV analyses were performed on each of the seven fractions of the samples and on an analytical blank.

4.2 Gas Chromatographic Analyses

4.2.1 Process Gases Analyses

Samples of the combined gas, treated and raw top product gas, treated and raw bottom product gas, and the combustion emissions from the product gas boilers and Boiler No. 3 were characterized on site for major components, hydrocarbons, and sulfur species by gas chromatography. In addition, samples were collected for off-site analysis for benzene, toluene, and xylene (BTX). Analyses were performed according to the procedures in Sections 5 and 6.4 of Appendix D.

4.2.2 Gas Chromatography - Hall Detector

Composited sample extracts were prepared from the gasifier ash, cyclone dust, ash pan water, and ESP oil to Boiler No. 3. These, and sample extracts of organic vapors and condensable organics in the combined product gas as well as combustion emissions from the product gas boilers and Boiler No. 3 firing ESP oil, were screened by gas chromatography. The analyses were performed using a gas chromatograph equipped with a Hall 700A electrolytic conductivity detector (HECD). The HECD may be operated in the nitrogen-, halogen-, or sulfur-specific mode. Using the HECD in each of the three modes provides data analogous to TCO data for nitrogen-, halogen-, and sulfur-containing organic compounds. A screening analysis for these compound classes was conducted because many organic compounds of health or ecological significance contain these heteroatoms.

4.2.3 Gas Chromatography - Electron Capture Detector

The sample extracts listed in 4.2.2 were screened for polychlorinated biphenyls (PCBs) using a gas chromatograph equipped with an electron capture

detector (ECD). This method provides quantitation and identification of various PCBs present in the extract.

4.2.4 Gas Chromatography - Mass Spectrometry

Gas chromatography - mass spectrometric (GC-MS) analyses of the composited extracts listed in 4.2.2 were performed to provide identification and quantitation of specific organic compounds. The analyses used a computer-assisted GC-MS system for the extractable EPA Priority Pollutants. Priority pollutant analyses were performed to satisfy the data needs of EPA's Effluent Guideline Division. It additionally provided a rapid and cost-effective means of identifying and quantitating other compounds such as polynuclear aromatics (PNAs).

In addition to priority pollutant compounds, the major non-priority pollutant peaks in each sample were quantitated relative to d₁₀-anthracene. This method provided a more complete characterization of organic compounds in the sample.

Finally, the total unidentified chromatographable mass in each extract was quantitated relative to d₁₀-anthracene.

4.3 Aerosol Organics and Particulate Loading

Aerosol organics and particulate loadings were determined by gravimetric analysis of filter catches. Particulate is defined as the material remaining following extraction with methylene chloride and diethyl ether. The aerosol loading is the sum of the aerosol (extractable) organic loading and the particulate loading.

4.4 Trace Elements Analysis

Analyses were performed for 17 elements by inductively coupled argon plasma emission spectrometry (ICPES) and 4 elements by atomic absorption spectrophotometry (AAS). In addition, the samples, except leachates, were

subcontracted to Commercial Testing and Engineering Co., Chicago, Illinois, for spark source mass spectral (SSMS) analysis for 75 elements. A list of the elements and methods used for analyses are given in Table 7. The following samples were analyzed:

- coal,
- gasifier ash,
- cyclone dust,
- ESP tar to Boiler No. 3,
- ash pan water,
- trace metals impinger catches,
- XAD-2 resin extracts (SSMS only),
- RCRA leachate of gasifier ash, and
- RCRA leachate of cyclone dust.

4.5 Water Quality and Ionic Species Analysis

Selected water quality parameters in ash pan water samples and ammonia and cyanide in the impinger catches were analyzed on site. More stable parameters were analyzed in preserved samples at Radian's Austin laboratory. Table 8 lists the water quality parameters analyzed and preservation techniques. Preservation procedures used were recommended by EPA in Methods for Chemical Analysis for Waters and Wastes (5).

4.6 Oxides of Nitrogen and Sulfur

Combustion emissions from the product gas boilers and Boiler No. 3 firing both ESP oil and natural gas were analyzed for oxides of nitrogen (NO_x) and sulfur (SO_x). Nitrogen oxide samples were collected in a dilute sulfuric acid-hydrogen peroxide solution and analyzed spectrophotometrically. Sulfur oxide samples were collected in hydrogen peroxide solutions and were analyzed titrimetrically.

TABLE 7. ANALYTICAL METHODS USED FOR TRACE ELEMENT ANALYSIS

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	X	X		Neodymium	X		
Antimony	X		X	Nickel	X	X	
Arsenic	X		X	Niobium	X		
Barium	X	X		Osmium	X		
Beryllium	X			Palladium	X		
Bismuth	X			Phosphorus	X		
Boron	X	X		Platinum	X		
Bromine	X			Potassium	X	X	
Cadmium	X			Praseodymium	X		
Calcium	X	X		Rhenium	X		
Cerium	X			Rhodium	X		
Cesium	X			Rubidium	X		
Chlorine	X			Ruthenium	X		
Chromium	X	X		Samarium	X		
Cobalt	X	X		Scandium	X		
Copper	X	X		Selenium	X		X
Dysprosium	X			Silicon	X		
Erbium	X			Silver	X		
Europium	X			Sodium	X	X	
Fluorine	X			Strontium	X		
Gadolinium	X			Sulfur	X		
Gallium	X			Tantalum	X		
Germanium	X			Tellurium	X		
Gold	X			Terbium	X		
Hafnium	X			Thallium	X		
Holmium	X			Thorium	X		
Iodine	X			Thulium	X		
Iridium	X			Tin	X		
Iron	X	X		Titanium	X	X	
Lanthanum	X			Tungsten	X		
Lead	X			Uranium	X		
Lithium	X			Vanadium	X	X	
Lutetium	X			Ytterbium	X		
Magnesium	X	X		Yttrium	X		
Manganese	X	X		Zinc	X	X	
Mercury	X		X	Zirconium	X		
Molybdenum	X	X					

TABLE 8. WATER QUALITY PARAMETERS & PRESERVATION TECHNIQUES

Parameter	Location	Method	Preservation
pH	OS*	Corning Model 125 pH meter	None
Conductivity	OS	YSI Model 33 conductivity meter	None
Phosphate	Lab†	Colorimetric	H ₂ SO ₄ , pH <2, cool
Chloride	Lab	Colorimetric	Cool
Fluoride	Lab	Specific Ion Electrode	Cool
Sulfate	Lab	Turbidimetric	Cool
Sulfide	OS	Titrimetric	Zinc acetate, pH >12
Sulfite	OS	Titrimetric	Zinc acetate, pH >12
Nitrate	Lab	Colorimetric	Cool
Cyanide	OS	Distillation/colorimetric	S ²⁻ removed, pH >12, cool
Thiocyanate	OS	Extraction, colorimetric	S ²⁻ removed, pH >12, cool
Ammonia	OS	Distillation, titrimetric	H ₂ SO ₄ , cool
COD	OS	Reflux, titrimetric	pH <2, cool
BOD	Lab	Digestion, titrimetric	Cool
TOC	Lab	Beckman 915B TOC analyzer	pH <2, cool
TSS	Lab	Gravimetric	Cool
TDS	Lab	Gravimetric	Cool
Phenol	Lab	Distillation, colorimetric	H ₂ PO ₄ , pH <4, cool
Trace Metals	Lab	SSMS, ICPES, AAS	HNO ₃ , pH <2

*on site

†Radian's Austin laboratory

4.7 Leaching Studies

Composited samples of gasifier ash and cyclone dust were leached using Resource Conservation and Recovery Act Extraction Procedure (RCRA-EP) protocol (6). Elemental analyses were performed as described in Section 4.4.

4.8 Proximate/Ultimate Analysis

Samples of coal, gasifier ash, cyclone dust, and ESP oil were subcontracted to Commercial Testing and Engineering Co., Chicago, Illinois, for proximate/ultimate determinations in accordance with ASTM methods (7).

4.9 Analytical Quality Assurance

Certified gas standards and standard solutions were analyzed on site during the sampling effort to provide calibration data and quality assurance.

Samples that were not analyzed on site were preserved and returned for laboratory analysis within preservation time limits. All samples were logged on site, and logged entries were cross-verified upon receipt in the laboratory. The on-site log is presented in Appendix A.

Solvent blanks, impinger solution blanks, charcoal tube blanks, and blank XAD-2 resins for trace elements were analyzed in the laboratory and included in the subcontracted analyses. Data for blank XAD-2 resins are presented in Appendices A and B. Blanks were obtained from LC column fractionation for all eluents used. The same solvent lots were used throughout the organic analysis, and blanks were analyzed for TCO and GRAV.

SECTION 5

RESULTS

Summarized results of the evaluation of major process and discharge streams are discussed in this section. The discharge and process streams addressed are as follows:

- discharge streams
 - emissions from the product gas boilers
 - emissions from the ESP oil boiler (Boiler No. 3)
 - gasifier ash
 - cyclone dust
 - ash pan water
- process streams
 - combined product gas
 - ESP oil to Boiler No. 3

Detailed analytical results are presented in Appendix B. The results are discussed in this section in the following order:

- Process Flow Rates and Material and Component Balance
- Elemental Mass Flow Rates
- Elemental Behavior
- Extractable Organics
- Gas Composition
- Efficiency of Gas Cleanup Devices
- Water Quality
- RCRA Leaching

5.1 Process Flow Rates and Material and Component Balances

Material and component balances have been established for gasification and combustion processes at the UMD facility. The combustion processes include combustion of product gas and combustion of ESP oil. Figure 4 is a schematic of the UMD facility; the three process areas for which balances have been determined are outlined. The balance around Boiler No. 3 is independent of balances around other process areas. The balances around the gasifier and the product gas boilers support each other because the combined product gas is a major outlet stream of the gasifier (~90 percent of the mass of the outlet streams of the gasifier) and is the fuel component to the product gas boilers (~17 percent of the mass of inlet streams to the boilers).

5.1.1 Gasification Process

Material and Component Balances

Figure 5 is a schematic of the gasification process and indicates the inlet and outlet streams that were considered in establishing the material and component balances around the gasifier. The material and component balances during the period 12:01 a.m. February 23, 1981 to 12:00 midnight February 27, 1981 are presented in Table 9. Closure of the balance is evaluated by the ratio of total mass flow of a component leaving the process to total mass flow of a component entering the process, with closure defined as a ratio of 1.0. The limits of the ratio are a measure of the degree of uncertainty associated with the ratio. The material and component balances close within 3 to 14 percent, with sulfur and carbon displaying the greatest difference between mass out and mass in. Less sulfur and carbon was accounted for in the outlet streams than in the inlet streams.

Flow rates for a number of the streams were determined from plant operation log data or were measured by UMD plant personnel. For example, the flow rate for dry coal was determined by multiplying the number of coal

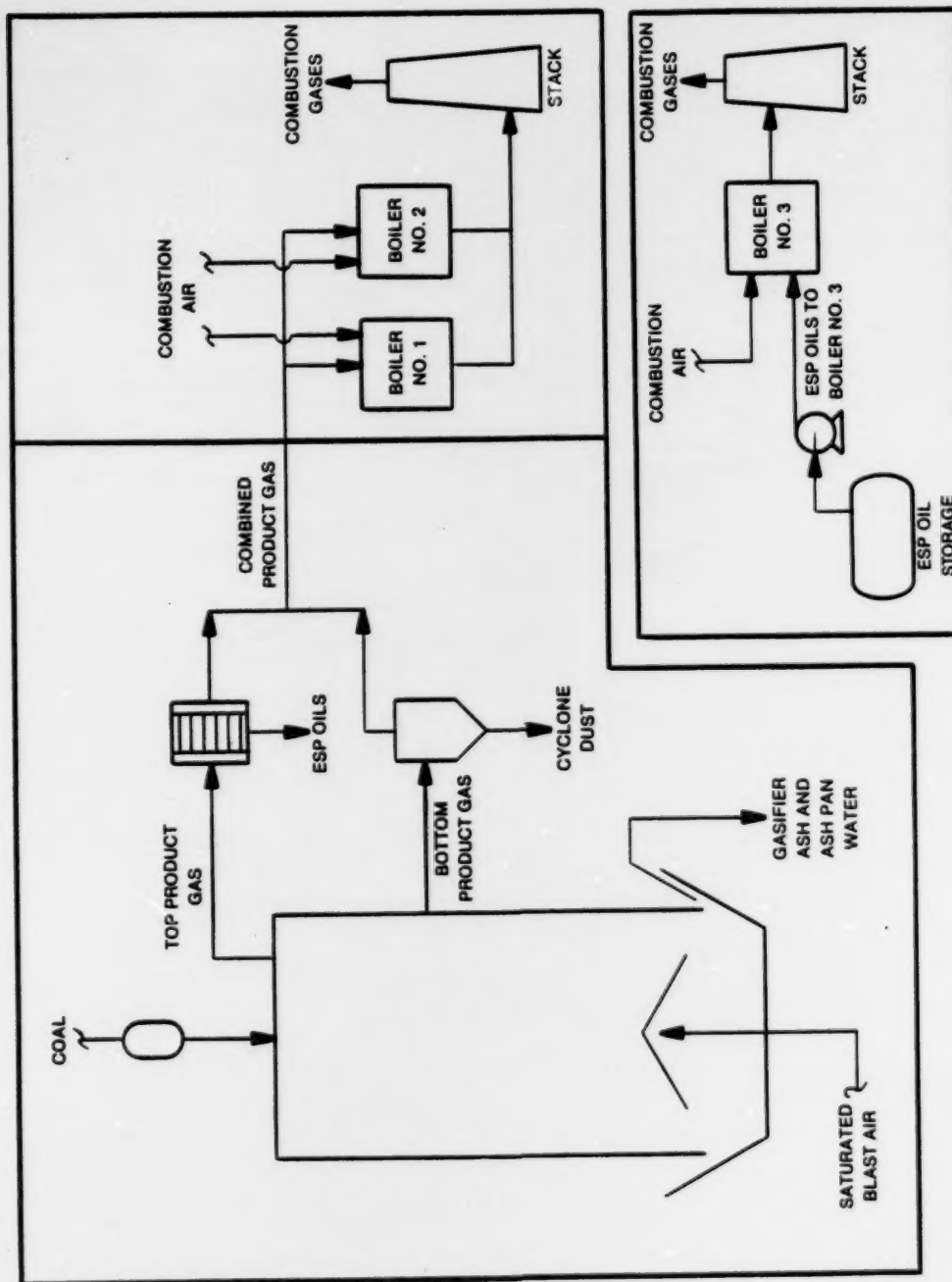


Figure 4. Schematic of UMD gasification and heating facility indicating balance boundaries.

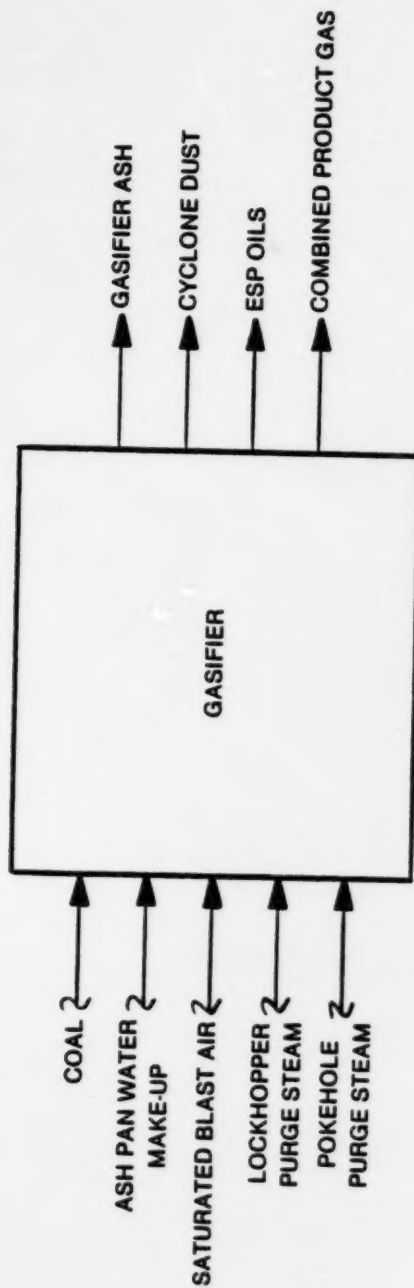


Figure 5. Schematic of mass balance streams for the Foster Wheeler/Stoic two-stage gasifier at UMD.

TABLE 9. MATERIAL AND COMPONENT BALANCES OF THE FOSTER WHEELER/STOIC TWO-STAGE GASIFIER - UMD

	Total (g/s)		Ash (g/s)		Carbon (g/s)		Hydrogen (g/s)		Oxygen + Argon (g/s)		Nitrogen (g/s)		Sulfur (g/s)	
	Rate	$\pm \sigma$	Rate	$\pm \sigma$	Rate	$\pm \sigma$	Rate	$\pm \sigma$	Rate	$\pm \sigma$	Rate	$\pm \sigma$	Rate	$\pm \sigma$
<u>Inlet Streams</u>														
Coal (dry)	270±27†		26±6		190±19		12±1.2		38±4		3.9±0.4		1.4±0.2	
Molature in Coal	30±3#						3±0.3		27±3					
Ash Pan Water-Makeup	74±1						1±0.5		6±3					
Blast Air	620±120†						8±0.8		145±28		475±95			
Blast Air Steam	70±7†						2±0.5		62±6					
Lockhopper Purge Stream	16±4††						1±0.3		14±4					
Pokehole Purge Stream	6±2††						1±0.3		5±1					
Σ In	1019±123		26±6		190±19		27±1.7		297±30		478±95		1.4±0.2	
<u>Discharge Streams</u>														
Gasifier Ash (dry)	35±4††		27±4		7.2±2.3		0.06±0.02		0.37±0.51		0.07±0.03		0.1±0.02	
Water in Gasifier Ash	9±0.9#						1±0.1		8±0.8					
Cyclone Dust	0.2±0.1††		0.05±0.03		0.1±0.07		0.4±0.10		0.1±0.03		0.02±0.01		1.1±0.2	
ESP Oil	4±1				3.4±0.9		17±4		230±48		523±109			
Combined Product Gas (dry)	930±190#				160±34		8±1.6		67±13					
Water in Combined Product Gas	75±15#													
Σ Out	1053±191		27±4		170±34		26±4		305±50		523±109		1.2±0.2	
Σ Out/Σ In	1.03		1.04		0.89		0.96		1.03		1.09		0.86	

†Flow rate determined from plant process data.

††Flow rate measured by UMD plant personnel.

#Flow rate measured by Radian.

*Error limit based on judgement value.

Note: All flow rates and error limits not footnoted have been determined empirically.

All error limits are one standard deviation (σ).

lockhopper dumps by the scale calibration factor (weight/dump) and correcting the resultant value for the moisture content of the coal (which was determined by proximate/ultimate analysis). Other flow rates were determined from measurements made by Radian personnel. For example, the flow rate of the combined product gas was calculated from velocity measurements and molecular weight determinations made by Radian personnel.

Confidence limits in any stream indicated in the material balance table are the result of calculations which take into consideration analytical and stream flow rate variability. The confidence limits were calculated from

$$\sigma_c = \sqrt{[\sigma_m(c)]^2 + [\sigma_f(c)]^2}$$

where:

Confidence limit = c = standard deviation for flow rate of a component "c"

$m(c)$ = analytical standard deviation for component "c"

$f(c)$ = flow rate standard deviation for stream containing component "c"

The confidence limits for the total process mass flow rate of each component were calculated by taking the square root of the sum of the squares of the standard deviations of the component mass flow in each stream.

Gasifier Efficiency

Based on the data taken during the test period, the UMD gasifier had cold gas efficiency of approximately 80 percent. Cold gas efficiency is a measure of a gasifier's ability to convert potential combustion energy in coal to potential combustion energy of the dry, tar-free product gas (higher heating value basis). The calculated cold gas efficiency value ($\approx 80\%$) should not be taken as representative of this gasifier's typical performance. This value is not supported by a detailed energy balance since heat balance and efficiency calculations were not within the scope of this environmental characterization effort.

5.1.2 Combustion Processes

Material and component balances were established around the two combustion processes at the UMD heating plant. Boilers No. 1 and No. 2 fire the combined product gas and exhaust the flue gases through a common stack. Boiler No. 3 fires the ESP oil; test data were also gathered during firing of natural gas in Boiler No. 3. Figures 6 and 7 present schematics of the inlet and outlet streams of the two combustion processes.

Product Gas Boilers

Boilers No. 1 and No. 2 fire the combined product gas and are termed the "product gas boilers". Material and component balances have been determined for the product gas boilers and are presented in Table 10. The flow rate of the combustion air was determined by closing the nitrogen balance, i.e., the total nitrogen in the inlet streams was set equal to the total nitrogen in the outlet streams by adjusting the combustion air flow rate. All other flow rates were determined experimentally. Because of equipment limitations, sampling on inlet and outlet streams could not be performed concurrently.

Hydrogen exhibits the poorest balance with a closure of 1.5. Sulfur also exhibits poor closure, with one-third more sulfur found in the emissions from the boilers than was found in the combined product gas. Data for oxides of nitrogen are also presented and correlate reasonably well with the concentration of ammonia and hydrogen cyanide in the product gas streams (see Section 5.5). Based on the limitations that necessarily must be placed on the data as a result of sampling inlet and outlet streams at different times, the calculated balances appear to be reasonable.

Boiler No. 3

Material and component balances were determined around Boiler No. 3 during combustion of ESP oils and during combustion of natural gas. Emissions from the combustion of natural gas were measured to provide a basis

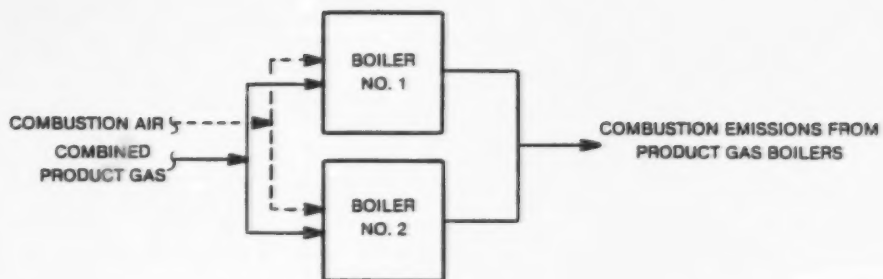


Figure 6. Schematic of mass balance streams for product gas boilers at UMD facility.

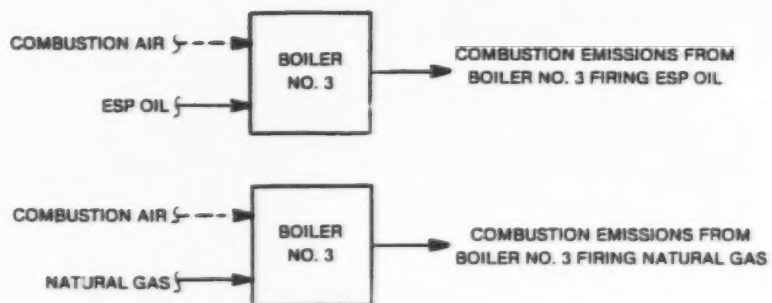


Figure 7. Schematic of mass balance streams for Boiler No. 3 at UMD facility.

TABLE 10. MATERIAL AND COMPONENT BALANCES OF THE PRODUCT GAS BOILERS - UMD

	Total (g/s)		Carbon (g/s)		Hydrogen (g/s)		Oxygen + Argon (g/s)		Nitrogen (g/s)		Sulfur (g/s)		NH ₃ , HCN (g/s)		NO _x (as NO ₂) (g/s)	
	Rate ± 0		Rate ± 0		Rate ± 0		Rate ± 0		Rate ± 0		Rate ± 0		Rate ± 0		Rate ± 0	
Inlet Stream																
Combined Product Gas (dry)	910±190		160±35		17±4		230±48		52±109		1.1±0.2		0.80±0.16		NA	
Water In Combined Product Gas	75±15				6±1.6		6±1.3									
Combustion Air In	4800±1200		160±34		4±1		115±280		3600±920		1.1±0.2		0.80±0.16		NA	
	5805±1215				29±4		1449±284		4200±926							
Discharge Stream																
Product Gas Boilers	5592±560		190±21		0.1		1200±132		4200±460		1.5±0.2		NA**		0.72±0.08	
Combustion Emissions	340±24				30±4		300±23									
Water In Combustion Emissions																
Out	5932±560		190±21		30±4		1500±136		4200±460		1.5±0.2		NA		0.72±0.08	
Out/In	1.02±0.23		1.19		1.31		1.03		1.00		1.36					

*Combustion air flow rate was determined by closing nitrogen balance.

**Flow rate measured by Radian.

***Error limit based on judgement value.

****NA - not applicable.

Note: All flow rates and error limits not footnoted have been determined empirically.

All error limits are one standard deviation (σ).

for comparison with the emissions during combustion of ESP oils. The material and component balances are presented in Table 11. The flow rate of ESP oil was determined from plant process data. The flow rate of the combustion air was determined by closing the nitrogen balance, i.e., by adjusting the value for combustion air flow rate to make the total nitrogen in the inlet streams equal to the total nitrogen in the outlet streams. All other flow rates were determined by Radian personnel. All components except sulfur exhibit closure; over one and one-half times more sulfur was measured in the flue gas than in the ESP oil fed to Boiler No. 3. Sulfur also exhibits the greater uncertainty associated with the determination of closure.

5.2 Elemental Mass Flow Rates

The mass flow rates of trace elements in feed streams and discharge streams of the gasifier, product gas boilers, and the ESP oil boiler were determined from the elemental concentration and flow rate of each stream. Elemental concentrations are presented in Tables B-29 through B-36 (Appendix B). The elemental mass flow rates of gasification streams are presented in Table 12; the elemental mass flow rates of product gas boilers streams and streams of Boiler No. 3 during combustion of ESP oils are presented in Table 13.

5.3 Elemental Behavior

5.3.1 Predictive Value of Elemental Behavior Information

Elements enter the gasification process with the coal and other feed streams, including water, and leave in the various product, by-product, and discharge streams. The distribution of elements in the discharge streams can provide a basis for predicting how and where elements of concern might enter the environment. Such predictions may be useful to designers of gas clean-up and pollution control systems.

Several discharge streams from the two-stage gasification process are vehicles for elemental emissions. The gasifier ash carries the greatest

TABLE 11. MATERIAL AND COMPONENT BALANCES OF BOILER NO. 3 FIRING ESP OIL - UMD

	Total (g/s)		Carbon (g/s)		Hydrogen (g/s)		Oxygen + Argon (g/s)		Nitrogen (g/s)		Sulfur (g/s)		NO _x (as NO ₂) (g/s)	
	Rate + σ		Rate + σ		Rate + σ		Rate + σ		Rate + σ		Rate + σ		Rate + σ	
<u>Inlet Streams</u>														
ESP Oil to Boiler No. 3	250±25†		210±21		27±3		8±1		1.2±0.7		0.60±0.16			
Combustion Air	5100±1275††		0.7±0.2				1200±300		3900±975					
ΣIn	5350±1275		211±21		27±3		1200±300		3900±975		0.60±0.16			
<u>Discharge Streams</u>														
Boiler No. 3 Firing ESP Oil	5200±520#		230±25				1100±121		3900±429		0.95±0.10		0.1	
Water in Combustion Effluents	230±23#				26±3		200±20							
ΣOut	5430±521		230±25		26±3		1300±123		3900±429		0.95±0.10		0.1	
ΣOut/ΣIn	1.01		1.09		0.96		1.08		1.00		1.58			

††Combustion air flow rate was determined by closing nitrogen balance.

†Flow rate determined from plant process data.

#Flow rate measured by Radian.

*Error limit based on judgement value.

Note: All flow rates and error limits not footnoted have been determined empirically.

All error limits are one standard deviation (σ).

TABLE 12. FOSTER WHEELER/STOIC TWO STAGE GASIFIER - UMD
ELEMENTAL MASS FLOW RATES - GASIFICATION PROCESS

Element	Coal ($\mu\text{g/s}$)	Gasifier Ash ($\mu\text{g/s}$)	Cyclone Dust ($\mu\text{g/s}$)	ESP Oils ($\mu\text{g/s}$)	Ash Pan Water ($\mu\text{g/s}$)	Combined Product Gas††† ($\mu\text{g/s}$)
Aluminum*	3.5E6	2.7E6	5.2E3	220	3.2	5.0E3
Antimony**	<54	<7.0	3.8	0.8	<0.02	11
Arsenic**	490	850	6.2	3.3	<0.02	15
Barium*	1.3E5	1.2E5	64	21	14	60
Beryllium	81	31	0.16	ND	ND	0.15
Bismuth	ND	<3.5	0.80	0.12	ND	0.75
Boron*	6.8E3	2.4E4	20	4.0	40	170
Bromine	540	100	1.4	0.40	0.9	40
Cadmium	270	180	0.60	ND	ND	0.55
Calcium*	9.0E5	1.7E6	2.4E3	120	5.5E3	2.2E3
Cerium	3.0E3	3.5E3	5.6	0.24	ND	5.2
Cesium	81	280	0.14	ND	<0.01	0.13
Chlorine	7.3E3	350	17	1.2	6.3	400
Chromium*	1.4E3	1.5E3	4.0	<0.4	0.41	78
Cobalt*	430	420	0.86	<2.4	0.16	0.80
Copper*	1.3E3	1.1E3	4.4	0.64	0.53	280
Dysprosium	ND	210	0.80	ND	ND	0.74
Erbium	ND	100	0.16	ND	ND	0.15
Europium	ND	35	0.20	ND	ND	0.18
Fluorine	1.9E3	3.9E3	>200	16	0.45	180
Gadolinium	ND	140	0.40	ND	ND	0.37
Gallium	1.4E3	800	56	0.28	0.18	52
Germanium	54	25	3.8	<0.08	ND	3.5
Gold	ND	ND	ND	ND	ND	ND
Hafnium	ND	70	0.40	ND	ND	<0.37
Holmium	ND	140	0.20	ND	ND	0.18
Iodine	110	10	0.80	0.24	0.06	0.74
Iridium	ND	ND	ND	ND	ND	ND
Iron*	9.7E5	1.4E6	2.4E3	680	<0.07	2.6E3
Lanthanum	2.4E3	6.3E3	4.0	0.20	ND	3.7
Lead	810	420	170	1.6	0.09	160
Lithium	3.2E3	2.2E3	5.4	0.12	1.8	43
Lutetium	ND	10	ND	ND	ND	ND
Magnesium*	2.0E5	4.2E5	480	<22	2.5	440
Manganese*	6.5E3	1.5E4	9.5	2.2	0.07	17
Mercury**	<19	10	0.15	0.64	0.38	2.2
Molybdenum*	780	560	1.5	2.6	1.1	6.0
Neodymium	240	660	0.40	ND	ND	0.37

(Continued)

TABLE 12. (Continued)

Element	Coal ($\mu\text{g/s}$)	Gasifier Ash ($\mu\text{g/s}$)	Cyclone Dust ($\mu\text{g/s}$)	ESP Oil ($\mu\text{g/s}$)	Ash Pan Water ($\mu\text{g/s}$)	Combined Product Gas††† ($\mu\text{g/s}$)
Nickel	1.2E3	1.1E3	3.2	<1.2	0.16	120
Niobium	1.1E3	1.3E3	2.0	ND	0.02	1.8
Osmium	ND	ND	ND	ND	ND	ND
Palladium	ND	ND	ND	ND	ND	ND
Phosphorus	>2.7E5	3.5E4	>200	12	0.9	260
Platinum	ND	ND	ND	ND	ND	ND
Potassium*	2.6E5	2.9E5	540	112	120	500
Praseodymium	540	560	1.0	<0.04	ND	0.9
Rhenium	ND	ND	ND	ND	ND	ND
Rhodium	ND	ND	ND	ND	ND	ND
Rubidium	1.1E3	2.1E3	1.8	0.08	0.27	2.0
Ruthenium	ND	ND	ND	ND	ND	ND
Samarium	540	210	0.80	ND	ND	0.74
Scandium	20	70	0.20	<0.04	<0.02	0.18
Selenium**	97	21	0.52	<1.2	<0.03	1.6
Silicon	>2.7E5	>3.5E4	>200	16	3.6	93
Silver	220	11	<0.40	0.80	ND	0.37
Sodium*	4.9E4	5.3E4	140	200	40	1.1E3
Strontium	2.7E5	3.5E4	150	1.6	>90	140
Sulfur	1.4E6†	9.8E4†	2.0E3†	9.6E3†	840††	1.16***
Tantalum	ND	21	ND	ND	ND	ND
Tellurium	110	7.0	0.20	ND	0.04	0.18
Terbium	ND	35	0.16	ND	ND	0.15
Thallium	ND	14	2.4	ND	ND	2.2
Thorium	1.1E3	1.0E3	1.6	ND	<0.09	1.5
Thulium	ND	7.0	ND	ND	ND	ND
Tin	110	18	15	0.15	ND	14
Titanium*	1.2E5	6.4E4	160	13	1.3	140
Tungsten	ND	70	0.80	ND	ND	0.74
Uranium	810	670	1.4	ND	<0.09	1.3
Vanadium*	2.6E3	2.3E3	4.8	10	0.99	4.5
Ytterbium	ND	70	ND	ND	ND	ND
Yttrium	3.8E3	1.8E3	3.2	ND	0.45	3.0
Zinc*	1.4E3	530	120	2.9	0.08	160
Zirconium	8.4E3	7.0E3	11	0.32	0.09	14

*Analysis performed by ICPES.

**Analysis performed by AAS.

***Analysis performed by FPD/GC.

†Ultimate analytical result.

††Water quality analytical result.

†††Includes particulate, impinger, and oil analyses.

ND - not detected.

aEb = a x 10^b

TABLE 13. FOSTER WHEELER/STOIC TWO STAGE GASIFIER - UMD
ELEMENTAL MASS FLOW RATES - COMBUSTION PROCESSES

Element	Combined Product Gas††† (µg/s)	Combustion Emissions from Product Gas Boilers††† (µg/s)	ESP Oil to Boiler No. 3 (µg/s)	Combustion Emissions from Boiler No. 3 Firing ESP Oil††† (µg/s)
Aluminum*	5.0E3	6.7E3	1.4E4	7.8E3
Antimony**	11	<4.2	50	<4.5
Arsenic**	15	<6.4	200	<7.0
Barium*	60	53	1300	<48
Beryllium	0.15	0.15	ND	0.16
Bismuth	0.75	0.73	75	0.80
Boron*	170	16	<250	380
Bromine	40	33	25	1.3
Cadmium	0.55	2.0	ND	0.61
Calcium*	2.2E3	2.0E3	7.5E3	2.6E3
Cerium	5.2	5.2	15	5.6
Cesium	0.13	0.13	ND	0.14
Chlorine	400	16	75	17
Chromium*	78	14	<25	1600
Cobalt*	0.80	0.71	150	27
Copper*	280	3.7	40	12
Dysprosium	0.74	0.74	ND	0.80
Erbium	0.15	0.15	ND	0.16
Europium	0.18	0.18	ND	0.20
Fluorine	180	170	1.0E3	190
Gadolinium	0.37	0.37	ND	0.40
Gallium	52	54	18	<57
Germanium	3.5	3.5	<5.0	3.8
Gold	ND	ND	ND	ND
Hafnium	<0.37	<0.37	ND	<0.40
Holmium	0.18	0.18	ND	0.20
Iodine	0.74	0.98	15	0.80
Iridium	ND	ND	ND	ND
Iron*	2.6E3	2.3E3	4.3E4	1.0E4
Lanthanum	3.7	3.7	13	4.0
Lead	160	160	100	0.53
Lithium	43	<4.8	7.5	10
Lutetium	ND	ND	ND	ND
Magnesium	440	440	<1400	480
Manganese*	17	10	130	170
Mercury**	2.2	45	40	110
Molybdenum*	6.0	9.2	163	120

(Continued)

TABLE 13. (Continued)

Element	Combined Product Gas††† (µg/s)	Combustion Emissions from Product Gas Boilers††† (µg/s)	ESP Oil to Boiler No. 3 (µg/s)	Combustion Emissions from Boiler No. 3 Firing ESP Oil††† (µg/s)
Neodymium	0.37	ND	ND	0.40
Nickel*	<120	14	<75	1.1E3
Niobium	1.8	1.7	ND	2.0
Osmium	ND	ND	ND	ND
Palladium	ND	ND	ND	ND
Phosphorus	260	170	750	200
Platinum	ND	ND	ND	ND
Potassium*	500	85	7.0E3	1.4E3
Praseodymium	0.9	0.9	<2.5	1.0
Rhenium	ND	ND	ND	ND
Rhodium	ND	ND	ND	ND
Rubidium	2.0	1.5	5.0	1.8
Ruthenium	ND	ND	ND	ND
Samarium	0.75	0.75	ND	0.80
Scandium	0.18	<0.80	<2.5	0.19
Selenium**	1.6	1.5	75	1.6
Silicon	93	180	1.0E3	>200
Silver	0.37	0.37	50	<0.40
Sodium*	1.1E3	140	1.3E4	300
Strontium	140	130	100	150
Sulfur	1.1E6***	1.5E6††	6.0E5†	8.0E5††
Tantalum	ND	ND	ND	ND
Tellurium	0.18	0.18	ND	0.20
Terbium	0.15	0.15	ND	0.16
Thallium	2.2	2.2	ND	2.4
Thorium	1.5	1.5	ND	1.6
Thulium	ND	ND	ND	ND
Tin	14	14	10	15
Titanium*	140	140	830	150
Tungsten	0.74	0.74	ND	0.80
Uranium	1.3	1.3	ND	1.4
Vanadium*	4.5	4.0	650	8.7
Ytterbium	ND	ND	ND	ND
Yttrium	3.0	2.7	ND	3.2
Zinc*	160	130	180	150
Zirconium	14	9.1	20	10

*Analysis performed by ICPES.

**Analysis performed by AAS.

***Analysis performed by FPD/GC.

ND - not detected.

aEb = a x 10^b

†Ultimate analytical result.

††Method 6 analytical result.

†††Includes particulate, impinger and oil analyses.

mass of trace elements from the process. Cyclone dust, also a solid waste stream, is a second vehicle for discharge of trace elements from the gasifier. The ultimate fate of trace elements carried from the process in the solid waste streams is a function of the method of disposal. For example, trace elements in the gasifier ash and cyclone dust may enter the environment as a result of leaching by run-off waters from surface disposal sites or by leaching from landfills into ground waters.

The ash pan water, because of its contact with the gasifier ash, is another vehicle for discharge of trace elements from the gasifier. The concentration of trace elements in the ash pan water is dependent upon the form of the element in the gasifier ash, the length of time the ash is in contact with the water, and various physical characteristics of the ash (e.g., particle size).

Two other streams, the product gas and ESP oil, are vehicles for the transport of trace elements, but are not discharge streams from the process. Any trace elements borne by these streams can, however, be released to the environment in the discharges from downstream combustion processes.

Data on elemental distribution in discharge streams from several low- and medium-Btu coal gasification processes have been generated as a result of similar source test and evaluation programs (4, 12).

5.3.2 Guidelines for Evaluation of Elemental Behavior

Three factors can generally be expected to influence the probable behavior of an element in a coal gasification system:

- the reactivity of the element,
- the elemental boiling point, and
- the temperature to which the element is exposed in the system.

Based on these factors, several general statements may be made that describe the behavior of elements:

- Nonvolatile elements or elements which do not react to form volatile compounds tend to remain in the mineral fraction of the coal ash and are discharged primarily in the gasifier bottom ash or wastewaters which have been in contact with the gasifier bottom ash.
- Moderately volatile trace elements can vaporize in the gasifier and condense by deposition upon available aerosol material or by nucleation. These elements, while found in the gasifier bottom ash, can also appear at increased concentrations in the solid wastes or organic condensates removed from the product gas by downstream gas clean-up systems.
- Volatile elements or elements which react to form volatile compounds generally remain in the product gases, and can be emitted to the environment in the emissions from product gas combustion.

Such elemental behavior was observed during characterization of single-stage gasification processes (4, 12), at which time certain criteria for evaluation of volatility were established. Using these criteria, elements were considered to exhibit volatile behavior if 25 percent or more of the total mass of the element discharged was emitted in the product gas. Elements were considered to exhibit moderately volatile behavior if 25 percent or more of their total mass was discharged in solid wastes or organic condensates from the product gas, and less than 25 percent of their total mass in the product gas.

Elements that do not discharge 25 percent or more of their total mass in either the product gas stream or solid matter borne in the product gas (e.g., cyclone dust or ESP oil) are considered to exhibit non-volatile behavior.

5.3.3 Elemental Distribution

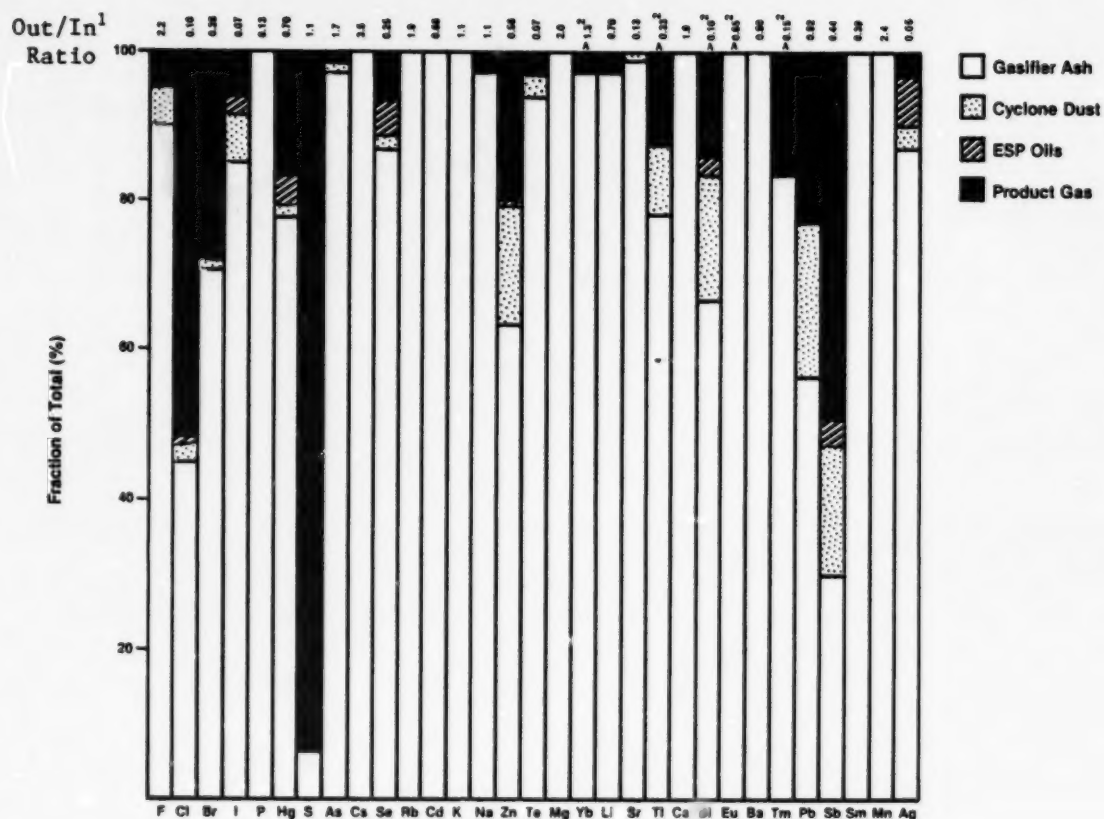
The distribution of minor and trace elements in the product, by-product, and discharge streams from the gasification process were calculated from the elemental mass flow rates of the respective streams. Figures 8 and 9 present the elemental distribution in the product, by-product, and discharge streams of the gasification process as a percent of the total mass of the element discharged. The elements are ordered according to ascending elemental boiling point. The ratio of the element in the product and discharge streams to the sum of the element in the coal is presented at the top of each bar.

Five elements have more than 25 percent of their total mass in the combined product gas and may be considered to exhibit volatile behavior. These elements are antimony, bromine, chlorine, sulfur, and tin. The vast majority of elements exhibit nonvolatile behavior and are discharged primarily in the gasifier ash. Of the 65 elements for which analytical data allowed an evaluation of elemental behavior, 57 are found primarily in the gasifier bottom ash and 43 of these have 95 percent or more of their mass discharged in the gasifier bottom ash.

While bismuth, lead, and zinc are discharged primarily in the bottom ash, 30 to 45 percent of the total mass discharged is found in the cyclone dust, ESP oil, and product gas. Tin, mentioned above as having more than 25 percent of its discharged mass in the product gas, is actually distributed almost equally across the bottom ash, cyclone dust, and product gas.

5.3.4 Elemental Enrichment

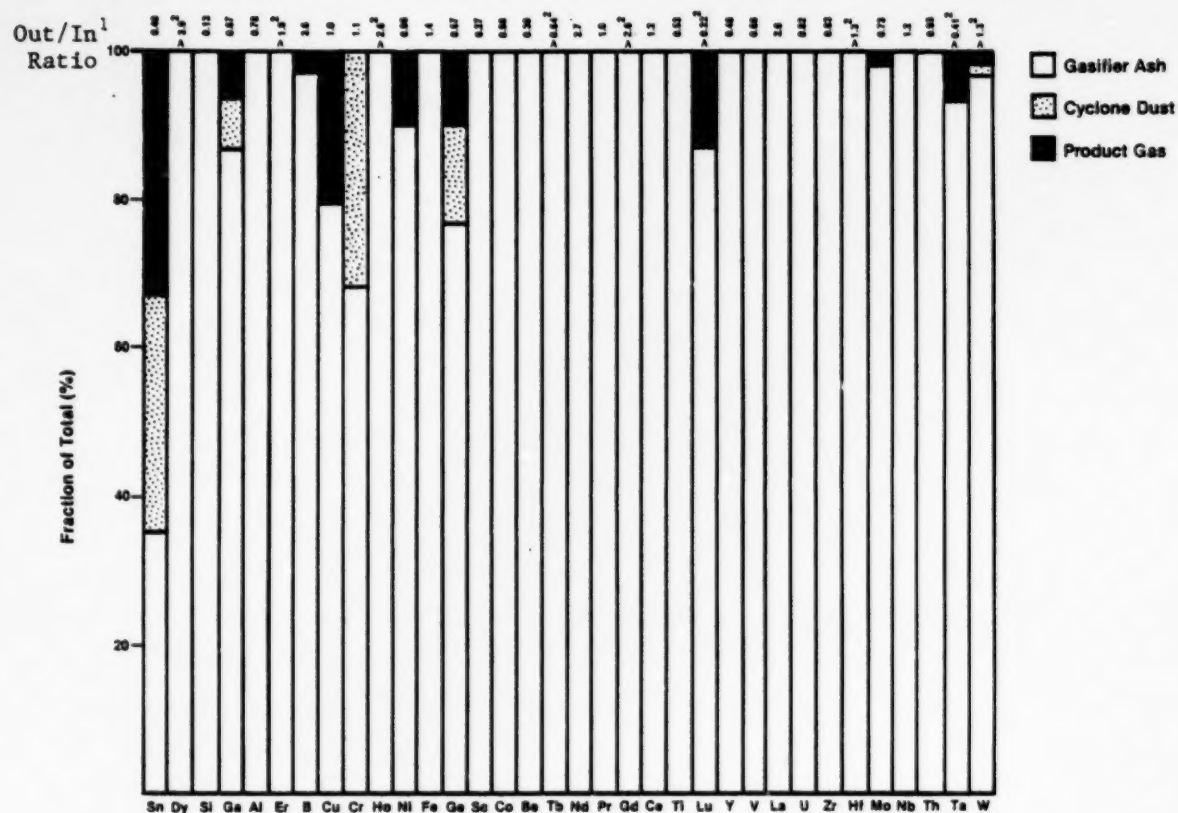
Another approach to evaluation of elemental behavior in the gasification process is the use of elemental concentration quotients. A concentration quotient is defined here as the ratio of the concentration of an element in a specific stream to the concentration of that same element in the coal, according to the following equation:



¹Out/in ratio is calculated as $\frac{\text{total mass flow out}}{\text{total mass flow in (coal)}}$

²Concentration in coal is below the detection limit; total mass flow for coal is calculated as the detection limit concentration x flow rate.

Figure 8. Elemental distribution ordered by increasing elemental boiling point (up to 2212°C) - gasification process.



¹Out/in ratio is calculated as $\frac{\text{total mass flow out}}{\text{total mass flow in (coal)}}$.

²Concentration in coal is below the detection limit; total mass flow for coal is calculated as the detection limit concentration x flow rate.

Figure 9. Elemental distribution ordered by increasing elemental boiling point (2270 to 5927°C) - gasification process.

$$\text{Concentration Quotient} = \frac{\text{Concentration of Element in Specific Stream}}{\text{Concentration of Element in Coal}}$$

Table 14 presents the concentrations and concentration quotients for elements in the gasifier ash, cyclone dust, and ESP oil. For the purpose of this discussion, concentrations and concentration quotients are also presented for the ash mineral fraction of each stream. The ESP oil is a by-product of gasification at UMD and is combusted in a boiler. However, other applications might increase the possibility of release of components in the ESP oil to the environment. For this reason, the ESP oil 's included with the wastes from the process for this study of elemental enrichment.

Elements enter the gasifier in the feed coal. The bulk of the carbonaceous material in the coal, as well as many volatile components, react during the gasification process to form gaseous products, while the bulk of the inorganic elemental material remains in the ash mineral fraction from the coal. If elemental enrichment is due to nothing more than the concentration of an element in the ash mineral fraction because of the loss of mass of carbonaceous and volatile organic components, then the change in elemental concentration of the element would be equal to the change of concentration of the ash mineral fraction in a stream.

However, due to various chemical or physical characteristics (e.g., boiling point, reactivity, native chemical form, etc.) some of the elements appear in waste streams at concentrations different from that which might be expected as a result of changes in the concentration of the ash mineral fraction (i.e., their concentration quotients, calculated from the equation given above, are greater or smaller than the concentration quotients calculated for the ash mineral fraction). If their concentration quotient is greater than that for the ash mineral fraction, they are considered to be enriched with respect to the ash mineral fraction. Conversely, if their concentration quotient is smaller, they are considered to be depleted with respect to the ash mineral fraction.

TABLE 14. FOSTER WHEELER/STOIC TWO-STAGE GASIFIER - ELEMENTAL CONCENTRATIONS (MASS/MASS) AND CONCENTRATION QUOTIENTS (BASED ON COAL) FOR MAJOR PROCESS STREAMS

Element	Coal	Gasifier Ash		Cyclone Dust		ESP oils	
	($\mu\text{g/g}$)	($\mu\text{g/g}$)	Q*	($\mu\text{g/g}$)	Q*	($\mu\text{g/g}$)	Q*
Aluminum	1.3 E4	7.8 E4	6	2.6 E4	2	54	(0.01)
Antimony	<0.2	<0.2	1	19	95	0.2	1
Arsenic	1.8	24	13	31	17	0.82	0.45
Barium	480	3300	6.9	320	0.7	5.2	0.01
Beryllium	0.3	0.9	3	0.8	2.7	ND	—
Bismuth	**	<0.1	0.5	4	20	0.03	0.2
Boron	25	690	28	98	4	<1	0.04
Bromine	2	3	1.5	7	3.5	0.1	0.05
Cadmium	1	5	5	3	3	ND	—
Calcium	3350	5.0 E4	15	1.2 E4	3.6	30	(0.01)
Cerium	11	99	9	28	2.5	0.06	(0.01)
Cesium	0.3	8	27	0.7	2.3	ND	—
Chlorine	27	10	0.37	86	3.2	0.3	0.01
Chromium	5.2	42	8	20	3.8	<0.1	0.02
Cobalt	1.6	12	7.5	4.3	2.7	<0.6	0.4
Copper	4.8	32	6.6	22	4.6	0.16	0.03
Dysprosium	**	6	30	4	20	ND	—
Erbium	**	3	15	0.8	4	ND	—
Europium	**	1	5	1	5	ND	—
Fluorine	7	110	16	>1000	>143	4	0.6
Gadolinium	**	4	20	2	10	ND	—
Gallium	5	23	4.6	280	56	0.07	0.01
Germanium	<0.2	0.7	3.5	19	95	<0.02	0.1
Gold	**	ND	—	ND	—	ND	—
Hafnium	**	2	10	<2	10	ND	—
Holmium	**	4	20	1	5	ND	—
Iodine	<0.4	0.3	1	4	10	0.06	0.15
Iridium	**	ND	—	ND	—	ND	—
Iron	3600	4.0 E4	11	1.2 E4	3.3	170	0.05
Lanthanum	9	180	20	20	2.2	.05	(0.01)
Lead	3	12	4	870	290	0.4	0.13
Lithium	12	64	5.3	27	2.3	0.03	(0.01)
Lutetium	**	0.3	1.5	ND	—	ND	—
Magnesium	760	1.2 E4	16	2400	3.2	<5.4	(0.01)
Manganese	24	440	18	48	2	0.54	0.02
Mercury	<0.07	0.30	4.3	0.75	10.7	0.16	2.3
Molybdenum	2.9	16	5.5	7.6	2.6	0.65	0.2
Neodymium	0.9	19	21	2	2.2	ND	—
Nickel	4.6	31	6.7	16	3.5	<0.3	0.2
Niobium	4	38	9.5	10	2.5	ND	—
Osmium	**	ND	—	ND	—	ND	—

(Continued)

TABLE 14. (Continued)

Element	Coal	Gasifier Ash		Cyclone Dust		ESP oils	
	($\mu\text{g/g}$)	($\mu\text{g/g}$)	Q*	($\mu\text{g/g}$)	Q*	($\mu\text{g/g}$)	Q*
Palladium	**	ND	—	ND	—	ND	—
Phosphorus	>1000	>1000	1	>1000	1	3	(0.01)
Platinum	**	ND	—	ND	—	ND	—
Potassium	950	8200	8.6	2700	2.8	28	0.03
Praseodymium	2	16	8	5	2.5	<0.01	(0.01)
Rhenium	**	ND	—	ND	—	ND	—
Rhodium	**	ND	—	ND	—	ND	—
Rubidium	4	59	15	9	2.2	0.02	(0.01)
Ruthenium	**	ND	—	ND	—	ND	—
Samarium	2	6	3	4	2	ND	—
Scandium	1	2	2	1	1	<0.01	(0.01)
Selenium	0.36	0.6	1.7	2.6	7.2	<0.3	0.8
Silicon	>1000	>1000	1	>1000	1	4	(0.01)
Silver	0.8	<0.3	0.4	<2	2.5	0.2	0.25
Sodium	180	1500	8.3	690	3.8	49	0.27
Strontium	>1000	>1000	1	760	0.8	0.4	(0.01)
Sulfur	5200	2800	0.53	1.0 E4	1.9	2400	0.46
Tantalum	**	0.6	3	ND	—	ND	—
Tellurium	0.4	0.2	0.5	1	2.5	ND	—
Terbium	**	1	5	0.8	4	ND	—
Thallium	**	0.4	2	12	60	ND	—
Thorium	4	29	7.3	8	2	ND	—
Thulium	**	0.2	1	ND	—	ND	—
Tin	0.4	0.5	1.2	77	192	0.04	0.1
Titanium	440	1800	4.1	780	1.8	3.3	(0.01)
Tungsten	**	2	10	4	20	ND	—
Uranium	3	19	6.3	7	2.3	ND	—
Vanadium	9.7	66	6.8	24	2.5	2.6	0.27
Ytterbium	**	2	10	ND	—	ND	—
Yttrium	14	50	3.6	16	1.1	ND	—
Zinc	5.1	15	2.9	620	121	0.71	0.14
Zirconium	31	200	6.4	55	1.8	0.08	(0.01)
Ash minerals	9.51 E4	7.77 E5	8.2	2.7 E5	2.8	<1800	0.02

*Concentration quotient = $\frac{\mu\text{g element/g stream}}{\mu\text{g element/g coal}}$ or $\frac{\mu\text{g ash/g stream}}{\mu\text{g ash/g coal}}$

**Not detected in coal at levels equal to detection limit for spark source mass spectrometry. Value set at detection limit (0.2 $\mu\text{g/g}$) for calculation of concentration quotient (Q). Calculated value presented as concentration quotient for these elements can be considered as an apparent minimum concentration quotient.

ND = Not detected at levels equal to detection limit of technique.

— = Quotient not calculated for elements not detected.

< = AA/ICPES detection limit.

< = SSMS value.

> = Major component (>1000 ppm) by SSMS analysis.

(0.01) = For ESP oil, calculated quotient is less than 0.01.

This relationship may be expressed as:

$$\text{Enrichment Factor} = \frac{\text{Concentration Quotient of Element in Stream}}{\text{Concentration Quotient of Ash Minerals in Stream}}$$

An enrichment factor greater than one is indicative of enrichment with respect to ash minerals; conversely, a factor less than one is indicative of depletion.

Many volatile elements become enriched with respect to the ash mineral fraction in aerosolized organic or particulate matter entrained in gaseous process streams. The theoretical mechanism by which this enrichment may occur has been described previously, in Section 5.3.

The elemental concentration quotients of selected minor and trace elements in the gasifier ash, cyclone dust, and ESP oil, are presented graphically in Figures 10 through 15. The elements are listed in order of ascending elemental boiling point.

The gasifier bottom ash is the primary vehicle for elemental discharge of the majority of elements. However, only eleven elements detected in the coal and also in the gasifier ash were found to be enriched with respect to the ash mineral fraction in the gasifier ash stream. These elements were:

Arsenic	Fluorine	Manganese
Boron	Iron	Neodymium
Calcium	Lanthanum	Rubidium
Cesium	Magnesium	

An additional fourteen elements can possibly be considered to be enriched in the gasifier ash because they were found in the gasifier ash and had not been found at detectable levels in the feed coal. These were:

Bismuth	Hafnium	Thallium
Dysprosium	Holmium	Thulium
Erbium	Lutetium	Tungsten
Europium	Tantalum	Ytterbium
Gadolinium	Terbium	

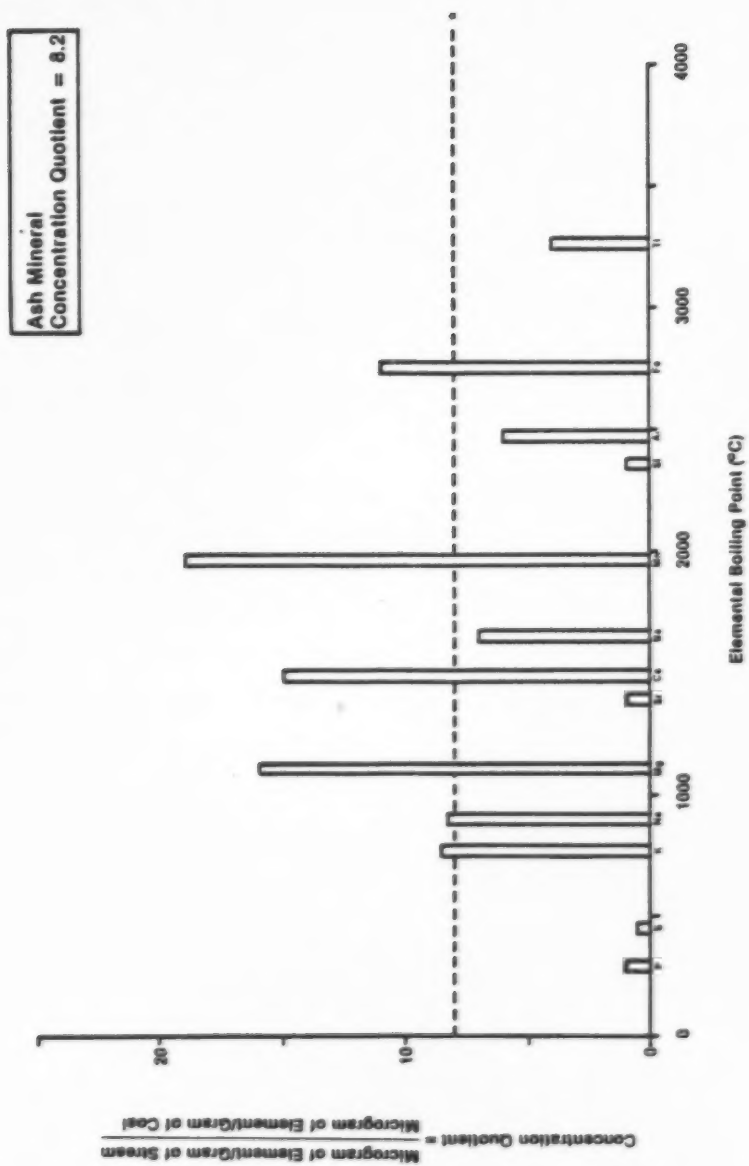


Figure 10. Concentration quotients for major elements in gasifier ash.

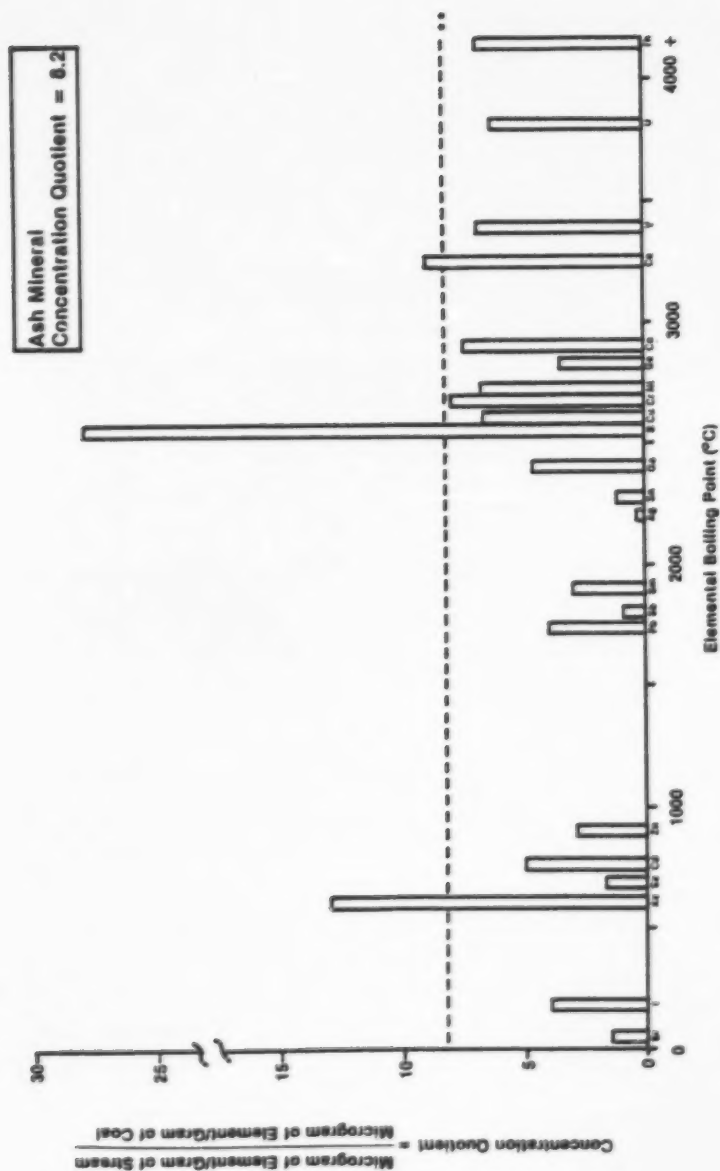


Figure 11. Concentration quotients for trace elements in gasifier ash.

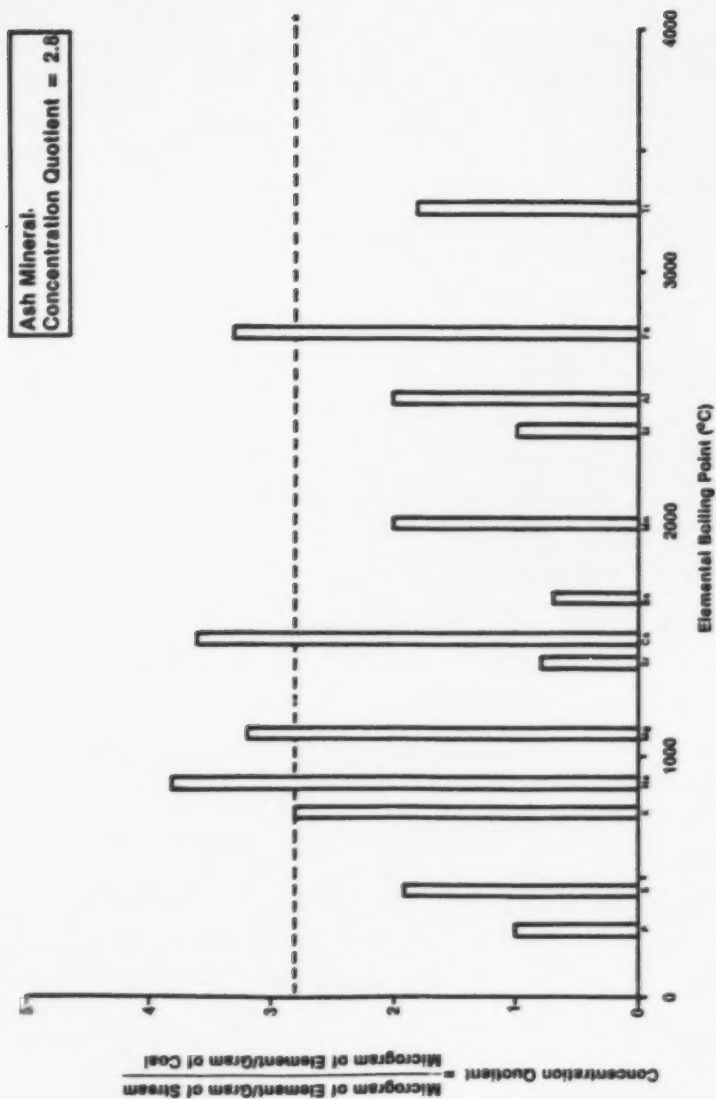


Figure 12. Concentration quotients for major elements in cyclone dust.

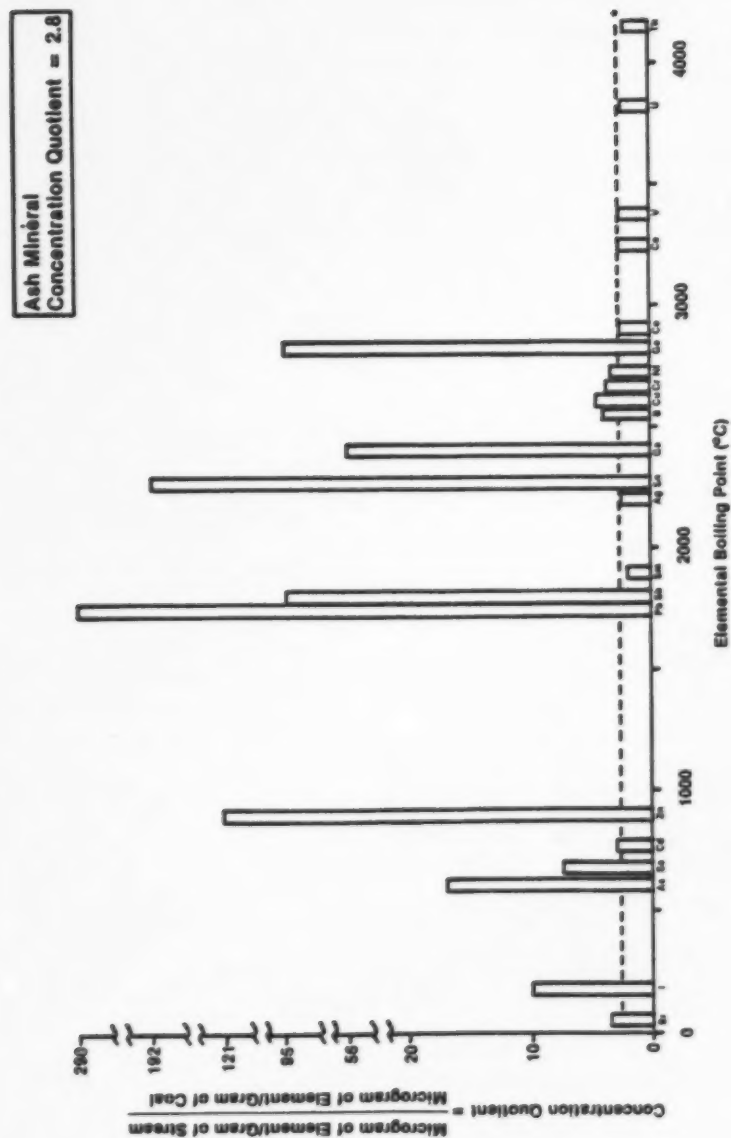


Figure 13. Concentration quotients for trace elements in cyclone dust.

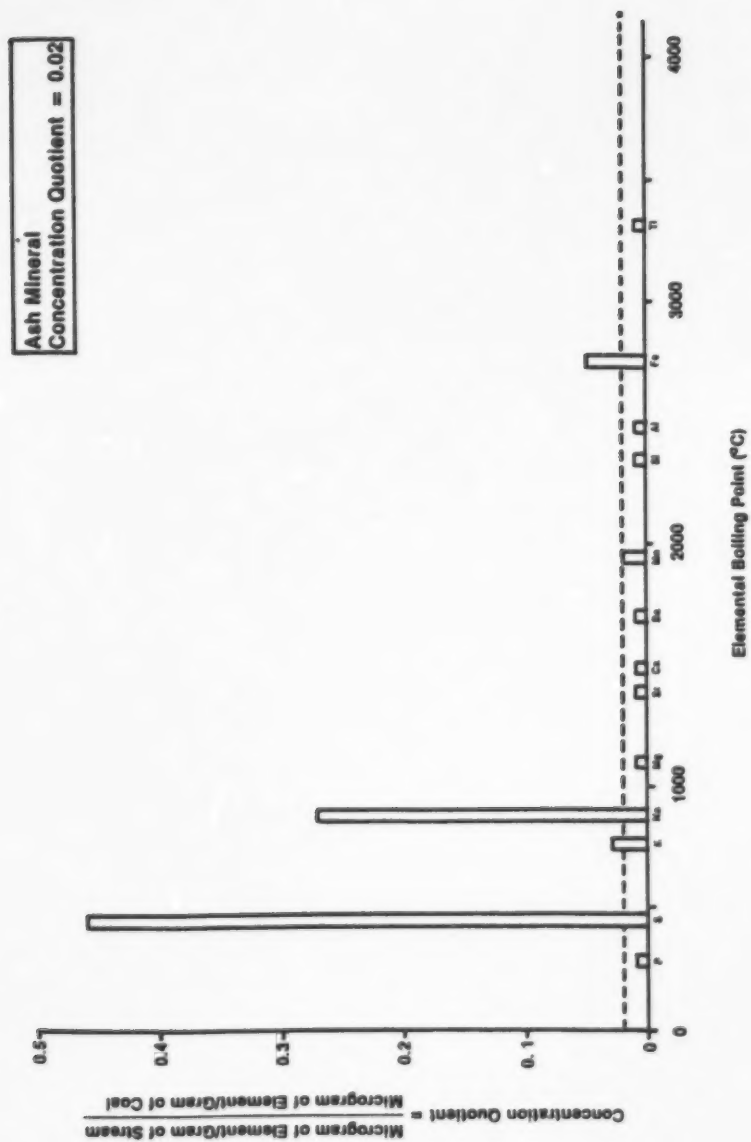


Figure 14. Concentration quotients for major elements in ESP oil.

Ash Mineral
Concentration Quotient = 0.02

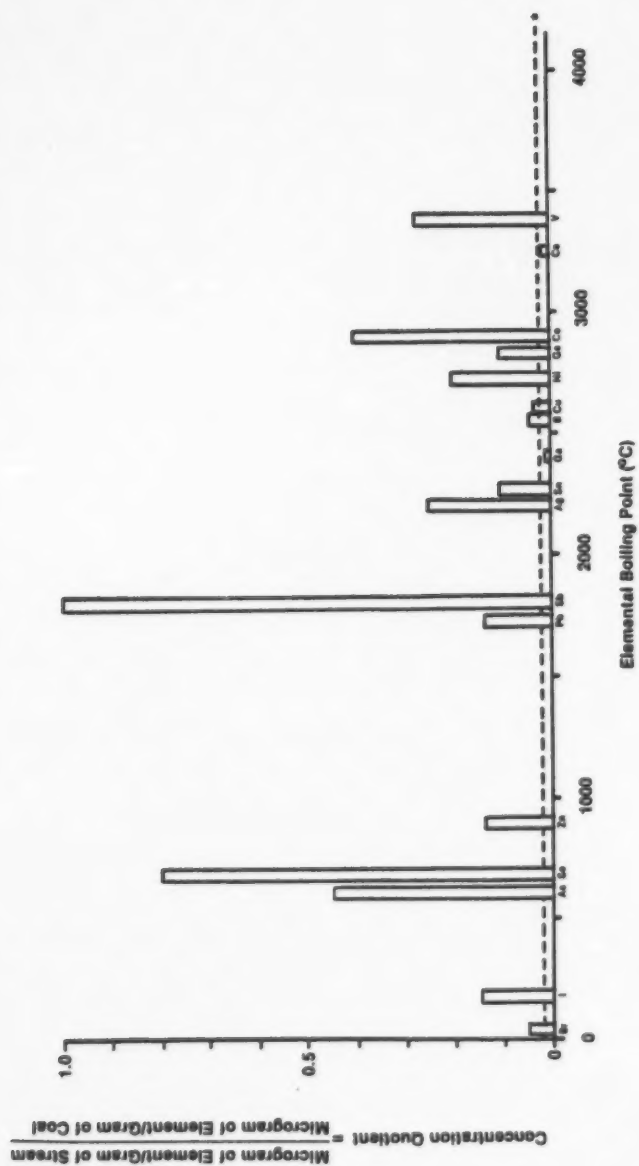


Figure 15. Concentration quotients for trace elements in ESP oil.

Based on the criteria established for the evaluation of elemental volatility (Section 5.3.2), all of these elements could be considered to be non-volatile and therefore would not be expected to be depleted in the gasifier ash. Four more elements, also considered to be non-volatile according to the criteria for assessing volatility, were found to show concentration quotients only slightly higher than the concentration quotient for the ash mineral fraction in the gasifier ash. These elements were cerium, niobium, potassium, and sodium. All the other elements (see Table 14), including many elements which, according to the criteria would be considered to be non-volatile, were found to be depleted in the gasifier ash with respect to the ash mineral fraction.

The possible influence of other factors such as reactivity or chemical form (e.g., the presence of organometallic compounds in the coal or the formation of similar complex materials in the volatile organic stream at the lower temperature encountered in the volatilization zone of a two-stage gasifier) may be indicated by the behavior of these elements.

Ten elements exhibited significant enrichment with respect to the ash mineral fraction of the cyclone dust. These were:

Antimony	Germanium	Mercury
Arsenic	Iodine	Tin
Fluorine	Lead	Zinc
Gallium		

An additional ten elements were found in the cyclone dust which were not found at detectable levels in the coal.

Bismuth	Gadolinium	Terbium
Dysprosium	Hafnium	Thallium
Erbium	Holmium	Tungsten
Europium		

Only lead and zinc could be considered to be moderately volatile to volatile based upon criteria for evaluation of volatility according to elemental

distribution (percentage of mass discharged in a stream). Antimony and tin were considered to be volatile according to the same criteria.

Thirteen elements were found in the cyclone dust at concentration quotients equal to or only slightly higher than the concentration quotient for the ash mineral fraction. These were:

Boron	Chromium	Nickel
Bromine	Copper	Potassium
Cadmium	Iron	Selenium
Calcium	Magnesium	Sodium
Chlorine		

All other elements (see Table 14) showed depletion with respect to the ash mineral fraction of the cyclone dust.

While a number of elements appeared to be enriched with respect to the ash mineral fraction of the ESP oil, the apparent elemental enrichment may be an artifact of the extremely low ash mineral content of the ESP oil, or may be due to some factor or factors unrelated to elemental boiling points. Mercury was present in the ESP oil at a concentration (mass/mass) exceeding its concentration in the feed coal. The concentration of antimony in the ESP oil was essentially identical with its concentration in the feed coal, and the concentration of selenium was within 20 percent of its concentration in the feed coal. All other elements in the ESP oil were found at concentrations significantly lower than their concentrations in the feed coal.

5.4 Extractable Organics

Extractable organics in product gas and discharge streams were characterized by total organic loading and by gas chromatographic analyses. The total organic loading is the sum of TCO and GRAV values. Table 15 summarizes the results of the organic loadings for the process and discharge streams.

TABLE 15. EXTRACTABLE ORGANICS SUMMARY

Stream	Extractable Organics	
	Concentration (TCO & GRAV)	Mass Flow (g/sec)
Ash Pan Water	1.0 E3* $\mu\text{g/L}$	9.0 E-6
Cyclone Dust	1.2 E2 $\mu\text{g/g}$	2.4 E-5
Gasifier Ash	1.0 E2 $\mu\text{g/g}$	3.5 E-3
ESP Oil	9.3 E5 $\mu\text{g/g}$	3.7 E0
Combined Product Gas	4.0 E6 $\mu\text{g/Nm}^3$	3.8 E0
Aerosols**	2.1 E5 $\mu\text{g/Nm}^3$	2.0 E-1
Vapors	3.8 E6 $\mu\text{g/Nm}^3$	3.6 E0
Cleaned Top Product Gas	7.4 E6 $\mu\text{g/Nm}^3$	4.1 E0
Aerosols**	1.5 E5 $\mu\text{g/Nm}^3$	8.3 E-2
Vapors	7.2 E6 $\mu\text{g/Nm}^3$	4.0 E0
Raw Top Product Gas	1.3 E7 $\mu\text{g/Nm}^3$	7.0 E0
Aerosols**	7.4 E6 $\mu\text{g/Nm}^3$	4.1 E0
Vapors	5.2 E6 $\mu\text{g/Nm}^3$	2.9 E0
Cleaned Bottom Product Gas	8.9 E4 $\mu\text{g/Nm}^3$	3.9 E-2
Raw Bottom Product Gas	1.2 E4 $\mu\text{g/Nm}^3$	7.6 E-3
Combustion Emissions from Product Gas Boiler	2.7 E1 $\mu\text{g/Nm}^3$	1.3 E-4
Combustion Emissions from Boiler No. 3 Firing ESP Oil	2.7 E3 $\mu\text{g/Nm}^3$	1.1 E-2

*aEb = a x 10^b.

**That portion of aerosol loading extractable with methylene chloride and diethyl ether.

As indicated in the table, the concentrations of extractable organics in the gasifier ash and cyclone dust are comparable. Approximately 98 percent (mass flow) of the extractable organics found as aerosols in the raw top gas are removed by the ESP. The mass flows of the extractable organics in the ESP oil, the cleaned top gas, and the combined product gas are essentially the same. However, the concentration of extractable organics in the combustion emissions from Boiler No. 3 firing ESP oil is about two orders of magnitude higher than in the emissions from the product gas boilers.

Extracts of composited samples of the gasifier ash, ash pan water, cyclone dust, ESP oil to Boiler No. 3, combined product gas, and combustion emissions from the product gas boilers and Boiler No. 3 were characterized by GC/MS analysis. Priority pollutants were identified and quantitated to provide data for EPA's Effluent Guideline Division. Table 16 presents the analytical results for priority pollutants identified in the streams. In addition, major peaks were identified and quantitated against d₁₀ anthracene. A list of the extractable organics of each sample identified by GC/MS is presented in Tables B-52 to B-58 (Appendix B). The data presented in these tables indicate that the majority of compounds analyzed by GC/MS are identified. Cyclone dust had the lowest percentage of identified mass (34 percent).

Table 17 gives the mass flows of priority pollutants found in the product and emissions streams on a mass flow basis. The combined product gas and ESP oil contain the greatest mass flow of priority pollutants of any stream leaving the gasification process; however, both streams are combusted and are not discharged directly. The mass flow of the priority pollutants identified in the emissions from combustion of product gas is 0.15 percent of the product gas loading, as shown below:

$$\frac{1.1\text{E}2 \text{ } \mu\text{g/sec emissions from product gas boiler}}{7.2\text{E}4 \text{ } \mu\text{g/sec combined product gas}} \times 100 = 0.15\%$$

For the ESP oil fired in Boiler No. 3, the loading of priority pollutants in the combustion emissions was 0.02 percent of that in the ESP oil:

TABLE 16. CONCENTRATION OF PRIORITY POLLUTANTS IDENTIFIED IN PRODUCT AND DISCHARGE STREAMS - UMD

Compound	Gasifier Ash ($\mu\text{g/g}$)	Ash Pan Water ($\mu\text{g/L}$)	Cyclone Dust ($\mu\text{g/g}$)	ESP Oil to Boiler #3 ($\mu\text{g/g}$)	Emissions from		Emissions from Boiler No. 3 ($\mu\text{g/Nm}^3$)
					Combined Product Gas ($\mu\text{g/Nm}^3$)	Product Gas Boiler ($\mu\text{g/Nm}^3$)	
Phenol	ND	3.3	ND	2400	4200	3.7	30
2,4-dimethylphenol	ND	ND	ND	ND	2.4E4	ND	ND
Naphthalene	0.01	0.33	0.15	310	7800	9.4	16
1-methylnaphthalene	ND	ND	0.01	540	7500	1.4	1.8
2-methylnaphthalene	ND	ND	0.01	370	6300	1.2	1.7
Dimethylnaphthalene	ND	ND	0.05	2500	2.3E4	1.4	1.3
Phenanthrene	ND	ND	0.01	320	310	0.32	0.72
Anthracene	ND	ND	ND	240	220	ND	ND
Fluorene	ND	ND	ND	320	29	0.36	0.48
Pyrene	ND	ND	0.04	ND	120	0.14	0.46
Benzo(a)anthracene	ND	ND	ND	240	ND	ND	ND
Bis(2-ethylhexyl) phthalate*	0.04	6.7	ND	20	1200	5.0	20

*Probable artifact of sample handling.

TABLE 17. MASS FLOW OF PRIORITY POLLUTANTS IN PRODUCT AND DISCHARGE STREAMS - UMD*

Compound	Product Streams		Discharge Streams				
	ESP Oil to Boiler No. 3	Combined Product Gas	Cyclone Dust	Gasifier Ash	Ash Pan Water	Product Gas Boiler	Combustion Emissions from Boiler No. 3
Phenol	6.0E5	3990	ND	ND	0.03	17.3	125
2,4-dimethylphenol	ND	2.4E4	ND	ND	ND	ND	ND
Naphthalene	7.8E4	7410	0.03	0.35	0.003	43.9	66.7
1-methylnaphthalene	1.3E5	7125	0.002	ND	ND	6.54	7.5
2-methylnaphthalene	9.2E4	5990	0.002	ND	ND	5.60	7.1
Dimethylnaphthalene	6.2E5	2.2E4	0.01	ND	ND	6.54	5.4
Phenanthrene	8.8E4	295	.002	ND	ND	1.49	3
Anthracene	6.0E4	209	ND	ND	ND	ND	ND
Fluorene	8.0E4	27	ND	ND	ND	1.68	2.0
Pyrene	ND	114	0.008	ND	ND	0.65	1.92
Benzo(a)anthracene	6.0E4	ND	ND	ND	ND	ND	ND
Bis(2-ethylhexyl) phthalate	5000	1140	ND	1.4	0.06	23.4	83.4
TOTAL	1.8E6	7.2E4	0.054	1.75	0.093	107	302
							411

*All mass flows are given as $\mu\text{g}/\text{sec}$.
ND = not detected.

$$\frac{3.0E2 \text{ } \mu\text{g/sec emissions from Boiler No. 3}}{1.86E6 \text{ } \mu\text{g/sec ESP oil}} \times 100 = 0.02\%$$

Noticeably absent from the analytical results are the polynuclear aromatic compounds such as acenaphthalene, benzo(a)pyrene, and fluoranthene. These compounds were not detected in any of the product or discharge streams at the detection limits of the technique.

Analyses by gas chromatography-Hall electrolytic conductivity detection (GC-HECD) of extracts of the combined product gas, ESP oil to Boiler No. 3, and the combustion emissions from the product gas boilers and Boiler No. 3 firing ESP oil are summarized in Table 18.

TABLE 18. RESULTS OF GC-HECD ANALYSIS FOR ORGANIC SULFUR, NITROGEN AND HALOGEN IN FEED AND COMBUSTION STREAMS* - UMD

	Product Gas Boilers		ESP Oil Boiler	
	Feed	Emissions	Feed	Emissions
Organic Sulfur Species	3.94E4	2.07E2	1.41E6	1.38E2
Organic Nitrogen Species	2.32E4	8.12E0	1.65E5	1.63E1
Organic Halogen Species	1.15E2	ND	ND	4.79E0

*Mass flow in $\mu\text{g/sec}$. Results are for sulfur, nitrogen, and halogen atoms bound to an extractable organic component.

ND = Not detected.

The levels of organic sulfur, nitrogen and halogen species in the combustion emissions relative to the levels in the feed streams indicate a significant destruction of the organic molecules during combustion. The apparent generation of a small ($<5 \text{ } \mu\text{g/sec}$) flow of organic halogen during combustion of ESP oil is considered to be an artifact of the sampling and analytical procedures used (non-concurrent sampling and analyses of gaseous and heavy organic liquid streams).

The total organic sulfur for ESP oil to Boiler No. 3 is about two orders of magnitude greater than the concentration of organic sulfur in the combined product gas. The emissions of organic sulfur for both boilers is essentially the same. Sulfur present in both feedstocks may be converted to inorganic sulfur species and measured as emissions of sulfur oxides. Nitrogen present in the form of organic nitrogen species in the combined product gas and ESP oil is combusted in the boilers and may be emitted as nitrogen oxide. Organic halogen present in the feed streams may be emitted as an inorganic halogen complex. A further discussion of nitrogen and sulfur oxides is found in Section 5.5. Results of the GC-HECD analyses for all discharge streams are found in Tables B-48 through B-51 (Appendix B).

5.5 Gas Compositions

Eight gaseous process and emission streams were characterized for concentrations of major gases, hydrocarbons, sulfur species, moisture, extractable organics, ammonia, cyanide, and aerosols and particulates. Also, emissions of nitrogen and sulfur oxides were measured in the combustion streams. Tables 19 to 22 present the analytical results. Results of the individual analyses are presented in Tables B-39 through B-47 and Table B-67 (Appendix B).

The C₁ through C₅ hydrocarbons present in the combined product gas appeared totally destroyed upon combustion in the product gas boilers. The mass flow rate for C₆ hydrocarbons appears to be higher in the emissions from the product gas boilers ($4.7\text{E}8 \pm 4.2\text{E}8$ $\mu\text{g}/\text{min}$) than in the combined product gas ($5.0\text{E}7 \pm 1.2\text{E}7$ $\mu\text{g}/\text{min}$). However, the emissions mass flow rate is based upon analysis of only two samples. The first sample, collected at 3:40 p.m. on February 24, 1981 and subjected to a single analysis, was found to contain 784 ppmv of C₆ hydrocarbons. The second sample was collected at 8:15 p.m. on the same date. Duplicate analyses of this sample resulted in C₆ concentrations of 179 and 180 ppmv. The value reported for C₆ flow rate in the emissions is calculated based on the analyses of both samples and thus shows a large standard deviation.

TABLE 19. AVERAGE COMPOSITION AND COMPONENT MASS FLOW OF THE COMBINED PRODUCT GAS AND COMBUSTION EMISSIONS FROM THE PRODUCT GAS BOILERS - UMD*

Parameter	Combined Product Gas		Combustion Emissions from Product Gas Boilers	
	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$
CO ₂	7.6 E7 + 1.1 E7	4.3 E9 + 6.2 E8	1.5 E8 + 3.6 E6	4.2 E1 + 1.0 E9
H ₂	1.3 E7 + 2.0 E6	7.4 E8 + 1.1 E8	3.3 E4 + 4.1 E4	9.2 E6 + 1.1 E7
O ₂ + Ar	9.8 E7 + 2.5 E6	5.5 E9 + 1.4 E8	1.6 E8 + 5.2 E6	4.4 E10 + 1.4 E9
N ₂	5.4 E8 + 3.6 E7	3.0 E10 + 2.0 E9	9.3 E8 + 4.6 E6	2.6 E11 + 1.2 E9
CO	3.2 E8 + 2.3 E7	1.8 E10 + 1.3 E9	ND	ND
Moisture	7.1 E7 + 6.3 E6	4.0 E9 + 3.5 E8	1.4 E8 + 3.3 E6	3.9 E10 + 9.2 E8
CH ₄	1.2 E7 + 2.4 E6	6.8 E8 + 1.3 E8	ND	ND
C ₂ H ₆	1.7 E6 + 3.7 E5	9.6 E7 + 2.1 E7	ND	ND
C ₂ H ₄	2.9 E5 + 7.6 E4	1.6 E7 + 4.3 E6	ND	ND
C ₃	1.3 E6 + 2.7 E5	7.4 E7 + 1.5 E7	ND	ND
C ₄	7.8 E5 + 1.7 E5	4.4 E7 + 9.6 E6	ND	ND
C ₅	6.2 E5 + 1.3 E5	3.5 E7 + 7.4 E6	ND	ND
C ₆	3.8 E5 + 2.2 E5	5.0 E7 + 1.2 E7	1.7 E6 + 1.5 E6	4.7 E8 + 4.2 E8
Benzene	2.4 E5 + 2.5 E4	1.3 E7 + 1.4 E6	<4.6 E2	<1.2 E5
Toluene	4.3 E5 + 1.6 E5	2.4 E7 + 9.1 E6	2.1 E4 + 2.2 E4	5.8 E6 + 6.1 E6
Xylene	3.0 E5 + 1.2 E5	1.7 E7 + 6.8 E6	ND	ND
H ₂ S	1.0 E6 + 1.1 E5	5.7 E7 + 6.2 E6	ND	ND
COS	1.9 E5 + 1.1 E4	1.0 E7 + 6.2 E5	ND	ND
CH ₃ SH	4.3 E4 + 1.1 E4	2.4 E6 + 6.2 E5	ND	ND
C ₂ H ₅ SH	<4.8 E3 + 2.3 E3	<2.7 E5 + 1.3 E5	ND	ND
CS ₂	<3.1 E3	<1.7 E5	ND	ND
C ₄ H ₄ S	<3.4 E3	<1.9 E5	ND	ND
NH ₃	8.3 E5 + 5.7 E4	4.7 E7 + 3.2 E6	<4.3 E2	<1.2 E5
HCN	2.4 E5 + 3.2 E4	1.3 E7 + 1.8 E6	1.7 E3 + 5.7 E1	4.7 E5 + 1.5 E4
SO ₂	ND	ND	7.0 E5 + 2.5 E4	1.9 E8 + 7.0 E6
NO _x (as NO ₂)	NA	NA	5.1 E5 + 1.3 E5	1.4 E8 + 3.6 E7
Extractable Organics†	4.0 E6	2.3 E8	2.7 E1	7.5 E3
Aerosol Organics‡	2.0 E4	1.1 E6	ND	ND
Particulate	1.9 E5	1.1 E7	3.9 E4	1.0 E7

*Results presented as concentration in dry gas.

†Sum of TCO and GRAV values.

‡That portion of aerosol extractable with methylene chloride and diethyl ether.

ND - not detected.

NA - not analyzed.

TABLE 20. AVERAGE COMPOSITION AND COMPONENT MASS FLOW OF THE TOP PRODUCT GAS - UMD*

Parameter	Raw Top Product Gas		Cleaned Top Product Gas	
	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$
CO ₂	6.7 E7 + 8.1 E6	2.2 E9 + 2.6 E8	7.2 E7 + 6.5 E6	2.3 E9 + 2.1 E8
H ₂	1.3 E7 + 1.1 E6	4.2 E8 + 3.6 E7	1.3 E7 + 1.7 E6	4.2 E8 + 5.6 E7
O ₂ + Ar	9.9 E6 + 1.2 E6	3.2 E8 + 3.9 E7	1.1 E7 + 3.9 E6	3.6 E8 + 1.2 E8
N ₂	4.9 E8 + 2.3 E7	1.6 E10 + 7.5 E8	5.0 E8 + 2.4 E7	1.6 E10 + 7.9 E8
CO	3.2 E8 + 1.1 E7	1.0 E10 + 3.6 E8	3.2 E8 + 1.1 E7	1.0 E10 + 3.6 E8
Moisture	1.4 E8 + 2.2 E7	4.6 E9 + 7.2 E8	1.4 E8 + 3.3 E6	4.6 E9 + 1.0 E8
CH ₄	2.4 E7 + 2.1 E6	7.2 E9 + 5.9 E7	2.0 E7 + 4.8 E6	6.6 E8 + 1.5 E8
C ₂ H ₆	3.2 E6 + 3.8 E5	1.0 E8 + 1.2 E7	2.7 E6 + 6.4 E5	8.9 E7 + 2.1 E7
C ₂ H ₄	6.0 E5 + 3.3 E4	1.9 E7 + 1.0 E6	3.8 E5 + 2.4 E5	1.2 E7 + 7.9 E6
C ₃	2.5 E6 + 1.7 E5	8.2 E7 + 5.6 E6	2.2 E6 + 4.9 E6	7.2 E7 + 1.6 E8
C ₄	1.5 E6 + 1.0 E5	4.9 E7 + 3.3 E6	1.3 E6 + 2.8 E5	4.2 E7 + 9.2 E6
C ₅	1.1 E6 + 7.7 E4	3.6 E7 + 2.5 E6	9.7 E5 + 2.3 E5	3.2 E7 + 7.5 E6
C ₆	1.6 E6 + 1.2 E5	5.2 E7 + 3.9 E6	1.3 E6 + 3.3 E5	4.2 E7 + 1.0 E7
Benzene	5.0 E5	1.6 E7	3.5 E5 + 4.8 E5	1.1 E7 + 1.5 E7
Toluene	7.4 E5	2.4 E7	9.5 E5 + 3.3 E4	3.1 E7 + 1.0 E6
Xylene	4.8 E5	1.5 E7	8.4 E5 + 7.6 E4	2.7 E7 + 2.5 E6
H ₂ S	1.0 E6 + 4.2 E4	3.3 E7 + 1.3 E6	1.0 E6 + 5.6 E4	3.3 E7 + 1.8 E6
COS	1.4 E5 + 2.9 E3	4.6 E6 + 9.5 E4	1.4 E5 + 2.5 E3	4.6 E6 + 8.2 E4
CH ₃ SH	7.9 E4 + 1.1 E3	2.6 E6 + 3.6 E4	7.7 E4 + 4.3 E3	2.5 E6 + 1.4 E5
C ₂ H ₅ SH	1.0 E4	3.3 E5	7.6 E3 + 0	2.5 E5 + 0
CS ₂	<3.1 E3	<1.0 E5	<3.1 E3	<1.0 E5
C ₄ H ₄ S	3.4 E3	1.1 E5	<3.4 E3	<1.1 E5
SO ₂	ND	ND	ND	ND
NH ₃	1.7 E6 + 1.6 E5	5.6 E7 + 5.2 E6	1.5 E6 + 3.9 E5	4.9 E7 + 1.2 E7
HCN	2.8 E5 + 3.2 E4	9.2 E6 + 1.0 E6	2.5 E5 + 4.6 E4	8.2 E6 + 1.5 E6
Extractable Organics†	1.3 E7	4.2 E8	7.4 E6	2.4 E8
Aerosol Organics‡	7.3 E6	2.4 E8	1.5 E5	4.9 E6
Particulate	1.2 E5	3.9 E6	3.1 E4	1.0 E6

*Results presented as concentration in dry gas.

†Sum of TCO and GRAV values.

‡That portion of aerosol loading extractables with methylene chloride and diethyl ether.

ND - not detected.

TABLE 21. AVERAGE COMPOSITION AND COMPONENT MASS FLOW OF BOTTOM PRODUCT GAS - UMD*

Parameter	Raw Bottom Product Gas		Cleaned Bottom Product Gas	
	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$
CO ₂	1.1 E8 + 5.8 E6	4.1 E9 + 2.2 E8	9.2 E7 + 1.3 E7	2.3 E9 + 3.3 E8
H ₂	8.2 E6 + 1.6 E6	3.1 E8 + 6.0 E7	9.8 E6 + 4.0 E5	2.5 E8 + 1.0 E7
O ₂ + Ar	1.1 E7 + 5.2 E5	4.1 E8 + 1.9 E7	1.0 E7 + 1.3 E5	2.6 E8 + 3.3 E6
N ₂	5.8 E8 + 4.1 E7	2.2 E10 + 1.5 E9	6.2 E8 + 4.0 E6	1.6 E10 + 1.0 E8
CO	3.2 E8 + 3.9 E7	1.2 E10 + 1.4 E9	3.6 E8 + 8.1 E6	9.3 E9 + 2.1 E8
Moisture	3.4 E7 + 8.1 E6	1.2 E9 + 3.0 E8	3.1 E7 + 5.0 E6	8.0 E8 + 1.3 E8
CH ₄	3.7 E5 + 3.9 E4	1.4 E7 + 1.4 E6	3.2 E5 + 6.5 E4	8.3 E6 + 1.6 E6
C ₂ H ₆	ND	ND	ND	ND
C ₂ H ₄	ND	ND	<1.1 E3	<2.8 E4
C ₃	ND	ND	ND	ND
C ₄	ND	ND	ND	ND
C ₅	ND	ND	ND	ND
C ₆	ND	ND	<1.4 E4	<3.6 E5
Benzene	2.4 E3	9.1 E4	<5.3 E3	<1.3 E5
Toluene	4.7 E4	1.7 E6	1.1 E5 + 8.6 E4	2.8 E6 + 2.2 E6
Xylene	1.4 E3	5.3 E4	9.5 E3 + 4.5 E2	2.4 E5 + 1.1 E4
H ₂ S	1.2 E6 + 5.0 E4	4.4 E7 + 1.9 E6	1.2 E5 + 7.9 E3**	3.1 E6 + 2.0 E5**
COS	2.4 E5 + 2.1 E4	9.2 E6 + 7.9 E5	7.9 E5 + 1.6 E4**	2.0 E7 + 4.1 E5**
CH ₃ SH	ND	ND	ND	ND
C ₂ H ₅ SH	ND	ND	ND	ND
CS ₂	3.1 E3	1.2 E5	<3.1 E3	<8.0 E4
C ₄ H ₄ S	ND	ND	ND	ND
SO ₂	ND	ND	ND	ND
NH ₃	2.9 E4 + 9.0 E3	1.1 E6 + 3.4 E5	4.7 E4 + 2.1 E4	1.2 E6 + 5.4 E5
HCN	4.3 E4 + 2.9 E3	1.6 E6 + 1.1 E5	3.8 E4 + 1.7 E4	9.8 E5 + 4.4 E5
Extractable Organics†	1.2 E4	4.5 E5	8.9 E4	2.3 E6
Aerosol Organics‡	ND	ND	ND	ND
Particulate	5.7 E5	2.1 E7	6.7 E5	1.7 E7

*Results presented as concentration in dry gas.

**Values are considered to be in error based on flow rates and concentrations in other product gas streams.

†Sum of TCO and GRAV values.

‡That portion of aerosol loading extractable with methylene chloride and diethyl ether.

ND - not detected.

TABLE 22. AVERAGE COMPOSITION AND COMPONENT MASS FLOW OF COMBUSTION
EMISSIONS FROM BOILER NO. 3 - UMD*

Parameter	Combustion Emissions from Boiler No. 3 Firing ESP Oils		Combustion Emissions from Boiler No. 3 Firing Natural Gas	
	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$	$\mu\text{g}/\text{Nm}^3 \pm \sigma$	$\mu\text{g}/\text{min} \pm \sigma$
CO ₂	1.8 E8 + 9.0 E6	4.5 E10 + 2.2 E9	1.8 E8	3.2 E10
H ₂	ND	ND	ND	ND
O ₂ + Ar	1.2 E8 + 1.8 E7	3.0 E10 + 4.5 E9	6.4 E7	1.1 E10
N ₂	8.9 E8 + 4.1 E7	2.2 E11 + 1.0 E10	9.6 E8	1.7 E11
CO	ND	ND	ND	ND
Moisture	5.2 E7 + 7.4 E5	1.3 E10 + 1.8 E8	1.1 E8 + 6.3 E6	1.9 E10 + 1.1 E9
CH ₄	ND	ND	ND	ND
C ₂ H ₆	ND	ND	ND	ND
C ₂ H ₄	ND	ND	ND	ND
C ₃	ND	ND	ND	ND
C ₄	ND	ND	ND	ND
C ₅	ND	ND	ND	ND
C ₆	ND	ND	ND	ND
Benzene	<8.4 E2	<2.1 E5	NA	NA
Toluene	2.5 E4 + 2.4 E4	6.2 E6 + 6.0 E6	NA	NA
Xylene	ND	ND	NA	NA
H ₂ S	ND	ND	ND	ND
COS	2.5 E3	6.2 E5	ND	ND
CH ₃ SH	ND	ND	ND	ND
C ₂ H ₅ SH	ND	ND	ND	ND
CS ₂	ND	ND	ND	ND
C ₄ H ₄ S	ND	ND	ND	ND
NH ₃	2.3 E3 + 1.6 E3	5.7 E5 + 4.0 E5	NA	NA
HCN	5.3 E3 + 1.5 E3	1.3 E6 + 3.7 E5	NA	NA
SO ₂	3.8 E5 + 3.0 E4	9.5 E7 + 7.5 E6	9.6 E3 + 2.9 E3	1.7 E6 + 5.2 E5
NO _x (as NO ₂)	7.9 E5 + 4.5 E4	1.9 E8 + 1.1 E7	2.1 E5 + 4.5 E3	3.7 E7 + 8.1 E5
Extractable	2.7 E3	6.7 E5	NA	NA
Organics†				
Aerosol Organics‡	1.0 E3	2.5 E5	NA	NA
Particulate	4.8 E4	1.2 E7	NA	NA

*Results presented as concentration in dry gas.

†Sum of TCO and GRAV values.

‡That portion of aerosol loading extractable with methylene chloride and diethyl ether.

ND - not detected.

NA - not analyzed.

The mass flow rate for toluene in the combustion emissions was lower ($5.9\text{E}6 \mu\text{g}/\text{min}$) than in the combined product gas ($2.4\text{E}7 \mu\text{g}/\text{min}$), while xylene was not detected in the combustion emissions. The mass flow rate of extractable organics was reduced from $2.3\text{E}8 \mu\text{g}/\text{min}$ in the combined product gas to $7.6\text{E}3 \mu\text{g}/\text{min}$ in the combustion emissions from the product gas boilers. Mass flow rates for ammonia and cyanide ($4.7\text{E}7 \mu\text{g}/\text{min}$ and $1.3\text{E}7 \mu\text{g}/\text{min}$, respectively) in the combined product gas were reduced by approximately two orders of magnitude in the combustion emissions.

The combined product gas stream and the emissions from the product gas boiler were sampled on the same day and analyzed to determine a material balance for the product gas boiler. The combined mass flow of organic nitrogen, ammonia, and cyanide in the combined product gas is equivalent to 0.78 g/s nitrogen. Nitrogen enters the gasifier with the coal at a rate of 3.9 g/s (no contribution from nitrogen in the blast air is included). The gasification process effectively removes 80 percent of the fuel nitrogen and converts approximately 20 percent to gaseous components that may contribute to emissions of nitrogen oxides (2.3 g/s as NO_2 or 0.72 g/s as N) from the boiler. The measured emission rate of NO_x from the product gas boilers represents a 92 percent conversion of nitrogen compounds in the fuel gas to NO_x in the combustion emissions and is equivalent to 310 ng/J , or $0.72 \text{ lb/E}^6 \text{ Btu}$. Somewhat more S (1.5 g/s) as SO_2 was found in the combustion emission stream from the product gas boilers than was accounted for in the combined product gas feed (1.1 g/s). The data indicate that greater than 100 percent conversion of fuel gas sulfur to SO_2 occurs during the combustion process. This is the equivalent of 440 ng/J or $0.99 \text{ lb/E}^6 \text{ Btu}$ (see Table 23). This discrepancy between in and out is not inconsistent within the limitations imposed by the sampling and analytical techniques which were used.

There was no change in the gas composition for the top product gases as a result of the gas clean-up, except for aerosol organics and particulates. The effect of the ESP on aerosols in the top product gas is discussed in Section 5.6. Samples of bottom gas collected downstream of the cyclone showed an apparent reduction of H_2S flow rate by an order of magnitude and

an apparent tripling of the COS flow rate. Those values are reported in Table 21. Based on relative flow rates of H_2S and COS in the raw bottom gas, raw and cleaned top gas, and combined product gas, it is concluded that the values reported for H_2S and COS flow rates are in error. This error may have resulted from sample degradation prior to analysis, perhaps because of inadequate drying of the gaseous sample during collection in the sample bomb. No other changes in gas composition were observed in the bottom gas stream (2).

The majority of the gas-phase organics exits the gasification process in the top gas. The mass flow of $\text{C}_1\text{-C}_6$ hydrocarbons ($9.5\text{E}8 \mu\text{g}/\text{min}$) in the treated top gas is about two orders of magnitude higher than the mass flow rate for $\text{C}_1\text{-C}_6$ hydrocarbons in the treated bottom gas ($8.7\text{E}6 \mu\text{g}/\text{min}$). The mass flow rates of benzene, toluene, and xylene were about an order of magnitude higher in the treated top gas ($6.9\text{E}7 \mu\text{g}/\text{min}$) than in the treated bottom gas ($6.1\text{E}6 \mu\text{g}/\text{min}$). The gaseous sulfur species in the combined product gas are comprised of H_2S (88%), COS (9%) and CH_3SH (3%) with trace amounts of $\text{C}_2\text{H}_5\text{SH}$, CS_2 , and $\text{C}_4\text{H}_4\text{S}$. The mass flow rates of organic sulfur species associated with a hydrocarbon moiety were higher in the treated top gas than in the treated bottom gas, as expected. The treated top gas was found to contain methyl mercaptan ($2.5\text{E}5 \mu\text{g}/\text{min}$), ethyl mercaptan ($2.5\text{E}5 \mu\text{g}/\text{min}$), and thiophene ($1.1\text{E}5 \mu\text{g}/\text{min}$), which were not detected in the treated bottom gas. Mass flow rates for extractable organics in the treated top gas ($2.4\text{E}8 \mu\text{g}/\text{min}$) were two orders of magnitude higher than in the treated bottom gas ($2.3\text{E}6 \mu\text{g}/\text{min}$). The combined concentrations of NH_3 and HCN in the treated top gas ($\sim 1.8\text{E}6 \mu\text{g}/\text{Nm}^3$) is over an order of magnitude higher than their combined concentrations in the treated bottom gas ($\sim 8.5\text{E}4 \mu\text{g}/\text{Nm}^3$) (2).

The amount of HCN measured in the combined gases appears high in light of the low levels of CO and NH_3 . This may be the result of contamination or the presence of an interferent. However, since no blank runs were made this cannot be confirmed.

The oxygen levels cited in Tables 20 through 22 and in the appendix tables appear high because they represent combined O₂ and Ar. The analytical technology did not provide for separation of O₂ and Ar. In any case, O₂ originating from air and from the coal cannot be distinguished.

The combustion emissions from Boiler No. 3 firing ESP oil and from Boiler No. 3 during the combustion of natural gas were collected to identify any differences that may be observed when firing natural gas and liquid synfuels in the same boiler. These results are presented in Table 22. Individual analytical results and the time of collection of samples are presented in Appendices A and B.

Table 23 presents calculated emission rates for SO₂, NO_x (as NO₂), and particulates in the discharge streams (boiler flue gases). Emission rates were calculated for emissions from the combustion of product gas, ESP oil, and natural gas.

TABLE 23. EMISSIONS OF SO₂, NO_x, and PARTICULATE: CALCULATED VALUES FOR UMD BOILERS

Fuel to Boiler	Combustion Emissions		
	SO ₂ lb/E6 Btu (ng/J)	NO _x (as NO ₂) lb/E6 Btu (ng/J)	Particulate lb/E6 Btu (ng/J)
Product Gas*	1.29 (560)	0.94 (400)	0.072 (31)
Product Gas†	0.99 (440)	0.72 (310)	0.055 (24)
ESP Oil‡	0.34 (150)	0.70 (300)	0.042 (18)
Natural Gas**	0.01 (4)	0.02 (90)	NA††

*Based on calculated heat content of product gas.

†Based on average high heating value of coal from proximate/ultimate analyses (moisture free).

‡Based on high heating value of ESP oil to Boiler No. 3 from proximate/ultimate analyses.

**Based on high heating value of natural gas from a reference manual.

††NA - not analyzed.

At maximum capacity, the UMD gasifier has a feed rate of approximately 72E6 Btu/hr and produces enough low-Btu gas to fire a 55E6 Btu/hr steam boiler (44,000 lb/hr steam). The current Federal New Source Performance Standards (NSPS) for fossil-fuel-fired steam generators apply to boilers with a heat input of 250E6 Btu/hr or greater; therefore, the NSPS would not apply to a facility of this size. The U.S. EPA is expected to present a revised NSPS which may have a lower applicability threshold and more stringent emission standards.

Emissions from the UMD boilers firing gasifier product gas can, however, be compared to the emission limitations in the current NSPS for coal combustion. Table 24 presents this comparison. The SO₂ and particulate emissions meet the NSPS limitations if the emission rate is calculated according to the gas firing rate to the boiler. The NO_x emissions are slightly above the NSPS level. If the emission rates are calculated according to the heating value of the coal used to produce the gas, the SO₂ and NO_x emissions exceed the NSPS limit, while particulate emissions remain below the standard.

TABLE 24. COMPARISON OF NSPS LIMITATIONS FOR FOSSIL-FUEL FIRED WITH UMD BOILER EMISSIONS

Pollutant	NSPS*	Product Gas†	Product Gas#
SO ₂ (lb/E6Btu)	1.2	1.29	0.99
NO _x (lb/E6Btu)	0.70	0.94	0.72
Particulate (lb/E6Btu)	0.10	0.072	0.055

*40 CFR60, SUBPART D, emission limitation for solid fossil fuel.

†Based on calculated heat content of product gas.

#Based on average high heating value of coal from proximate/ultimate analysis (moisture free).

5.6 Efficiency of Gas Clean-up Devices

The efficiencies of control devices for removal of aerosolized oils and particulate from the product gas were evaluated by obtaining gravimetric measurements of aerosol and particulate loadings on both sides of the clean-up devices. The results of the analyses are summarized in Tables 20 and 21.

The collection efficiency of the top product gas ESP was calculated to be 98 percent for aerosolized oils and 74 percent for particulates based on loadings. Correcting for differences in measured gas volumes through the clean-up device, the collection efficiency was 98 percent for aerosolized oils and 72 percent for particulates based on mass flow. The efficiency of the cyclone was not evaluated because of difficulties in determining a consistent gas flow at the sampling points. The sampling valve was located approximately 30 cm upstream of a butterfly control valve which may disrupt the flow pattern of the bottom gas as it leaves the cyclone. The efficiency determined for the ESP may also be questionable due to the possible flow disruptions caused by the injection of treated bottom gas into raw top gas. This disruption could not be defined because only one sampling port was available and only a single traverse could be performed at the sampling location for the raw top gas. In addition, the configuration of the pipe transporting the treated top gas might result in the collection of a non-representative sample.

5.7 Water Quality

The ash pan water was analyzed for water quality parameters in addition to trace elements and extractable organics. The presence and concentration of components in the ash pan water are probably due primarily to the dissolution of components from the gasifier ash into the ash pan water. No analysis was performed on the service water. The flow rate of the ash pan water is quite low because the waters are not discharged directly. Ash pan water losses result from carry over with the discharged gasifier ash and, to a small extent, from evaporation.

Table B-38 in Appendix B presents the analytical results for water quality parameters in the ash pan water. Elemental concentrations in the ash pan water are presented in Table B-33. The ash pan water composition was compared to National Interim Primary Drinking Water Standards (NIPDWS) which are federally enforceable standards for drinking water. The ash pan water was also compared to National Secondary Drinking Water Regulations (NSDWR) guidelines, which are intended to serve as non-enforceable guidelines relating to aesthetic qualities and public acceptance of drinking water. The comparison is presented in Table 25. Note that this comparison is between a process wastewater and drinking water standards. The impact of parameters exceeding the drinking water standards would be lessened upon dilution of the ash pan water in a receiving water stream as well as by other factors specific to a given site. Concentrations of barium and mercury in the ash pan water exceed NIPDWS, and the concentration of chromium equals NIPDWS. The pH of the ash pan water is not within the guidelines of NSDWR, and the concentration of total dissolved solids exceeds NSDWR. The concentration of mercury in the ash pan water may not be typical of process conditions. During initial start-up of the gasifier, a large quantity of mercury serving as a seal was spilled into the ash pan. The mercury was cleaned up, and the mercury seal was replaced with a mechanical seal, but traces of mercury probably still contaminate the ash pan.

5.8 RCRA Leaching

The ultimate environmental fate of trace elements found in gasifier ash and cyclone dust depends upon the method of disposal and leaching of materials from the solid wastes by groundwater or run-off waters. In response to the Resource Conservation and Recovery Act (RCRA), EPA has promulgated regulations which define solid and hazardous wastes and has specified procedures for handling and disposal of them. The regulations include five criteria which define the characteristics of a hazardous waste. The category most pertinent to solid waste from gasification processes is Extraction Procedures (EP) toxicity.

TABLE 25. COMPARISON OF ASH PAN WATER TO DRINKING WATER STANDARDS†

Parameter	Ash Pan Water		
	NIPDWS*	NDSWR**	Concentration Range
As	50		<2
Ba	1,000		1,500
Cd	10		<1
Cl ⁻		250,000	200,000***
Cr	50		50
Cu		1,000	60
F ⁻	1,800		560***
Fe		300	<8
Pb	50		10
Mn		50	8
Hg****	2		4.2
Se	10		<3
Ag	50		<1
Zn		5,000	9
NO ₃ ⁻ (as N)		10,000	2,700***
SO ₄ ⁼		250,000	58,000***
pH		6.5-8.5	12.51***
TDS		500,000	1,800,000***
			1,100 - 4,400
			56,000 - 61,000
			12.48 - 12.54
			1,700,000 - 2,000,000

†Concentrations in µg/L.

*National Interim Primary Drinking Water Standards.

**National Secondary Drinking Water Regulations.

***Values given are averages of two analyses.

****Concentrations of Hg may not be typical because of spill contamination.

The test for EP toxicity is intended to simulate the release of leachate from landfills under conditions of improper management. The extraction procedure is defined, and maximum concentrations of contaminants that can be permitted in the resulting leachate are specified. Fourteen contaminants (eight trace elements, four pesticides, and two herbicides) are specified. The results of RCRA EP toxicity testing are presented in Table 26. Analyses were performed only for the eight trace elements, since little rationale exists for the presence of the listed pesticides and/or herbicides in the coal or for their production by the process. A previous study (3) has verified this concept.

TABLE 26. RESULTS OF RCRA EP TOXICITY TESTING IN SOLID WASTES FROM FOSTER WHEELER/STOIC TWO-STAGE GASIFIER - UMD*

Element	Gasifier Ash	Cyclone Dust	Maximum Permissible Concentration†
Arsenic	<3	120	5,000
Barium	650	6,900	100,000
Cadmium	100	<100 •	1,000
Chromium	<200	<200	5,000
Lead	<200	<200	5,000
Mercury#	<0.2	2.9	200
Selenium	<4	<4	1,000
Silver	<60	<60	5,000

*Concentrations in µg/L.

†RCRA Extraction Procedure toxicity.

#Concentrations of Hg may not be typical because of spill contamination.

According to the results of EP toxicity testing, the gasifier ash and cyclone dust can be considered as non-hazardous solid wastes. None of the eight trace elements were found in the extracts at levels greater than ten percent of the maximum allowable concentration, as specified in RCRA.

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APPENDIX A

SAMPLING DATA

A.1.0 SAMPLING DATA

The sampling data for the UMD gasification facility are presented in this appendix. Table A-1 gives the physical parameters for the gaseous streams. Tables A-2 and A-3 give sample volumes, stream concentration, and collection efficiencies for gaseous ammonia and cyanide samples, respectively. Tables A-4 and A-5 give sample volumes for gaseous trace metals and organic samples, respectively. Tables A-6 and A-7 give sample volumes and stream concentrations for NO_x and SO_x , respectively. The sampling log is presented in Table A-8.

TABLE A-1. PHYSICAL PARAMETERS DATA FOR THE UMD GASEOUS PROCESS
AND EMISSION STREAMS

Combined Product Gas:

Flow Rate:	62.9 Nm ³ /min (760 mm Hg and 25°C, dry) (2150 scfm, 934 g/s)
Velocity:	11.2 m/s (stream conditions) (23.7 fps)
Temperature:	242°C (468°F)
Pressure:	805 mm/Hg (31.7 in Hg)
% Moisture:	9.6
Molecular Weight:	23.2
Sampling Date:	February 24, 1981 (NH ₃ , HCN only); February 25, 1981.

Cleaned Bottom Gas:

Flow Rate:	26.3 Nm ³ /min (760 mm Hg, 25°C, dry) (900 scfm, 378 g/s)
Velocity:	2.70 m/s (stream conditions) (8.87 fps)
Temperature:	418°C (785°F)
Pressure:	774 mm Hg (30.48 in Hg)
% Moisture:	4.4
Molecular Weight:	24.8
Sampling Date:	March 5, 1981

Raw Bottom Gas:

Flow Rate:	37.7 Nm ³ /min (760 mm Hg, 25°C, dry) (1290 scfm, 673 g/s)
Velocity:	3.04 m/s (stream conditions) (9.97 fps)
Temperature:	514°C (958°F)
Pressure:	774 mm Hg (30.48 in. Hg.)
% Moisture:	4.5
Molecular Weight:	25.9
Sampling Date:	March 5, 1982

Cleaned Top Gas:

Flow Rate:	33.6 Nm ³ /min (760 mm Hg, 26°C, dry) (1150 scfm, 559 g/s)
Velocity:	6.19 m/s (stream conditions) (20.7 fps)
Temperature:	58.9°C (192°F)
Pressure:	762 mm Hg (30.01 in Hg)
% Moisture:	19.2
Molecular Weight:	23.2
Sampling Date:	February 27, 1981

(Continued)

TABLE A-1. (Continued)

Raw Top Gas:

Flow Rate:	32.8 Nm ³ /min (760 mm Hg, 25°C, dry) (1120 scfm, 534 g/s)
Velocity:	9.14 m/s (stream conditions) (30.0 fps)
Temperature:	90°C (194°F)
Pressure:	762 mm Hg (30.01 in Hg)
% Moisture:	19.3
Molecular Weight:	22.8
Sampling Date:	February 27, 1981

Boiler No. 3 Firing ESP Oil:

Flow Rate:	251 Nm ³ /min (760 mm Hg, 25°C, dry) (8530 scfm, 5150 g/s)
Velocity:	8.47 m/s (stream conditions) (27.8 fps)
Temperature:	331°C (628°F)
Pressure:	719 mm Hg (28.3 in Hg)
% Moisture:	7.0
Molecular Weight:	29.2
Sampling Date:	March 3, 1981

Boiler No. 3 Firing Natural Gas:

Flow Rate:	181 Nm ³ /min (760 mm Hg, 25°C, dry) (6190 scfm, 3680 g/s)
Velocity:	6.55 m/s (stream conditions) (21.5 fps)
Temperature:	286°C (547°F)
Pressure:	720 mm Hg (28.36 in Hg)
% Moisture:	15
Molecular Weight:	28.0
Sampling Date:	March 4, 1981

Product Gas Boiler:

Flow Rate:	274 Nm ³ /min (760 mm Hg, 25°C, dry) (9370 scfm, 5570 g/s)
Velocity:	4.69 m/s (stream conditions) (15.4 fps)
Temperature:	241°C (465°F)
Pressure:	759 mm Hg (29.90 in Hg)
% Moisture:	9.2
Molecular Weight:	28.2
Sampling Date:	February 25, 1981

TABLE A-2. UMD GASIFICATION FACILITY: SAMPLING DATA FOR GASEOUS AMMONIA SAMPLES

Description	Sample Volume			1st Imp mg NH ₃	2nd Imp mg NH ₃	Total NH ₃	ppm NH ₃	Collection Efficiency (%)
	SCF 60°F, 1 atm.	Nm ³ 25°C 1 atm.						
Combined Gas Run #1 2/24	3.794	0.1110		98.6	2.05	100.6	1310	98.0
2024-2036 hrs								
Combined Gas Run #2 2/25	4.403	0.1288		100.5	2.13	102.6	1150	97.9
0513-0529 hrs								
Combined Gas Run #3 2/25	3.480	0.1018		86.6	2.20	88.8	1260	97.5
0937-0952 hrs								
Product Gas Boiler Run #1 2/25	22.547	0.6594		<0.15	<0.13	<0.28	<0.6	--
0222-0312 hrs								
Product Gas Boiler Run #2 2/25	21.127	0.6179		<0.15	<0.12	<0.27	<0.6	--
0820-0910 hrs								
Cleaned Top Gas Run #1 2/27	3.610	0.1056		189.0	0.56	189.5	2590	99.7
1055-1111 hrs								
Cleaned Top Gas Run #2 2/27	6.554	0.1917		319.3	1.38	320.7	2410	99.6
1655-1737 hrs								
Cleaned Top Gas Run #3 2/27	3.069	0.0898		164.0	2.18	166.2	2670	98.7
2045-2057 hrs								
Raw Top Gas Run #1 2/27	3.430	0.1003		175.8	1.32	177.1	2540	99.3
1047-1057 hrs								
Raw Top Gas Run #2 2/27	2.395	0.0700		125.9	0.56	126.5	2600	99.6
1405-1415 hrs								
Raw Top Gas Run #3 2/27	5.008	0.1465		217.2	1.23	218.4	2150	99.4
1706-1726 hrs								
ESP Oil Boiler Run #1 3/3	19.024	0.5564		0.62	<0.12	<0.74	1.6	>83.8
ESP Oil Boiler Run #2 3/3	18.957	0.5544		0.18	0.12	<0.30	0.5	>60.0

(Continued)

TABLE A-2. (Continued)

Description	Sample Volume			1st Imp mg NH ₃	2nd Imp mg NH ₃	Total NH ₃	ppm NH ₃	Collection Efficiency (%)
	SCF 60°F, 1 atm.	Nm ³ 25°C 1 atm.						
Cleaned Bottom Gas Run #1 3/5 1437-1452 hrs	5.728	0.1675		9.51	0.92	10.4	89.7	91.2
Cleaned Bottom Gas Run #2 3/5 1923-1937 hrs	3.814	0.1115		3.23	0.38	3.61	46.6	89.5
Cleaned Bottom Gas Run #3 3/6 0103-0113 hrs	4.002	0.1170		97.8	0.33	98.1	1210	99.7
Raw Bottom Gas Run #1 3/5 1508-1532 hrs	3.525	0.1031		3.69	<0.12	<3.81	51.5	>96.9
Raw Bottom Gas Run #2 3/5 2111-2128 hrs	3.781	0.1106		2.29	0.22	2.51	32.7	91.2
Raw Bottom Gas Run #3 3/6 0114-0129 hrs	3.617	0.1058		47.3	<0.06	<47.4	644	>99.9

TABLE A-3. UMD GASIFICATION FACILITY: SAMPLING DATA FOR GASEOUS CYANIDE SAMPLES

Description	Sample Volume			1st Imp mg NH ₃	2nd Imp mg NH ₃	Total NH ₃	ppm NH ₃	Collection Efficiency (%)
	SCF 60°F, 1 atm.	Nm ³ 25°C 1 atm.						
Combined Gas Run #1 2/24	4.165	0.1218		24.2	0.40	24.6	183	98.4
2041-2055 hrs								
Combined Gas Run #2 2/25	4.353	0.1273		33.5	0.25	33.8	240	99.3
0537-0551 hrs								
Combined Gas Run #3 2/25	3.170	0.0927		22.4	0.16	22.6	221	99.3
0955-1010 hrs								
Product Gas Boiler Run #1 2/25	24.192	0.7075		1.23	<0.40	<1.63	1.6	>75.0
0323-0415 hrs								
Product Gas Boiler Run #2 2/25	18.389	0.5378		0.89	0.005	0.89	1.5	99.4
0618-0658 hrs								
Cleaned Top Gas Run #1 2/27	4.104	0.1200		33.4	0.39	33.8	255	98.8
1015-1050 hrs								
Cleaned Top Gas Run #2 2/27	3.958	0.1158		32.3	0.14	32.4	254	99.6
1617-1652 hrs								
Cleaned Top Gas Run #3 2/27	3.373	0.0986		19.2	0.36	19.6	180	98.2
2028-2040 hrs								
Raw Top Gas Run #1 2/27	3.938	0.1152		34.9	0.38	35.3	278	98.9
1033-1043 hrs								
Raw Top Gas Run #2 2/27	2.720	0.0795		20.7	0.28	21.0	239	98.7
1348-1358 hrs								
Raw Top Gas Run #3 2/27	3.784	0.1107		26.8	0.31	27.1	222	98.9
1643-1703 hrs								
ESP Oil Boiler Run #1 3/3	19.107	0.5588		0.93	1.42	2.35	3.8	39.6
0945-1047								
ESP Oil Boiler Run #2 3/3	17.968	0.5255		2.88	0.43	3.31	5.7	87.0
1626-1736								

(Continued)

TABLE A-3. (Continued)

Description	Sample Volume		1st Imp mg NH ₃	2nd Imp mg NH ₃	Total NH ₃	ppm NH ₃	Collection Efficiency (%)
	SCF 60°F, 1 atm.	Nm ³ 25°C 1 atm.					
Cleaned Bottom Gas Run #1 3/5 1417-1434 hrs	4.550	0.1331	6.67	0.04	6.71	45.7	99.4
Cleaned Bottom Gas Run #2 3/5 1905-1919 hrs	3.976	0.1163	4.98	<0.37	<5.35	38.8	>93.1
Cleaned Bottom Gas Run #3 3/6 0117-0127 hrs	4.091	0.1196	2.25	<0.38	<2.63	17.0	>85.6
Raw Bottom Gas Run #1 3/5 1440-1504 hrs	3.726	0.1090	5.07	0.03	5.10	42.4	98.2
Raw Bottom Gas Run #2 3/5 2052-2106 hrs	4.022	0.1176	4.90	0.05	4.95	38.2	98.9
Raw Bottom Gas Run #3 3/6 0057-0111 hrs	3.704	0.1083	4.46	0.02	4.48	37.5	99.6

TABLE A-4. UMD GASIFICATION FACILITY: SAMPLING DATA FOR
GASEOUS TRACE METAL SAMPLES

Stream	Sample Volume	
	SCF 600°F 1 atm.	Nm ³ 25°C 1 atm.
Combined Product Gas 2/25 0125-0415 hrs	50.9	1.49
Product Gas Boiler 2/25 0057-0701 hrs	548.0	16.0
Cleaned Top Gas 2/27 1116-1615 hrs	42.2	1.23
Raw Top Gas 2/27 1102-0332 hrs	41.6	1.22
ESP Oil Boiler 2/27 1052-1730 hrs	573.0	16.8
Cleaned Bottom Gas 3/5 1456-1856 hrs	54.3	1.59
Raw Bottom Gas 3/5 1534-2040 hrs	47.7	1.39

TABLE A-5. UMD GASIFICATION FACILITY: SAMPLING DATA FOR
GASEOUS ORGANIC SAMPLES

Stream	Sample Volume	
	SCF 600°F 1 atm.	Nm ³ 25°C 1 atm.
Combined Product Gas 2/25 0647-0927 hrs	43.6	1.28
Product Gas Boiler 2/25 0057-0701 hrs	548.0	16.0
Cleaned Top Gas 2/27 1802-2022 hrs	52.9	1.55
Raw Top Gas 2/27 1435-1635 hrs	27.4	0.80
ESP Oil Boiler 2/27 1052-1730 hrs	573.0	16.8
Cleaned Bottom Gas 3/5 2004-0056 hrs	100.0	2.93
Raw Bottom Gas 3/5 2137-0046 hrs	46.9	1.37

TABLE A-6. UMD GASIFICATION FACILITY: SAMPLING DATA FOR
GASEOUS NO_x SAMPLES

Description	Sample Volume		μg NO ₂	ppm NO _x
	SCF 60°F, 1 atm.	Nm ³ 25°C, 1 atm.		
Product Gas Boiler 2/26 Bomb 21 0525 hrs	0.0497	0.001454	932	341
Product Gas Boiler 2/26 Bomb 34 0525 hrs	0.0500	0.001462	750	273
Product Gas Boiler 2/27 Bomb 28 1005 hrs	0.0501	0.001465	571	207
Boiler #3 ESP Oil 3/3 Bomb 28 1522 hrs	0.0509	0.001489	1,120	400
Boiler #3 ESP Oil 3/3 Bomb 34 1522 hrs	0.0502	0.001468	1,240	449
Boiler #3 ESP Oil 3/3 Bomb 27 1910 hrs	0.0478	0.001398	1,115	424
Boiler #3 ESP Oil 3/3 Bomb 21 1910 hrs	0.0477	0.001395	1,045	399
Boiler #3 Natural Gas 3/4 Bomb 24 1022 hrs	0.0466	0.001363	277	108
Boiler #3 Natural Gas 3/4 Bomb 26 1022 hrs	0.0442	0.001293	272	112
Boiler #3 Natural Gas 3/4 Bomb 20 1127 hrs	0.0432	0.001263	264	111
Boiler #3 Natural Gas 3/4 Bomb 22 1127 hrs	0.0443	0.001296	260	107

TABLE A-7. UMD GASIFICATION FACILITY: SAMPLING DATA FOR GASEOUS SO_x SAMPLES

Description	Sample Volume			1st Imp Mg SO ₂	2nd Imp Mg SO ₂	Total Mg SO ₂	ppm SO ₂	Collection Efficiency (%)
	SCF 60°F 1 atm.	Nm ³ 25°C, 1 atm.						
Product Gas Boiler Run #1 2/25 0422-0500 hrs	16.159	0.4726		336.5	8.07	344.6	279	97.6
Product Gas Boiler Run #2 2/25 0723-0800 hrs & 0915-0945 hrs	26.181	0.7657		510.1	10.2	520.3	260	98.0
Boiler #3 ESP Oil Run #1 3/3 1209-1251 hrs	15.253	0.4461		183.1	1.61	184.7	158	99.1
Boiler #3 ESP Oil Run #2 3/3 1358-1430 hrs	11.711	0.3425		139.7	1.06	140.8	157	99.2
Boiler #3 ESP Oil Run #3 3/3 1601-1623 hrs	9.314	0.2724		95.5	1.00	96.5	135	98.9
Boiler #3 ESP Oil Run #4 3/3 1739-1810 hrs	5.683	0.1662		59.6	1.31	60.9	140	97.8
Boiler #3 Natural Gas Run #1 3/4 1115-1145 hrs	12.945	0.3786		4.04	0.68	4.72	4.8	85.6
Boiler #3 Natural Gas Run #2 3/4 1148-1218 hrs	11.637	0.3403		2.90	0.35	3.25	3.7	89.2
Boiler #3 Natural Gas Run #3 3/4 1233-1303 hrs	10.005	0.2926		1.61	0.36	1.97	2.6	81.7

TABLE A-8. SAMPLE LOGBOOK

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-001	02-24-81	Wastewater	Ash Pan Water 1100
UMD-002	02-24-81	Gas	Combined Gas - CN Run #1 Imp. #1
UMD-003	02-24-81	Gas	Combined Gas - CN Run #1 Imp. #2
UMD-004	02-24-81	Gas	Combined Gas NH ₃ Run #1 Imp. #1
UMD-005	02-24-81	Gas	Combined Gas NH ₃ Run #1 Imp. #2
UMD-006	02-25-81	Gas	Main Stack CN Run #1 Imp. #1
UMD-007	02-25-81	Gas	Main Stack CN Run #1 Imp. #2
UMD-008	02-25-81	Gas	Main Stack NH ₃ Run #1 Imp. #1
UMD-009	02-25-81	Gas	Main Stack NH ₃ Run #1 Imp. #2
UMD-010	02-25-81	Gas	Combined Gas CN Run #2 Imp. #1
UMD-011	02-25-81	Gas	Combined Gas CN Run #2 Imp. #2
UMD-012	02-25-81	Gas	Combined Gas NH ₃ Run #2 Imp. #1
UMD-013	02-25-81	Gas	Combined Gas NH ₃ Run #2 Imp. #2
UMD-014	02-25-81	Gas	SO ₂ Run #1 Imp. #1 Main Stack
UMD-015	02-25-81	Gas	Main Stack SO ₂ Run #1 Imp. #2
UMD-016	02-25-81	Gas	Main Stack CN Run #2 Imp. #1
UMD-017	02-25-81	Gas	Main Stack CN Run #2 Imp. #2
UMD-018	02-25-81	Wastewater	Ash Pan Composite--See also 056,304 (02-24-81) 1600, 2020 hrs (02-25-81) 0200, 0620 hrs
UMD-019	02-25-81	Filter	Main Stack SASS #1
UMD-020	02-25-81	Gas	Combined Gas CN Run #3 Imp. #1
UMD-021	02-25-81	Gas	Combined Gas CN Run #3 Imp. #2
UMD-022	02-25-81	Gas	Main Stack NH ₃ Run #2 Imp. #1
UMD-023	02-25-81	Gas	Main Stack NH ₃ Run #2 Imp. #1
UMD-024	02-25-81	Gas	Combined Gas Run #3 NH ₃ Imp. #1
UMD-025	02-25-81	Gas	Combined Gas NH ₃ Run #3 Imp. #2
UMD-026	02-25-81	Gas	Main Stack SO ₂ Run #2 Imp. #1
UMD-027	02-25-81	Gas	Main Stack SO ₂ Run #2 Imp. #2
UMD-028	02-24-81	Solid	Coal - 1030 hrs
UMD-029	02-24-81	Solid	Coal - 1630 hrs
UMD-030	02-24-81	Solid	Coal - 1945 hrs
UMD-031	02-25-81	Solid	Coal - 0145 hrs
UMD-032	02-25-81	Solid	Coal - 1615 hrs
UMD-033	--	--	No sample
UMD-034	02-27-81	Gas	Cleaned Top Gas Cn Run #1 Imp. #1
UMD-035	02-27-81	Gas	Cleaned Top Gas CN Run #1 Imp. #2
UMD-036	02-27-81	Gas	Raw Top Gas CN Run #1 Imp. #1
UMD-037	02-27-81	Gas	Raw Top Gas CN Run #1 Imp. #2
UMD-038	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #1 Imp. #1
UMD-039	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #1 Imp. #2

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-040	02-27-81	Gas	Raw Top Gas NH ₃ Run #1 Imp. #1
UMD-041	02-27-81	Gas	Raw Top Gas NH ₃ Run #1 Imp. #2
UMD-042	02-27-81	Gas	Raw Top Trace Metals Imp. #1
UMD-043	02-27-81	Gas	Raw Top Trace Metals Imp. #2
UMD-044	02-27-81	Gas	Raw Top Gas CN Run #2 Imp. #1
UMD-045	02-27-81	Gas	Raw Top Gas CN Run #2 Imp. #2
UMD-046	02-27-81	Gas	Raw Top Gas NH ₃ Run #2 Imp. #1
UMD-047	02-27-81	Gas	Raw Top Gas NH ₃ Run #2 Imp. #2
UMD-048	02-27-81	Gas	Raw Top Trace Metals Imp. Rinse Methylene Chloride
UMD-049	02-25-81	Gas	Combined Gas Trace Metals Imp. #1
UMD-050	02-27-81	Gas	Combined Gas Trace Metals Imp. #2
UMD-051	02-25-81	Gas	Combined Gas Metals Impingers MeCl ₂ Rinse
UMD-052	02-27-81	Gas	Cleaned Top Gas CN Run #2 Imp. #1
UMD-053	02-27-81	Gas	Cleaned Top Gas CN Run #2 Imp. #2
UMD-054	02-27-81	Gas	Cleaned Top Gas Trace Metals Imp. #1
UMD-055	02-27-81	Gas	Cleaned Top Gas Trace Metals Imp. #2
UMD-056	02-27-81	Water	Ash Pan Water Composite Grabs Taken at 11:15, 14:20, 17:35 hours.
UMD-057	02-27-81	Gas	Raw Top Gas CN Run #3 Imp. #1
UMD-058	02-27-81	Gas	Raw Top Gas CN Run #3 Imp. #2
UMD-059	02-27-81	Gas	Cleaned Top Gas CN Run #3 Imp. #1
UMD-060	02-27-81	Gas	Cleaned Top Gas CN Run #3 Imp. #2
UMD-061	02-27-81	Gas	Raw Top Gas NH ₃ Run #3 Imp. #1
UMD-062	02-27-81	Gas	Raw Top Gas NH ₃ Run #3 Imp. #2
UMD-063	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #3 Imp. #1
UMD-064	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #3 Imp. #2
UMD-065	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #2 Imp. #1
UMD-066	02-27-81	Gas	Cleaned Top Gas NH ₃ Run #2 Imp. #2
UMD-067	02-25-81	Gas	SASS #1 Condensate, Main Stack
UMD-068	02-25-81	Gas	SASS #1, 1st Impinger, Main Stack
UMD-069	02-25-81	Gas	SASS #1, 2nd & 3rd Impingers, Main Stack
UMD-070	02-25-81	Gas	SASS #1, Resin & Rinse #1, Main Stack
UMD-071	02-25-81	Gas	Organics, Resin & Rinse, Combined Gas
UMD-072	02-24-81	Gas	Trace Metals, Filter 2.2830 gm, Combined Gas
UMD-073	02-24-81	Gas	Organics, NH ₃ , CN, Filter 2.1039, Combined Gas
	02-25-81		

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-074	02-24-81	Gas	Particle Load, Filter 2.6364, Main Stack
	02-25-81		
UMD-075	02-25-81	Gas	SASS #1, Condenser Rinse Main Stack
UMD-076	02-25-81	Gas	SASS #1, Probe Cyclone Filter Rinse, Main Stack
UMD-077	02-25-81	Gas	Organics, Condenser Rinse, Combined Gas
UMD-078	02-25-81	Gas	SASS #1, Resin & Rinse #2, Main Stack
UMD-079	02-25-81	Tar	ESP Tar Grab, 11:15 hrs
UMD-080	02-25-81	Solid	Gasifier Ash, 11:05 hrs
UMD-081	03-03-81	Gas	Stub Stack CN Run #1 Imp. #1
UMD-082	03-03-81	Gas	Stub Stack CN Run #1 Imp. #2
UMD-083	03-03-81	Gas	Stub Stack SO ₂ Run #1 Imp. #1
UMD-084	03-03-81	Gas	Stub Stack SO ₂ Run #1 Imp. #2
UMD-085	03-03-81	Gas	Stub Stack NH ₃ Run #1 Imp. #1
UMD-086	03-03-81	Gas	Stub Stack NH ₃ Run #1 Imp. #2
UMD-087	03-03-81	Gas	Stub Stack SO ₂ Run #2 Imp. #1
UMD-088	03-03-81	Gas	Stub Stack SO ₂ Run #2 Imp. #2
UMD-089	03-03-81	Gas	Stub Stack SO ₂ Run #2 Imp. #1
UMD-090	03-03-81	Gas	Stub Stack SO ₂ Run #2 Imp. #2
UMD-091	03-03-81	Gas	Stub Stack NH ₃ Run #2 Imp. #1
UMD-092	03-03-81	Gas	Stub Stack NH ₃ Run #2 Imp. #2
UMD-093	03-03-81	Gas	Stub Stack CN Run #2 Imp. #1
UMD-094	03-03-81	Gas	Stub Stack CN Run #2 Imp. #2
UMD-095	03-03-81	Gas	Stub Stack SASS Run Imp. #1 and Wash
UMD-096	03-03-81	Gas	Stub Stack SASS Run Imp. #2
UMD-097	03-03-81	Gas	Stub Stack SO ₂ Run #4 Imp. #1
UMD-098	03-03-81	Gas	Stub Stack SO ₃ Run #4 Imp. #2
UMD-099	02-24-81	Tar	ESP Tar 1100 hrs
UMD-100	02-24-81	Tar	ESP Tar 1640 hrs
UMD-101	02-24-81	Tar	ESP Tar 2005 hrs
UMD-102	02-24-81	Tar	ESP Tar 0155 hrs
UMD-103	02-28-81	Tar	ESP Tar 0615 hrs
UMD-104	02-26-81	Tar	ESP Tar 1235 hrs
UMD-105	02-27-81	Tar	ESP Tar 1430 hrs
UMD-106	02-27-81	Tar	ESP Tar 1805 hrs
UMD-107	03-02-81	Tar	ESP Tar to Boiler 1615 hrs
UMD-108	03-02-81	Tar	ESP Tar to Boiler 1115 hrs
UMD-109	03-03-81	Tar	ESP Tar to Boiler 1240 hrs
UMD-110	03-03-81	Tar	ESP Tar to Boiler 1924 hrs
UMD-111	03-03-81	Tar	ESP Tar to Boiler 1140 hrs
UMD-112	02-24-81	Solid	Cyclone Dust 1050 hrs

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-113	02-24-81	Solid	Cyclone Dust 2000 hrs
UMD-114	02-24-81	Solid	Cyclone Dust 1345 hrs
UMD-115	02-25-81	Solid	Cyclone Dust app. 0100 hrs
UMD-116	02-25-81	Solid	Cyclone Dust app. 0100 hrs
UMD-117	02-26-81	Solid	Cyclone Dust 0630 hrs
UMD-118	02-26-81	Solid	Cyclone Dust 0240 hrs
UMD-119	02-27-81	Solid	Cyclone Dust 1120 hrs
UMD-120	02-27-81	Solid	Cyclone dust 1425 hrs
UMD-121	02-27-81	Solid	Cyclone Dust 1810 hrs
UMD-122	02-24-81	Solid	Gasifier Ash 1045 hrs
UMD-123	02-24-81	Solid	Gasifier Ash 1635 hrs
UMD-124	02-24-81	Solid	Gasifier Ash 2005 hrs
UMD-125	02-25-81	Solid	Gasifier Ash 0210 hrs
UMD-126	02-25-81	Solid	Gasifier Ash 0625 hrs
UMD-127	02-26-81	Solid	Gasifier Ash 0620 hrs
UMD-128	02-26-81	Solid	Gasifier Ash 1225 hrs
UMD-129	02-27-81	Solid	Gasifier Ash 1110 hrs
UMD-130	02-27-81	Solid	Gasifier Ash 1430 hrs
UMD-131	02-27-81	Solid	Gasifier Ash 1815 hrs
UMD-132	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #1 Imp. #1
UMD-133	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #1 Imp. #2
UMD-134	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #1 Imp. #1
UMD-135	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #2 Imp. #2
UMD-136	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #3 Imp. #1
UMD-137	03-04-81	Gas	#3 Boiler Natural Gas SO ₂ Run #3 Imp. #2
UMD-138	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #1 Imp. #1
UMD-139	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #1 Imp. #2
UMD-140	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #1 Imp. #1
UMD-141	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #1 Imp. #2
UMD-142	03-05-81	Gas	Raw Bottom Gas CN Run #1 Imp. #1
UMD-143	03-05-81	Gas	Raw Bottom Gas CN Run #1 Imp. #2
UMD-144	03-05-81	Gas	Cleaned Bottom Gas CN Run #1 Imp. #1
UMD-145	03-05-81	Gas	Cleaned Bottom Gas CN Run #1 Imp. #2
UMD-146	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #2 Imp. #1
UMD-147	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #2 Imp. #2

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-148	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #2 Imp. #1
UMD-149	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #2 Imp. #2
UMD-150	03-05-81	Gas	Raw Bottom Gas CN Run #2 Imp. #1
UMD-151	03-05-81	Gas	Raw Bottom Gas CN Run #2 Imp. #2
UMD-152	03-05-81	Gas	Cleaned Bottom Gas CN Run #2 Imp. #1
UMD-153	03-05-81	Gas	Cleaned Bottom Gas CN Run #2 Imp. #2
UMD-154	03-05-81	Gas	Cleaned Bottom Gas Trace Metals Imp. #1
UMD-155	03-05-81	Gas	Cleaned Bottom Gas Trace Metals Imp. #2
UMD-156	03-05-81	Gas	Raw Bottom Gas Trace Metals Imp. #1
UMD-157	03-05-81	Gas	Raw Bottom Gas Trace Metals Imp. #2
UMD-158	03-05-81	Water	Composite Ash Pan Water 1250, 1555, 2100 hrs
UMD-159	03-05-81	Gas	Cleaned Bottom Gas Trace Metals Imp. MeCl ₂ Rinse
UMD-160	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #3 Imp. #1
UMD-161	03-05-81	Gas	Raw Bottom Gas NH ₃ Run #3 Imp. #2
UMD-162	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #3 Imp. #1
UMD-163	03-05-81	Gas	Cleaned Bottom Gas NH ₃ Run #3 Imp. #2
UMD-164	03-05-81	Gas	Raw Bottom Gas CN Run #3 Imp. #1
UMD-165	03-05-81	Gas	Raw Bottom Gas CN Run #3 Imp. #2
UMD-166	03-05-81	Gas	Cleaned Bottom Gas CN Run #3 Imp. #1
UMD-167	03-05-81	Gas	Cleaned Bottom Gas CN Run #3 Imp. #2
UMD-168	02-25-81	Gas	Organics, Combined Gas, Probe and Filter Rinse
UMD-169	02-25-81	Gas	Organics, Combined Gas, Condensate
UMD-170	02-27-81	Gas	ESP Precipitate - Raw Top Gas
UMD-171	02-27-81	Gas	Cleaned Top Gas, Condensate/Organics
UMD-172	02-27-81	Gas	Raw Top Gas, Probe Rinse
UMD-173	02-27-81	Gas	Raw Top Gas, ESP Rinse
UMD-174	02-27-81	Gas	Cleaned Top Gas, Probe Rinse/Organics
UMD-175	02-27-81	Gas	Cleaned Top Gas, Condenser Rinse/Organics
UMD-176	02-27-81	Gas	Raw Top Gas, Condenser Rinse/Organics
UMD-177	02-27-81	Gas	Raw Top Gas, Resin/Organics
UMD-178	02-27-81	Gas	Raw Top Gas, Condensate/Organics
UMD-179	02-27-81	Gas	Cleaned Top Gas, XAD-2 Resin/Organics
UMD-180	02-27-81	Gas	Cleaned Top Gas, Filter/TM Run 2.9093 Tare
UMD-181	02-27-81	Gas	Cleaned Top Gas, Filter/Organics 2.8933 Tare
UMD-182	02-27-81	Gas	Cleaned Top Gas, XAD-2 Resin/Blank
UMD-183	02-27-81	Gas	Cleaned Top Gas, Impinger Rinse-MeCl ₂ /TM

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-184	03-03-81	Gas	Tar Stack, SASS Condensate 1 of 2
UMD-185	03-03-81	Gas	Tar Stack, SASS Condensate 2 of 2
UMD-186	03-03-81	Gas	Tar Stack, SASS Condenser Rinse
UMD-187	03-03-81	Gas	Tar Stack, SASS Probe Rinse
UMD-188	03-03-81	Gas	Tar Stack, SASS Probe Rinse
UMD-189	03-03-81	Gas	Tar Stack, SASS XAD Resin (Organics: 1 of 3)
UMD-190	03-03-81	Gas	Tar Stack, SASS Filter
UMD-191	03-03-81	Gas	Tar Stack, SASS Thimble
UMD-192	03-03-81	Gas	Tar Stack, SASS XAD Resin (2 of 3)
UMD-193	03-03-81	Gas	Tar Stack, SASS XAD Resin (3 of 3)
UMD-194	03-05-81	Gas	Cleaned Bottom Gas, Condensate/Organics
UMD-195	03-05-81	Gas	Raw Bottom Gas, Condenser Rinse/ Organics
UMD-196	03-05-81	Gas	Cleaned Bottom Gas
UMD-197	03-05-81	Gas	Raw Bottom Gas, XAD Resin Blank
UMD-198	03-05-81	Gas	Cleaned Bottom Gas, XAD
UMD-199	03-05-81	Gas	Cleaned Bottom Gas, Condenser Rinse
UMD-200	03-05-81	Gas	Raw Bottom Gas, XAD
UMD-201	03-06-81	Gas	Raw Bottom Gas #1, Nozzle and Probe Rinse
UMD-202	03-06-81	Gas	Cleaned Bottom Gas #14, Nozzle and Probe Rinse
UMD-203	03-06-81	Solid	Feed Coal 2040 hrs
UMD-204	02-26-81	Solid	Feed Coal 1250 hrs
UMD-205	02-26-81	Solid	Feed Coal 1810 hrs
UMD-206	03-05-81	Solid	Feed Coal 1240 hrs
UMD-207	03-05-81	Solid	Feed Coal 1540 hrs
UMD-208	03-05-81	Gas	XAD Resin Blank
UMD-209	03-05-81	Gas	Raw Bottom Gas Thimble
UMD-210	03-05-81	Gas	Cleaned Bottom Gas Thimble
UMD-211	03-05-81	Tar	ESP Tars 1550 hrs
UMD-212	03-05-81	Tar	ESP Tars 1245 hrs
UMD-213	03-05-81	Tar	ESP Tars 2045 hrs
UMD-214	03-05-81	Solid	Cyclone Dust 2055 hrs
UMD-215	03-05-81	Solid	Cyclone Dust 1600 hrs
UMD-216	03-05-81	Solid	Cyclone Dust 1255 hrs
UMD-217	03-05-81	Solid	Gasifier Ash 1300 hrs
UMD-218	03-05-81	Solid	Gasifier Ash 1603 hrs
UMD-219	03-05-81	Solid	Gasifier Ash 2100 hrs
UMD-220	03-05-81	Water	Service Water 1535 hrs

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-221		Gas	Trace Metals SASS Blank .2M Ammonium Persulfate, .02M Silver Nitrate
UMD-222		Gas	SO ₂ Blank 6% H ₂ O ₂
UMD-223	02-25-81	Gas	Main Stack Hg #1
UMD-224	02-25-81	Gas	Main Stack Hg #2
UMD-225	02-25-81	Gas	Combined Gas Hg #1
UMD-226	02-25-81	Gas	Combined Gas Hg #2
UMD-227	02-25-81	Gas	Tar Stack Hg #1
UMD-228	03-03-81	Gas	Tar Stack Hg #2
UMD-229	02-27-81	Solid	Coal 1035 hrs
UMD-230	02-27-81	Solid	Coal 1350 hrs
UMD-231	02-27-81	Solid	Coal 1825 hrs
UMD-232		Gas	Raw Top Gas, Thimble behind ESP 4.5878 Tare
UMD-233		Gas	Raw Bottom Gas, Condensate/Organics 26 mL
UMD-301	Aug. 81	Coal Composite	Composite Sample - Coal, 18.0 g dry, pulverized UMD-030, + 18.0 g dry, pulverized UMD-031, + 18.0 g dry, pulverized UMD-032
UMD-302	Aug. 81	Coal Composite	Composite Sample - Coal, 18.0 g UMD-229, + 18.0 g UMD-230, + 18.0 g UMD-231
UMD-303	Aug. 81	Coal Composite	Composite Sample - Coal, 18.0 g UMD-206 18.0 g UMD-207 18.0 g UMD-203
UMD-304	Aug. 81	Solids	Ash Pan Water 018,056 Filtered to Yield Solids Weighing 0.41478 g. Solids removed by Filtration (Millipore® Polycarbonate)
UMD-305	Aug. 81	Tar Composite	45 g UMD-105 + 45 g UMD-106
UMD-306	Aug. 81	Tar Composite	30 g UMD-212 + 30 g UMD-211 + 30 g UMD-213
UMD-307	Aug. 81	Tar Composite	40.5 g UMD-108 + 40.5 g UMD-109 + 40.5 g UMD-110
UMD-308	Aug. 81	Tar Composite	30.0 g UMD-100 + 30.0 g UMD-101 + 30.0 g UMD-102

(Continued)

TABLE A-8. (Continued)

Sample Number	Date	Sample Type	Sample Description or Parameter
UMD-309	Aug. 81	Cyclone Dust Composite	20 g each of UMD-114, 113, and 115/116
UMD-310	Aug. 81	Cyclone Dust Composite	20 g each of UMD-119, 120, 121
UMD-311	Aug. 81	Cyclone Dust Composite	20 g each of UMD-216, 215, 214
UMD-312	09-16-81	Gasifier Ash Composite	20 g each of UMD-124, 125, 126
UMD-313	09-16-81	Gasifier Ash Composite	20 g each of UMD-129, 130, 131
UMD-314	09-16-81	Gasifier Ash Composite	20 g each of UMD-217, 218, 219
UMD-315	10-01-81	Coal Composite	Composite 20 g each of UMD-030, 031, 032, 229, 230, 231
UMD-316	10-01-81	Cyclone Dust Composite	Composite 20 g each of UMD-113, 114, 115/116, 119, 120, 121
UMD-317	10-01-81	Gasifier Ash Composite	Composite 20 g each of UMD-124, 125, 126, 129, 130, 131
UMD-318	10-01-81	Impinger Composite	Trace Metals Combined 17, UMD-051 Digested and added to UMD-049 Composite UMD-049 (with 051) 232 mL, 1/2 of total and UMD-050 72 mL, 1/2 of total
UMD-319	10-01-81	Impinger Composite	SASS #1 Main Stack UMD-068 239 mL, 1/4 of total; UMD-069 64 mL, 1/4 of total
UMD-320	10-01-81	Impinger Composite	Stub Stack SASS — UMD-095 282 mL 1/4 of total UMD-096 74 mL 1/4 of total Main Stack
UMD-321	10-02-81	Solvent Extract Composite	SASS #1, MeCl_2 + Et_2O UMD-070 Resin & Rinse #1 UMD-078 Resin & Rinse #2
UMD-322	10-02-81	Solvent Extract Composite	Tar Stack SASS MeCl_2 + Et_2O UMD-189 XAD Resin/Organics UMD-192 XAD Resin/Organics UMD-193 XAD Resin/Organics

APPENDIX B

ANALYTICAL RESULTS

TABLE B-1. COAL MOISTURE

Sample		Concentration ¹
Date	Time	%
2-24	1945	11.9
2-25	0145	11.2
2-25	0615	11.4
2-27	1035	11.0
2-27	1550	10.3
2-27	1825	10.8
3-5	1240	10.6
3-5	1540	11.1
3-5	2040	10.6
AVERAGE		11.0 ± 0.5

¹Moisture determined by drying sample at 110°C and calculating weight loss.

TABLE B-2. GASIFIER ASH MOISTURE

Sample		Concentration ¹
Date	Time	%
2-24	2005	23.1
2-25	0210	26.4
2-25	0625	24.9
2-27	1110	17.5
2-27	1430	21.3
2-27	1815	24.6
3-5	1300	28.1
3-5	1603	31.3
3-5	2100	27.0
AVERAGE		24.9 ± 4.0

¹Moisture determined by drying sample at 110°C and calculating weight loss.

TABLE B-3. ESP OIL DENSITY

Composite Samples		Density ¹
Date	Time	g/mL
2-24	1640	0.905
2-24	2005	
2-25	0155	
2-27	1430	0.912
2-27	1805	
3-3	1115	0.911
3-3	1249	
3-3	1825	
3-5	1245	0.925
3-5	1550	
3-5	2045	
AVERAGE		0.913 $\pm$ 0.008

¹Measured at 38°C (100°F).

Composite sample.

TABLE B-4. RESULTS OF SSMS ANALYSIS OF XAD-2 RESIN BLANK

COMMERCIAL TESTING & ENGINEERING CO.

Reply to

GENERAL OFFICES: 328 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 738-8434
INSTRUMENTAL ANALYSIS DIVISION, 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401. PHONE: 303-278-9321

To: Mr. Michael R. Fuchs
Radian Corporation
P. O. Box 9948
Austin, TX 78766



Date: October 19, 1981

Analyst: J. Oldham

P. O. No.: 8075

Sample No.: 208

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN $\mu\text{g/mL}$

IAD No. 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	<0.001
Thorium		Gadolinium		Molybdenum	0.005	Titanium	0.05
Bismuth		Europium		Niobium	<0.002	Scandium	<0.001
Lead	0.008	Samarium		Zirconium	<0.003	Calcium	0.7
Thallium		Neodymium		Yttrium	<0.003	Potassium	MC
Mercury	NR	Praseodymium		Strontium	0.005	Chlorine	1
Gold		Cerium		Rubidium	<0.001	Sulfur	1
Platinum		Lanthanum		Bromine	0.03	Phosphorus	0.02
Iridium		Barium	0.3	Selenium	0.006	Silicon	MC
Osmium		Cesium		Arsenic	0.002	Aluminum	0.01
Rhenium		Iodine	<0.001	Germanium		Magnesium	0.09
Tungsten		Tellurium		Gallium	0.001	Sodium	>2
Tantalum		Antimony		Zinc	0.08	Fluorine	=0.003
Hafnium		Tin		Copper	0.01	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.002	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.001	Carbon	NR
Thulium		Silver		Iron	0.02	Boron	<0.001
Erbium		Palladium		Manganese	<0.001	Beryllium	
Holmium		Rhodium		Chromium	0.003	Lithium	0.001
Dysprosium						Hydrogen	NR

STD – Internal Standard

NR – Not Reported

All elements not detected < 0.001 µg/mL

MC – Major Component > 10 µg/mL

INT - Interference

Approved:

TABLE B-5. RESULTS OF SSMS ANALYSIS OF TRACE METALS IMPINGERS SOLUTION

COMMERCIAL TESTING & ENGINEERING CO.GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 - AREA CODE 312 728-8434
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401 PHONE 303-278-9531

Reply to

To: Mr. Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766

Date: October 26, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #221

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No.: 97-H492-113-12

CONCENTRATION IN $\mu\text{g/ml}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	<0.001
Thorium		Gadolinium		Molybdenum		Titanium	0.06
Bismuth		Europium		Niobium		Scandium	<0.002
Lead	0.01	Samarium		Zirconium	0.01	Calcium	0.6
Thallium		Neodymium		Yttrium	<0.002	Potassium	3
Mercury	NR	Praseodymium		Strontium	0.009	Chlorine	1
Gold		Cerium		Rubidium		Sulfur	MC
Platinum		Lanthanum		Bromine	0.2	Phosphorus	0.3
Iridium		Barium	0.009	Selenium	<0.01	Silicon	7
Osmium		Cesium		Arsenic		Aluminum	0.2
Rhenium		Iodine	0.002	Germanium		Magnesium	0.02
Tungsten		Tellurium		Gallium	<0.002	Sodium	0.6
Tantalum		Antimony		Zinc	0.03	Fluorine	=3
Hafnium		Tin		Copper	0.6	Oxygen	NR
Lutetium		Indium	STD	Nickel	*0.3	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.001	Carbon	NR
Thulium		Silver	MC	Iron	*MC	Boron	0.002
Erbium		Palladium		Manganese	*0.04	Beryllium	
Holmium		Rhodium		Chromium	0.2	Lithium	<0.001
Dysprosium		* Heterogeneous				Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.002 $\mu\text{g/ml}$ MC - Major Component > 10 $\mu\text{g/ml}$

INT - Interference

Approved:

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TABLE B-6. RESULTS OF SSMS ANALYSIS OF ASH PAN WATER COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 - AREA CODE 312 726-8434
INSTRUMENTAL ANALYSIS DIVISION, 14325 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9521

Reply to

To: Mr. Michael R. Fuchs
Radian Corporation
P. O. Box 9948
Austin, TX 78766



Date: October 19, 1981

Analyst: J. Oldham

P. O. No.: 8075

Sample No.: 304

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No. 97-H492-113-12

CONCENTRATION IN $\mu\text{g/mL}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium	<0.01	Terbium		Ruthenium		Vanadium	0.001
Thorium	<0.01	Gadolinium		Molybdenum	0.02	Titanium	0.2
Bismuth		Europium		Niobium	0.002	Scandium	<0.002
Lead	0.01	Samarium		Zirconium	0.01	Calcium	MC
Thallium		Neodymium		Yttrium	0.05	Potassium	6
Mercury	NR	Praseodymium		Strontium	MC	Chlorine	0.7
Gold		Cerium		Rubidium	0.03	Sulfur	>9
Platinum		Lanthanum		Bromine	0.1	Phosphorus	0.1
Iridium		Barium	4	Selenium	0.04	Silicon	0.4
Osmium		Cesium	<0.001	Arsenic	0.03	Aluminum	0.03
Rhenium		Iodine	0.007	Germanium		Magnesium	0.3
Tungsten		Tellurium	0.004	Gallium	0.02	Sodium	0.2
Tantalum		Antimony		Zinc	0.04	Fluorine	=0.05
Hafnium		Tin		Copper	0.02	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.002	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	0.005	Carbon	NR
Thulium		Silver		Iron	9	Boron	0.4
Erbium		Palladium		Manganese	<0.001	Beryllium	
Holmium		Rhodium		Chromium	0.04	Lithium	0.2
Dysprosium						Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.001 $\mu\text{g/mL}$

MC - Major Component > 10 $\mu\text{g/mL}$

INT - Interference

Approved: *[Signature]*

TABLE B-7. RESULTS OF SSMS ANALYSIS OF ESP TAR TO BOILER NO. 3 COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 - AREA CODE 312 728-6434
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9321

Reply to



To: Mr. Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766

Date: October 20, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #307

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN PPM WEIGHT

IAD No. 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	0.2
Thorium		Gadolinium		Molybdenum	0.02	Titanium	0.6
Bismuth	0.03	Europium		Niobium		Scandium	<0.01
Lead	0.4	Samarium		Zirconium	0.08	Calcium	12
Thallium		Neodymium		Yttrium		Potassium	3
Mercury	NR	Praseodymium	<0.01	Strontium	0.4	Chlorine	0.3
Gold		Cerium	0.06	Rubidium	0.02	Sulfur	8
Platinum		Lanthanum	0.05	Bromine	0.1	Phosphorus	3
Iridium		Barium	4	Selenium	<0.03	Silicon	4
Osmium		Cesium		Arsenic	0.8	Aluminum	*5
Rhenium		Iodine	0.06	Germanium	<0.02	Magnesium	0.2
Tungsten		Tellurium		Gallium	0.07	Sodium	2
Tantalum		Antimony	<0.01	Zinc	*MC	Fluorine	=4
Hafnium		Tin	0.04	Copper	0.07	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.09	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.01	Carbon	NR
Thulium		Silver	*0.2	Iron	MC	Boron	0.02
Erbium		Palladium		Manganese	0.6	Beryllium	
Holmium		Rhodium		Chromium	0.6	Lithium	0.03
Dysprosium			* Heterogeneous			Hydrogen	NR

STD - Internal Standard
NR - Not Reported
All elements not detected < 0.01ppm
MC - Major Component > 100ppm
INT - Interference

Approved: *[Signature]*
[Signature]

TABLE B-8. RESULTS OF SSMS ANALYSIS OF COAL COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 238 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 • AREA CODE 312 726-8434
INSTRUMENTAL ANALYSIS DIVISION, 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9521

Reply to

To: Mr. Michael R. Fuchs
Radian Corporation
P. O. Box 9948
Austin, TX 78766



Date: October 19, 1981

Analyst: J. Oldham

P. O. No.: 8075

Sample No.: 315

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN PPM WEIGHT

IAD No.: 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium	3	Terbium		Ruthenium		Vanadium	14
Thorium	4	Gadolinium		Molybdenum	2	Titanium	890
Bismuth		Europium	0.2	Niobium	4	Scandium	1
Lead	3	Samarium	2	Zirconium	31	Calcium	MC
Thallium		Neodymium	0.9	Yttrium	14	Potassium	MC
Mercury	NR	Praseodymium	2	Strontium	MC	Chlorine	27
Gold		Cerium	11	Rubidium	4	Sulfur	MC
Platinum		Lanthanum	9	Bromine	2	Phosphorus	MC
Iridium		Barium	MC	Selenium	≤0.6	Silicon	MC
Osmium		Cesium	0.3	Arsenic	2	Aluminum	MC
Rhenium		Iodine	≤0.4	Germanium	≤0.2	Magnesium	870
Tungsten		Tellurium	0.4	Gallium	5	Sodium	>480
Tantalum		Antimony	0.9	Zinc	3	Fluorine	≈7
Hafnium		Tin	0.4	Copper	5	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.8	Nitrogen	NR
Ytterbium		Cadmium	1	Cobalt	1	Carbon	NR
Thulium		Silver	0.8	Iron	MC	Boron	8
Erbium		Palladium		Manganese	24	Beryllium	0.3
Holmium		Rhodium		Chromium	2	Lithium	12
Dysprosium						Hydrogen	NR

STD - Internal Standard
NR - Not Reported
All elements not detected < 0.2 ppm
MC - Major Component > 1,000 ppm
INT - Interference

Approved: *[Signature]*
10/20/81

TABLE B-9. RESULTS OF SSMS ANALYSIS OF CYCLONE DUST COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 338 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 - AREA CODE 312 738-8434
INSTRUMENTAL ANALYSIS DIVISION: 14325 WEST 44TH AVENUE, GOLDEN, COLORADO 80401. PHONE: 303-278-9521

Reply to

To: Mr. Michael R. Fuchs
Radian Corporation
P. O. Box 9948
Austin, TX 78766



Date: October 19, 1981

Analyst: J. Oldham

P. O. No.: 8075

Sample No.: 316

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN PPM WEIGHT

IAD No.: 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium	7	Terbium	0.8	Ruthenium		Vanadium	30
Thorium	8	Gadolinium	2	Molybdenum	9	Titanium	MC
Bismuth	4	Europium	1	Niobium	10	Scandium	1
Lead	870	Samarium	4	Zirconium	55	Calcium	MC
Thallium	12	Neodymium	2	Yttrium	16	Potassium	MC
Mercury	NR	Praseodymium	5	Strontium	760	Chlorine	86
Gold		Cerium	28	Rubidium	9	Sulfur	MC
Platinum		Lanthanum	20	Bromine	7	Phosphorus	MC
Iridium		Barium	MC	Selenium	10	Silicon	MC
Osmium		Cesium	0.7	Arsenic	40	Aluminum	MC
Rhenium		Iodine	4	Germanium	19	Magnesium	MC
Tungsten	4	Tellurium	1	Gallium	280	Sodium	>480
Tantalum		Antimony	4	Zinc	560	Fluorine	MC
Hafnium	<2	Tin	77	Copper	64	Oxygen	NR
Lutetium		Indium	STD	Nickel	3	Nitrogen	NR
Ytterbium		Cadmium	3	Cobalt	1	Carbon	NR
Thulium		Silver	<2	Iron	MC	Boron	400
Erbium	0.8	Palladium		Manganese	97	Beryllium	0.8
Holmium	1	Rhodium		Chromium	220	Lithium	27
Dysprosium	4					Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.2 ppm

MC - Major Component > 1,000 ppm

INT - Interference

Approved:

TABLE B-10. RESULTS OF SSMS ANALYSIS OF GASIFIER ASH COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 238 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 736-8484
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9531

Reply to

To: Mr. Michael R. Fuchs
Radian Corporation
P. O. Box 9948
Austin, TX 78766



Date: October 19, 1981

Analyst: J. Oldham

P. O. No.: 8075

Sample No.: 317

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN PPM WEIGHT

IAD No.: 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium	19	Terbium	1	Ruthenium		Vanadium	68
Thorium	29	Gadolinium	4	Molybdenum	3	Titanium	MC
Bismuth	<0.1	Europium	1	Niobium	*38	Scandium	2
Lead	12	Samarium	6	Zirconium	*200	Calcium	MC
Thallium	0.4	Neodymium	19	Yttrium	50	Potassium	MC
Mercury	NR	Praseodymium	16	Strontium	MC	Chlorine	10
Gold		Cerium	99	Rubidium	59	Sulfur	>420
Platinum		Lanthanum	180	Bromine	3	Phosphorus	MC
Iridium		Barium	MC	Selenium	0.8	Silicon	MC
Osmium		Cesium	8	Arsenic	1	Aluminum	MC
Rhenium		Iodine	0.3	Germanium	0.7	Magnesium	MC
Tungsten	2	Tellurium	0.2	Gallium	23	Sodium	>170
Tantalum	0.6	Antimony	1	Zinc	2	Fluorine	≈110
Hafnium	2	Tin	0.5	Copper	23	Oxygen	NR
Lutetium	0.3	Indium	STD	Nickel	3	Nitrogen	NR
Ytterbium	2	Cadmium	5	Cobalt	1	Carbon	NR
Thulium	0.2	Silver	≤0.3	Iron	MC	Boron	64
Erbium	3	Palladium		Manganese	230	Beryllium	0.9
Holmium	4	Rhodium		Chromium	8	Lithium	64
Dysprosium	6	*Heterogeneous				Hydrogen	NR

STD - Internal Standard
NR - Not Reported
All elements not detected < 0.1 ppm
MC - Major Component > 1,000 ppm
INT - Interference

Approved: [Signature]

TABLE B-11. RESULTS OF SSMS ANALYSIS OF COMBINED GAS TRACE METALS
IMPINGERS COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

Reply to: GENERAL OFFICES: 338 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 738-8434
INSTRUMENTAL ANALYSIS DIVISION: 14325 WEST 64TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9521

To: Mr. Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766



Date: October 26, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #318

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No.: 97-H492-113-12

CONCENTRATION IN $\mu\text{g/ml}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	0.002
Thorium		Gadolinium		Molybdenum		Titanium	0.04
Bismuth		Europium		Niobium		Scandium	<0.001
Lead		Samarium		Zirconium	0.02	Calcium	2
Thallium		Neodymium		Yttrium		Potassium	MC
Mercury	NR	Praseodymium		Strontium	0.006	Chlorine	2
Gold		Cerium		Rubidium	0.003	Sulfur	2
Platinum		Lanthanum		Bromine	0.3	Phosphorus	0.5
Iridium		Barium		Selenium	<0.02	Silicon	4
Osmium		Cesium		Arsenic	0.06	Aluminum	0.4
Rhenium		Iodine		Germanium		Magnesium	0.6
Tungsten		Tellurium		Gallium	<0.003	Sodium	3
Tantalum		Antimony		Zinc	0.5	Fluorine	=2
Hafnium		Tin		Copper	0.4	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.2	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.005	Carbon	NR
Thulium		Silver	MC	Iron	0.7	Boron	0.3
Erbium		Palladium		Manganese	0.06	Beryllium	
Holmium		Rhodium		Chromium	0.3	Lithium	0.1
Dysprosium						Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.003 $\mu\text{g/ml}$

MC - Major Component > 10 $\mu\text{g/ml}$

INT - Interference

Approved:

M. J. Jacobs

TABLE B-12. RESULTS OF SSMS ANALYSIS OF MAIN STACK SASS IMPINGERS COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 728-8434
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401 PHONE 303-278-9321

Reply to

To: Mr. Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766



Date: October 26, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #319

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS
CONCENTRATION IN $\mu\text{g/ml}$

IAD No.: 97-H492-113-12

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	
Thorium		Gadolinium		Molybdenum		Titanium	0.06
Bismuth		Europium		Niobium		Scandium	<0.002
Lead	0.009	Samarium		Zirconium	0.008	Calcium	2
Thallium		Neodymium		Yttrium		Potassium	MC
Mercury	NR	Praseodymium		Strontium	0.01	Chlorine	0.6
Gold		Cerium		Rubidium		Sulfur	>8
Platinum		Lanthanum		Bromine	0.3	Phosphorus	0.2
Iridium		Barium	0.03	Selenium	0.008	Silicon	4
Osmium		Cesium		Arsenic		Aluminum	1
Rhenium		Iodine	0.003	Germanium		Magnesium	0.06
Tungsten		Tellurium		Gallium	0.007	Sodium	>3
Tantalum		Antimony		Zinc	0.4	Fluorine	=0.5
Hafnium		Tin		Copper	0.02	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.02	Nitrogen	NR
Ytterbium		Cadmium	0.004	Cobalt	<0.001	Carbon	NR
Thulium		Silver	MC	Iron	0.3	Boron	0.003
Erbium		Palladium		Manganese	0.005	Beryllium	
Holmium		Rhodium		Chromium	0.01	Lithium	<0.001
Dysprosium						Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.001 $\mu\text{g/ml}$

MC - Major Component > 10 $\mu\text{g/ml}$

INT - Interference

Approved:

M. Jacobs

TABLE B-13. RESULTS OF SSMS ANALYSIS OF BOILER NO. 3 SASS IMPINGERS
COMPOSITE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 338 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 738-8434
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9521

Reply to

To: Mr. Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766



Date: October 26, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #320

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No.: 97-H492-113-12

CONCENTRATION IN $\mu\text{g/ml}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	0.001
Thorium		Gadolinium		Molybdenum		Titanium	0.04
Bismuth		Europium		Niobium		Scandium	<0.001
Lead		Samarium		Zirconium		Calcium	1
Thallium		Neodymium		Yttrium		Potassium	1
Mercury	NR	Praseodymium		Strontium	0.003	Chlorine	0.09
Gold		Cerium		Rubidium		Sulfur	3
Platinum		Lanthanum		Bromine	0.1	Phosphorus	0.1
Iridium		Barium		Selenium	0.002	Silicon	4
Osmium		Cesium		Arsenic		Aluminum	0.2
Rhenium		Iodine		Germanium		Magnesium	0.7
Tungsten		Tellurium		Gallium	<0.001	Sodium	0.8
Tantalum		Antimony		Zinc	0.1	Fluorine	=0.05
Hafnium		Tin		Copper	0.02	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.2	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	≤ 0.002	Carbon	NR
Thulium		Silver	MC	Iron	2	Boron	0.001
Erbium		Palladium		Manganese	0.04	Beryllium	
Holmium		Rhodium		Chromium	0.9	Lithium	0.002
Dysprosium						Hydrogen	NR

STD - Internal Standard

NR - Not Reported

All elements not detected < 0.001 $\mu\text{g/ml}$

MC - Major Component > 10 $\mu\text{g/ml}$

INT - Interference

Approved:

TABLE B-14. RESULTS OF SSMS ANALYSIS OF MAIN STACK SASS XAD-2 RESIN AND RINSE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 338 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 · AREA CODE 312 738-8434
INSTRUMENTAL ANALYSIS DIVISION, 14335 WEST 44TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9321

Reply to

To: Michael Fuchs
Radian Corporation
P.O. Box 9943
Austin, TX 78766



Date: October 20, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #321

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No. 97-H492-113-12

CONCENTRATION IN $\mu\text{g/ml}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	0.006
Thorium		Gadolinium		Molybdenum	0.01	Titanium	0.03
Bismuth		Europium		Niobium	<0.003	Scandium	<0.002
Lead		Samarium		Zirconium	0.003	Calcium	MC
Thallium		Neodymium		Yttrium	<0.002	Potassium	*9
Mercury	NR	Praseodymium		Strontium	0.08	Chlorine	*1
Gold		Cerium		Rubidium	0.002	Sulfur	*0.2
Platinum		Lanthanum		Bromine	0.5	Phosphorus	0.04
Iridium		Barium	2	Selenium		Silicon	MC
Osmium		Cesium		Arsenic	0.002	Aluminum	0.03
Rhenium		Iodine	<0.001	Germanium		Magnesium	0.09
Tungsten		Tellurium		Gallium	0.002	Sodium	*3
Tantalum		Antimony		Zinc	0.06	Fluorine	=0.3
Hafnium		Tin		Copper	*MC	Oxygen	NR
Lutetium		Indium	STD	Nickel	0.02	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.002	Carbon	NR
Thulium		Silver		Iron	0.1	Boron	0.008
Erbium		Palladium		Manganese	0.003	Beryllium	
Holmium		Rhodium		Chromium	0.001	Lithium	0.008
Dysprosium		* Heterogeneous				Hydrogen	NR

STD — Internal Standard

NR – Not Reported

All elements not detected < 0.002 µg/ml

MC – Major Component > 10 µg/ml

INT = Interference

Approved: W L Joubert by S K Taylor
24 Oct 81

TABLE B-15. RESULTS OF SSMS ANALYSIS OF BOILER NO. 3 STACK SASS XAD-2
RESIN AND RINSE

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 238 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 - AREA CODE 312 726-8434
INSTRUMENTAL ANALYSIS DIVISION: 14335 WEST 46TH AVENUE, GOLDEN, COLORADO 80401, PHONE: 303-278-9521

Reply to

To: Michael Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766



Date: October 20, 1981

Analyst: J. Oldham

P. O. No.:

Sample No.: #322

SPARK SOURCE MASS SPECTROGRAPHIC ANALYSIS

IAD No.: 97-H492-113-12

CONCENTRATION IN $\mu\text{g/ml}$

ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.	ELEMENT	CONC.
Uranium		Terbium		Ruthenium		Vanadium	<0.001
Thorium		Gadolinium		Molybdenum	*0.002	Titanium	0.09
Bismuth		Europium		Niobium		Scandium	<0.001
Lead	0.01	Samarium		Zirconium	0.01	Calcium	0.7
Thallium		Neodymium		Yttrium		Potassium	0.06
Mercury	NR	Praseodymium		Strontium	0.004	Chlorine	0.3
Gold		Cerium		Rubidium		Sulfur	0.03
Platinum		Lanthanum		Bromine	0.04	Phosphorus	0.03
Iridium		Barium	0.005	Selenium	0.006	Silicon	0.07
Osmium		Cesium		Arsenic	0.001	Aluminum	0.03
Rhenium		Iodine		Germanium		Magnesium	0.02
Tungsten		Tellurium		Gallium	<0.003	Sodium	0.01
Tantalum		Antimony		Zinc	0.02	Fluorine	=0.2
Hafnium		Tin		Copper	0.006	Oxygen	NR
Lutetium		Indium	STD	Nickel	<0.001	Nitrogen	NR
Ytterbium		Cadmium		Cobalt	<0.001	Carbon	NR
Thulium		Silver		Iron	0.01	Boron	<0.001
Erbium		Palladium		Manganese	<0.001	Beryllium	
Holmium		Rhodium		Chromium	0.002	Lithium	<0.001
Dysprosium		* Heterogeneous				Hydrogen	NR

STD - Internal Standard
NR - Not Reported
All elements not detected < 0.002 $\mu\text{g/ml}$
MC - Major Component > 10 $\mu\text{g/ml}$
INT - Interference

Approved: *[Signature]*
J. Oldham

TABLE B-16. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF COAL

COMMERCIAL TESTING & ENGINEERING CO.
 GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 728-6434



Reply to
 Instrument Analysis Division
 490 Orchard Street
 Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
 Radian Corporation
 P.O. Box 9948
 Austin, TX 78766
 IAD No.: 97-H491-113-13

Date: November 9, 1981
 Purchase Order: 8075
 Sample I.D.: #301 Solid

PROXIMATE ANALYSIS			ULTIMATE ANALYSIS		
	As Received	Dry Basis		As Received	Dry Basis
% Moisture	3.33	xxxx	% Moisture	3.33	xxxx
% Ash	8.58	8.88	% Carbon	67.95	70.29
% Volatile	35.98	37.22	% Hydrogen	4.49	4.64
% Fixed			% Nitrogen	1.38	1.43
Carbon	52.11	53.90	% Chlorine	0.00	0.00
	100.00	100.00	% Sulfur	0.47	0.49
Btu/lb	11740	12144	% Ash	8.58	8.88
% Sulfur	0.47	0.49	% Oxygen	13.80	14.27
			(diff)		
				100.00	100.00
Moisture, Ash-free Btu	13327				
% Moisture, Ash-free Volatile	40.85				

M. L. Jacobs
 M. L. Jacobs, Ph.D., Mgr.
 Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
 TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE E-17. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF COAL

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 229 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 726-8434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 9, 1981
Purchase Order: 8075
Sample I.D.: #302 Solid

PROXIMATE ANALYSIS			ULTIMATE ANALYSIS		
	As Received	Dry Basis		As Received	Dry Basis
% Moisture	4.18	xxxx	% Moisture	4.18	xxxx
% Ash	11.33	11.82	% Carbon	65.09	67.93
% Volatile	35.11	36.64	% Hydrogen	4.24	4.43
% Fixed			% Nitrogen	1.36	1.42
Carbon	49.38	51.54	% Chlorine	0.00	0.00
	100.00	100.00	% Sulfur	0.53	0.55
Btu/lb	11184	11672	% Ash	11.33	11.82
% Sulfur	0.53	0.55	% Oxygen	13.27	13.85
			(diff)		
			100.00	100.00	
Moisture, Ash-free Btu	13237				
% Moisture, Ash-free Volatile	41.55				

M. L. Jacobs
M. L. Jacobs, Ph.D., Mngr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-10 RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF COAL

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 726-6434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 9, 1981
Purchase Order: 8075
Sample I.D.: #303 Solid

PROXIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	4.08	xxxx
% Ash	7.50	7.82
% Volatile	35.07	36.56
% Fixed Carbon	53.35	55.62
	100.00	100.00
Btu/lb	11721	12220
% Sulfur	0.49	0.51
Moisture, Ash-free Btu	13257	
% Moisture, Ash-free Volatile	39.66	

ULTIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	4.08	xxxx
% Carbon	68.52	71.43
% Hydrogen	4.35	4.53
% Nitrogen	1.45	1.51
% Chlorine	0.00	0.00
% Sulfur	0.49	0.51
% Ash	7.50	7.82
% Oxygen (diff)	13.61	14.20
	100.00	100.00

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-19. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF ESP OIL

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 738-8484



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766

Date: November 10, 1981

Purchase Order: 8075

Sample I.D.: #305 Tar

IAD No.: 97-H491-113-13

PROXIMATE ANALYSIS

*% Weight Loss	0.60
% Ash	0.18
% Volatile	99.38
% Fixed Carbon	0.44
	100.60

ULTIMATE ANALYSIS

* % Weight Loss	0.60
% Carbon	85.11
% Hydrogen	10.88
% Nitrogen	0.44
% Sulfur	0.20
% Ash	0.18
% Oxygen	3.43
	100.84

% Sulfur 0.20
** Btu/lb 18959

* Weight loss: Sample dried at 105°C for one hour
** Btu calculated by Dulong formula

M. L. Jacobs
M. L. Jacobs, Ph.D., Mngr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-20. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF ESP OIL

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 725-6434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 10, 1981

Purchase Order: 8075

Sample I.D.: #306 Tar

PROXIMATE ANALYSIS

% Weight Loss	0.60
% Ash	0.11
% Volatile	94.99
% Fixed Carbon	4.90
	100.60

ULTIMATE ANALYSIS

% Weight Loss	0.60
% Carbon	83.85
% Hydrogen	9.91
% Nitrogen	0.97
% Sulfur	0.31
% Ash	0.11
% Oxygen	4.34
	100.09

% Sulfur	0.31
** Btu/lb	18101

* Weight loss: Sample dried at 105°C for one hour.

** Btu calculated by Dulong formula

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-21. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF ESP OIL

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 726-8484



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 10, 1981

Purchase Order: 8075

Sample I.D.: #308 Tar

PROXIMATE ANALYSIS

** Weight Loss	0.45
% Ash	<0.10
% Volatile	99.52
% Fixed Carbon	0.48
	100.55

ULTIMATE ANALYSIS

** Weight Loss	0.45
% Carbon	85.06
% Hydrogen	10.84
% Nitrogen	0.32
% Sulfur	0.21
% Ash	<0.10
% Oxygen	3.41
	100.39

% Sulfur	0.21
** Btu/lb	18928

* Weight loss: Sample dried at 105°C for one hour.
** Btu calculated by Dulong formula

M. L. Jacobs
M. L. Jacobs, Ph.D., Mngr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-22. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF ESP OIL TO
BOILER NO. 3

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 725-8434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766

Date: November 10, 1981

Purchase Order: 8075

Sample I.D.: #307 Tar

IAD No.: 97-H491-113-13

PROXIMATE ANALYSIS

% Weight Loss	0.45
% Ash	0.18
% Volatile	97.99
% Fixed Carbon	1.83
	100.45

ULTIMATE ANALYSIS

% Weight Loss	0.45
% Carbon	85.13
% Hydrogen	10.82
% Nitrogen	0.48
% Sulfur	0.24
% Ash	0.18
% Oxygen	3.36
	100.66

% Sulfur	0.24
** Btu/lb	18930

* Weight loss: Sample dried at 105°C for one hour.

** Btu calculated by Dulong formula

M. L. Jacobs
M. L. Jacobs, Ph.D., Mngr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-23. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF CYCLONE DUST

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 736-8434



Reply to

Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 6, 1981
Purchase Order: 8075
Sample I.D.: #309 Solid

PROXIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.55	xxxx
% Ash	23.98	24.36
% Volatile	2.54	2.58
% Fixed Carbon	71.93	73.06
	100.00	100.00
Btu/lb	10479	10644
% Sulfur	0.84	0.85

ULTIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.55	xxxx
% Carbon	71.94	73.07
% Hydrogen	0.77	0.78
% Nitrogen	0.80	0.81
% Chlorine	0.01	0.01
% Sulfur	0.84	0.85
% Ash	23.98	24.36
% Oxygen (diff)	0.11	0.12
	100.00	100.00

Moisture, Ash-free Btu 14075
% Moisture, Ash-free Volatile 3.41

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-24. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF CYCLONE DUST

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 328 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 725-6434



Reply to
Instrument Analysis Division
400 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 6, 1981
Purchase Order: 8075
Sample I.D.: #310 Sol1d

PROXIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.25	xxxx
% Ash	28.48	28.84
% Volatile	2.03	2.06
% Fixed Carbon	68.24	69.10
	<u>100.00</u>	<u>100.00</u>
Btu/lb	9922	10048
% Sulfur	0.87	0.88

Moisture, Ash-free Btu 14120
% Moisture, Ash-free Volatile 2.89

ULTIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.25	xxxx
% Carbon	69.51	70.39
% Hydrogen	0.55	0.56
% Nitrogen	0.66	0.67
% Chlorine	0.01	0.01
% Sulfur	0.87	0.88
% Ash	28.48	28.84
% Oxygen (diff)	-1.33	-1.35
	<u>100.00</u>	<u>100.00</u>

M. L. Jacobs
M. L. Jacobs, Ph.D., Mngr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-25. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF CYCLONE DUST

COMMERCIAL TESTING & ENGINEERING CO.
 GENERAL OFFICES: 225 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 726-8434



Reply to
 Instrument Analysis Division
 490 Orchard Street
 Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
 Radian Corporation
 P.O. Box 9948
 Austin, TX 78766
 IAD No.: 97-H491-113-13

Date: November 9, 1981
 Purchase Order: 8075
 Sample I.D.: #311 Solid

PROXIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.00	xxxx
% Ash	27.53	27.81
% Volatile	2.53	2.56
% Fixed Carbon	68.94	69.63
	100.00	100.00
Btu/lb	9883	9983
% Sulfur	1.25	1.26

Moisture, Ash-free Btu 13829
 % Moisture, Ash-free Volatile 3.55

ULTIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.00	xxxx
% Carbon	69.74	70.44
% Hydrogen	0.64	0.65
% Nitrogen	0.68	0.69
% Chlorine	0.01	0.01
% Sulfur	1.25	1.26
% Ash	27.53	27.81
% Oxygen (diff)	-0.85	-0.86
	100.00	100.00

M. L. Jacobs
 M. L. Jacobs, Ph.D., Mngr.
 Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
 TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-26. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF GASIFIER ASH

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 738-8434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 9, 1981
Purchase Order: 8075
Sample I.D.: #312 Solid

PROXIMATE ANALYSIS			ULTIMATE ANALYSIS		
	As Received	Dry Basis		As Received	Dry Basis
% Moisture	1.13	xxxx	% Moisture	1.13	xxxx
% Ash	76.21	77.08	% Carbon	20.40	20.63
% Volatile	3.25	3.29	% Hydrogen	0.11	0.11
% Fixed			% Nitrogen	0.24	0.24
Carbon	19.41	19.63	% Chlorine	0.00	0.00
	100.00	100.00	% Sulfur	0.27	0.27
Btu/lb	2950	2984	% Ash	76.21	77.08
% Sulfur	0.27	0.27	% Oxygen	1.64	1.67
			(diff)		
				100.00	100.00
Moisture, Ash-free Btu	13020				
% Moisture, Ash-free Volatile	14.35				

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



Charter Member

OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-27. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF GASIFIER ASH

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 729-9434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 9, 1981
Purchase Order: 8075
Sample I.D.: #313 Solid

PROXIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.80	xxxx
% Ash	69.08	70.35
% Volatile	3.25	3.31
% Fixed Carbon	25.87	26.34
	100.00	100.00
Btu/lb	3380	3442
% Sulfur	0.25	0.25
Moisture, Ash-free Btu	11609	
% Moisture, Ash-free Volatile	11.16	

ULTIMATE ANALYSIS

	<u>As Received</u>	<u>Dry Basis</u>
% Moisture	1.80	xxxx
% Carbon	26.37	26.85
% Hydrogen	0.20	0.20
% Nitrogen	0.24	0.24
% Chlorine	0.01	0.01
% Sulfur	0.25	0.25
% Ash	69.08	70.35
% Oxygen (diff)	2.05	2.10
	100.00	100.00

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-28. RESULTS OF PROXIMATE AND ULTIMATE ANALYSIS OF GASIFIER ASH

COMMERCIAL TESTING & ENGINEERING CO.

GENERAL OFFICES: 228 NORTH LA SALLE STREET, CHICAGO, ILLINOIS 60601 AREA CODE 312 726-8434



Reply to
Instrument Analysis Division
490 Orchard Street
Golden, CO 80401

Phone: 303-278-9521

To: Mr. Michael R. Fuchs
Radian Corporation
P.O. Box 9948
Austin, TX 78766
IAD No.: 97-H491-113-13

Date: November 9, 1981
Purchase Order: 8075
Sample I.D.: #314 Solid

PROXIMATE ANALYSIS			ULTIMATE ANALYSIS		
	As Received	Dry Basis		As Received	Dry Basis
% Moisture	0.81	xxxx	% Moisture	0.81	xxxx
% Ash	85.00	85.69	% Carbon	14.20	14.32
% Volatile	4.80	4.84	% Hydrogen	0.18	0.18
% Fixed			% Nitrogen	0.10	0.10
Carbon	9.39	9.47	% Chlorine	0.00	0.00
	100.00	100.00	% Sulfur	0.32	0.32
Btu/lb	1695	1709	% Ash	85.00	85.69
% Sulfur	0.32	0.32	% Oxygen		
			(diff)	-0.61	-0.61
				100.00	100.00
Moisture, Ash-free Btu	11943				
% Moisture, Ash-free Volatile	33.82				

M. L. Jacobs
M. L. Jacobs, Ph.D., Mgr.
Instrumental Analysis Div.



OVER 40 BRANCH LABORATORIES STRATEGICALLY LOCATED IN PRINCIPAL COAL MINING AREAS,
TIDEWATER AND GREAT LAKES PORTS, AND RIVER LOADING FACILITIES

TABLE B-29. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF COAL¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	>1000	1.3%		Neodymium	0.9		
Antimony	0.9		<0.2	Nickel	0.8	4.6	
Arsenic	2		1.8	Niobium	4		
Barium	>1000	480		Osmium	ND		
Beryllium	0.3			Palladium	ND		
Bismuth	ND			Phosphorus	>1000		
Boron	8	25		Platinum	ND		
Bromine	2			Potassium	>1000	950	
Cadmium	1			Praseodymium	2		
Calcium	>1000	3300		Rhenium	ND		
Cerium	11			Rhodium	ND		
Cesium	0.3			Rubidium	4		
Chlorine	27			Ruthenium	ND		
Chromium	2	5.2		Samarium	2		
Cobalt	1	1.6		Scandium	1		
Copper	5	4.8		Selenium	<0.6		0.36
Dysprosium	ND			Silicon	>1000		
Erbium	ND			Silver	0.8		
Europium	ND			Sodium	>480	180	
Fluorine	7			Strontium	>1000		
Gadolinium	ND			Sulfur ²	>1000		
Gallium	5			Tantalum	ND		
Germanium	<0.2			Tellurium	0.4		
Gold	ND			Terbium	ND		
Hafnium	ND			Thallium	ND		
Holmium	ND			Thorium	4		
Iodine	<0.4			Thulium	ND		
Iridium	ND			Tin	0.4		
Iron	>1000	3600		Titanium	890	440	
Lanthanum	9			Tungsten	ND		
Lead	3			Uranium	3		
Lithium	12			Vanadium	14	9.7	
Lutetium	ND			Ytterbium	ND		
Magnesium	870	760		Yttrium	14		
Manganese	24	24		Zinc	3	5.1	
Mercury	NR		<0.07	Zirconium	31		
Molybdenum	2	2.9					

¹Concentration in µg/g.²Ultimate analysis value 0.52%.

NOTE: ICPES and AAS analyses performed in duplicate except for mercury.

SSMS results of single analysis.

ND - not detected.

NR - no record.

TABLE B-30. RESULTS OF SSMS, ICPES, AND AAS ANALYSIS OF GASIFIER ASH¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	>1000	7.8%		Neodymium	19		
Antimony	1		<0.2	Nickel	3	31	
Arsenic	1		24	Niobium	38		
Barium	>1000	3300		Osmium	ND		
Beryllium	0.9			Palladium	ND		
Bismuth	<0.1			Phosphorus	>1000		
Boron	64	690		Platinum	ND		
Bromine	3			Potassium	>1000	8200	
Cadmium	5			Praseodymium	16		
Calcium	>1000	5%		Rhenium	ND		
Cerium	99			Rhodium	ND		
Cesium	8			Rubidium	59		
Chlorine	10			Ruthenium	ND		
Chromium	8	42		Samarium	6		
Cobalt	1	12		Scandium	2		
Copper	23	32		Selenium	0.8		0.6
Dysprosium	6			Silicon	>1000		
Erbium	3			Silver	<0.3		
Europium	1			Sodium	>170	1500	
Fluorine	110			Strontium	>1000		
Gadolinium	4			Sulfur ²	>420		
Gallium	23			Tantalum	0.6		
Germanium	0.7			Tellurium	0.2		
Gold	ND			Terbium	1		
Hafnium	2			Thallium	0.4		
Holmium	4			Thorium	29		
Iodine	0.3			Thulium	0.2		
Iridium	ND			Tin	0.5		
Iron	>1000	4.0%		Titanium	>1000	1800	
Lanthanum	180			Tungsten	2		
Lead	12			Uranium	19		
Lithium	64			Vanadium	68	66	
Lutetium	0.3			Ytterbium	2		
Magnesium	>1000	1.2%		Yttrium	50		
Manganese	230	440		Zinc	2	15	
Mercury	NR		0.30	Zirconium	200		
Molybdenum	3	16					

¹Concentration in µg/g.²Ultimate analysis value 0.28%.

NOTE: All ICPES and AAS analyses performed in duplicate except for mercury.

SSMS results of single analysis.

ND - not detected.

NR - no record.

TABLE B-31. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF CYCLONE DUST¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	>1000	2.6%		Neodymium	2		
Antimony	4		19	Nickel	3	16	
Arsenic	40		31	Niobium	10		
Barium	>1000	320		Osmium	ND		
Beryllium	0.8			Palladium	ND		
Bismuth	4			Phosphorus	>1000		
Boron	400	98		Platinum	ND		
Bromine	7			Potassium	>1000	2700	
Cadmium	3			Praseodymium	5		
Calcium	>1000	1.2%		Rhenium	ND		
Cerium	28			Rhodium	ND		
Cesium	0.7			Rubidium	9		
Chlorine	86			Ruthenium	ND		
Chromium	220	20		Samarium	4		
Cobalt	1	4.3		Scandium	1		
Copper	64	22		Selenium	10		2.6
Dysprosium	4			Silicon	>1000		
Erbium	0.8			Silver	<2		
Europium	1			Sodium	>480	690	
Fluorine	>1000			Strontium	760		
Gadolinium	2			Sulfur ²	>1000		
Gallium	280			Tantalum	ND		
Germanium	19			Tellurium	1		
Gold	<0.2			Terbium	0.8		
Hafnium	<2			Thallium	12		
Holmium	1			Thorium	8		
Iodine	4			Thulium	ND		
Iridium	ND			Tin	77		
Iron	>1000	1.2%		Titanium	>1000	780	
Lanthanum	20			Tungsten	4		
Lead	870			Uranium	7		
Lithium	27			Vanadium	30	24	
Lutetium	ND			Ytterbium	ND		
Magnesium	>1000	2400		Yttrium	16		
Manganese	97	48		Zinc	560	620	
Mercury	NR		0.75	Zirconium	55		
Molybdenum	9	7.6					

¹Concentration in µg/g.

²Ultimate analysis result 1.0%.

NOTE: ICPES and AAS analyses performed in duplicate except for mercury. SSMS results of single analysis.

ND - not detected.

NR - no record.

TABLE B-32. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF
ESP OIL¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	>5	54		Neodymium	ND		
Antimony	<0.01		0.2	Nickel	0.09	<0.3	
Arsenic	0.8		0.82	Niobium	ND		
Barium	4	5.2		Osmium	ND		
Beryllium	ND			Palladium	ND		
Bismuth	0.03			Phosphorus	3		
Boron	0.02	<1		Platinum	ND		
Bromine	0.1			Potassium	3	28	
Cadmium	ND			Praseodymium	<0.01		
Calcium	12	30		Rhenium	ND		
Cerium	0.06			Rhodium	ND		
Cesium	ND			Rubidium	0.02		
Chlorine	0.3			Ruthenium	ND		
Chromium	0.6	170		Samarium	ND		
Cobalt	<0.01	<0.6		Scandium	<0.01		
Copper	0.07	0.17		Selenium	<0.03		<0.3
Dysprosium	ND			Silicon	4		
Erbium	ND			Silver	0.2		
Europium	ND			Sodium	2	49	
Fluorine	4			Strontium	0.4		
Gadolinium	ND			Sulfur ²	8		
Gallium	0.07			Tantalum	ND		
Germanium	<0.02			Tellurium	ND		
Gold	ND			Terbium	ND		
Hafnium	ND			Thallium	ND		
Holmium	ND			Thorium	ND		
Iodine	0.06			Thulium	ND		
Iridium	ND			Tin	0.04		
Iron	>100	170		Titanium	0.6	3.3	
Lanthanum	0.05			Tungsten	ND		
Lead	0.4			Uranium	ND		
Lithium	0.03			Vanadium	0.2	2.6	
Lutetium	ND			Ytterbium	ND		
Magnesium	0.2	<5.4		Yttrium	ND		
Manganese	0.6	0.54		Zinc	>100	0.72	
Mercury	NR		0.16	Zirconium	0.08		
Molybdenum	0.02	0.65					

¹Concentration in µg/g.

²Ultimate analysis value is 0.24%.

NOTE: ICPES and AAS analyses performed in duplicate except for mercury.

SSMS results of single analysis.

ND - not detected.

NR - no record.

TABLE B-33. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF
ASH PAN WATER¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	0.03	0.36		Neodymium	ND		
Antimony	ND		<0.002	Nickel	0.002	0.02	
Arsenic	0.03		<0.002	Niobium	0.002		
Barium	4	1.5		Osmium	ND		
Beryllium	ND			Palladium	ND		
Bismuth	ND			Phosphorus	0.1		
Boron	0.4	4.4		Platinum	ND		
Bromine	0.1			Potassium	6	13	
Cadmium	ND			Praseodymium	ND		
Calcium	>10		610	Rhenium	ND		
Cerium	ND			Rhodium	ND		
Cesium	<0.001			Rubidium	0.03		
Chlorine	0.7			Ruthenium	ND		
Chromium	0.04	0.05		Samarium	ND		
Cobalt	0.005	0.02		Scandium	<0.002		
Copper	0.02	0.06		Selenium	0.04		<0.003
Dysprosium	ND			Silicon	0.4		
Erbium	ND			Silver	ND		
Europium	ND			Sodium	0.2	4.5	
Fluorine	0.05			Strontium	>10		
Gadolinium	ND			Sulfur	>9		
Gallium	0.02			Tantalum	ND		
Germanium	ND			Tellurium	0.004		
Gold	ND			Terbium	ND		
Hafnium	ND			Thallium	ND		
Holmium	ND			Thorium	<0.01		
Iodine	0.007			Thulium	ND		
Iridium	ND			Tin	ND		
Iron	9	<0.008		Titanium	0.2	0.14	
Lanthanum	ND			Tungsten	ND		
Lead	0.01			Uranium	<0.01		
Lithium	0.2			Vanadium	0.001	0.11	
Lutetium	ND			Ytterbium	ND		
Magnesium	0.3	0.28		Yttrium	0.05		
Manganese	<0.001	0.008		Zinc	0.04	0.009	
Mercury	NR		0.0042	Zirconium	0.01		
Molybdenum	0.02	0.12					

¹Concentration in µg/mL.

NOTE: ICPES and AAS analyses performed in duplicate except for mercury.

SSMS results of single analysis.

ND - not detected.

NR - no record.

B.34 Elemental Analysis of the Combined Product Gas

Table B-34 presents the results of SSMS, AAS, and ICPEs analyses of the combined product gas. The results for each method of analysis includes contributions from the impinger solutions, particulate, and oil loadings. The contribution of elements in the particulate was calculated by equating the elemental concentrations in the particulate load ($1.95E5 \mu\text{g}/\text{Nm}^3$) with the elemental concentrations in an identical mass of cyclone dust. The contribution of elements in the oil was calculated by equating the elemental concentrations in the oil load ($2.0E4 \mu\text{g}/\text{Nm}^3$) with the elemental concentrations in an identical mass of ESP oil. These results were added to the analytical results of the impinger solutions to determine the elemental concentrations in the combined product gas.

B.35 Elemental Analysis of Combustion Emissions from the Product Gas Boilers

The results of the analysis of impinger solutions, XAD-2 resin, and particulate loading for the combustion emissions from the product gas boilers are presented in Table B-35. The elemental concentrations for the particulate loading were calculated by equating the particulate concentration in the emission stream ($3.9E4 \mu\text{g}/\text{Nm}^3$) to cyclone dust, as described above.

B.36 Elemental Analysis of the Combustion Emissions from Boiler No. 3 Firing ESP Oils

Table B-36 presents the results of elemental analysis of impinger solutions, XAD-2 resin, and aerosol loading for the combustion emissions from Boiler No. 3 firing ESP oils. The elemental concentrations for the aerosol loading were calculated by equating the particulate concentration in the emission stream ($4.8E4 \mu\text{g}/\text{Nm}^3$) to cyclone dust and the oil loading ($1.0E3 \mu\text{g}/\text{Nm}^3$) to ESP oil, as described above. The contribution from the oil loading was insignificant.

TABLE B-34. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF COMBINED PRODUCT GAS¹

	SSMS	ICPES	AAS		SSMS	ICPES	AAS
Aluminum	270	5300		Neodymium	0.39		
Antimony	0.78		12	Nickel	0.59	120	
Arsenic	32		16	Niobium	2		
Barium	>195	62		Osmium	ND		
Beryllium	0.16			Palladium	ND		
Bismuth	0.78			Phosphorus	>270		
Boron	200	180		Platinum	ND		
Bromine	42			Potassium	>3100	530	
Cadmium	0.59			Praseodymium	0.98		
Calcium	>770	2300		Rhenium	ND		
Cerium	5.5			Rhodium	ND		
Cesium	0.14			Rubidium	>2.2		
Chlorine	430			Ruthenium	ND		
Chromium	84	82		Samarium	0.78		
Cobalt	<1.8	0.84		Scandium	0.20		
Copper	12	290		Selenium	5.3		<1.7
Dysprosium	0.78			Silicon	>190		
Erbium	0.16			Silver	<0.39		
Europium	0.20			Sodium	>1100	1100	
Fluorine	>190			Strontium	150		
Gadolinium	0.39			Sulfur**	>190		
Gallium	55			Tantalum	ND		
Germanium	3.7			Tellurium	0.20		
Gold	ND			Terbium	0.16		
Hafnium	<0.39			Thallium	2.3		
Holmium	0.20			Thorium	1.6		
Iodine	0.78			Thulium	ND		
Iridium	ND			Tin	15		
Iron	>200	2700		Titanium	>190	150	
Lanthanum	3.9			Tungsten	0.78		
Lead	170			Uranium	1.4		
Lithium	45			Vanadium	>6.3	4.7	
Lutetium	ND			Ytterbium	ND		
Magnesium	>430	470		Yttrium	3.1		
Manganese	27	18		Zinc	300	170	
Mercury*	NR		2.4	Zirconium	15		
Molybdenum	1.8	6.4					

¹Concentration in $\mu\text{g}/\text{Nm}^3$; includes impinger solutions, particulate and oil loadings.

*Value determined by gold amalgamation method is $4.6 \mu\text{g}/\text{Nm}^3$.

**Value determined by FPD/GC is $1.1\text{E}6 \mu\text{g}/\text{Nm}^3$.

ND - not detected.

NR - no record.

TABLE B-35. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS OF COMBUSTION EMISSIONS FROM THE PRODUCT GAS BOILERS¹

Element	Impingers	XAD-2 Resin	Particulates
Aluminum	570 ²	1.5	1000 ²
Antimony	<0.15 ³	ND	0.74 ³
Arsenic	<0.15 ³	BC	1.2 ³
Barium	<0.08 ²	128	12 ²
Beryllium	ND	ND	0.03
Bismuth	ND	ND	0.16
Boron	BC ²	0.60	3.8 ²
Bromine	7.5	35	0.27
Cadmium	0.30	ND	0.12
Calcium	BC ²	>700	470 ²
Cerium	ND	ND	1.1
Cesium	ND	ND	0.03
Chlorine	BC	BC	3.4
Chromium	2.3 ²	BC	0.78 ²
Cobalt	BC ²	<0.15	0.17 ²
Copper	BC ²	>750	0.86 ²
Dysprosium	ND	ND	0.16
Erbium	ND	ND	0.03
Europium	ND	ND	0.04
Fluorine	BC	22	>39
Gadolinium	ND	ND	0.08
Gallium	>0.38	0.08	11
Germanium	ND	ND	0.74
Gold	ND	ND	ND
Hafnium	ND	ND	<0.08
Holmium	ND	ND	0.04
Iodine	0.08	<0.08	0.16
Iridium	ND	ND	ND
Iron	21 ²	6	470
Lanthanum	ND	ND	0.78
Lead	BC	ND	34
Lithium	<0.08	0.53	1.1
Lutetium	ND	ND	ND
Magnesium	BC ²	BC	94 ²
Manganese	0.53 ²	0.23	1.9 ²
Mercury	9.7 ³	NR	NR
Molybdenum	1.9 ²	0.38	0.30 ²

(Continued)

TABLE B-35. (Continued)

Element	Impingers	XAD-2 Resin	Particulates
Neodymium	ND	ND	0.08
Nickel	2.6 ²	1.4	0.62 ²
Niobium	ND	≤0.23	0.39
Osmium	ND	ND	ND
Palladium	ND	ND	ND
Phosphorus	BC	1.5	>39
Platinum	ND	ND	ND
Potassium	75 ²	BC	100 ²
Praseodymium	ND	ND	0.20
Rhenium	ND	ND	ND
Rhodium	ND	ND	ND
Rubidium	ND	0.15	0.35
Ruthenium	ND	ND	ND
Samarium	ND	ND	0.16
Scandium	≤0.15	≤0.15	0.04
Selenium	0.23 ³	ND	0.10 ³
Silicon	BC	BC	>39
Silver	BC	ND	≤0.08
Sodium	6 ²	<75	27 ²
Strontium	0.08	5.6	30
Sulfur**	BC	BC	>39
Tantalum	ND	ND	ND
Tellurium	ND	ND	0.04
Terbium	ND	ND	0.03
Thallium	ND	ND	0.47
Thorium	ND	ND	0.31
Thulium	ND	ND	ND
Tin	ND	ND	3
Titanium	0.76 ²	BC	30 ²
Tungsten	ND	ND	0.16
Uranium	ND	ND	0.27
Vanadium	BC ²	0.45	0.94 ²
Ytterbium	ND	ND	ND
Yttrium	ND	≤0.15	0.62
Zinc	2.6 ²	BC	24 ²
Zirconium	BC	0.23	2.1

NR - No record.

BC - Value corrected for blank.

ND - Not detected.

¹All concentrations in $\mu\text{g}/\text{Nm}^3$.²Analyses performed by ICPES.³Analyses performed by AAS.*19.4 $\mu\text{g}/\text{Nm}^3$ by gold amalgamation method.**3.25E5 $\mu\text{g}/\text{Nm}^3$ by EPA Method 6.

Note: Analysis performed by SSMS except where indicated.

TABLE B-36. RESULTS OF SSMS, ICPES, AND AAS ELEMENTAL ANALYSIS
OF COMBUSTION EMISSIONS OF ESP OIL¹

Element	Impingers	XAD-2 Resin	Particulates
Aluminum	600 ²	1.4	1200 ²
Antimony	<0.17 ³	ND	0.91 ³
Arsenic	<0.17 ³	BC	1.5 ³
Barium	<0.08 ²	BC	15 ²
Beryllium	ND	ND	0.04
Bismuth	ND	ND	0.19
Boron	85 ²	0.07	4.7 ²
Bromine	BC	0.68	0.34
Cadmium	ND	ND	0.15
Calcium	BC ²	BC	570 ²
Cerium	ND	ND	1.3
Cesium	ND	ND	0.03
Chlorine	BC	BC	4.1
Chromium	390 ²	BC	1.1 ²
Cobalt	6.1 ²	<0.07	0.21 ³
Copper	1.8 ²	BC	1.1 ²
Dysprosium	ND	ND	0.19
Erbium	ND	ND	0.04
Europium	ND	ND	0.05
Fluorine	BC	13	>48
Gadolinium	ND	ND	0.09
Gallium	<0.09	BC	13
Germanium	ND	ND	0.91
Gold	ND	ND	ND
Hafnium	ND	ND	<0.09
Holmium	ND	ND	0.05
Iodine	BC	ND	0.19
Iridium	ND	ND	ND
Iron	1800 ²	BC	570 ²
Lanthanum	ND	ND	0.96
Lead	BC	0.14	42
Lithium	0.17	BC	1.3
Lutetium	ND	ND	ND
Magnesium	BC ²	BC	110 ²
Manganese	37 ²	<0.07	2.3 ²
Mercury*	0.19 ³	NR	0.04 ³
Molybdenum	28 ²	BC	0.37 ²

(Continued)

TABLE B-36. (Continued)

Element	Impingers	XAD-2 Resin	Particulates
Neodymium	ND	ND	0.09
Nickel	260 ²	BC	0.77 ²
Niobium	ND	ND	0.48
Osmium	ND	ND	ND
Palladium	ND	ND	ND
Phosphorus	BC	0.68	>48
Platinum	ND	ND	ND
Potassium	200 ²	BC	130 ²
Praseodymium	ND	ND	0.24
Rhenium	ND	ND	ND
Rhodium	ND	ND	ND
Rubidium	ND	ND	0.43
Ruthenium	ND	ND	ND
Samarium	ND	ND	0.19
Scandium	BC	<0.07	0.05
Selenium	<0.26 ³	BC	0.12 ³
Silicon	BC	BC	>48
Silver	BC	ND	<0.09
Sodium	38 ²	BC	33 ²
Strontium	BC	BC	37
Sulfur**	BC	BC	>48
Tantalum	ND	ND	ND
Tellurium	ND	ND	0.05
Terbium	ND	ND	0.04
Thallium	ND	ND	0.57
Thorium	ND	ND	0.39
Thulium	ND	ND	ND
Tin	ND	ND	3.7
Titanium	BC ²	2.7	38 ²
Tungsten	ND	ND	0.19
Uranium	ND	ND	0.34
Vanadium	0.85 ²	<0.07	1.2 ²
Ytterbium	ND	ND	ND
Yttrium	ND	ND	0.77
Zinc	5.3 ²	BC	30 ²
Zirconium	ND	0.68	2.6

NR - No record.

BC - Value corrected for blank.

ND - Not detected.

¹All concentrations in $\mu\text{g}/\text{Nm}^3$.²Analyses performed by ICPE.³Analyses performed by AAS.*19.4 $\mu\text{g}/\text{Nm}^3$ by gold amalgamation method.**3.25E5 $\mu\text{g}/\text{Nm}^3$ by EPA Method 6.

Note: Analysis performed by SSMS except where indicated.

TABLE B-37. ASH PAN WATER PH AND CONDUCTIVITY ANALYTICAL RESULTS

Date	Time	pH	Conductivity mho/cm at 25°C
2/24/81	1100	12.16	5560
	1600	12.30	5050
	2020	12.36	5080
2/25/81	0200	12.37	5210
	0620	12.32	5610
2/26/81		12.41	5470
2/27/81	1115	12.39	5640
	1420	12.39	6020
	1735	12.39	6040
3/5/81	1250	12.38	6000
	1555	12.32	6320
	2100	12.42	6140

TABLE B-38. RESULTS OF WATER QUALITY ANALYSES OF ASH PAN WATER¹

Parameters	2/24-1600 hrs, 2020 hrs 2/25-0200 hrs, 0620 hrs	2/27-1115 hrs, 1420 hrs, 1735 hrs	3/5-1250 hrs, 1555 hrs, 2100 hrs
Phosphate (P)	0.66 ¹	1.3	1.3
Chloride	205	195	200
Fluoride	0.48	0.56	0.64
Sulfide (S)	43	64	18
Sulfite (SO ₃)	143	5.4	94
Nitrate (N)	1.1	4.4	2.5
Cyanide	0.19	<0.02	<0.02
Thiocyanate	<4	<4	<4
Sulfate (SO ₄)	56	56	61
Ammonia (N)	<0.5	0.78	<0.5
COD	81	114	105
BOD	2.8	<1	1.8
TOC	3.8	3.2	3.2
Total Suspended Solids	136	151	151
Total Dissolved Solids	1740	1680	1950
pH (pH units)	12.5	12.5	12.5
Conductivity (average $\mu\text{mho/cm}$ at 25°C)	5240 $\pm$ 260	5900 $\pm$ 320	6150 $\pm$ 160
Phenol	<0.006	<0.006	<0.006

¹Concentration in $\mu\text{g/mL}$ except where noted.

TABLE B-39. ANALYTICAL RESULTS: GASEOUS NH₃

Description	Sample Volume		1st Imp. mg NH ₃	2nd Imp. mg NH ₃	Total mg NH ₃	ppm NH ₃	µg/Nm ³ NH ₃
	SCF 60°F 1 atm.	Nm ³ 25°C 1 atm.					
Combined Gas Run No. 1 2/24/81, 2024-2036 hrs.	3.794	0.1110	98.6	2.05	100.6	1310	9.07E5
Combined Gas Run No. 2 2/25/81, 0513-0529 hrs.	4.403	0.1288	100.5	2.13	102.6	1150	7.97E5
Combined Gas Run No. 3 2/25/81, 0937-0952 hrs.	3.480	0.1018	86.6	2.20	88.8	1260	8.71E2
Product Gas Boilers Run No. 1 2/25/81, 0222-0312 hrs.	22.547	0.6594	<0.15	<0.13	<0.28	<0.6	<4.25E2
Product Gas Boilers Run No. 2 2/25/81, 0820-0910 hrs.	21.127	0.6179	<0.15	<0.12	<0.27	<0.6	<4.37E2
Cleaned Top Gas Run No. 1 2/27/81, 1055-1111 hrs.	3.610	0.1056	189.0	0.56	189.6	1590	1.80E6
Cleaned Top Gas Run No. 2 2/27/81, 1655-1737 hrs.	6.554	0.1917	319.3	1.38	320.7	2410	1.67E6
Cleaned Top Gas Run No. 3 2/27/81, 1655-2057 hrs.	3.069	0.0898	164.0	2.18	166.2	2670	1.85E6
Raw Top Gas Run No. 1 2/27/81, 1047-1057 hrs.	3.430	0.1033	175.8	1.32	177.1	2450	1.77E6
Raw Top Gas Run No. 2 2/27/81, 1405-1415 hrs.	2.395	0.0700	125.9	0.56	126.5	2600	1.81E6
Raw Top Gas Run No. 3 2/27/81, 1706-1726 hrs.	5.008	0.1465	217.2	1.23	218.4	2150	1.49E6
Boiler No. 3 ESP Oils Run No. 1 3/3/81	19.024	0.5564	0.62	<0.12	0.62	1.6	1.11E3
Boiler No. 3 ESP Oils Run No. 2 3/3/81	18.957	0.5544	0.19	<0.13	0.19	0.5	3.43E3
Cleaned Bottom Gas Run No. 1 3/5/81, 1437-1452 hrs.	5.728	0.1675	9.51	0.92	10.4	89.7	6.23E4
Cleaned Bottom Gas Run No. 2 3/5/81, 1923-1937 hrs.	3.814	0.1115	3.23	0.38	3.61	46.6	3.24E4
Cleaned Bottom Gas Run No. 3 3/6/81, 0103-0113 hrs.	4.002	0.1170	97.8	0.33	98.1	1210*	8.39E5
Raw Bottom Gas Run No. 1 3/5/81, 0103-0113 hrs.	3.525	0.1030	3.69	<0.12	3.69	51.5	3.58E4
Raw Bottom Gas Run No. 2 3/5/81, 2111-2128 hrs.	3.781	0.1106	2.29	0.22	2.51	32.7	3.58E4
Raw Bottom Gas Run No. 3 3/6/81, 0114-0129 hrs.	3.617	0.1058	47.3	<0.06	47.3	644*	4.47E5

*Data rejected due to process upset.

aEb = a x 10^b

TABLE B-40. ANALYTICAL RESULTS: GASEOUS HCN

Description	Sample Volume		1st Imp. mg HCN ¹	2nd Imp. mg HCN ¹	Total mg HCN	ppm HCN	µg/Nm ³ HCN
	SCF 60°F 1 atm.	Nm ³ 25°C 1 atm.					
Combined Gas Run No. 1 2/24/81, 2041-2055 hrs.	4.165	0.1218	24.2	0.40	24.6	183	2.02E5
Combined Gas Run No. 2 2/25/81, 0537-0551 hrs.	4.353	0.1273	33.5	0.25	33.8	240	2.65E5
Combined Gas Run No. 3 2/25/81, 0955-1010 hrs.	3.170	0.0927	22.4	0.16	22.6	221	2.43E5
Product Gas Boilers Run No. 1 2/25/81, 0323-0415 hrs.	24.192	0.7075	1.23	<0.40	1.23	1.6	1.74E3
Product Gas Boilers Run No. 2 2/25/81, 0618-0658 hrs.	18.389	0.5378	0.89	0.005	0.89	1.5	1.66E3
Cleaned Top Gas Run No. 1 2/27/81, 1015-1050 hrs.	4.104	0.1200	33.4	0.39	33.8	255	2.82E5
Cleaned Top Gas Run No. 2 2/27/81, 1617-1652 hrs.	3.958	0.1158	32.3	0.14	32.4	252	2.80E5
Cleaned Top Gas Run No. 3 2/27/81, 2028-2040 hrs.	3.373	0.0986	19.2	0.36	19.6	180	1.98E5
Raw Top Gas Run No. 1 2/27/81, 1033-1043 hrs.	3.938	0.1152	34.9	0.38	35.3	278	3.06E5
Raw Top Gas Run No. 2 2/27/81, 1348-1358 hrs.	2.720	0.0795	20.7	0.28	21.0	239	2.64E5
Raw Top Gas Run No. 3 2/27/81, 1348-1358 hrs.	3.784	0.1107	26.8	0.31	27.1	222	2.45E5
Boiler #3 ESP Oil Run No. 1 3/3/81, 0945-1047 hrs.	19.107	0.5588	0.93	1.42	2.35	3.8	4.21E3
Boiler #3 ESP Oil Run No. 2 3/3/81, 1626-1736 hrs.	17.968	0.5255	2.88	0.43	3.31	5.7	6.30E3
Cleaned Bottom Gas Run No. 1 3/5/81, 1417-1434 hrs.	4.550	0.1331	6.67	0.04	6.71	45.7	5.04E4
Cleaned Bottom Gas Run No. 2 3/5/81, 1905-1919 hrs.	3.976	0.1163	4.98	<0.37	4.98	38.8	4.28E4
Cleaned Bottom Gas Run No. 3 3/6/81, 0117-0127 hrs.	4.091	0.1196	2.25	<0.38	2.25	17.0*	1.88E4
Raw Bottom Gas Run No. 1 3/5/81, 1440-1504 hrs.	3.726	0.1090	5.07	0.03	5.10	42.4	4.68E4
Raw Bottom Gas Run No. 2 3/5/81, 2052-2106 hrs.	4.022	0.1176	4.90	0.05	4.95	38.2	4.21E4
Raw Bottom Gas Run No. 3 3/6/81, 0057-0111 hrs.	3.704	0.1083	4.46	0.02	4.48	37.5*	4.14E4

¹Includes both CN- and SCN- analyses converted to HCN. Adjusted for CO₂ absorbed by impinger solutions.

*Data rejected due to process upset.

aEb = a x 10^b

TABLE B-41. ANALYTICAL RESULTS: GASEOUS NO_x

Description	Sample Volume		µg NO ₂	ppm NO _x	g/Nm ³ NO _x
	SCF 60°F 1 atm.	Nm ³ 25°C 1 atm.			
Product Gas Boilers 2/26/81, 0525 hrs.	0.0497	0.001454	932	341	6.41E5
Product Gas Boilers 2/26/81, 0525 hrs.	0.0500	0.001462	750	273	5.13E5
Product Gas Boilers 2/27/81, 1005 hrs.	0.0501	0.001465	571	207	3.90E5
Boiler No. 3 ESP Oils 3/3/81, 1522 hrs.	0.0509	0.001489	1120	400	7.52E5
Boiler No. 3 ESP Oils 3/3/81, 1522 hrs.	0.0502	0.001468	1240	449	8.45E5
Boiler No. 3 ESP Oils 3/3/81, 1910 hrs.	0.0478	0.001398	1115	424	7.98E5
Boiler No. 3 ESP Oils 3/3/81, 1910 hrs.	0.0477	0.001395	1045	399	7.49E5
Boiler No. 3 Natural Gas 3/4/81, 1022 hrs.	0.0466	0.001363	277	108	2.03E5
Boiler No. 3 Natural Gas 3/4/81, 1022 hrs.	0.0442	0.001293	272	112	2.10E5
Boiler No. 3 Natural Gas 3/4/81, 1127 hrs.	0.0432	0.001263	264	111	2.09E5
Boiler No. 3 Natural Gas 3/4/81, 1127 hrs.	0.0443	0.001296	260	107	2.01E5

aEb = a x 10^b

TABLE B-42. ANALYTICAL RESULTS: GASEOUS SO_x

Description	Sample Volume			1st Imp. mg SO ₂	2nd Imp. mg SO ₂	Total mg SO ₂	ppm SO ₂	$\mu\text{g}/\text{Nm}^3$ SO ₂
	SCF 60°F 1 atm.	Nm ³ 25°C 1 atm.						
Product Gas Boilers Run No. 1, 2/25/81 0422-0500 hrs	16.159	0.4726		336.5	8.07	344.6	279	7.29E5
Product Gas Boilers Run No. 2, 2/25/81 0723-0800 hrs. & 0915-0945 hrs.	26.181	0.7657		510.1	10.2	520.3	260	6.80E5
Boiler No. 3 ESP Oils No. 1, 3/3/81, 1209-1251 hrs.	15.253	0.4461		183.1	1.61	184.7	158	414E5
Boiler No. 3 ESP Oils No. 2, 3/3/81, 1458-1430 hrs.	11.711	0.3425		139.7	1.06	140.8	157	411E5
Boiler No. 3 ESP Oils No. 3, 3/3/81, 1601-1623 hrs.	9.314	0.2724		95.5	1.00	96.5	135	3.54E5
Boiler No. 3 ESP Oils No. 4, 3/3/81, 1739-1810 hrs.	5.683	0.1662		59.6	1.31	60.9	140	3.66E5
Boiler No. 3 Natural Gas Run No. 1, 3/4/81, 1115-1145 hrs.	12.945	0.3786		4.04	0.68	4.72	4.8	1.25E4
Boiler No. 3 Natural Gas Run No. 2, 3/4/81, 1148-1218 hrs.	11.637	0.3403		2.90	0.35	3.25	3.7	9.55E3
Boiler No. 3 Natural Gas Run No. 3, 3/4/81, 1233-1303 hrs.	10.005	0.2926		1.61	0.36	1.97	2.6	6.73E3

aEb = a x 10^b

TABLE B-43. WATER CONTENT OF GASEOUS PROCESS AND EMISSION STREAMS

Stream	Date	Time	(%)
Combined Product Gas	2/24	2024-2036	10.0
	2/25	0513-0529	9.24
	2/25	0937-0952	9.50
	2/26-27	2327-0015	10.5
	2/27	0315-0355	10.0
	2/27	0550-0635	8.47
	2/27	2230-2235	8.22
	2/28	0035-0130	10.4
	2/28	0225-0335	10.4
Combustion Emissions from Product Gas Boilers	2/25	0222-0312	9.81
	2/25	0422-0500	9.87
	2/25	0723-0800/0915-0945	8.70
	2/25	0820-0910	8.82
	2/25	0057-0701	8.66
Cleaned Top Product Gas	2/27	1055-1111	19.7
	2/27	1116-1615	19.4
	2/27	1655-1737	18.5
	2/27	1802-2022	19.3
	2/27	2045-2057	19.0
Raw Top Product Gas	2/27	1047-1057	20.0
	2/27	1102-1332	19.6
	2/27	1405-1415	19.7
	2/27	1435-1635	22.8
	2/27	1706-1726	14.4
Cleaned Bottom Product Gas	3/5	1437-1452	3.20
	3/5	1456-1856	4.71
	3/5	1923-1937	4.5
	3/5-6	2004-0056	4.25
Raw Bottom Product Gas	3/5	1508-1532	4.52
	3/5	1534-2046	3.62
	3/5	2111-2128	4.22
	3/5-6	2137-0046	6.24
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3/3	1049-1205	7.05
	3/3	1209-1251	7.12
	3/3	1430-1458	6.89
	3/3	1601-1623	6.98
Combustion Emissions from Boiler No. 3 Firing Natural Gas	3/4	1115-1145	14.4
	3/4	1148-1218	14.7
	3/4	1233-1303	16.0

TABLE B-44. RESULTS OF ANALYSES FOR MAJOR GASES IN GASEOUS PROCESS
AND EMISSION STREAMS¹

STREAM	DATE	TIME	COMPONENT						TOTAL
			CO ₂	H ₂	O ₂ + Ar	N ₂	CH ₄	CO	
Combined Product Gas	2/24/81	1015	3.44	18.8	0.62	48.6	2.16	30.6	104
	2/24/81	1145	3.91	19.9	0.60	48.7	2.39	30.6	106
	2/24/81	1325	3.35	16.3	0.61	49.8	1.92	30.3	102
	2/24/81	2330	4.13	19.8	1.03	54.0	2.26	32.0	113
	2/25/81	0115	3.90	19.3	0.53	52.6	2.20	31.9	110
	2/25/81	0320	4.87	18.3	1.30	46.7	1.80	24.2	97.2
	2/25/81	0550	3.90	17.8	0.67	46.6	1.56	26.1	96.6
	2/26/81	0905	5.26	16.4	0.53	46.2	2.61	28.0	99.0
	2/26/81	1015	4.2	13.6	0.57	44.9	2.01	28.2	93.5
	2/26/81	2225	5.06	14.6	0.80	50.2	1.93	28.0	100
	2/27/81	0050	4.76	15.0	0.78	48.2	1.94	28.4	99.1
	2/27/81	0415	4.74	15.0	0.97	49.0	1.97	28.9	100
	2/27/81	2050	4.22	13.6	0.74	47.1	1.84	29.2	96.7
	2/27/81	2335	3.53	13.2	0.71	43.3	1.92	27.7	90.4
	2/28/81	0145	3.99	14.0	0.78	43.9	1.91	27.7	92.3
	2/28/81	0345	4.14	12.2	0.68	43.7	1.72	26.7	89.2
	2/28/81	0545	3.93	13.6	0.76	42.8	1.87	26.8	89.8
	3/5/81	1704	5.24	15.7	0.80	45.4	2.61	28.5	98.3
Combustion Emissions from Product Gas Boiler	2/24/81	1540	8.24	0.07	12.3	80.9	ND	ND	101
	2/24/81	2040	8.52	ND	11.8	80.3	ND	ND	101
Cleaned Top Product Gas	2/26/81	1155	3.89	16.7	1.26	45.0	3.71	28.0	98.6
	2/26/81	1410	4.32	17.8	0.62	43.7	4.05	28.6	98.1
	2/27/81	1003	4.18	15.0	0.67	41.0	3.08	26.6	90.5
	2/27/81	1725	3.51	12.9	0.72	45.8	2.66	28.5	96.1
Raw Top Product Gas	2/26/81	1755	3.67	15.7	0.71	44.2	3.69	27.8	95.8
	2/26/81	2120	4.40	18.2	0.72	45.7	3.88	29.8	103
	2/27/81	1328	3.44	15.3	0.70	42.8	3.16	28.4	93.8
	2/27/81	1500	3.46	15.3	0.89	41.1	3.76	27.7	92.2
Cleaned Bottom Product Gas	3/5/81	1338	4.63	12.2	0.79	54.0	0.39	31.8	104
	3/5/81	1521	5.63	11.5	0.80	54.5	ND	30.8	103
Raw Bottom Product Gas	3/5/81	2021	6.34	11.9	0.89	53.7	ND	30.8	104
	3/5/81	2252	5.89	9.08	0.84	48.8	ND	26.0	90.6
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3/2/81	1135	9.74	ND	8.43	74.4	ND	ND	92.6
	3/2/81	1555	9.77	ND	9.81	75.6	ND	ND	95.2
	3/2/81	1825	10.1	ND	12.3	85.1	ND	ND	107
	3/3/81	0908	10.4	ND	8.38	74.8	ND	ND	93.6
	3/3/81	1150	10.9	ND	8.61	77.3	ND	ND	96.8
	3/3/81	1355	10.8	ND	8.72	77.3	ND	ND	96.9
Combustion Emissions for Boiler No. 3 Firing Natural Gas	3/4/81	1155	9.79	ND	4.87	83.5	ND	ND	98.2

¹Concentration in volume percent

TABLE B-45. RESULTS OF ANALYSES FOR HYDROCARBONS IN GASEOUS PROCESS AND EMISSION STREAMS¹

STREAM	DATE	TIME	CH ₄	COMPONENT						
				C ₂ H ₆	C ₂ H ₄	C ₃	C ₄	C ₅	C ₆	
Combined Product Gas	2/24/81	1015	1.92E4	1.58E3	2.68E2	8.02E2	3.65E2	2.36E2	2.46E2	
	2/24/81	1145	9.50E3	7.63E3	1.28E2	3.92E2	1.82E2	1.16E2	1.19E2	
	2/24/81	1325	1.71E4	1.40E3	2.37E2	7.10E2	3.27E2	2.10E2	2.30E2	
	2/24/81	2325	1.92E4	1.60E3	2.65E2	8.12E2	3.71E2	2.40E2	3.16E2	
	2/25/81	0115	1.80E4	1.49E3	2.48E2	7.58E2	3.45E2	2.21E2	2.72E2	
	2/25/81	0320	2.00E4	1.62E3	2.72E2	8.25E2	3.75E2	2.44E2	3.23E2	
	2/25/81	0540	1.62E4	1.25E3	2.06E2	6.35E2	2.88E2	1.87E2	2.58E2	
	2/26/81	1015	1.57E4	1.20E3	2.18E2	6.46E2	2.88E2	1.74E2	2.07E2	
	2/26/81	1712	2.41E4	1.84E3	3.92E2	9.48E2	4.61E2	2.77E2	3.12E2	
	2/24/81	1540	ND	ND	ND	ND	ND	ND	7.84E2	
Combustion Emissions from Product Gas Boilers	2/24/81	2015	ND	ND	ND	ND	ND	ND	1.80E2	
Cleansed Top Product Gas	2/26/81	1135	3.38E4	2.52E3	4.44E2	1.33E3	6.03E2	3.93E2	4.61E2	
	2/26/81	1400	3.54E4	2.66E3	4.68E2	1.40E3	6.28E2	3.80E2	4.40E2	
	2/27/81	0940	3.00E4	2.20E3	3.91E2	1.16E3	5.20E2	3.18E2	3.81E2	
	2/27/81	1715	1.91E4	1.50E3	2.76E2	7.94E2	3.56E2	2.19E2	2.49E2	
Raw Top Product Gas	2/26/81	1810	3.51E4	2.80E3	4.88E2	1.47E3	6.68E2	4.04E2	4.54E2	
	2/26/81	2140	3.05E4	2.21E3	4.98E2	1.30E3	5.83E2	3.59E2	4.70E2	
	2/27/81	1334	3.68E4	2.86E3	5.26E2	1.52E3	6.78E2	4.18E2	4.74E2	
	2/27/81	1447	3.43E4	2.41E3	5.53E2	1.44E3	6.34E2	3.81E2	4.01E2	
Cleansed Bottom Product Gas	3/5/81	1345	5.64E2	ND	1	ND	ND	ND	7	
	3/5/81	1531	4.22E2	ND	ND	ND	ND	ND	ND	
Raw Bottom Product Gas	3/5/81	2027	5.98E2	ND	ND	ND	ND	ND	ND	
	3/5/81	2300	5.13E2	ND	ND	ND	ND	ND	ND	
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3/2/81	1150	ND	ND	ND	ND	ND	ND	ND	
	3/2/81	1605	ND	ND	ND	ND	ND	ND	ND	
	3/2/81	1813	ND	ND	ND	ND	ND	ND	ND	
	3/3/81	0934	ND	ND	ND	ND	ND	ND	ND	
	3/3/81	1158	ND	ND	ND	ND	ND	ND	ND	
	3/3/81	1408	ND	ND	ND	ND	ND	ND	ND	
	3/3/81	1806	ND	ND	ND	ND	ND	ND	ND	
Combustion Emissions from Boiler No. 3 Firing Natural Gas	3/4/81	1205	ND	ND	ND	ND	ND	ND	ND	

APB = $\mu \times 10^6$

ND = Not detected (<0.1 ppm)

¹ Concentrations in parts per million (volume) - ppm

TABLE B-46. RESULTS OF ANALYSES FOR SULFUR SPECIES IN GASEOUS
PROCESS AND EMISSIONS STREAMS¹

STREAM	DATE	TIME	COMPONENT						
			H ₂ S	COS	SO ₂	CH ₃ SH	C ₂ H ₄ SH	CS ₂	C ₄ H ₄ S
Combined Product Gas	2/24/81	1015	794	77	ND	24	<1	<1	<1
	2/24/81	1145	867	74	ND	21	<1	<1	<1
	2/24/81	1325	746	78	ND	22	<1	<1	<1
	2/24/81	2325	663	70	ND	20	2	>1	ND
	2/25/81	0115	778	72	ND	21	2	<1	<1
	2/25/81	0230	726	72	ND	21	2	<1	<1
	2/25/81	0540	697	75	ND	16	2	<1	ND
	2/26/81	1015	713	75	ND	20	2	<1	ND
	3/5/81	1712	856	79	ND	38	4	<1	<1
	3/5/81	1329	612	86	ND	21	2	<1	ND
Combustion Emissions from Product Gas Boilers	2/24/81	1540	ND	ND	240	ND	ND	ND	ND
	2/24/81	2015	ND	ND	231	ND	ND	ND	ND
Cleaned Top Product Gas	2/26/81	1135	757	57	ND	37	3	<1	1
	2/26/81	1400	781	57	ND	38	3	<1	ND
	2/27/81	0940	695	55	ND	40	3	ND	1
	2/27/81	1715	708	57	ND	42	ND	ND	ND
Raw Top Product Gas	2/26/81	1810	774	58	ND	41	4	<1	1
	2/27/81	1334	715	56	ND	40	4	<1	1
	2/27/81	1447	751	56	ND	40	4	ND	
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3/2/81	1150	ND	5	*	ND	ND	ND	ND
	3/2/81	1605	ND	<1	77	ND	ND	ND	ND
	3/2/81	1833	ND	ND	160	ND	ND	ND	ND
	3/3/81	0904	ND	ND	220	ND	ND	ND	ND
	3/3/81	1158	ND	ND	219	ND	ND	ND	ND
	3/3/81	1408	ND	ND	245	ND	ND	ND	ND
	3/3/81	1808	ND	ND	206	ND	ND	ND	ND
Combustion Emissions from Boiler No. 3 Firing Natural Gas	3/4/81	1205	ND	ND	DR	ND	ND	ND	ND
Cleaned Bottom Product Gas	3/5/81	1345	84	322	ND	ND	ND	<1	ND
	3/5/81	1531	92	313	ND	ND	ND	ND	ND
Raw Bottom Product Gas	3/5/81	2027	809	92	ND	ND	ND	<1	ND
	3/5/81	2300	860	104	ND	ND	ND	<1	ND

*Results not reproducible.

<1 = Quantifiable detection limit.

ND = Not detected (<0.1 ppm).

¹Concentrations in parts per million (volume) - ppm.

DR = Data rejected due to analytical error.

TABLE B-47. RESULTS OF ANALYSES FOR ORGANIC LOADING

STREAM	TCO CONCENTRATION	CRAV CONCENTRATION	ORGANIC LOADING
Combustion Emissions from Product Gas Boilers	2.7 E1 $\mu\text{g}/\text{Nm}^3$	BC	2.7 E1 $\mu\text{g}/\text{Nm}^3$
Combustion Emissions from Boiler No. 3	5.8 E2 $\mu\text{g}/\text{Nm}^3$	2.1 E3 $\mu\text{g}/\text{Nm}^3$	2.7 E3 $\mu\text{g}/\text{Nm}^3$
Piring ESP Oil	3.7 E6 $\mu\text{g}/\text{Nm}^3$	3.1 E5 $\mu\text{g}/\text{Nm}^3$	4.0 E6 $\mu\text{g}/\text{Nm}^3$
Combined Product Gas	1.2 E5 $\mu\text{g}/\text{Nm}^3$	9.0 E4 $\mu\text{g}/\text{Nm}^3$	2.1 E5 $\mu\text{g}/\text{Nm}^3$
Aerosols	3.6 E6 $\mu\text{g}/\text{Nm}^3$	2.2 E5 $\mu\text{g}/\text{Nm}^3$	3.8 E6 $\mu\text{g}/\text{Nm}^3$
Vapors	6.9 E6 $\mu\text{g}/\text{Nm}^3$	4.8 E5 $\mu\text{g}/\text{Nm}^3$	7.4 E6 $\mu\text{g}/\text{Nm}^3$
Cleaned Top Product Gas	1.2 E5 $\mu\text{g}/\text{Nm}^3$	2.2 E4 $\mu\text{g}/\text{Nm}^3$	1.5 E5 $\mu\text{g}/\text{Nm}^3$
Aerosols	6.8 E6 $\mu\text{g}/\text{Nm}^3$	4.6 E5 $\mu\text{g}/\text{Nm}^3$	7.2 E6 $\mu\text{g}/\text{Nm}^3$
Vapors	8.8 E6 $\mu\text{g}/\text{Nm}^3$	3.8 E6 $\mu\text{g}/\text{Nm}^3$	1.3 E7 $\mu\text{g}/\text{Nm}^3$
Raw Top Product Gas	3.6 E6 $\mu\text{g}/\text{Nm}^3$	3.8 E6 $\mu\text{g}/\text{Nm}^3$	7.4 E6 $\mu\text{g}/\text{Nm}^3$
Aerosols	5.2 E6 $\mu\text{g}/\text{Nm}^3$	6.0 E3 $\mu\text{g}/\text{Nm}^3$	5.2 E6 $\mu\text{g}/\text{Nm}^3$
Vapors	8.8 E4 $\mu\text{g}/\text{Nm}^3$	7.1 E2 $\mu\text{g}/\text{Nm}^3$	8.9 E4 $\mu\text{g}/\text{Nm}^3$
Cleaned Bottom Product Gas	1.0 E4 $\mu\text{g}/\text{Nm}^3$	1.5 E3 $\mu\text{g}/\text{Nm}^3$	1.2 E4 $\mu\text{g}/\text{Nm}^3$
Raw Bottom Product Gas	7.3 E1 $\mu\text{g}/\text{g}$	3.0 E1 $\mu\text{g}/\text{g}$	1.0 E2 $\mu\text{g}/\text{g}$
Gasifier Ash	BC	1.2 E2 $\mu\text{g}/\text{g}$	1.2 E2 $\mu\text{g}/\text{g}$
Cyclone Dust	8.6 E4 $\mu\text{g}/\text{g}$	8.4 E5 $\mu\text{g}/\text{g}$	9.3 E5 $\mu\text{g}/\text{g}$
ESP Oils	1.0 E3 $\mu\text{g}/\text{L}$	BC	1.0 E3 $\mu\text{g}/\text{L}$
Ash Pan Water			

BC - Value equal to or less than blank.

aEb = a x 10^b

TABLE B-48. RESULTS OF GC-HECD ANALYSIS FOR ORGANIC SULFUR*

STREAM	TOTAL ORGANIC SULFUR	MAJOR PEAK**
Combustion Emissions from Product Gas Boilers	4.44 E1 $\mu\text{g}/\text{Nm}^3$	1.29 E1 $\mu\text{g}/\text{Nm}^3$
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3.30 E1 $\mu\text{g}/\text{Nm}^3$	5.43 E0 $\mu\text{g}/\text{Nm}^3$
Combined Product Gas	4.16 E4 $\mu\text{g}/\text{Nm}^3$	2.92 E3 $\mu\text{g}/\text{Nm}^3$
Gasifier Ash	3.42 E0 $\mu\text{g}/\text{g}$	3.32 E0 $\mu\text{g}/\text{g}$
Cyclone Dust	1.75 E1 $\mu\text{g}/\text{g}$	1.72 E1 $\mu\text{g}/\text{g}$
ESP Oils	5.63 E3 $\mu\text{g}/\text{g}$	3.06 E2 $\mu\text{g}/\text{g}$
Ash Pan Water	9.21 E2 $\mu\text{g}/\text{L}$	8.99 E2 $\mu\text{g}/\text{L}$

*Analysis performed by gas chromatograph with Hall electrolytic conductivity detector (HECD) in sulfur-specific mode.

**Quantitation of single major peak of chromatogram - indicates maximum concentration of a single sulfur-containing compound.

aEb = $a \times 10^b$

TABLE B-49. RESULTS OF GC-HECD ANALYSIS FOR ORGANIC NITROGEN*

STREAM	TOTAL ORGANIC NITROGEN	MAJOR PEAK**
Combustion Emissions from Product Gas Boilers	1.74 E0 $\mu\text{g}/\text{Nm}^3$	1.09 E0 $\mu\text{g}/\text{Nm}^3$ †
Combustion Emissions from Boiler No. 3 Firing ESP Oil	3.91 E0 $\mu\text{g}/\text{Nm}^3$	3.67 E0 $\mu\text{g}/\text{Nm}^3$
Combined Product Gas	2.44 E4 $\mu\text{g}/\text{Nm}^3$	8.76 E3 $\mu\text{g}/\text{Nm}^3$
Gasifier Ash	BC	2.82 E4 $\mu\text{g}/\text{g}$ †
Cyclone Dust	BC	BC
ESP Oils	6.58 E2 $\mu\text{g}/\text{g}$	7.09 E1 $\mu\text{g}/\text{g}$
Ash Pan Water	3.12 E-1 $\mu\text{g}/\text{L}$	2.46 E-1 $\mu\text{g}/\text{L}$

*Analysis of extract of XAD-2 resin or liquid or solid sample performed by gas chromatograph with Hall electrolytic conductivity detector (HECD) in nitrogen-specific mode.

**Quantitation of single major peak of chromatogram - indicates maximum concentration of a single nitrogen-containing organic compound.

BC - Value equal to or less than blank.

†No corresponding peak in blank, therefore, measured sample concentration not reduced by blank value.

aEb = $a \times 10^b$

TABLE B-50. RESULTS OF GC-HECD ANALYSIS FOR ORGANIC HALOGEN*

STREAM	TOTAL ORGANIC HALOGEN	MAJOR PEAK**
Combustion Emissions from Product Gas Boilers	BC	BC
Combustion Emissions from Boiler No. 3 Firing ESP Oil	1.15 E0 $\mu\text{g}/\text{Nm}^3$	4.29 E-1 $\mu\text{g}/\text{Nm}^3$
Combined Product Gas	1.21 E2 $\mu\text{g}/\text{Nm}^3$	1.21 E2 $\mu\text{g}/\text{Nm}^3$
Gasifier Ash	1.85 E-2 $\mu\text{g}/\text{g}$	5.00 E-3 $\mu\text{g}/\text{g}$
Cyclone Dust	4.80 E-2 $\mu\text{g}/\text{g}$	4.80 E-2 $\mu\text{g}/\text{g}$
ESP Oils	BC	BC
Ash Pan Water	3.05 $\mu\text{g}/\text{L}$	3.05 $\mu\text{g}/\text{L}$

*Analysis performed by gas chromatograph with Hall electrolytic conductivity detector (HECD) in halogen-specific mode.

**Quantitation of single major peak of chromatogram - indicates maximum concentration of a single halogen-containing organic compound.

BC - Value equal to or less than blank.

TABLE B-51. RESULTS OF GC-ECD ANALYSIS FOR POLYCHLORINATED BIPHENYLS*

STREAM	CONCENTRATION ¹
Combustion Emissions from Product Gas Boilers	<3.4 $\mu\text{g}/\text{Nm}^3$
Combustion Emissions from Boiler No. 3 Firing ESP Oil	<2.4 $\mu\text{g}/\text{Nm}^3$
Combined Product Gas	<290 $\mu\text{g}/\text{Nm}^3$
Gasifier Ash	<0.4 $\mu\text{g}/\text{g}$
Cyclone Dust	<0.13 $\mu\text{g}/\text{g}$
ESP Oils	<200 $\mu\text{g}/\text{g}$
Ash Pan Water	<6.7 $\mu\text{g}/\text{L}$

*Analysis performed by gas chromatograph with electron capture detector (ECD)

¹All results below analytical detection limit.

TABLE B-52. RESULTS OF GC/MS ANALYSIS OF GASIFIER ASH¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
1	8.10	NI ²	69208	0.76	0.151
2	8.15	NI	113851	1.25	0.249
3	8.23	NI	71800	0.79	0.157
4	8.33	NI	39129	0.43	0.086
5	8.43	Alkyl benzene	254590	2.79	0.555
6	8.63	Alkyl benzene	735384	8.06	1.60
7	8.78	NI	50614	0.56	0.111
8	8.93	NI	24411	0.27	0.054
9	9.05	Alkyl benzene	111243	1.22	0.243
10	9.10	m-Menthane	97060	1.06	0.211
11	9.17	NI	49080	0.54	0.107
12	9.27	NI	21180	0.23	0.048
13	9.37	Alkyl benzene	389617	4.27	0.850
14	9.55	C ₁₀	482822	5.29	1.05
15	9.72	NI	17947	0.20	0.039
16	9.78	NI	22754	0.25	0.049
17	9.83	NI	34028	0.37	0.074
18	9.98	NI	12079	0.13	0.026
19	10.10	Hydrocarbon	284832	3.12	0.621
20	10.22	Cyclic hydrocarbon	172528	1.89	0.376
21	10.27	NI	93643	1.03	0.205
22	10.32	Alkane	24375	0.27	0.053
23	10.43	Alkane	85661	0.94	0.187
24	10.53	C ₁₁ alkene	6878	0.08	0.016
25	10.60	C ₁₁ alkene	13187	0.14	0.028
26	10.77	NI	127938	1.40	0.279
27	10.80	Alkyl benzene	22928	0.25	0.049
28	10.90	Alkane	128623	1.41	0.281
29	10.95	Alkyl benzene	69902	0.77	0.153
30	11.07	Alkane	107890	1.18	0.235
31	11.15	Alkyl benzene	19600	0.21	0.042
32	11.20	Alkane	75334	0.83	0.165
33	11.30	Unsaturated hydrocarbon	18372	0.20	0.039
34	11.33	NI	27860	0.31	0.062
35	11.40	Alkyl benzene	50838	0.56	0.111
36	11.43	Alkyl benzene	41206	0.45	0.089
37	11.55	NI	31445	0.34	0.067
38	11.60	Alkyl benzene	92815	1.02	0.203
39	11.87	NI	41120	0.45	0.089

(Continued)

TABLE B-52. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
40	11.97	C ₁₁	448586	4.92	0.979
41	12.08	Alkyl benzene	23879	0.26	0.052
42	12.18	NI	31121	0.34	0.067
43	12.35	Alkyl benzene	69104	0.76	0.151
44	12.45	Alkyl benzene	16665	0.18	0.036
45	12.60	Hydrocarbon	69349	0.76	0.151
46	12.80	Cyclic C ₁₁	53321	0.58	0.115
47	12.90	Hydrocarbon	39496	0.43	0.086
48	13.08	Alkyl benzene	8534	0.09	0.016
49	13.25	NI	41488	0.45	0.089
50	13.32	NI	46870	0.51	0.101
51	13.52	Alkane	8879	0.10	0.019
52	13.65	Unsaturated hydrocarbon	34033	0.37	0.074
53	14.18	NI	18314	0.20	0.039
54	14.27	NI	33543	0.37	0.074
55	14.40	C ₁₂	159974	1.75	0.348
56	14.72	Hydrocarbon	34916	0.38	0.076
57	15.33	NI	12365	0.14	0.028
58	15.80	NI	7485	0.08	0.016
59	16.03	NI	33255	0.36	0.072
60	16.67	NI	8401	0.09	0.018
61	16.78	C ₁₃	336639	3.69	0.734
62	17.28	Aromatic compd.	31214	0.34	0.068
63	17.38	Aromatic compd.	12911	0.14	0.028
64	17.67	Hydrocarbon	19990	0.22	0.044
65	17.78	Aromatic compd.	43787	0.48	0.096
66	18.02	Aromatic compd.	17626	0.19	0.038
67	18.12	Alkyl benzene	32868	0.36	0.072
68	18.27	Alkyl benzene	86511	0.95	0.189
69	18.40	Alkyl benzene	89432	0.98	0.195
70	18.53	Hydrocarbon	41366	0.45	0.089
71	18.87	NI	8132	0.09	0.018
72	19.10	C ₁₄	579057	6.35	1.26
73	19.48	Aromatic compd.	14788	0.16	0.032
74	19.88	Aromatic compd.	23382	0.26	0.052
75	20.03	NI	97118	1.06	0.211
76	20.15	Aromatic compd.	71747	0.79	0.157
77	20.30	Aromatic compd.	14274	0.16	0.032
78	20.40	Hydrocarbon	44160	0.48	0.096

(Continued)

TABLE B-52. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
79	20.55	Aromatic compd.	128496	1.41	0.281
80	20.62	NI	57018	0.63	0.125
81	20.68	Aromatic compd.	146615	1.61	0.320
82	21.22	Hydrocarbon	183822	2.02	0.402
83	22.03	Aromatic compd.	101363	1.11	0.221
84	22.23	NI	62441	0.68	0.135
85	22.60	NI	39553	0.43	0.086
86	23.33	NI	558920	6.13	1.22
87	23.97	NI	22600	0.25	0.049
88	24.02	NI	59947	0.66	0.131
89	24.22	NI	49787	0.55	0.109
90	24.62	NI	38942	0.43	0.086
91	25.13	Unsaturated hydrocarbon	154920	1.70	0.338
92	25.23	NI	29285	0.32	0.064
93	25.33	NI	51747	0.68	0.135
94	25.83	NI	33212	0.36	0.072
95	25.93	NI	28146	0.31	0.062
96	26.15	NI	24558	0.27	0.054
97	26.55	NI	22725	0.25	0.049
98	26.97	d ₁₀ Anthracene	183750	2.01	--*
99	27.23	NI	23002	0.25	0.049
100	28.85	NI	13561	0.15	0.029
101	30.55	NI	13717	0.15	0.029
102	31.42	NI	73628	0.81	0.161
103	32.15	NI	14534	0.16	0.032
104	33.72	NI	14677	0.16	0.032
105	35.20	NI	15408	0.17	0.034
106	36.63	NI	17942	0.20	0.039
107	38.00	NI	17756	0.19	0.038
108	38.73	NI	29227	0.32	0.064
Total Eluted				19.478	$\mu\text{g/g}$
Total Identified				14.003	$\mu\text{g/g}$
% Identified				71.9	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

* - Internal reference standard.

TABLE B-53. RESULTS OF GC/MS ANALYSIS OF ASH PAN WATER¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/L}$
1	8.78	Phenol	9116	1.22	3.23
2	9.62	NI ²	14147	1.89	5.00
3	9.93	NI	13077	1.75	4.63
4	10.12	NI	8904	1.19	3.15
5	10.58	Chlorinated Alkane	24940	3.33	8.82
6	10.75	NI	8758	1.17	3.10
7	11.87	Methyl benzoate	13181	1.76	4.66
8	12.63	Dichlorophenol	33742	4.51	11.9
9	13.12	NI	11595	1.55	4.10
10	13.75	NI	3102	0.41	1.09
11	14.18	NI	10021	1.34	3.55
12	14.53	NI	2924	0.39	1.03
13	14.73	NI	4860	0.65	1.72
14	15.13	NI	14053	1.88	5.00
15	15.35	NI	5720	0.76	2.01
16	17.10	NI	20615	2.75	7.28
17	17.18	NI	11974	1.60	4.24
18	18.28	NI	2925	0.39	1.03
19	20.63	Alkyl benzene	35364	4.73	12.5
20	21.25	NI	5769	0.77	2.04
21	21.60	NI	8061	1.08	2.86
22	23.30	NI	16617	2.22	5.88
23	25.15	Unsaturated hydro-carbon	125614	16.79	44.5
24	26.97	d ₁₀ -Anthracene	188453	25.18	—*
25	30.05	NI	2112	0.28	0.74
26	31.42	Sulfur	84359	11.27	29.8
27	32.40	NI	4992	0.67	1.77
28	33.73	NI	1924	0.26	0.69
29	36.70	Dioctyl adipate	33373	4.46	11.8
30	38.77	Bis(2-ethylhexyl) phthalate	22403	2.99	7.92
31	44.18	NI	5588	0.75	1.99
Total Eluted				198.02 $\mu\text{g/L}$	
Total Identified				135.1 $\mu\text{g/L}$	
% Identified				68.2	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

* - Internal reference standard.

TABLE B-54. RESULTS OF GC/MS ANALYSIS OF CYCLONE DUST¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
1	3.57	NI ²	1082511	23.00	3.45
2	4.17	Chlorine compd.	77697	1.65	0.247
3	4.82	Toluene	142489	3.03	0.454
4	6.50	NI	17669	0.38	0.057
5	6.65	Xylene	87783	1.87	0.280
6	7.12	NI	32591	0.69	0.103
7	7.27	NI	26478	0.56	0.084
8	7.47	NI	25979	0.55	0.082
9	8.00	NI	21460	0.46	0.069
10	8.45	NI	19610	0.42	0.063
11	8.65	NI	17043	0.36	0.054
12	8.80	NI	10173	0.22	0.033
13	9.17	NI	10370	0.22	0.033
14	9.37	Alkyl benzene	48253	1.03	0.154
15	9.52	C ₁₀	46861	1.00	0.150
16	10.07	NI	29634	0.63	0.094
17	10.35	5-Methyl-1,3-benzenediol	138956	2.95	0.442
18	11.88	NI	17168	0.36	0.054
19	11.95	NI	57019	1.21	0.181
20	14.07	Naphthalene	34928	0.74	0.111
21	14.42	NI	33650	0.72	0.108
22	16.80	Alkyl naphthalene	30262	0.64	0.096
23	19.08	Dialkyl naphthalene	35558	0.76	0.114
24	21.10	Dialkyl naphthalene	14521	0.31	0.046
25	21.25	NI	47388	1.01	0.151
26	21.77	NI	18700	0.40	0.06
27	23.30	NI	9680	0.21	0.031
28	24.27	NI	20203	0.43	0.064
29	24.95	NI	26502	0.56	0.084
30	25.17	NI	48893	1.04	0.156
31	25.23	NI	44214	0.94	0.141
32	25.35	NI	42654	0.91	0.136
33	26.98	d ₁₀ -Anthracene	416816	8.86	—*
34	27.10	NI	73509	1.56	0.234
35	28.87	NI	67260	1.43	0.234
36	30.50	NI	18897	0.40	0.06
37	30.57	NI	89504	1.90	0.285

(Continued)

* - Internal reference standard.

TABLE B-54. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
38	30.67	NI	56978	1.21	0.181
39	31.40	NI	284568	6.05	0.907
40	31.53	NI	453228	9.63	1.44
41	32.15	C ₂₁	83593	1.78	0.267
42	33.70	C ₂₂	107835	2.29	0.343
43	35.18	C ₂₃	115987	2.46	0.369
44	36.60	C ₂₄	124497	2.65	0.397
45	37.98	C ₂₅	161607	3.43	0.514
46	38.72	NI	41121	0.87	0.130
47	39.30	C ₂₆	117062	2.49	0.373
48	40.58	C ₂₇	89512	1.90	0.285
49	41.82	NI	52732	1.12	0.168
50	43.00	NI	15731	0.33	0.049
51	45.10	NI	18594	0.40	0.06
Total Eluted				13.671	$\mu\text{g/g}$
Total Identified				4.600	$\mu\text{g/g}$
% Identified				33.6	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

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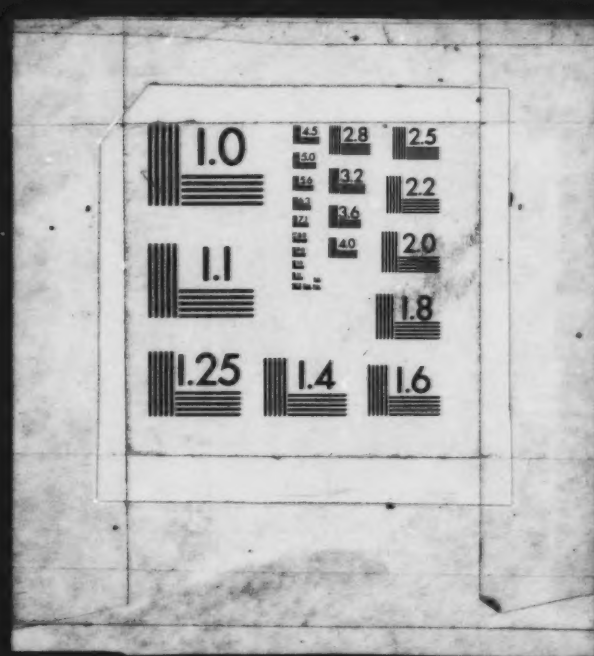


TABLE B-55. RESULTS OF GC/MS ANALYSIS OF COMBUSTION EMISSIONS
FROM BOILER NO. 3 FIRING ESP OIL¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g}/\text{Nm}^3$
1	8.30	Trichloropropane	93139	2.21	10.5
2	8.62	Benzaldehyde	33776	0.80	3.81
3	10.73	C ₄ -Alkyl benzene	29781	0.71	3.38
4	10.88	C ₄ -Alkyl benzene	18592	0.44	2.09
5	11.17	C ₄ -Alkyl benzene	29518	0.70	3.33
6	11.62	Methylindan	140864	3.34	15.9
7	11.80	Methylindan	44889	1.06	4.99
8	11.90	Methyl benzoate	63563	1.51	7.19
9	12.38	C ₂ Alkyl styrene	27718	0.66	3.14
10	13.00	NI ²	28785	0.68	3.24
11	13.57	Aliphatic aldehyde	50953	1.21	5.76
12	14.05	Naphthalene	259208	6.15	29.3
13	18.68	Biphenyl	34035	0.81	3.86
14	20.67	NI	63217	1.50	7.14
15	23.80	Alkyl biphenyl	32428	0.77	3.67
16	25.17	Unsaturated hydro- carbon	185984	4.41	26.0
17	26.87	Phenanthrene	26786	0.84	3.05
18	27.00	d ₁₀ -Anthracene	211249	5.01	--*
19	32.17	C ₂₁	28684	0.68	3.24
20	33.72	C ₂₂	78296	1.86	8.86
21	35.20	C ₂₃	125733	2.98	14.2
22	36.77	Di-n-octyl adipate	2138340	50.70	241
23	38.00	C ₂₅	129793	3.08	14.7
24	38.77	bis(2-ethylhexyl) phthalate	185477	4.40	20.9
25	39.33	C ₂₆	103701	2.46	11.7
26	40.62	C ₂₇	52845	1.25	5.95
Total Eluted				451.9	$\mu\text{g}/\text{Nm}^3$
Total Identified				441.52	$\mu\text{g}/\text{Nm}^3$
% Identified				97.7	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

* - Internal reference standard.

TABLE B-56. RESULTS OF GC/MS ANALYSIS OF COMBUSTION EMISSIONS
FROM PRODUCT GAS BOILERS¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc., $\mu\text{g}/\text{Nm}^3$
1	8.62	Benzaldehyde	68494	4.88	8.43
2	9.52	Phenol	24319	1.73	2.99
3	10.75	NI ²	17040	1.21	2.09
4	10.90	NI	15155	1.08	1.87
5	11.63	Ethyl styrene	98231	7.00	12.1
6	11.80	NI	35215	2.51	4.33
7	11.88	NI	30973	2.21	3.82
8	11.95	C ₁₁	24251	1.73	2.99
9	12.38	Dimethylindan	43976	3.13	5.41
10	13.57	NI	18283	1.30	2.24
11	13.93	NI	17494	1.25	2.16
12	14.06	Naphthalene	175704	12.52	21.6
13	14.22	NI	14908	1.06	1.83
14	16.72	Methyl naphthalene	8458	0.60	1.04
15	17.12	NI	17231	1.23	2.12
16	18.70	NI	13358	0.95	1.64
17	19.08	NI	14010	1.00	1.73
18	20.65	Aromatic compd.	76994	5.49	9.48
19	21.25	NI	13018	0.93	1.61
20	23.33	Phthalate	39865	2.84	4.90
21	23.82	NI	23764	1.69	2.92
22	24.10	NI	17304	1.23	2.12
23	25.17	Unsaturated hydro- carbon	214802	15.30	26.4
24	26.88	Phenanthrene	17329	1.23	2.12
25	27.00	d ₁₀ -Anthracene	202785	14.45	—*
26	28.87	NI	12319	0.94	1.62
27	30.07	NI	14288	1.02	1.76
28	30.55	NI	13352	0.95	1.64
29	33.73	NI	10793	0.77	1.33
30	35.20	NI	12339	0.88	1.52
31	36.70	Di-n-octyl adipate	52601	3.75	6.48
32	38.78	Bis(2-ethylhexyl) phthalate	44078	3.14	5.42
Total Eluted				147.71	$\mu\text{g}/\text{Nm}^3$
Total Identified				110.11	$\mu\text{g}/\text{Nm}^3$
% Identified				74.5	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

* - Internal reference standard.

TABLE B-57. RESULTS OF GC/MS ANALYSIS OF COMBINED GAS¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	µg/Nm ³
1	8.15	NI ²	165018	0.30	1517
2	8.30	NI	90674	0.16	809
3	8.43	Alkyl benzene	296287	0.54	2731
4	8.65	Alkyl benzene	909703	1.65	8345
5	8.78	Alkyl benzene	452650	0.82	4147
6	9.07	Alkyl benzene	361148	0.65	3287
7	9.12	NI	172794	0.31	1568
8	9.42	Phenol	2037577	3.69	1.86E4
9	9.48	NI	219377	0.40	2023
10	9.58	Decane	462952	0.84	4248
11	9.70	Decalin	118769	0.22	1113
12	9.88	Decene	175244	0.32	1618
13	10.12	Alkyl benzene	1132667	2.05	1.04E4
14	10.42	Indan	572858	1.04	5259
15	10.62	NI	228986	0.41	2073
16	10.78	Alkyl benzene	308088	0.56	2832
17	10.92	NI	172427	0.31	1568
18	10.98	Cresol	913907	1.66	8395
19	11.07	Alkyl benzene	516850	0.94	4754
20	11.18	NI	10951	0.02	101
21	11.35	NI	72503	0.13	657
22	11.50	NI	475567	0.86	4349
23	11.67	Cresol	2221164	4.02	2.03E4
24	12.05	Undecane	802445	1.45	7333
25	12.13	Alkyl benzene	261255	0.47	2377
26	12.27	Xylenol	604261	1.09	5512
27	12.42	Alkyl benzene	232865	0.42	2124
28	12.55	Alkyl benzene	235041	0.43	2174
29	12.63	NI	57408	0.10	506
30	13.00	Methyl indan	361848	0.66	3337
31	13.12	NI	131048	0.24	1213
32	13.27	NI	372756	0.68	3438
33	13.33	Xylenol	1093977	1.98	1.00E4
34	13.40	NI	1007258	1.83	9254
35	13.57	NI	182954	0.33	1668
36	13.67	NI	36230	0.07	354
37	13.88	Xylenol	839028	1.52	7686
38	13.95	NI	911551	1.65	8344
39	14.08	Naphthalene	713885	1.29	6523
40	14.20	Alkyl indan	275182	0.50	2528

(Continued)

TABLE B-57. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g}/\text{Nm}^3$
41	14.27	Hydrocarbon	338158	0.61	3084
42	14.33	NI	266180	0.48	2427
43	14.47	Alkyl indan	1290487	2.34	11833
44	14.60	Dodecane	208829	0.38	1921
45	14.67	Alkyl phenol	191898	0.35	1769
46	14.72	NI	56429	0.10	505
47	14.82	Hydrocarbon	391759	0.71	3590
48	14.90	NI	431095	0.78	3944
49	15.03	NI	29071	0.05	252
50	15.23	NI	295402	0.54	2730
51	15.30	Alkyl phenol	1040255	1.88	9507
52	15.50	Alkyl phenol	461069	0.84	4247
53	15.58	Alkyl phenol	360382	0.65	3289
54	15.70	Methyl tetralin	149103	0.27	1365
55	15.80	Alkyl indan	156945	0.28	1415
56	15.96	Alkyl indan	443236	0.80	4045
57	16.05	Alkyl phenol	694707	1.26	6372
58	16.18	Alkyl phenol	687432	1.25	6321
59	16.28	Methyl tetralin	299511	0.54	2730
60	16.48	NI	780712	1.41	7130
61	16.60	NI	111193	0.20	1011
62	16.67	Tridecane	611824	1.11	5613
63	16.82	Methyl naphthalene	1375502	2.49	12591
64	16.90	NI	628947	1.12	5663
65	16.98	NI	440270	0.80	4045
66	17.20	Methyl naphthalene	1224050	2.22	11226
67	17.38	Alkyl tetralin	844627	1.53	7737
68	17.57	Alkyl tetralin	502046	0.91	4601
69	17.70	NI	78378	0.14	708
70	17.85	Alkyl tetralin	647123	1.17	5916
71	17.93	NI	179189	0.32	1618
72	18.05	NI	92505	0.17	859
73	18.28	Hydrocarbon	489814	0.89	4500
74	18.43	NI	469868	0.85	4298
75	18.60	Alkyl tetralin	598308	1.08	5461
76	18.73	NI	73046	0.13	657
77	18.95	NI	390169	0.71	3590
78	19.03	Hydrocarbon	418611	0.76	3843

(Continued)

TABLE B-57. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g}/\text{Nm}^3$
79	19.12	NI	109299	0.20	1011
80	19.23	Tetradecane	22765	0.04	202
81	19.33	NI	488075	0.88	4450
82	19.47	NI	156602	0.28	1416
83	19.67	Dimethyl naphthalene	659547	1.20	6068
84	19.75	Dimethyl naphthalene	1101908	2.00	10114
85	19.95	NI	197696	0.36	1820
86	20.07	Dimethyl naphthalene	297240	0.54	2730
87	20.15	Dimethyl naphthalene	193953	0.35	1769
88	20.27	NI	95497	0.17	859
89	20.33	NI	48186	0.09	455
90	20.40	Dimethyl naphthalene	151622	0.27	1365
91	20.48	Hydrocarbon	408844	0.74	3742
92	20.62	Alkyl benzene	198499	0.36	1820
93	20.73	Alkyl benzene	157545	0.29	1466
94	20.85	NI	106484	0.19	960
95	21.05	NI	674020	1.22	6169
96	21.12	NI	845886	1.53	7737
97	21.22	Alkyl phenol	18249	0.03	151
98	21.30	Pentadecane	759866	1.38	6978
99	21.40	NI	142832	0.26	1314
100	21.58	NI	127365	0.23	1163
101	21.70	Dibenzofuran	160869	0.29	1466
102	21.83	NI	561232	1.02	5158
103	21.90	Alkyl naphthalene	230068	0.42	2124
104	22.02	Alkyl naphthalene	218117	0.40	2023
105	22.17	NI	5021	0.01	50
106	22.35	Alkyl naphthalene	282842	0.51	2579
107	22.68	Alkyl naphthalene	1001691	1.82	9203
108	22.75	NI	340454	0.62	3135
109	23.08	Fluorene	208096	0.38	1921

(Continued)

TABLE B-57. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g}/\text{Nm}^3$
110	23.22	Alkyl naphthalene	1029048	1.86	9406
111	23.35	NI	663401	1.20	6068
112	23.42	NI	233755	0.42	2124
113	23.57	Alkyl naphthalene	203370	0.37	1871
114	23.83	Methyl fluorene	120337	0.22	1112
115	24.10	Methyl fluorene	89021	0.16	809
116	24.27	Hydrocarbon	428916	0.78	3944
117	24.80	Alkyl naphthalene	73661	0.13	657
118	24.95	Alkyl naphthalene	161954	0.29	1466
119	25.10	NI	417447	0.76	3843
120	25.27	Hydrocarbon	500967	0.91	4601
121	25.37	Hydrocarbon	468450	0.85	4298
122	25.70	NI	134934	0.24	1213
123	25.83	Hydrocarbon	364820	0.66	3337
124	26.03	NI	487280	0.88	4450
125	26.98	Anthracene	340075	0.62	3135
126	27.08	d ₁₀ -Anthracene	317469	0.58	—*
127	28.95	C ₁₉	184001	0.33	1668
128	30.53	C ₂₀	104509	0.19	960
129	30.65	NI	95905	0.17	859
130	32.17	C ₂₁	74332	0.13	657
131	33.72	C ₂₂	60380	0.11	556
132	35.18	C ₂₃	62420	0.11	556
133	36.70	Di-n-octyl adipate	544497	0.99	5006
134	38.77	C ₂₅	109951	0.20	1011
Total Eluted				505875 $\mu\text{g}/\text{Nm}^3$	
Total Identified				369742 $\mu\text{g}/\text{Nm}^3$	
% Identified				73.1	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

* - Internal reference standard.

TABLE B-58. RESULTS OF GC/MS ANALYSIS OF ESP OIL TO BOILER NO. 3¹

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
1	11.40	Cresol	63002	0.28	270
2	11.93	Undecane	23291	0.10	96
3	13.20	Methyl cresol	7754	0.03	28.9
4	13.68	Methyl cresol	56833	0.25	241
5	14.03	Naphthalene	29696	0.13	125
6	14.40	C ₁₂	52773	0.23	222
7	14.73	NI ²	18863	0.08	77.0
8	15.88	NI	15603	0.07	67.4
9	16.42	Dimethyl indan	39007	0.17	164
10	16.58	NI	21255	0.09	86.7
11	16.68	1-Methylnaphthalene	124161	0.55	530
12	16.78	C ₁₃	113731	0.51	491
13	17.10	2-Methylnaphthalene	91029	0.41	395
14	17.30	C ₃ -Alkyl indan	28787	0.13	125
15	17.48	C ₃ -Alkyl indan	34418	0.15	144
16	17.78	C ₃ -Alkyl indan	18673	0.08	77.0
17	18.27	NI	19028	0.08	77.0
18	18.55	NI	58735	0.26	250
19	18.88	NI	35631	0.16	154
20	18.97	Dimethylnaphthalene	56768	0.25	241
21	19.08	C ₁₄	184670	0.82	790
22	19.23	Dimethylnaphthalene	118801	0.53	510
23	19.57	Dimethylnaphthalene	117813	0.52	500
24	19.63	NI	225335	1.00	963
25	20.00	Dimethylnaphthalene	41524	0.18	173
26	20.07	Dimethylnaphthalene	46440	0.21	202
27	20.35	Dimethylnaphthalene	59228	0.26	250
28	20.43	Alkane	109785	0.49	472
29	20.57	NI	45578	0.20	193
30	20.65	NI	109472	0.49	472
31	21.02	Diocylethane	118077	0.53	510
32	21.08	Aromatic compd.	225243	1.00	963
33	21.17	Hydrocarbon	30601	0.14	135
34	21.25	NI	314695	1.40	1350
35	21.35	NI	23463	0.10	96.3
36	21.58	NI	9283	0.04	28.5
37	21.65	Dibenzofuran	30450	0.14	135
38	21.78	C ₅ -alkyl indan	169874	0.76	732
39	21.87	C ₂ alkyl naphthalene	74812	0.33	318

(Continued)

TABLE B-58. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
40	21.97	C ₂ -alkyl naphthalene	67596	0.30	289
41	22.17	C ₅ -alkyl indan	44986	0.20	193
42	22.30	C ₃ -alkyl naphthalene	39864	0.18	173
43	22.50	NI	32133	0.14	135
44	22.65	NI	396874	1.77	1700
45	22.72	C ₃ -alkyl naphthalene	67157	0.30	289
46	23.05	Fluorene	68222	0.30	289
47	23.18	C ₃ -alkyl naphthalene	562926	2.51	2420
48	23.30	C ₁₆	405409	1.80	1730
49	23.42	Methyldibenzofuran	61960	0.28	270
50	23.57	NI	53628	0.24	231
51	23.83	Alkyl naphthalene	63148	0.28	270
52	24.10	NI	54526	0.24	231
53	24.27	Hydrocarbon	276958	1.23	1180
54	24.43	NI	70353	0.31	299
55	24.53	NI	24481	0.11	106
56	24.82	NI	66401	0.30	289
57	24.95	NI	46874	0.21	202
58	25.13	C ₄ -alkylnaphthalene	233160	1.04	1000
59	25.17	NI	159336	0.71	684
60	25.28	C ₁₇	371567	1.65	1590
61	25.40	Alkane	476749	2.12	2040
62	25.50	C ₄ -alkylnaphthalene	34042	0.15	144
63	25.60	NI	38590	0.17	164
64	25.70	NI	108323	0.48	462
65	25.87	C ₁₉ Alkene	437571	1.95	1880
66	26.05	C ₁₁ Alkene	544403	2.42	2330
67	26.22	Alkyl naphthalene	37947	0.17	164
68	26.38	NI	7890	0.04	38.5
69	26.70	NI	19356	0.09	86.7
70	26.78	NI	22952	0.10	96.3
71	26.90	Phenanthrene	61327	0.28	270
72	27.00	d ₁₀ -Anthracene	458951	2.04	—*
73	27.13	C ₁₈	571607	2.54	2450
74	27.20	NI	47006	0.21	202
75	27.28	NI	62759	0.28	270

(Continued)

* - Internal reference standard

TABLE B-58. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
76	27.45	NI	27812	0.12	116
77	27.73	NI	52662	0.23	222
78	27.95	NI	42417	0.19	183
79	28.22	NI	38116	0.17	164
80	28.52	NI	9942	0.04	38.5
81	28.65	NI	13128	0.06	57.8
82	28.70	NI	52851	0.24	231
83	28.78	NI	161357	0.72	693
84	28.92	C ₁₉	563125	2.51	2420
85	29.20	NI	87794	0.39	376
86	29.32	NI	34063	0.15	144
87	29.63	NI	16442	0.07	67.4
88	29.85	NI	29791	0.13	125
89	29.95	NI	16851	0.08	77.0
90	30.30	NI	44305	0.20	193
91	30.48	NI	114998	0.51	491
92	30.62	C ₂₀	666847	2.95	2860
93	30.72	NI	280178	1.25	1200
94	30.88	NI	69692	0.31	299
95	31.22	NI	19951	0.09	86.7
96	31.42	NI	32051	0.14	135
97	31.52	NI	30717	0.14	135
98	31.92	NI	30436	0.14	135
99	32.12	NI	388955	1.73	1670
100	32.23	C ₂₁	951449	4.24	4080
101	32.52	NI	54247	0.24	231
102	32.72	NI	10240	0.05	48.2
103	33.10	NI	13211	0.06	57.8
104	33.25	NI	26354	0.12	116
105	33.32	NI	20844	0.09	86.7
106	33.68	C ₂₂ alkene	335875	1.50	1440
107	33.80	n-C ₂₂	1094215	4.87	4690
108	34.07	NI	61454	0.27	260
109	34.22	NI	32716	0.15	144
110	34.60	NI	29190	0.13	125
111	34.68	NI	8210	0.04	38.5
112	35.17	C ₂₃ alkene	328425	1.45	1410
113	35.28	C ₂₃	1093270	4.87	4690
114	35.55	NI	81784	0.36	347
115	36.15	NI	21974	0.10	96.3
116	36.60	C ₂₄ -alkene	245626	1.09	1050

(Continued)

TABLE B-58. (Continued)

Peak Number	Retention Time	Identification	Peak Area	% Total Peak Area	Conc. $\mu\text{g/g}$
117	36.72	C ₂₄	1168295	5.20	5010
118	36.98	NI	50825	0.23	222
119	37.52	NI	49764	0.22	212
120	38.10	C ₂₅	1214471	5.41	5210
121	38.33	NI	47100	0.21	202
122	38.78	NI	5677	0.03	28.9
123	39.20	NI	52972	0.24	231
124	39.40	NI	1098823	4.89	4710
125	39.67	NI	19229	0.08	86.7
126	40.18	NI	15212	0.07	67.4
127	40.68	C ₂₇	918497	4.09	3940
128	40.95	NI	31424	0.14	135
129	41.90	C ₂₈	529158	2.36	2270
130	43.08	C ₂₉	413215	1.84	1770
131	43.48	NI	11332	0.05	48.2
132	43.62	NI	49239	0.22	212
133	44.05	NI	33105	0.15	144
134	44.22	NI	145400	0.65	626
135	45.18	NI	83127	0.37	356
136	45.33	NI	75594	0.34	327
137	45.77	NI	83923	0.37	356
138	46.22	NI	39654	0.18	173
139	46.87	NI	29781	0.13	125
Total Eluted				94324	$\mu\text{g/g}$
Total Identified				73361	$\mu\text{g/g}$
% Identified				77.8	

¹Quantitated using the response factor for d₁₀-Anthracene because response factors for individual compounds were not available.

²NI - Not identified.

TABLE B-59. RESULTS OF LIQUID CHROMATOGRAPHY FRACTIONATION OF
GASIFIER ASH

LC Fraction	TCO ¹	GRAV ¹	
1	13	19	
2	2.4	1.1	
3	0.44	0	
4	0	0	
5	0	0	
6	0.80	7.6	
7	<u>0</u>	<u>3.3</u>	
	17	31	
	<u>TCO (mg)</u>	<u>GRAV (mg)</u>	<u>Total (mg)</u>
Total Sample ²	15.0	7.0	22.0
Taken for LC ³	11.8	5.5	17.3
Recovered ⁴	3.1	5.7	8.8
% Recovered	26.3	103	50.9

¹Concentration in µg/g.

²Quantity in sample prior to LC.

³Portion taken for LC.

⁴Quantity recovered from LC column.

TABLE B-60. RESULTS OF LIQUID CHROMATOGRAPHY FRACTIONATION OF
COMBINED GAS CONDENSABLE ORGANICS

LC Fraction	TCO ¹	GRAV ¹
1	6.8E5	9.8E3
2	3.2E5	0
3	1.5E4	0
4	1.1E4	0
5	1.7E5	0
6	4.9E5	3.6E5
7	<u>2.6E3</u>	<u>3.1E4</u>
Total	1.7E6	4.0E5

	<u>TCO (mg)</u>	<u>GRAV (mg)</u>	<u>Total (mg)</u>
Total Sample ²	4600	280	4880
Taken for LC ³	92	5.6	97.6
Recovered ⁴	43.4	10.2	53.6
% Recovered	47.1	182	54.9

¹Concentration in $\mu\text{g}/\text{Nm}^3$.

²Quantity in sample prior to LC.

³Portion taken for LC.

⁴Quantity recovered from LC column.

TABLE B-61. RESULTS OF LIQUID CHROMATOGRAPHY FRACTIONATION OF
ESP OILS TO BOILER NO. 3

LC Fraction	TCO ¹	GRAV ¹
1	2.0E4	3.1E4
2	6.3E3	6.7E4
3	3.9E3	9.0E4
4	0	0
5	0	7.9E3
6	2.4E3	1.6E5
7	0	0
TOTAL	3.2E4	6.4E5

	TCO (mg)	GRAV (mg)	Total (mg)
Total Sample ²	21.9	214	552
Taken for LC ³	4.4	42.8	47.2
Recovered ⁴	1.6	32.6	34.2
% Recovered	37.3	76.1	72.5

¹Concentration in $\mu\text{g/g}$.

²Quantity in sample prior to LC.

³Portion taken for LC.

⁴Quantity recovered from LC column.

TABLE B-62. RESULTS OF ANALYSES OF CHARCOAL TUBE EXTRACTS FOR
BENZENE, TOLUENE, AND XYLENE¹

Sample	Benzene	Toluene	Xylene
Combined Gas 1035-1045 300 sec	2.17E5	3.22E5	2.22E5
Combined Gas 1047-1052 150 sec	2.54E5	5.42E5	3.87E5
Cleaned Top Gas 1044-1059 450 sec	6.87E5	9.25E5	7.85E5
Cleaned Top Gas 1025-1030 1050 sec	5.03E3	9.71E5	8.93E5
Raw Top Gas 1616-1626 300 sec	4.95E5	7.40E5	4.80E5
Cleaned Bottom Gas 1758-1813 450 sec	5.28E3	5.08E4	9.83E3
Cleaned Bottom Gas 2319-2324 150 sec	ND	1.73E5	9.19E3
Raw Bottom Gas 0007-0022 450 sec	2.36E3	4.68E4	1.40E3
Product Gas Boilers 1447-1552 3200 sec (#2)	4.64E2	4.75E3	ND
Product Gas Boilers 1658-1708 500 sec (#2)	ND	3.69E4	ND
Boiler No. 3 Firing ESP Oils 1542-1722 2000 sec (#1)	8.39E2	8.08E3	ND
Boiler No. 3 Firing ESP Oils 1813-1823 500 sec (#2)	ND	4.25E4	ND

aEb = a x 10^b

¹Concentration in $\mu\text{g}/\text{Nm}^3$.

ND = Not detected.

APPENDIX C

SAMPLING TECHNIQUES

C.1.0 INTRODUCTION

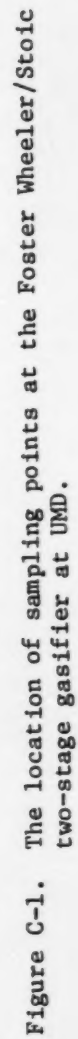
This appendix describes the sampling techniques used to collect samples for the environmental assessment of the Foster Wheeler/Stoic two-stage gasifier on the University of Minnesota-Duluth (UMD) campus. Figure C-1 presents a simplified schematic of the gasification facility showing the sampling points. The streams sampled were:

Gaseous Streams:

- Combined product gas
- Raw top product gas (upstream from electrostatic precipitator)
- Cleaned product gas (downstream from electrostatic precipitator)
- Raw bottom product gas (upstream from cyclone)
- Cleaned bottom product gas (downstream from cyclone)
- Combustion emissions from product gas boilers
- Combustion emissions from Boiler No. 3 firing ESP oil
- Combustion emissions from Boiler No. 3 firing natural gas

Solid Streams:

- Coal
- Gasifier ash
- Cyclone dust



Liquid Streams:

- ESP oil to storage
- ESP oil from storage to Boiler No. 3
- Ash pan water

The following sections describe the techniques used for sampling these streams.

C.2.0 PROCESS GASES

Five sampling points were used for the collection of process gas samples. The process streams sampled were the combined product gas, raw and treated top product gases, and raw and treated bottom product gases. These streams were sampled for the following parameters:

- aerosol and particulate,
- vapor phase organics,
- trace elements,
- ammonia,
- cyanide, and
- fixed gases, C₁-C₆ hydrocarbons, sulfur species, and benzene, toluene, and xylene (BTX).

Samples for all but the fixed gases, C₁-C₆ hydrocarbons, sulfur species, and BTX were taken isokinetically using a stainless steel probe fitted with the following:

- a packing gland system,
- a type S pitot,
- a type K thermocouple, and
- a gooseneck nozzle.

In all cases, because of sample port limitations, only a single traverse could be performed for flow calculations.

The sampling techniques for each of the above parameters are discussed in the following sections.

C.2.1 Aerosol and Particulate Collection

Because of the variety of process streams sampled, three different aerosol and particulate collection systems were used. These were:

- in-stream filtration using a glass fiber thimble (for the combined product gas and treated top gas),
- an electrostatic precipitator (ESP) followed by a glass fiber filter (for the raw top gas), and
- out-of-stream filtration using a glass fiber thimble in a heated oven (for the treated and raw bottom gases).

Samples recovered from the nozzle, probe, and collection device were weighed and used to calculate particulate loadings. The configurations of these collection systems are shown in Figure C-2 and discussed below.

C.2.1.1 Combined Product Gas and Cleaned Top Gas

The combined product gas and clean top gas streams contained cleaned gases at temperatures of 400°C (750°F), well above the dew point of the hydrocarbons of the process stream. They contained vaporized tars and oils as well as particulates. Because no aerosolized tars and oils were expected in the gas streams, and the process temperature did not exceed the limitations of the glass fiber material, the filtering system was placed on the in-stream portion of the probe. In-stream filtration was preferred for two reasons: 1) there was no need to provide external heat to the filter, and 2) no tars, oils, or particulates were deposited in the sample line prior to filtration. The aerosol and particulate collection system consisted of a

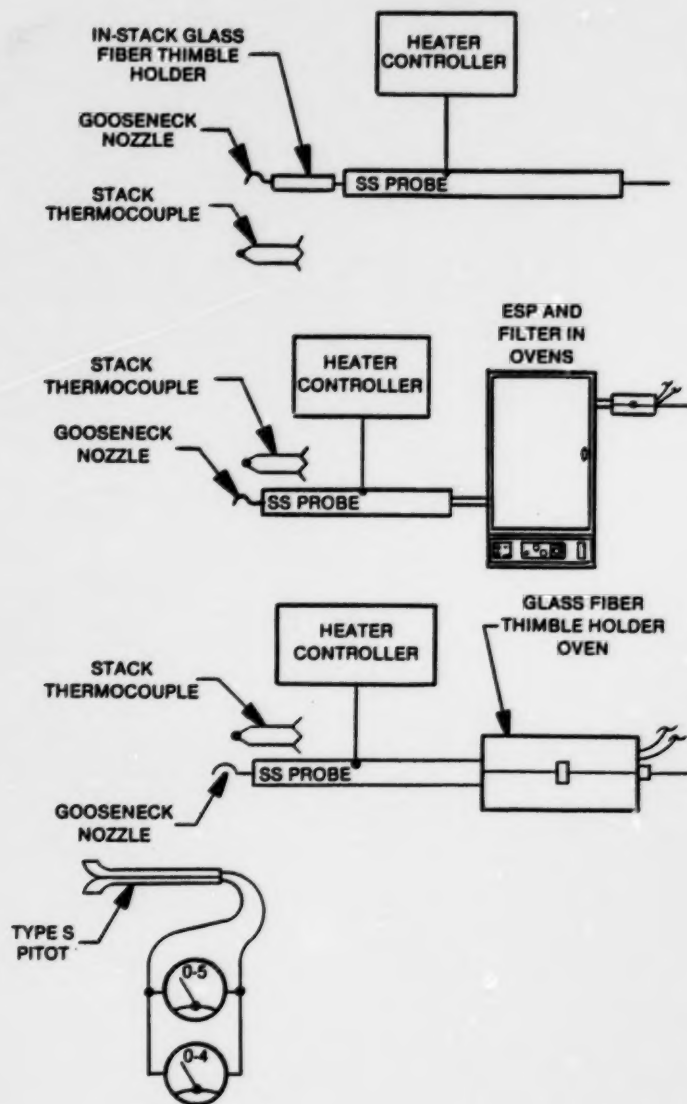


Figure C-2. Three sampling configurations for aerosol and particulate collection.

probe with a 30 x 100 mm glass fiber thimble and holder preceded by a gooseneck nozzle.

C.2.1.2 Raw Top Gas

The raw top gas stream contained a dispersion of aerosolized tars, oils, and particulate matter which had to be collected and assayed. An electrostatic precipitator was used to collect the aerosol from the raw top gas. The ESP was followed by a filter which captured any aerosol not collected by the ESP. The tar and oil loading and particulate loading were determined by combining the materials collected by the ESP and filter. The temperature of the top gas was approximately 95°C (200°F); this temperature did not present any significant problems to the sampling effort.

The collection system for the top gas consisted of:

- a probe with a gooseneck nozzle in-stream,
- an electrostatic precipitator in a heated oven, and
- a 43 x 123 mm pliable glass fiber thimble and holder in a heated oven.

The gas stream entered the lower portion of the ESP, shown in Figure C-3, and was directed toward a catch cup at the bottom of the ESP, thus permitting the collection of any easily removed material (particulate and organic aerosol) without fouling the upper portion of the ESP. As the gas stream moved upward through the ESP, electrostatic collection was induced by the 12 KV DC potential between the 0.025 mm center electrode and cylindrical shell. Colloidal sized aerosols were removed from the gas stream and forced to coalesce by their ultimate contact with the outer shell and other aerosol particles. Accumulated oils and tars flowed down the walls to the catch cup at the bottom of the ESP. The cleaned gas exited near the ESP top. Most collection occurred near the base of the ESP; the upper portions remained relatively free of material.

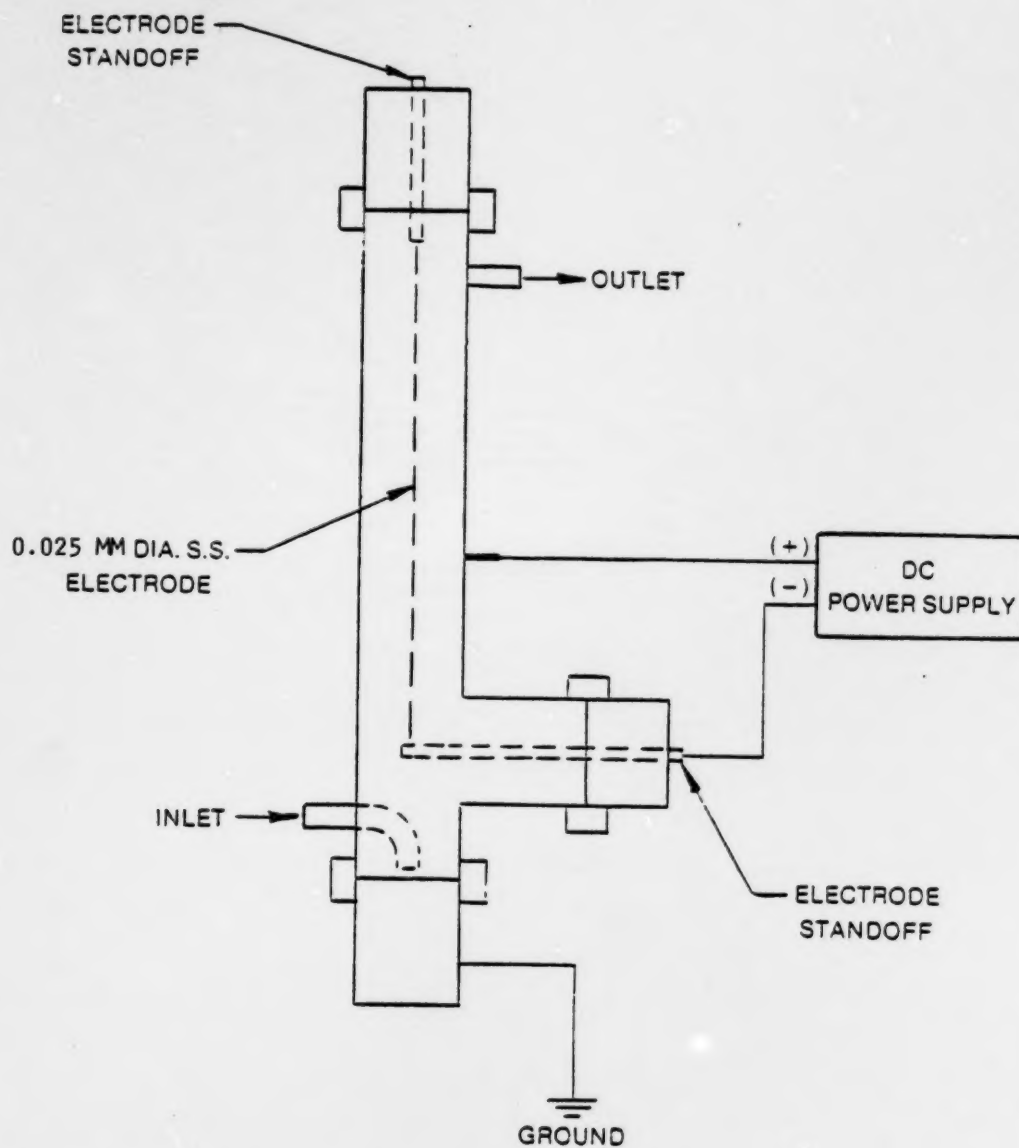


Figure C-3. A schematic of ESP used for aerosol and particulate removal.

ESPs of this configuration have been shown to be 99+% efficient for dry material of colloidal size. However, the primary reason for the use of the ESP was its ability to remove particulates and oil and tar aerosols without frequent stops for servicing or changing filtering elements. A glass fiber filter was used at the exit of the ESP as a polish and a back-up in the event of an ESP failure. The ESP temperature was held at 140°C (285°F) and the filter at 150°C (300°F) to avoid further condensation of any organic vapors remaining in the gas streams. Figure C-4 shows the ESP and filter in their respective ovens.

C.2.1.3 Bottom Gas

The bottom gas required a different sampling technique. No aerosolized tars and oils were expected in the bottom gas. Therefore, the ESP was not required. However, the temperature of the bottom gas, 650°C (1200°F), presented some problems in sampling.

Glass fiber filters have a temperature limitation of 480°C (900°F). For this reason, the filtering system could not be placed in the process stream. While it is desirable to have all parts of the probe and particulate sampling system at the temperature of the gas being sampled, the filter system, which was mounted on the outside end of the probe, could only be heated to 480°C. This was adequate since the condensable organics content of the bottom gas was low.

Both bottom gas sampling locations use the following system:

- a probe with a gooseneck nozzle, and
- a 43 x 123 mm pliable glass fiber filter and holder in a heated oven.

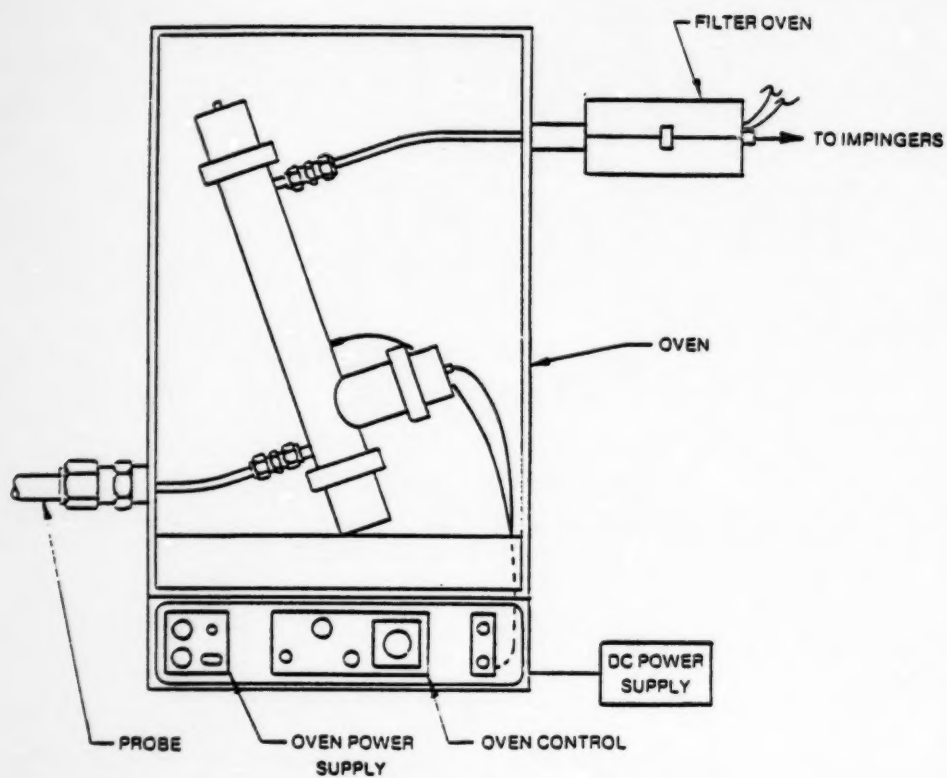


Figure C-4. ESP filter and assembly.

C.2.2 Organic Vapors

Organic vapors were collected using a sampling train having a condenser and XAD-2 resin cartridge for vapor collection. Figure C-5 shows the organic vapor sampling train used for the clean top gas and combined product gases. The sampling trains used for the other process gas streams were identical to the train shown in Figure C-5 with the exception of the aerosol and particulate collection devices (see Section 2.1).

Organic vapors were collected by passing the gas stream through a condenser and an organic resin canister. The condenser served two purposes: it cooled the sample gas, and it collected condensed vaporous tars and oils from the sample gas stream. Uncondensed organic vapors were then adsorbed on the resin in the canister. Results of the analysis of condensed tars and oils and the uncondensed vapors (including probe rinse) were combined with the aerosol/particulate catch to give a total profile of the organic constituents in the process gas stream.

The condenser, constructed of stainless steel, is of triple chamber design. Cooling water circulates through a central cylindrical chamber, or core. This core is concentric with a surrounding chamber, through which the sample gas is passed. This chamber is likewise surrounded by a third concentric chamber which contains circulating cooling water.

The sample gas entered the condenser through a tangential inlet tube near the bottom and exited near the top. The cooling water (ice water) entered the condenser at the top and exited at the bottom. The countercurrent flow of the sample gas and cooling water assured a maximum temperature differential between the sample and coolant throughout the condenser. Tars and oils condensed from the sample gas stream were removed from the condenser through a drain port at the bottom of the gas circulation chamber.

Uncondensed organic vapors leaving the top of the condenser flowed to the resin canister. As the vapors entered the canister, they struck a

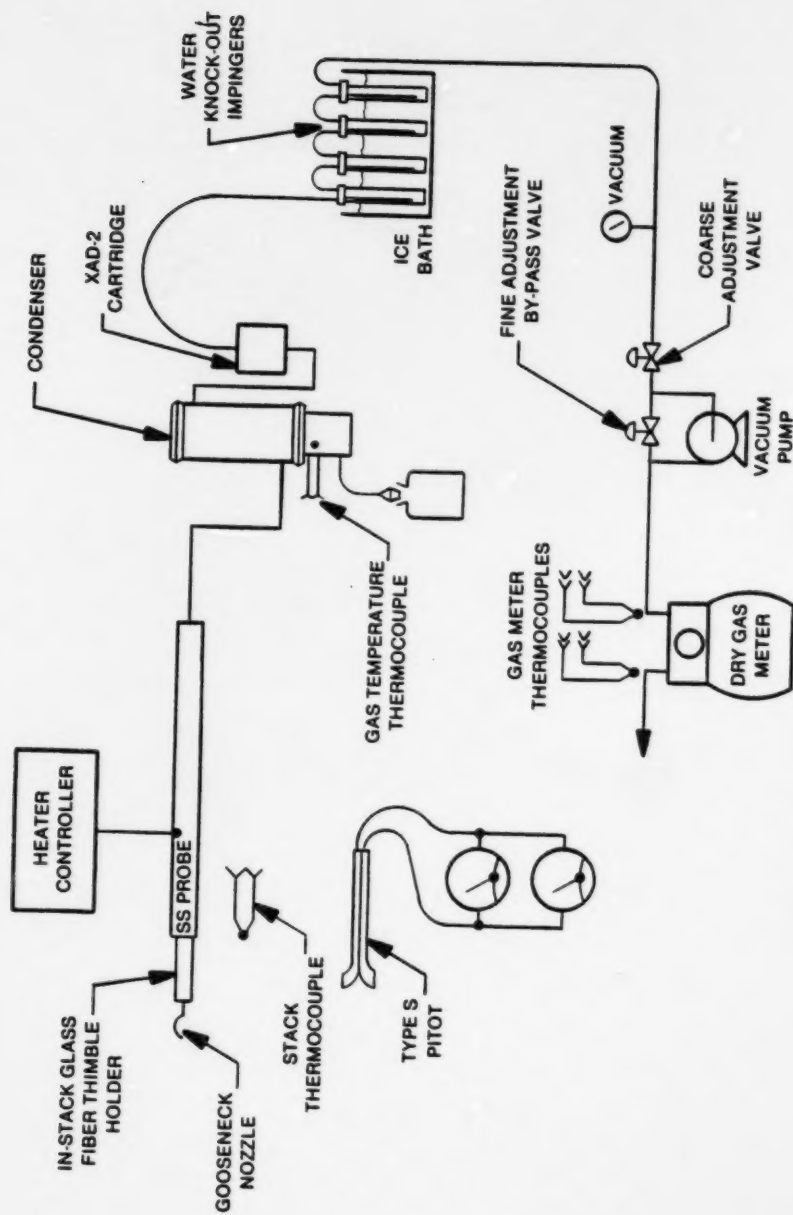


Figure C-5. Sampling system for organics -- cleaned top product gas and combined product gas.

perforated plate which is designed to distribute the gases evenly over the adsorption resin. The resin is held in the canister by discs of stainless steel wire mesh. Retaining rings are used to maintain the alignment of the distribution plates and wire mesh.

The pumping and metering devices used in the collection of organic vapors and in the collection of trace elements (as discussed in the following section) are contained in a commercially available console that has been modified. The two main components of the console are a double diaphragm pump and a dry gas meter. In addition, the console contains the following devices:

- an oil manometer and Magnehelic® gauge for measurement of pressure differential across the type S pitot,
- a digital temperature read-out with selector switch (for temperature measurement at several locations of the system),
- a Magnehelic® gauge for measurement of pressure drop across an orifice mounted behind the dry gas meter (a measurement of gas flow rate),
- heat controllers for the probe and filter hot box of the system, and
- two additional heat controllers.

C.2.3 Trace Elements, Ammonia and Cyanide

Gaseous inorganic components were collected by passing the sample gas stream through a series of Greenberg-Smith impingers which contained chemical solutions for removal of the species of interest from the gas stream. The species sought dictated the chemical solution or solutions to be used in the impinger and also the number of impingers. The final impinger contained a desiccant to remove moisture (which could have damaged the pumping and metering system) from the gas stream to provide dry gas for metering. During sampling, the impingers were housed in a cold box containing ice water.

This maintained the chemical solutions at a low temperature and increased the collection efficiency of the solutions. An example of the inorganic sampling train used for the bottom product gases is shown in Figure C-6. The sampling trains used for the other process gas streams were identical to the train shown in Figure C-6 with the exception of the aerosol and particulate collection devices (see Section 2.1).

C.2.3.1 Trace Elements

The impinger contents used for vapor phase trace element collection were:

- | | |
|--|--------|
| 1. 0.2M $(\text{NH}_4)_2\text{S}_2\text{O}_8$ + 0.2M AgNO_3 | 250 mL |
| 2. 0.2M $(\text{NH}_4)_2\text{S}_2\text{O}_8$ + 0.2M AgNO_3 | 250 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

The mass of aqueous condensate collected by the train plus the total volume of gas sampled were used to calculate moisture fractions for process gas streams.

Mercury is not collected efficiently in most liquid sorption systems. For more precise quantitation of vaporous mercury in a gas stream, a sample of the gas is passed over pure gold wire formed into a plug and placed in an enclosed glass tube. The mercury sampling tube is shown in Figure C-7. Sample volume is determined from time and flow rate measurements.

C.2.3.2 Ammonia

The impinger contents used for vapor phase ammonia collection were:

- | | |
|-------------------------------|--------|
| 1. 5% H_2SO_4 | 100 mL |
| 2. 5% H_2SO_4 | 100 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

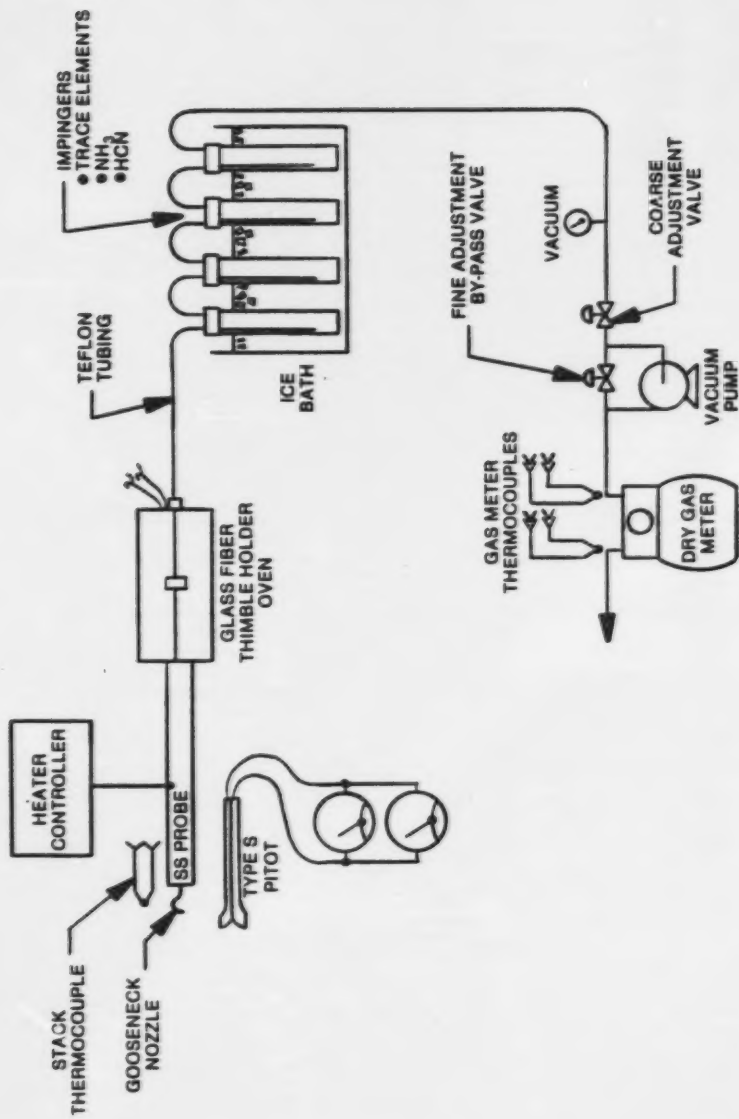
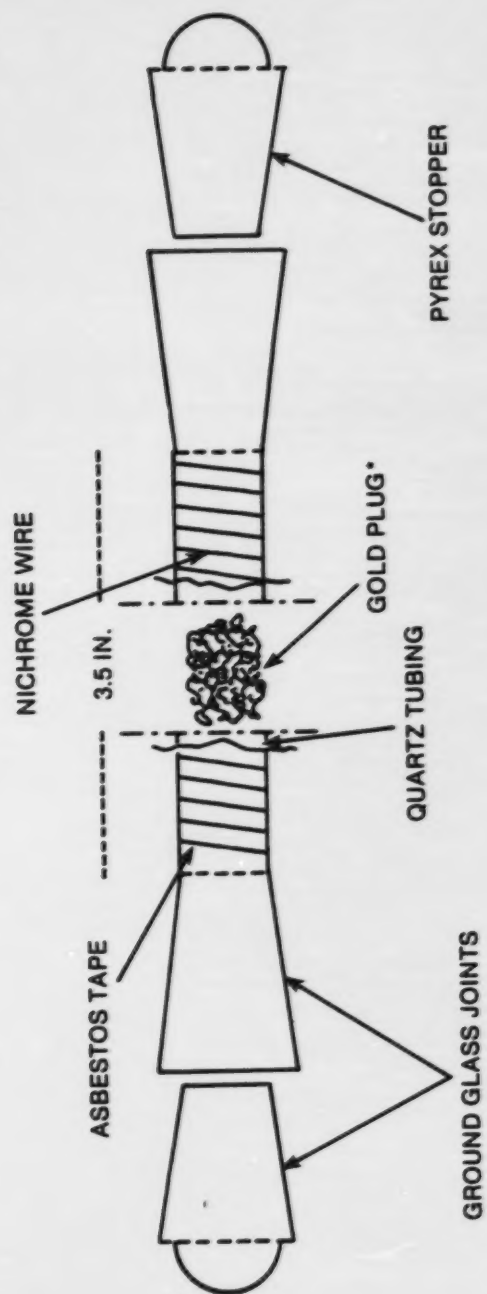


Figure C-6. Sampling system for trace elements, NH_3 , and HCN -- cleaned and raw bottom product gas.



*THE GOLD PLUG IS MADE FROM 36-GAUGE GOLD WIRE; 8-10 GRAMS OF WIRE IS SUFFICIENT.

Figure C-7. Mercury sampling tube.

The impinger solutions were analyzed on site by the analytical methods described in Appendix D.

The resulting impinger solutions were treated to remove sulfides and then analyzed on site, as described in Appendix D.

C.2.3.3 Cyanide

The impinger solutions used for cyanide collection were:

- | | |
|--------------------------|--------|
| 1. 10% NaOH | 100 mL |
| 2. 10% NaOH | 100 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

C.2.4 Samples for Gas Chromatographic Analyses and Benzene, Toluene, and Xylene Analysis

The process gases were analyzed on site by gas chromatography for major (fixed) gases, gaseous sulfur species, and C₁-C₆ hydrocarbons as described in Appendix D. The gas samples were collected with a sampling and conditioning apparatus consisting of 1.2 meters of heat-traced Teflon sample line followed by a Teflon membrane filter in a stainless steel filter holder. These were followed by a Perma-pure® drier, a Teflon-lined diaphragm pump, and a flow meter. Silanized glass sample bombs were used to collect the conditioned gas at the exit of the train. Sample bombs were purged, pressurized at >155 kPa (8 psig), and immediately transported to the laboratory for analysis. Benzene, toluene, and xylene (BTX) samples were collected on two-section charcoal tubes in place of the sample bombs using the same sampling system.

C.3.0 COMBUSTION GASES

Two combustion emissions streams were sampled during the test program. These were:

- the combined emissions from Boilers No. 1 and No. 2 firing combined product gas, and
- the emissions from Boiler No. 3 firing ESP oils only and natural gas only.

With the exception of Boiler No. 3 firing natural gas, these streams were sampled for:

- aerosol and particulate,
- vapor phase organics,
- trace elements,
- ammonia and cyanide,
- oxides of sulfur (SO_x),
- oxides of nitrogen (NO_x), and
- fixed gases, C_1 - C_6 hydrocarbons, sulfur species, and benzene, toluene, and xylenes.

Samples of emissions from Boiler No. 3 firing natural gas were analyzed for SO_x , NO_x , fixed gases, C_1 - C_6 hydrocarbons, and sulfur species.

The combined emissions from Boilers No. 1 and 2 are exhausted through a circular stack 30 meters in height. The combustion gases exit each boiler and are combined in a rectangular duct prior to the stack. Sampling was performed in this duct. Five velocity traverses were performed from five 3-inch sampling ports about 7 meters above the point where the two individual streams are combined. Sampling was performed from a single port at the point of average flue gas velocity. The emissions from Boiler No. 3 are exhausted through a 1.4 meter diameter duct extending about 2.5 meters above the building roof. Two velocity traverses were performed from two 4-inch ports at 90° to each other about 1 meter above the building roof. Sampling was performed on a traverse from each port. The techniques used for taking the various types of samples are discussed below.

C.3.1 Aerosol and Particulates, Vapor Phase Organics and Trace Elements

Aerosol and particulates, vapor phase organics, and trace elements were sampled using a modified EPA Source Assessment Sampling System (SASS train). The train configuration and method of operation meet current specifications (Ref. C-1), except for the location of the XAD-2 canister in the organic module and the use of a single, $>1\ \mu\text{m}$ cyclone instead of a series of three cyclones. These procedures currently require that the XAD-2 canister be positioned beneath the gas cooler. In this position, large quantities of aqueous and organic phase condensate flow through the XAD-2 resin and are subsequently trapped in a knock-out below the resin. This arrangement could cool the resin early in the sampling effort and result in reduced collection efficiencies for vapor phase organic material.

The modified configuration shown in Figure C-8 traps condensate material before it passes through the resin, thus reducing the possibility of reduced resin collection efficiency. It also simplifies sample recovery and analysis. The single cyclone was used because very low particle concentrations in the flue gases obviated the need for three cyclones for particle sizing.

Aerosol and particulate loadings were determined from the probe and nozzle rinse, cyclone catch, and filter catch. The organic sample consisted of the probe and nozzle rinses, particulate catch extract, and the sample taken from the organic module consisting of condensate and XAD-2 resin extract.

Trace element samples consisted of aliquots of the organic module samples and the samples collected in the trace element impingers. The impinger solutions used for vapor phase trace element collection were:

- | | |
|--|--------|
| 1. $0.2\text{M } (\text{NH}_4)_2\text{S}_2\text{O}_8 + 0.2\text{M AgNO}_3$ | 750 mL |
| 2. $0.2\text{M } (\text{NH}_4)_2\text{S}_2\text{O}_8 + 0.2\text{M AgNO}_3$ | 750 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

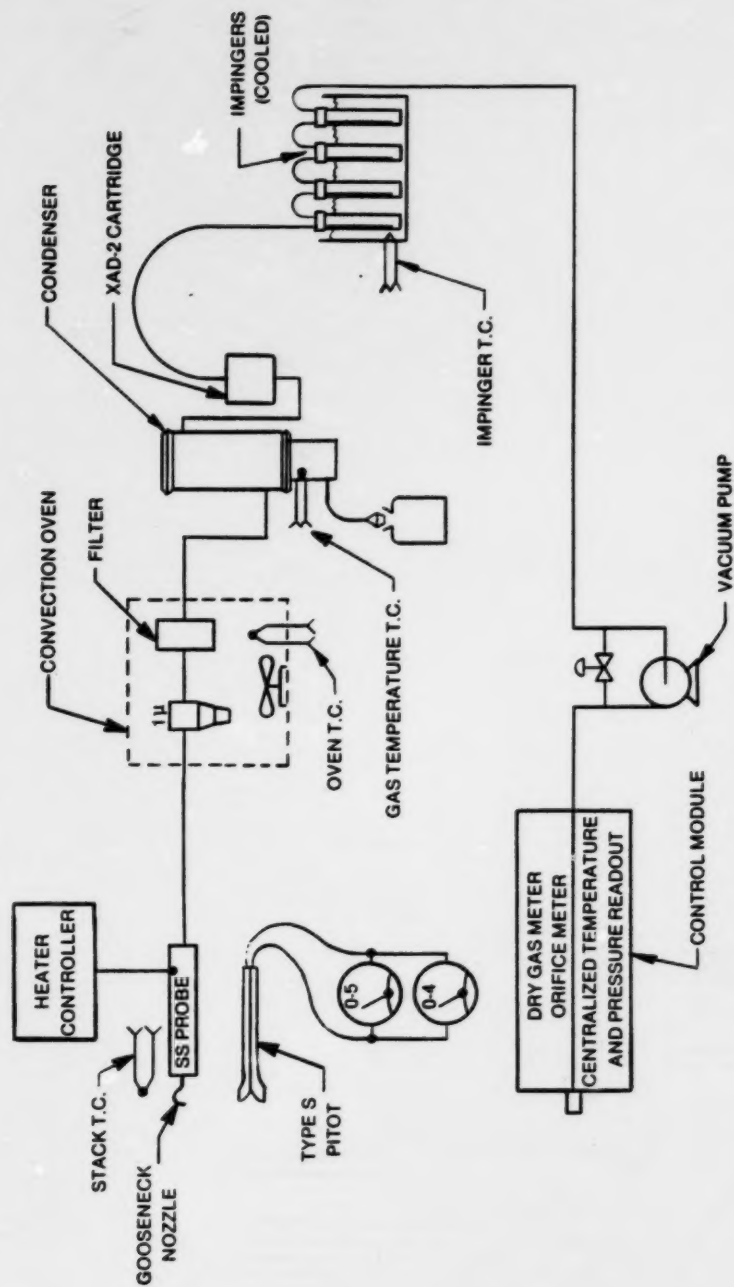


Figure C-8. High volume Source Assessment Sampling System (SASS) -- sampling system for organics and trace elements in combustion emissions from product gas boilers and Boiler No. 3 firing ESP oil.

C.3.2 Ammonia and Cyanide

Samples for analysis of ammonia and cyanide in the flue gas were collected during the SASS run. An auxiliary sampling train shown in Figure C-9 was used. It consisted of a stainless steel probe and a heated 47 mm glass fiber filter. The impinger contents for ammonia were:

- | | |
|---------------------------------------|--------|
| 1. 10% H ₂ SO ₄ | 100 mL |
| 2. 10% H ₂ SO ₄ | 100 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

The gas temperature was held at 120°C (250°F) until the gas had been filtered. At the completion of the run, the impinger contents were transferred to sample containers and analyzed on site by the analytical method described in Appendix D.

The auxiliary sampling assembly and sampling methodology used for ammonia collection was also used for cyanides. The impinger solutions used for cyanide collection were:

- | | |
|--------------------------|--------|
| 1. 10% NaOH | 100 mL |
| 2. 10% NaOH | 100 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

The resulting impinger solutions from the runs were treated to remove sulfides and then analyzed on site, as described in Appendix D.

C.3.3 Sulfur Oxides (SO_x)

Oxides of sulfur (SO_x) samples were collected using a combination of EPA Methods 5 and 6 (Ref. C-2). Particulate samples were also collected using this method. The sample gas was drawn isokinetically through a heated probe and heated glass fiber filter. SO_x was collected in an impinger

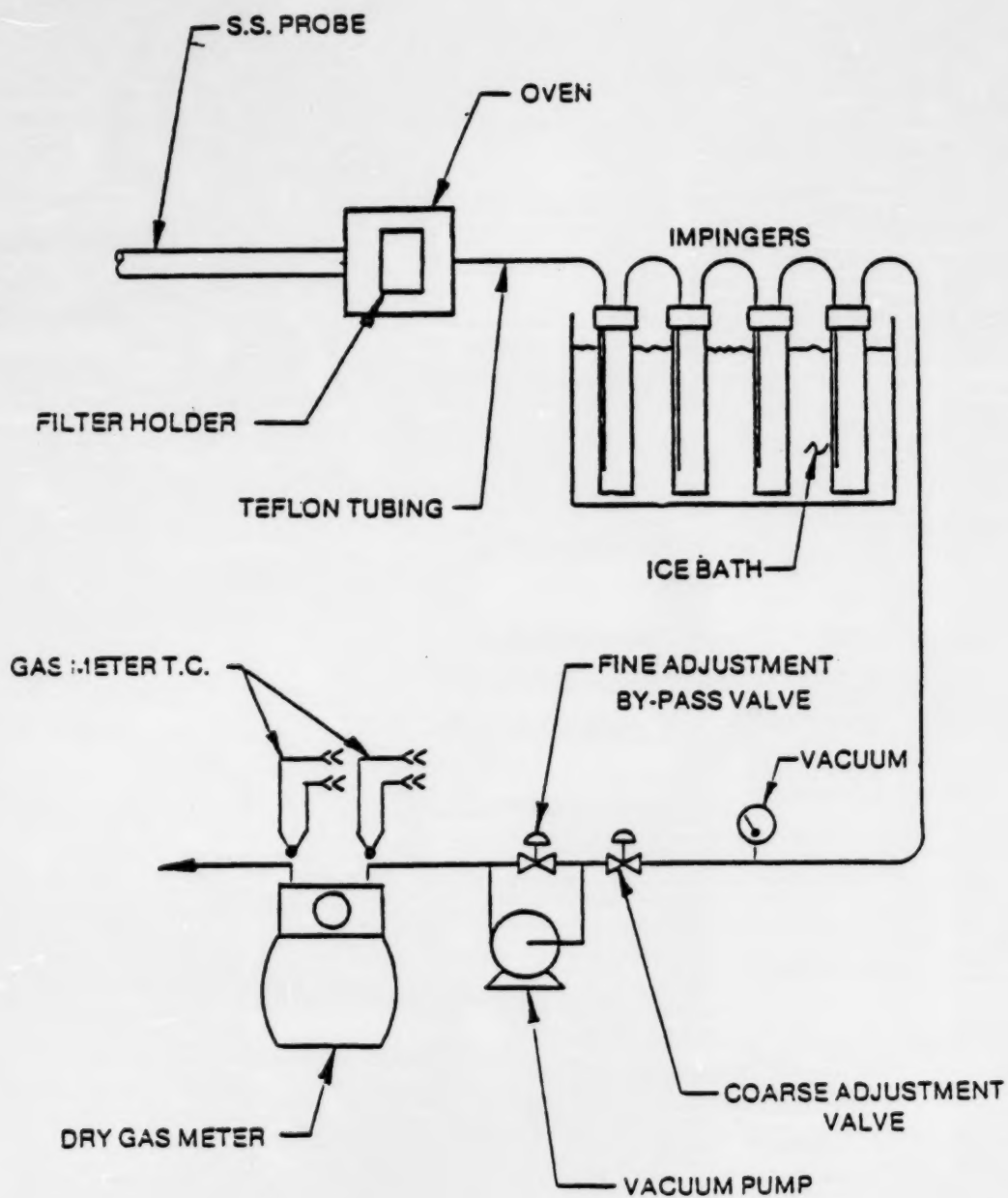


Figure C-9. Schematic drawing of impinger sampling train used for boiler emissions sampling.

train downstream of the filter. The impinger solutions used for collecting SO_x were:

- | | |
|------------------------------|--------|
| 1. 3% H_2O_2 | 100 mL |
| 2. 3% H_2O_2 | 100 mL |
| 3. Dry | |
| 4. Preweighed silica gel | |

The samples were transferred to plastic bottles and analyzed using the method given in Appendix D.

C.3.4 Nitrogen Oxides (NO_x)

Oxides of nitrogen were collected using EPA Method 7 (Ref. C-2). Grab samples of flue gas were taken using a heated glass probe fitted with a glass wool plug. The sample was pulled into an evacuated 2-L flask containing 25 mL of a dilute sulfuric acid-hydrogen peroxide absorbing solution. After taking the sample, the bomb was allowed to sit for 16 hours. The pressure in the flask was measured using a mercury U-tube manometer to determine sample volume. The absorbing solution and water rinses were transferred to a plastic bottle and the pH of the sample was adjusted to between 9 and 12 with 1 N sodium hydroxide. The samples were analyzed on site using the method given in Appendix D.

C.3.5 Fixed Gases, C_1 - C_6 Hydrocarbons, Sulfur Species, and Benzene, Toluene, and Xylenes (BTX)

The combustion emission gases were analyzed on site by gas chromatography for major (fixed) gases, gaseous sulfur species, and C_1 - C_6 hydrocarbons as described in Appendix D. The gas samples were collected with a sampling and conditioning apparatus consisting of 1.2 meters of heat-traced Teflon sample line followed by a Teflon membrane filter in a stainless steel filter holder. These were followed by a Perma-pure® Drier, a Teflon-lined diaphragm pump, and a flow meter. The Perma-pure® Drier removes moisture from the gas sample by highly selective permeation of water through Nafion® tubing. Silanized glass sample bombs were used to collect the conditioned

gas at the exit of the train. Sample bombs were purged, pressurized to >155 kPa (8 psig) and immediately transported to the laboratory for analysis. BTX samples were collected using charcoal tubes in place of the sample bombs.

C.4.0 SOLIDS

Three solid streams were sampled during the test period. These were:

- coal,
- gasifier ash, and
- cyclone dust.

The sampling techniques used for these streams are discussed below.

C.4.1 Coal

Coal is stored in a series of hoppers inside the heating plant and conveyed from the hoppers to sizing screens located above the coal feed lock-hopper. From the screens, the coal is conveyed to a weigh hopper. Coal samples were collected as the coal fell from the conveyor belt into the weigh hopper. Nine coal samples were taken, three each for three 24-hour periods. The samples were stored in 1-L polyethylene bottles. These samples were composited to form a total of three daily composites for proximate and ultimate analyses, and one weekly composite for trace element analysis. In addition, moisture determinations were performed on each of the nine samples.

C.4.2 Gasifier Ash

Two hydraulically operated buckets remove the gasifier ash from the ash pan and dump the ash onto belt conveyors which transport the ash to a storage area. The ash dewater through screens which constitute the bottoms of

the buckets. Gasifier ash samples were collected as the ash fell from the buckets onto the conveyor belt. Nine ash samples were collected at approximately the same time that the coal samples were collected. One-liter samples of gasifier ash were stored in two 500-mL glass bottles with Teflon-lined caps.

Moisture determinations were performed on each of the nine samples. The ash samples were composited to form a total of three daily composites for proximate and ultimate analyses, and one weekly composite for trace elements, RCRA leaching, and extractable organics analyses.

C.4.3 Cyclone Dust

Dust removed from the bottom product gas by the cyclone is collected in a lockhopper. The dust is periodically emptied from the lockhopper, augered to drums, and stored for disposal. Samples of the cyclone dust were collected from the auger and stored in 500-mL glass jars with Teflon-lined caps.

Nine cyclone dust samples were taken, three each for the same three 24-hour periods during which ash and coal samples were taken. The individual samples were composited to form three daily composites for proximate and ultimate analyses, and one weekly composite for trace elements, RCRA leaching, and extractable organics analyses.

C.5.0 LIQUIDS

Three liquid streams were sampled during the test period. These were:

- ESP oils from the top product gas,
- ESP oils to Boiler No. 3, and
- ash pan water.

The sampling techniques used for these streams are discussed below.

C.5.1 ESP Oils from Top Product Gas

Oils removed from the top product gas by the electrostatic precipitator (ESP) flow by gravity in heat-traced, insulated lines from the ESP to a heated, insulated, intermediate storage tank. Grab samples of the ESP oil (500 mL) were collected through a valve in the transport line between the ESP and storage tank and were stored in two 250-mL glass bottles with Teflon-lined caps.

Samples were taken during the same three 24-hour periods when the coal, ash, and dust samples were taken. These samples were composited to form three daily composites for proximate and ultimate analyses.

C.5.2 ESP Oils to Boiler No. 3

During the combustion of ESP oils in Boiler No. 3, ESP oils are pumped to the boiler from an underground heated storage tank. Samples of the ESP oils to Boiler No. 3 were collected through a valve mounted in the transfer line between the pump and boiler. Grab samples of the ESP oil to Boiler No. 3 (500 mL) were stored in two 250-mL glass bottles with Teflon-lined caps.

Three samples were taken during the 24-hour period when Boiler No. 3 was firing the oils. This time period was not concurrent with the rest of the three 24-hour sampling periods that took place. The three samples were composited to form a daily composite for density, proximate and ultimate, trace elements, and extractable organics analyses.

C.5.3 Ash Pan Water

Ash pan water samples were collected directly from the ash pan. Eleven samples were collected during the three 24-hour periods when all samples

except ESP oils to Boiler No. 3 were taken. Conductivity and pH were determined for each of these samples. Of the eleven samples, ten were aliquoted into 1-L glass bottles and subsequently composited to form three daily composites for water quality parameters analyses. Seven samples were aliquoted into 1-L glass bottles with Teflon-lined caps and 1-L polyethylene bottles and subsequently composited to form two weekly composites for extractable organics and trace elements analyses, respectively.

REFERENCES

- C-1 Lentzen, D.E., D.E. Wagoner, E.D. Estes, and W.F. Gutknecht. IERL-RTP Procedures Manual: Level 1 Environmental Assessment (Second Edition). NTIS No. PB-293 795. Research Triangle Institute, Research Triangle Park, North Carolina, 1978. (EPA-600/7-78-201)
- C-2 Reference Methods. Appendix A to 40 CFR 60. Environmental Reporter. Bureau of National Affairs, Washington, D.C., December 5, 1980.

APPENDIX D

ANALYTICAL TECHNIQUES

D.1.0 INTRODUCTION

This appendix describes the analytical techniques used for the analysis of samples collected for the assessment of the Foster Wheeler/Stoic gasifier. The analytical matrix for each sample is detailed in Figures D-1 through D-9.

D.2.0 WASTEWATER PARAMETERS AND IONIC SPECIES ANALYSES

D.2.1 Ammonia

Ammonia in the sulfuric acid impinger solutions and water samples was determined on-site by distillation/titrimetric procedures (Ref. D-1.) The sample was initially distilled into an indicating boric acid absorbing solution. The collected distillate was titrated with standard sulfuric acid. The detection limit for the procedure is 1.0 $\mu\text{g/mL}$ ammonia as nitrogen in 200 mL of impinger solution. This is equivalent to 1.1 ppm ammonia by volume collected from 0.28 SCM of sampled gas.

D.2.2 Cyanide

Cyanide was determined on-site by the distillation/colorimetric procedure (Ref. D-1). Interfering sulfide was precipitated from the caustic impinger solutions and water samples with lead nitrate. An acid reflux distillation procedure was used to remove interferences. Distilled hydrogen cyanide was collected in a caustic solution and converted to cyanogen chloride by reaction with chloramine-T. Following complete conversion, pyridine-barbituric acid reagent was added to produce a red-blue dye with

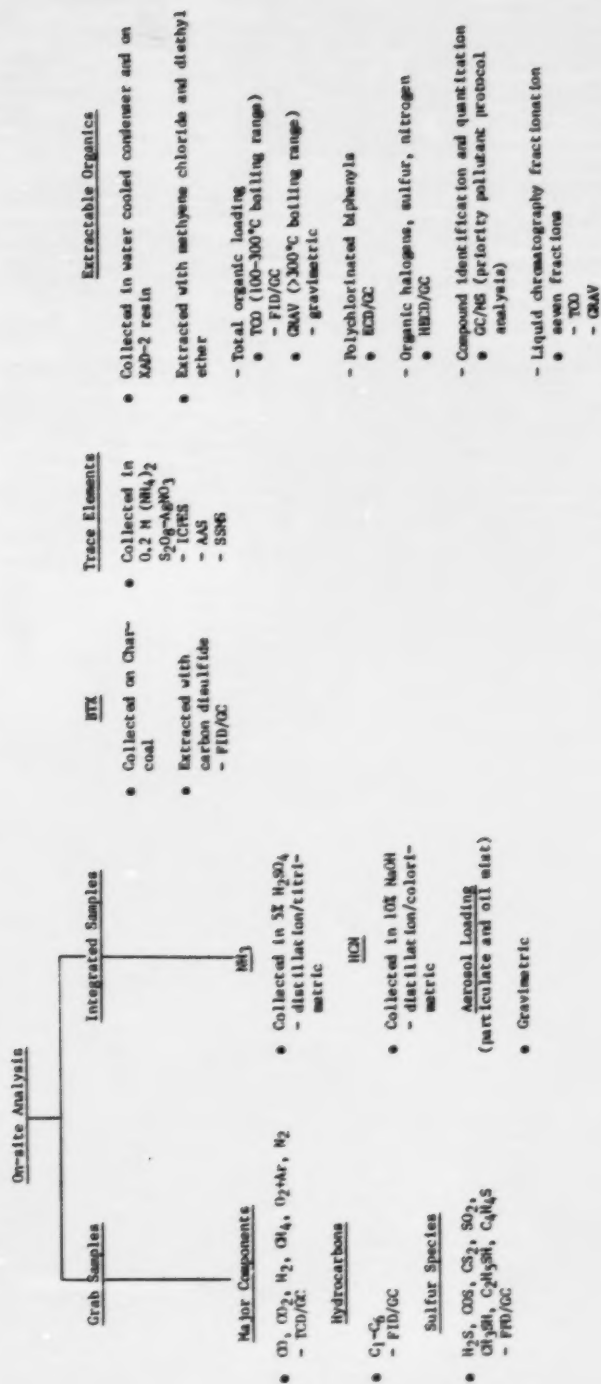


Figure D-1. An analytical matrix for combined product gas from the Foster Wheeler/Stoic two-stage gasifier (UMD).

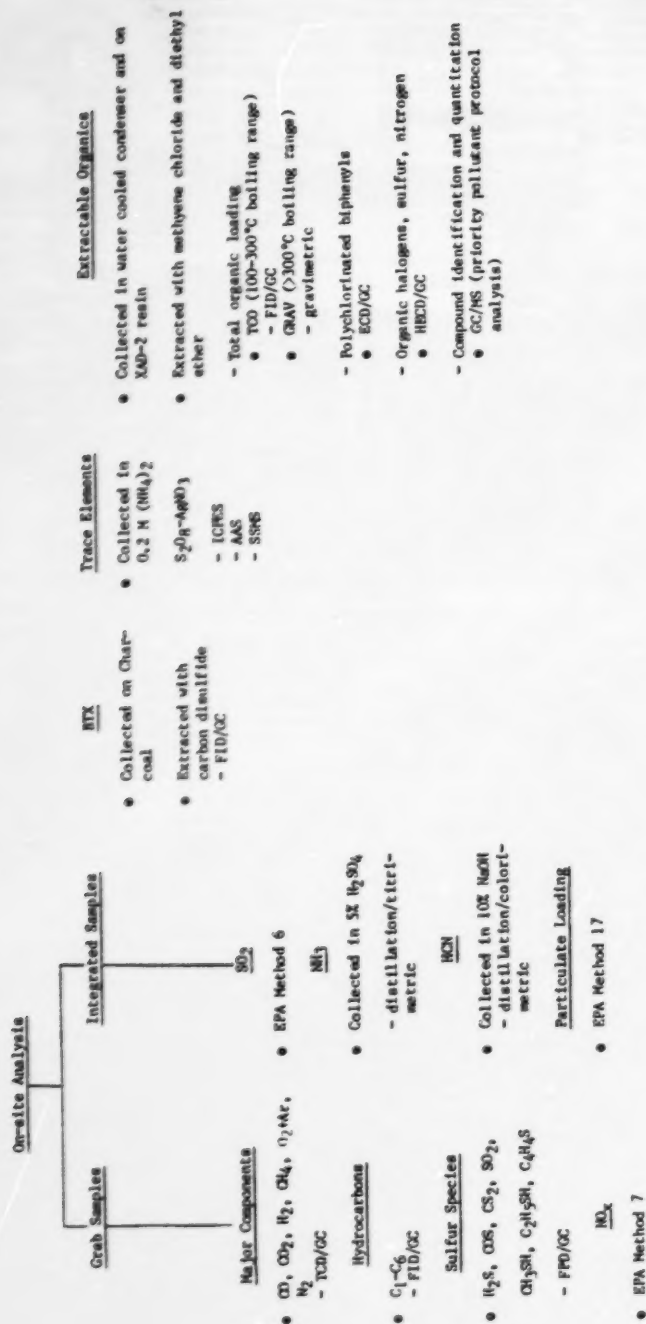


Figure D-2. An analytical matrix for combustion emissions from the product gas boilers and Boiler No. 3 firing ESP oils.

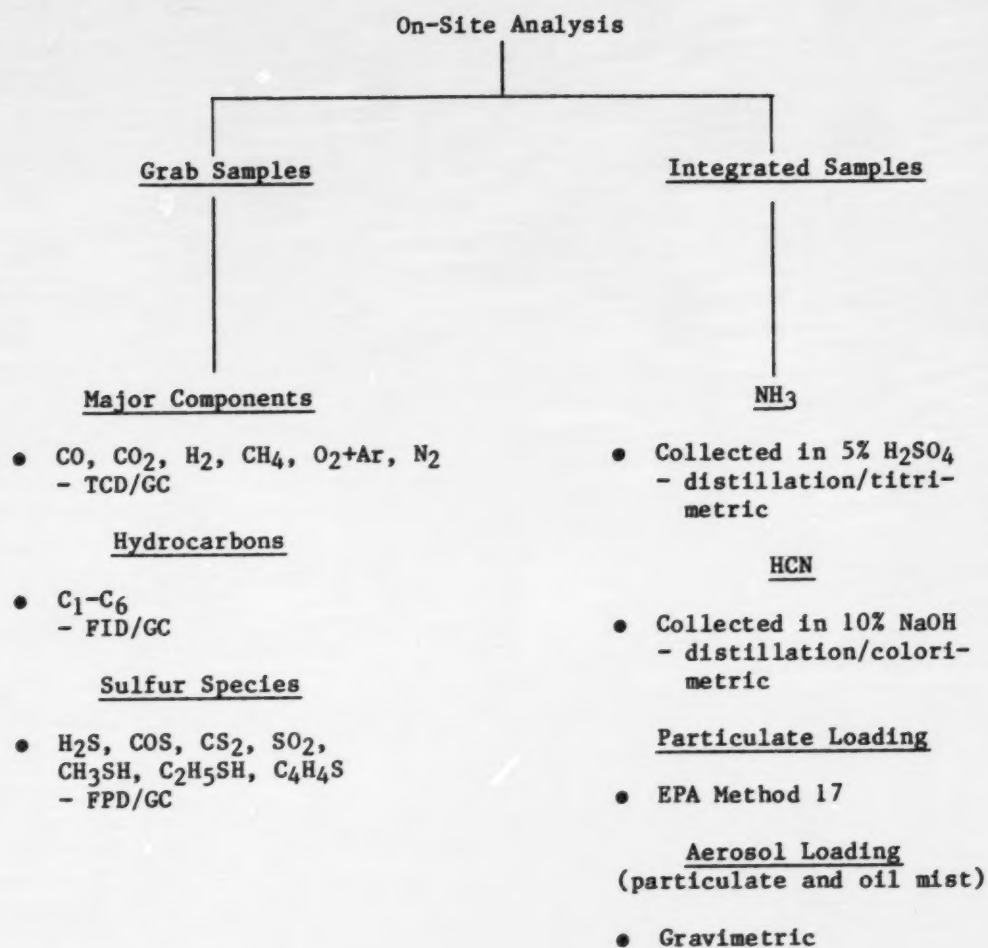


Figure D-3. An analytical matrix for cleaned and raw top product gas and cleaned and raw bottom product gas from the Foster Wheeler/Stoic two-stage gasifier (UMD).

<u>Individual Samples</u>		<u>Daily Composites</u>		<u>Weekly Composites</u>	
2/24	1945 hrs.	2/24	1945 hrs.}	2/24	1945 hrs.}
2/25	0145 hrs.	2/25	0145 hrs.}	2/25	0145 hrs.}
2/25	0615 hrs.	2/25	0615 hrs.}	2/25	0615 hrs.}
2/27	1035 hrs.	2/27	1035 hrs.}	2/27	1035 hrs.}
2/27	1350 hrs.	2/27	1350 hrs.}	2/27	1350 hrs.}
2/27	1825 hrs.	2/27	1825 hrs.}	2/27	1825 hrs.}
3/05	1240 hrs.	3/05	1240 hrs.}		
3/05	1540 hrs.	3/05	1540 hrs.}		
3/05	2040 hrs.	3/05	2040 hrs.}		
	Moisture		Proximate/ Ultimate		Trace Elements
					-ICPES -AAS -SSMS

Figure D-4. An analytical matrix for coal from the Foster Wheeler/Stoic two-stage gasifier (UMD).

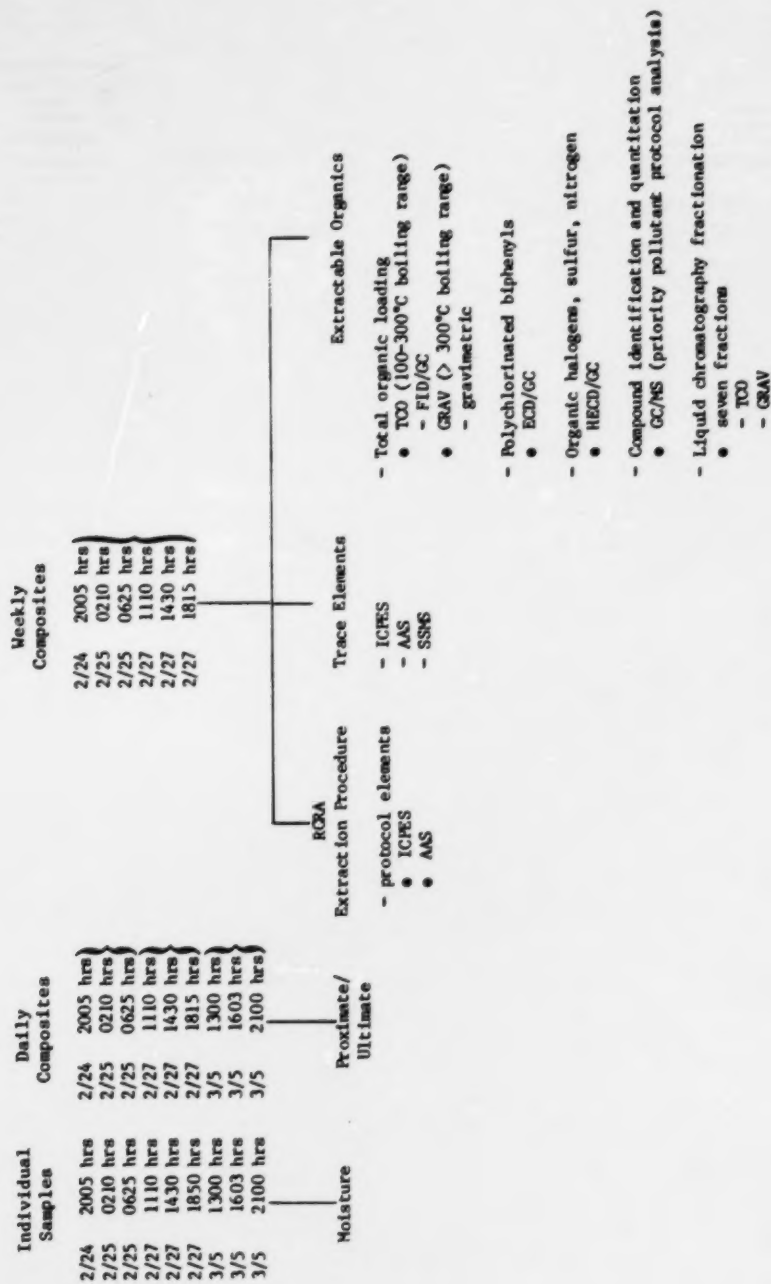


Figure D-5. An analytical matrix for gasifier ash from the Foster Wheeler/Stoic two-stage gasifier (UMD).

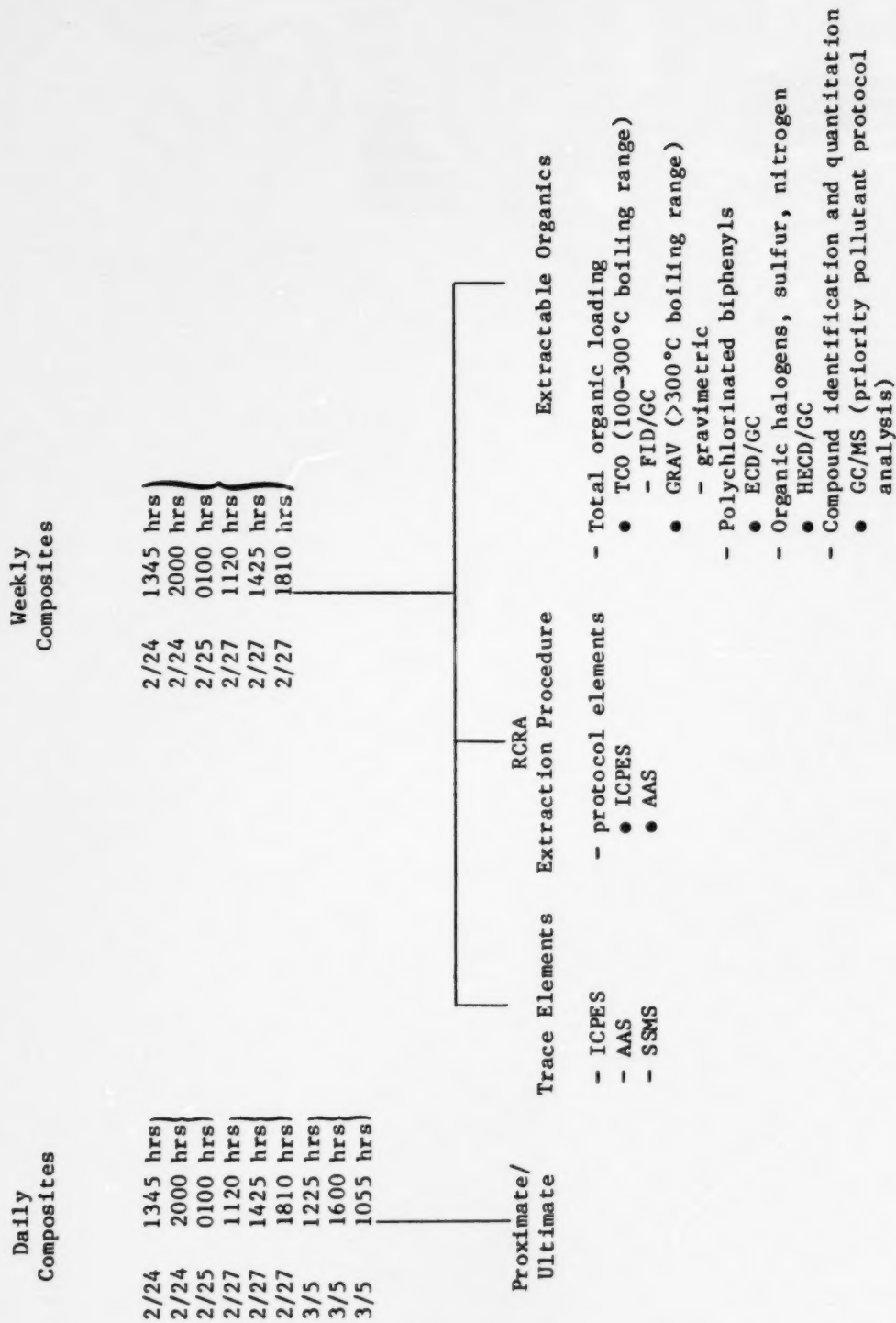


Figure D-6. An analytical matrix for cyclone dust from the Foster Wheeler/Stoic two-stage gasifier (UMD).

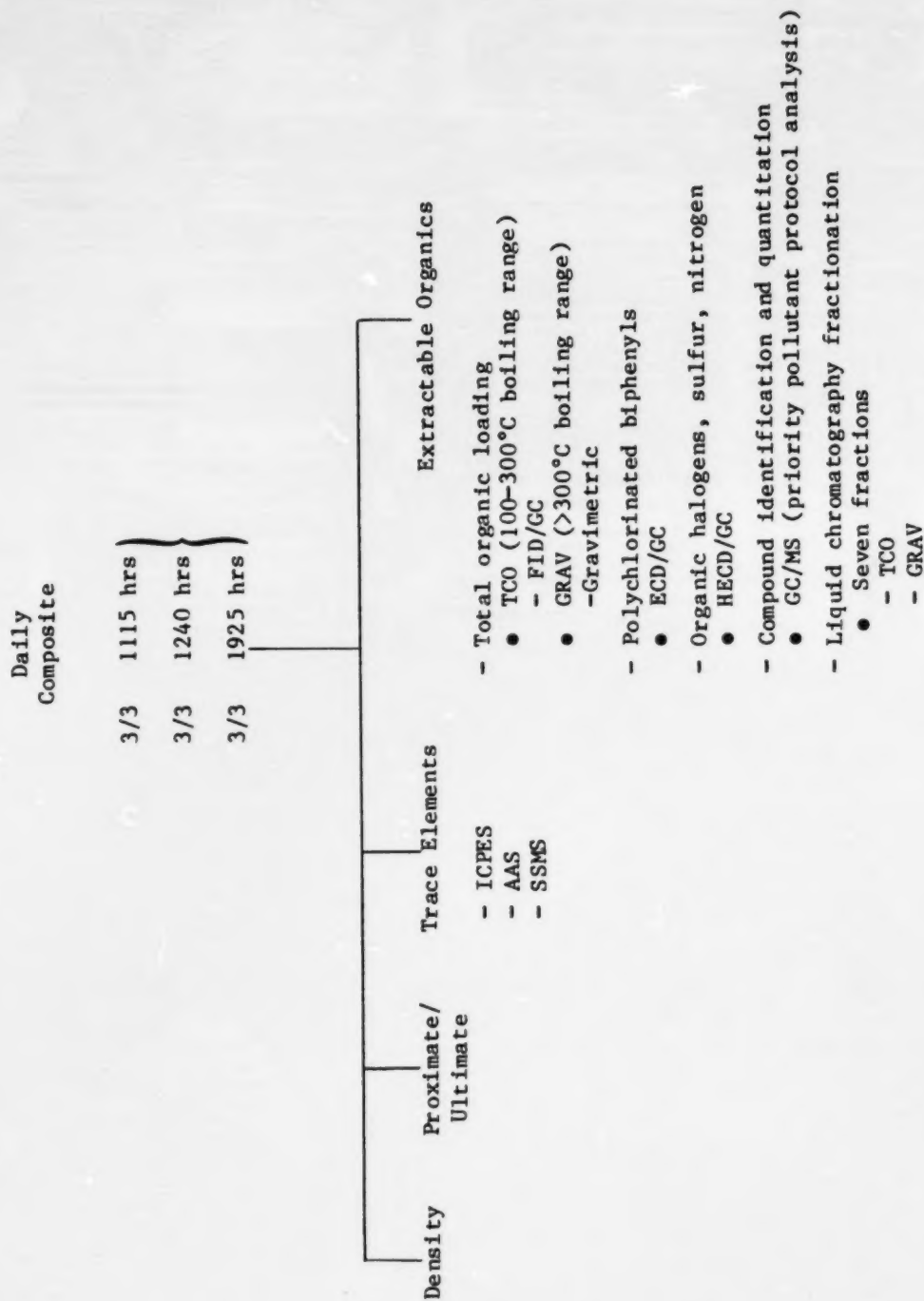


Figure D-7. An analytical matrix for ESP Oils to Boiler No. 3 from the Foster Wheeler/Stoic two-stage gasifier (UMD).

Daily
Composites

2/24	1640 hrs
2/24	2005 hrs
2/25	0155 hrs
2/27	1430 hrs
2/27	1805 hrs
3/5	1245 hrs
3/5	1550 hrs
3/5	2025 hrs

Proximate/
Ultimate

Figure D-8. An analytical matrix for ESP oils from the Foster Wheeler/Stoic two-stage gasifier (UMD).

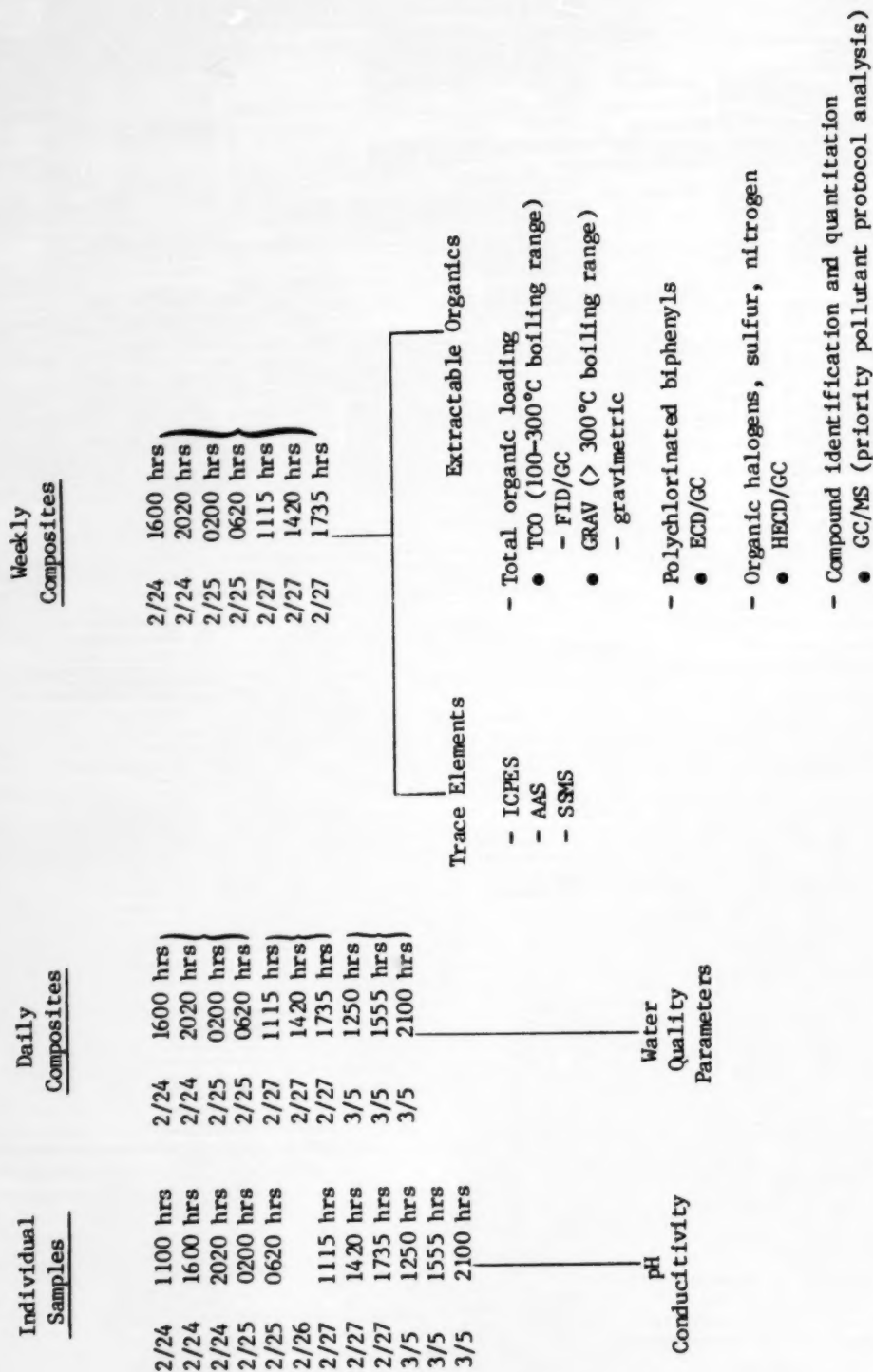


Figure D-9. Analytical results for ash pan water from the Foster Wheeler/Stoic two-stage gasifier (UMD).

absorbance at 578 nm. Cyanide in the range of 1-10 μg was determined spectrophotometrically by comparison with a standard calibration curve generated from potassium cyanide standard solutions. The detection limit for the procedure is .02 $\mu\text{g/mL}$ cyanide in the diluted distillate. This is equivalent to 20 ppb hydrogen cyanide by volume collected from 0.28 SCM of sampled gas in 200 mL of impinger solution.

D.2.3 Thiocyanate

Thiocyanate was determined colorimetrically using the same samples used for cyanide analysis (Ref. D-3). An aliquot of the sample solution was extracted with chloroform to remove organic compounds. A solution of copper and pyridine was added to the aqueous layer to form a copper dithiocyanato-pyridine complex. The complex was extracted with chloroform and determined spectrophotometrically at 407 nm by comparison to a calibration curve generated from standard solutions. The detection limit for the procedure is 4 $\mu\text{g/mL}$ thiocyanate. This is equivalent to 2.7 ppm thiocyanate by volume collected from 0.28 SCM of sampled gas in 200 mL of impinger solution.

D.2.4 Fluoride

Baker (Ref. D-3), Bauman (Ref. D-4), and Thomas (Ref. D-5), describe standard addition techniques using a specific ion electrode for the measurement of fluoride in aqueous samples. A citrate buffer was added to the sample to buffer both pH and ionic strength. Fluoride bound in metal complexes was released by the buffer. Fluoride concentration was related to potential changes which were measured using an Orion Model 94-09A solid-state fluoride electrode and a Corning Model 125 pH/millivolt meter. A detection limit of 0.05 $\mu\text{g/mL}$ is reported.

D.2.5 Phosphate

Total phosphate was determined by the persulfate digestion/single reagent colorimetric method (Ref. D-1). Aqueous samples were boiled with

ammonium persulfate and sulfuric acid to convert all oxidizable phosphorus compounds to the orthophosphate form. The resulting solutions were then treated with a single reagent containing ammonium molybdate, antimony, potassium tartrate, and ascorbic acid for color development. This reagent forms an antimony phosphomolybdate complex when reacted with orthophosphate solutions under acidic conditions. The complex was reduced by ascorbic acid to an intensely blue complex which is measured spectrophotometrically and compared to two standard calibration curves. The detection limit for the method was 0.5 $\mu\text{g PO}_4\text{-P/mL}$.

D.2.6 Chloride

Chloride was determined spectrophotometrically according to ASTM methodology [Designation D-512-67 (1974), Method C (Ref. D-6)]. An aliquot of sample was treated successively with a ferric ammonium sulfate solution and a mercuric thiocyanate solution. The intensity of the red ferric thiocyanate complex formed was measured at 463 nm using the Spectronic 20 and a 1 cm cell. Calibration curves were prepared by analyzing aliquots of standards containing 1-10 $\mu\text{g Cl/mL}$.

D.2.7 Nitrate

Nitrate was determined colorimetrically using the standard method (Ref. D.1). Nitrate reacts with chromotropic acid to form a yellow reaction product which was measured spectrophotometrically at 410 nm and compared to standards ranging from 1-10 $\mu\text{g NO}_3\text{-N/mL}$. A detection limit of 1.0 $\mu\text{g NO}_3\text{-N/mL}$ was reported.

D.2.8 Sulfate

In aqueous samples, sulfate was determined by the addition of barium chloride (Ref. D-1). The resulting turbid solution was analyzed spectrophotometrically at 420 nm. Standard curves were prepared for sample quantitation. A detection limit of 5 $\mu\text{g/mL}$ was reported.

Sulfur dioxide and sulfur trioxide in gaseous streams were analyzed as sulfate. Both gas species were adsorbed and oxidized to sulfate in hydrogen peroxide impinger solutions (Ref. D-7). The solutions were then titrated with standard barium perchlorate solution with thorin as an indicator. A detection limit of 3.4 mg SO₂/m³ was reported.

D.2.9 Sulfite-Sulfide

Total sulfite and sulfide in aqueous samples were analyzed by titrating with a standardized potassium iodide-iodine solution with starch as an end-point indicator (Ref. D-1). Sulfide in the impinger solutions was precipitated as zinc sulfide with zinc acetate under alkaline conditions. The zinc sulfide was removed by filtration, transferred to a flask, and an excess of a standard iodine solution was added. The excess iodine was then back-titrated with sodium thiosulfate to a starch end-point to determine the amount of iodine consumed by sulfide (Ref. D-1). The filtrate was titrated with a standard iodine-iodate solution. The detection limit was 2 µg S-/mL and 2.5 µg SO₃/mL.

D.2.10 Chemical Oxygen Demand (COD)

Chemical oxygen demand was determined by the standard method (Ref. D-1) by refluxing the sample with potassium dichromate and sulfuric acid for two hours. After cooling, the excess dichromate was titrated with ferrous ammonium sulfate. The amount of potassium dichromate consumed is proportional to the amount of oxidizable organic matter in the sample.

D.2.11 Total Organic Carbon (TOC)

Total organic carbon (TOC) was determined with a Beckman Model 915-B carbon analyzer. Total organic carbon in the sample was converted to CO₂ by a process of oxidative combustion, and the CO₂ concentration was measured by a non-dispersive infrared detector. Because all carbon in the sample is measured, procedural modifications are needed to limit the

determination of organic carbon. Inorganic carbonates were removed by acid sparging the sample prior to the analysis for TOC (Reference D-1).

D.2.12 Biological Oxygen Demand (BOD)

BOD is a measure of the change in the amount of dissolved oxygen in a sample when incubated in the dark at 20°C for five days (Ref. D-1). This change in dissolved oxygen was related to the amount of organic matter assimilated and oxidized by microorganisms. An initial dissolved oxygen concentration was determined, and after five days, a final concentration was determined.

D.2.13 Total Suspended Solids (TSS)

TSS was determined using the standard method (Ref. D-1). The sample was filtered through a fine glass fiber filter. The resulting residue on the filter was dried to constant weight at 105°C. The mass of solids per liquid volume was reported as total suspended solids in mg/L.

D.2.14 Total Dissolved Solids (TDS)

TDS was determined using the standard method (Ref. D-1). The filtrate from the TSS determination was evaporated to dryness. The resulting residue was heated to constant weight at 180°C. The mass of solids per liquid volume was reported as TDS at 180°C in units of mg/L.

D.2.15 pH Value

Hydrogen ion activity was determined electrometrically by the standard method (Ref. D-1) using a Corning Model 125 pH meter with a combination electrode.

D.2.16 Specific Conductance

The specific conductance of a sample was measured by the standard method (Ref. D-1) using a YSI Model 33 conductivity meter. Samples were either analyzed at 25°C, or temperature corrections were made to standardize reporting.

D.2.17 NO_x Analysis

The nitrogen oxide content of a sample was determined using EPA Method 7 (Ref D-7). The gas sample was collected in evacuated 2-L flasks containing a sulfuric acid-hydrogen peroxide absorbing solution which converts NO to NO₂. The sample was basified and the NO₂ concentration determined colorimetrically using phenoldisulfonic acid. Sample absorbance was compared to measured absorbance of potassium nitrate standards at 410 nm.

D.2.18 Phenols

Phenols were analyzed in aqueous samples by the standard method (Ref. D-1). Interferences were removed from the sample by distilling the phenols. The phenols in the distillate were reacted with 4-aminoantipyrine at pH 10 in the presence of potassium ferricyanide to form a colored antipyrine dye. This dye was extracted with chloroform and the absorbance was measured spectrophotometrically at 460 nm. A detection limit of 10 µg/mL was reported.

D.3.0 ELEMENTAL ANALYSIS

D.3.1 Spark Source Mass Spectrometry (SSMS)

Trace metals in the gas, liquid, and solid streams were analyzed by spark source mass spectrometry. Gas samples were collected by SASS train in impingers containing silver persulfate solution (Ref. D-8). The impinger volumes were determined, and aliquots that represented a constant volume of

gas were removed from each impinger in the sample train. Aliquots of the liquid and solid samples were sent directly for SSMS with no additional sample preparation. XAD-2 resin samples were extracted with methylene chloride and diethyl ether and analyzed like the liquid samples.

D.3.2 Antimony, Arsenic, Selenium, and Mercury

The volatile elements were analyzed directly from the composite impinger and aqueous samples by atomic absorption spectrophotometry (AAS). Solids were digested in perchloric acid, dried, and dissolved in hydrochloric acid; the resultant solutions were analyzed by graphite furnace atomic absorption.

Arsenic, antimony, and selenium were analyzed in liquid samples using a hydride generation attachment for the atomic absorption spectrophotometer (Ref. D-8). An aliquot of sample was introduced into the system and acidified. Sodium borohydride was added to reduce the species to the hydride. The generated gas was purged into a hydrogen-argon flame for analysis.

The absorbances and detection limits for the hydride generation method are show below:

<u>Element</u>	<u>Wavelength</u>	<u>Detection Limit</u> <u>(ng/mL of solution)</u>
As	193.7 nm	3 ppb
Sb	217.0 nm	5 ppb
Se	196.0 nm	4 ppb

Mercury in samples of coal, ash, cyclone dust, and tar were determined by ignition in an oxygen bomb with absorption in KMnO_4 . The resultant solution was analyzed according to "Chemical Analysis of Water and Wastes," U.S. EPA, March, 1979 (Ref. D-9), which involves treatment with potassium persulfate and subsequent reduction of the mercury with stannous chloride and analysis by cold vapor atomic absorption. The ash pan water composite was analyzed directly without the oxygen bomb preparation.

Mercury in gas streams was collected on gold plugs and analyzed by thermal desorption of the mercury from the gold plug. An argon carrier gas swept the mercury into an absorption cell of an atomic absorption spectrometer and the resultant peak was recorded. Gas phase mercury standards obtained by withdrawing various volumes of gas from a sealed bottle saturated with mercury vapor were used for calibration.

D.3.3 Inductively Coupled Argon Plasma Emission Spectrometry (ICPES)

Samples of ash pan water, ESP tar to Boiler No. 3, trace metal impingers, the impinger blank, and perchloric acid digestions of coal, gasifier ash, and cyclone dust were analyzed by ICPES. Solid samples for boron determination were prepared by sodium carbonate fusion and dissolution in dilute hydrochloric acid. The solutions were aspirated into an argon plasma; the high temperature of plasma (10,000°C) induces electronic excitation of the atoms. The emitted light was passed through a slit and diffracted by a grating. Photomultiplier tubes were set at the characteristic wavelengths of the elements measured.

D.3.4 Proximate/Ultimate Analysis

Proximate/ultimate analyses, subcontracted to Commercial Testing and Engineering Co., Chicago, Illinois, were performed on composites of coal, cyclone dust, gasifier ash, and ESP oil by ASTM methods (Ref. D-10).

D.4.0 LEACHING STUDIES

Gasifier ash and cyclone dust were extracted using the Resource Conservation and Recovery Act (RCRA) procedure as outlined in the Federal Register of December 18, 1978, "Hazardous Wastes, Proposed Guidelines and Regulations" (Ref. D-11). One hundred grams of the air-dried solid were weighed into a polyethylene bottle and 1.6 kg of deionized water was added. The pH was adjusted to 5.0 with Ultrex glacial acetic acid. The pH was monitored for one hour and adjusted as needed to maintain the pH 5.0 level.

The sample was agitated for 24 hours with a rotary shaker. At the end of the 24-hour leaching, the samples were filtered through a 0.45 μ m Millipore filter. The filtrate was diluted to 2 kg water. The dilute filtrates were analyzed for trace elements by ICPEs and AAS.

D.5.0 GAS CHROMATOGRAPHY

Grab samples of the combined gas, clean top gas, dirty top gas, clean bottom gas, dirty bottom gas, main stack combustion emissions, and combustion emissions from Boiler No. 3 firing ESP oil and natural gas were analyzed for the following components:

- fixed gases: H_2 , O_2+Ar , N_2 , CO_2 , CO , and CH_4 ,
- volatile sulfur species: H_2S , COS , SO_2 , CH_3SH , C_2H_5SH , C_4H_4S , and CS_2 ,
- light hydrocarbons: C_1 to C_6 , boiling range $-160^\circ C$ to $80^\circ C$

The samples were taken in silanized glass sample bombs and analyzed immediately.

D.5.1 Volatile Sulfur Species

The instrument operating parameters for the analysis of volatile sulfur species were:

Instrument:	Tracor Model 560 FPD-GC
Column:	10 ft x 1/8 inch OD Teflon, 1% TCEP [1,2,3-tris(cyanoethoxy) propane] and 0.5% H_3PO_4 on Carbopack B, 60/80 mesh
Carrier:	Nitrogen at 20 mL/min
Detector Temperature:	$150^\circ C$
Oven Program:	$30^\circ C$ for 7 min.; $10^\circ C/min$ to $120^\circ C$; $120^\circ C$ for 10 min.

Standard samples were generated on site using Metronics permeation tubes with certified permeation rates and zero grade nitrogen at known flows. Figures D-10 through D-16 show calibration curves for H₂S, COS, SO₂, CH₃SH, C₂H₅SH, C₄H₄S, and CS₂.

D.5.2 Fixed Gases

The fixed gas analyses were conducted under the following conditions:

Instrument:	Fisher Gas Partitioner Model 1200-TCD-GC
Column 1:	6 1/2 ft. x 1/8 inch OD, Porapak PQ
Column 2:	11 ft x 3/16 inch OD, Molecular Sieve 13X
Carrier:	8.5% Hydrogen in Helium at 28 mL/min
Oven Temperature:	50°C
Bridge Current:	225 mA

The instrument calibration for the fixed gases was performed on site using certified gas mixtures containing the species of interest. The certified gas mixtures supplied by Scientific Gas Products had the following composition:

<u>Species</u>	<u>STD #1</u>	<u>STD #2</u>	<u>STD #3</u>
CO ₂	19.7%	10.1%	1.99%
H ₂	29.8%	9.90%	19.8%
O ₂ +Ar	0.85%	0.97%	1.97%
N ₂	24.33%	64.99%	34.06%

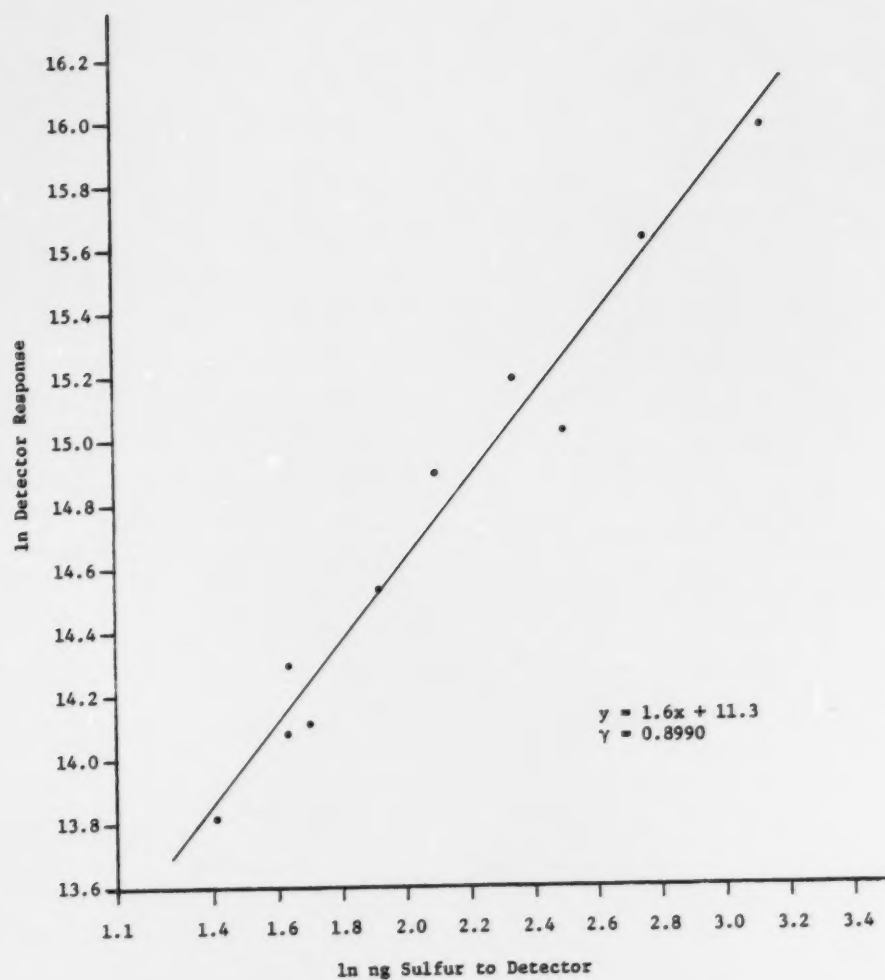


Figure D-10. Hydrogen sulfide (H₂S) calibration curve.

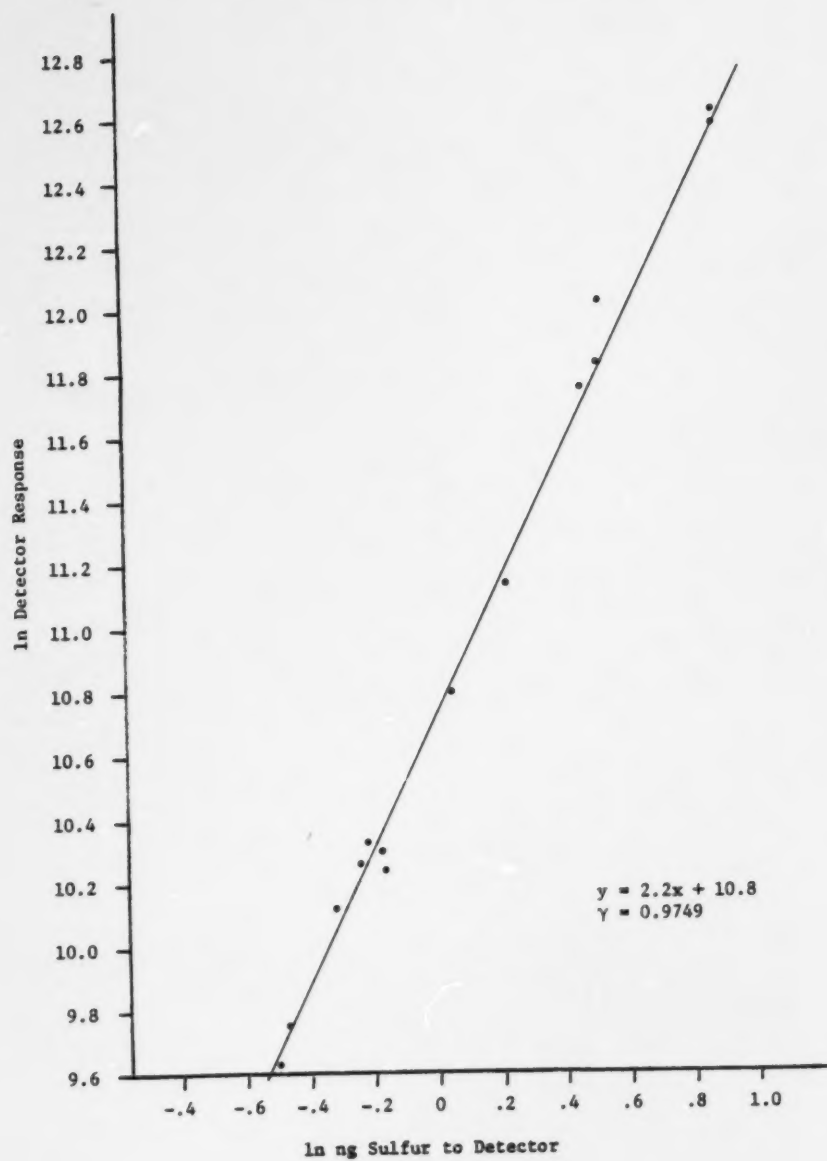


Figure D-11. Carbonyl sulfide (COS) calibration curve.

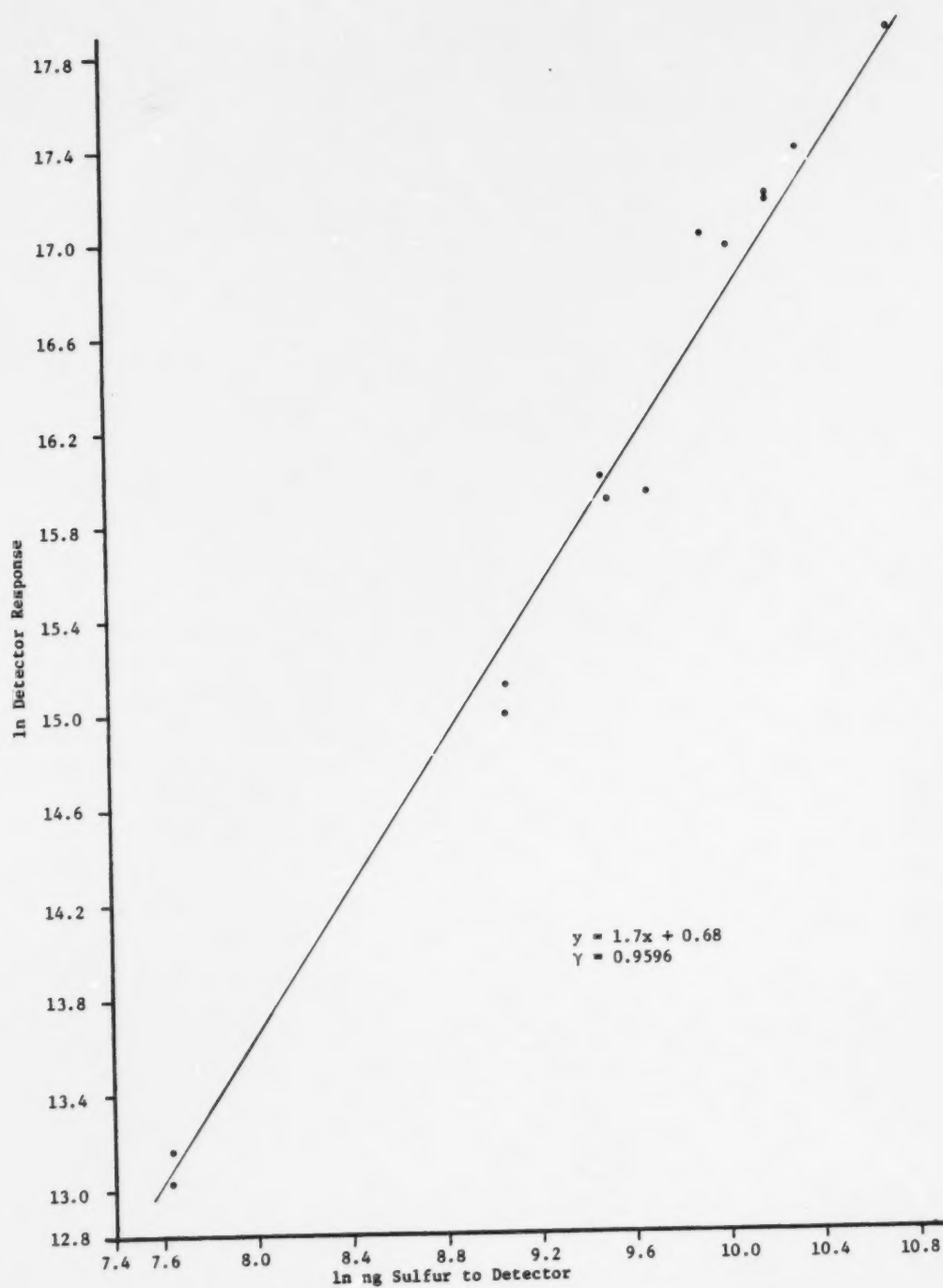


Figure D-12. Sulfur dioxide (SO₂) calibration curve.

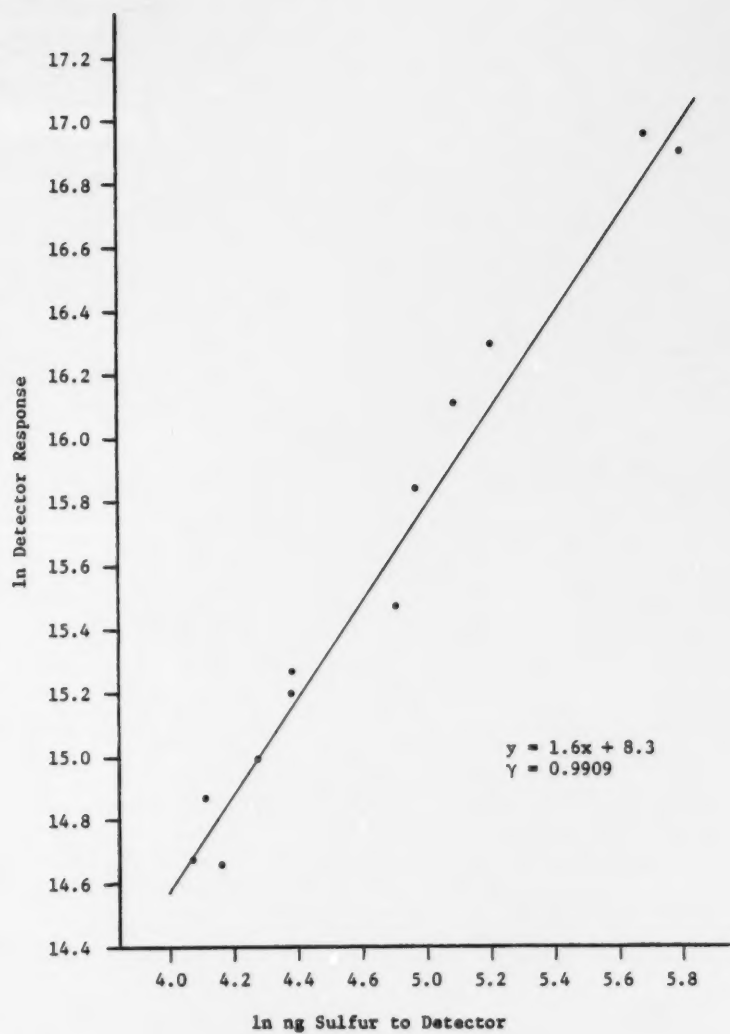


Figure D-13. Methyl mercaptan (CH_3SH) calibration curve.

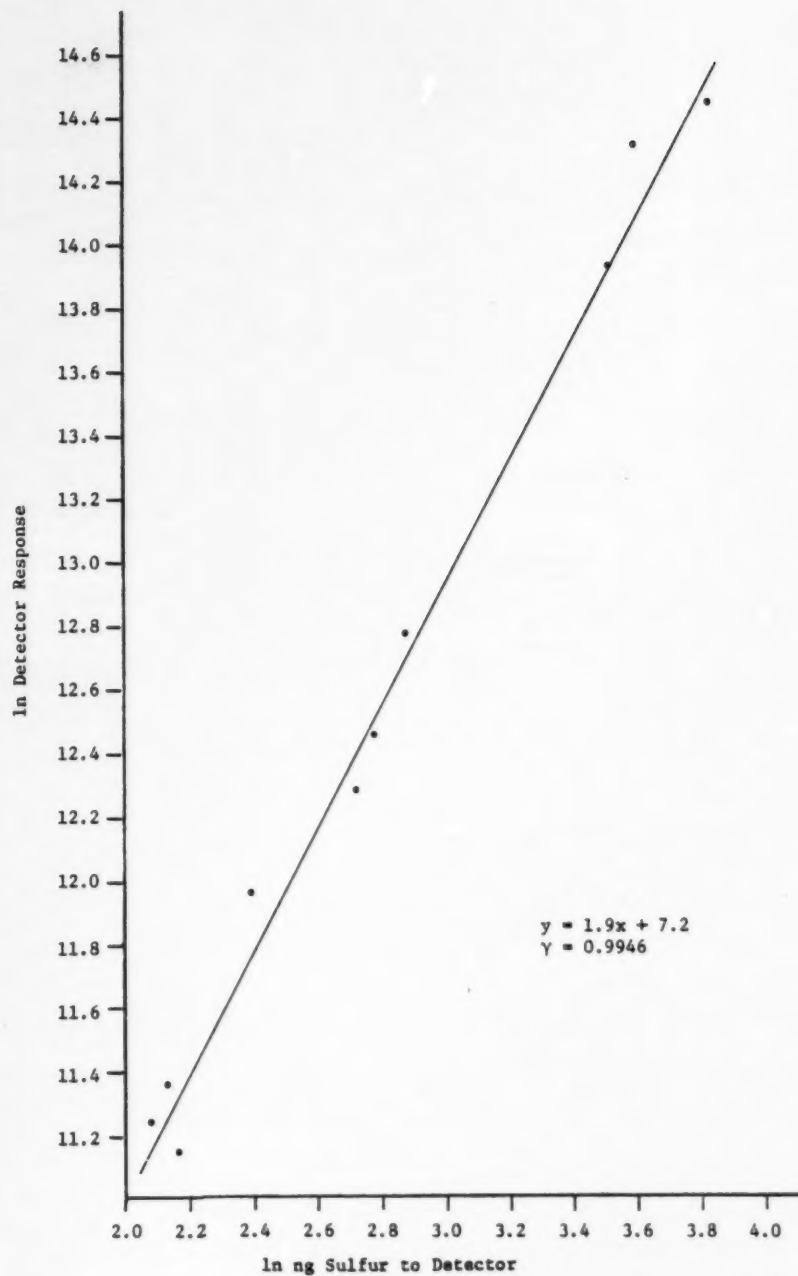


Figure D-14. Ethyl mercaptan (C_2H_5SH) calibration curve.

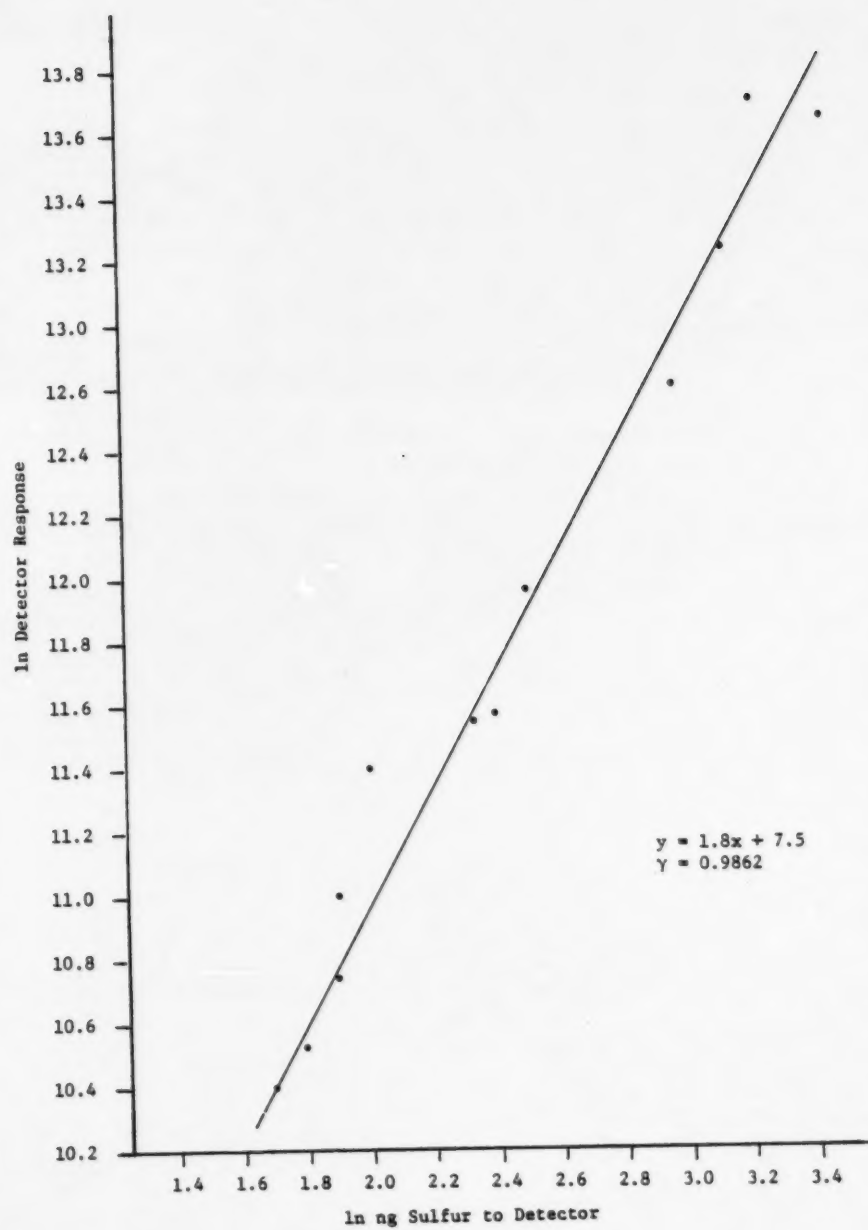


Figure D-15. Thiophene (C_4H_4S) calibration curve.

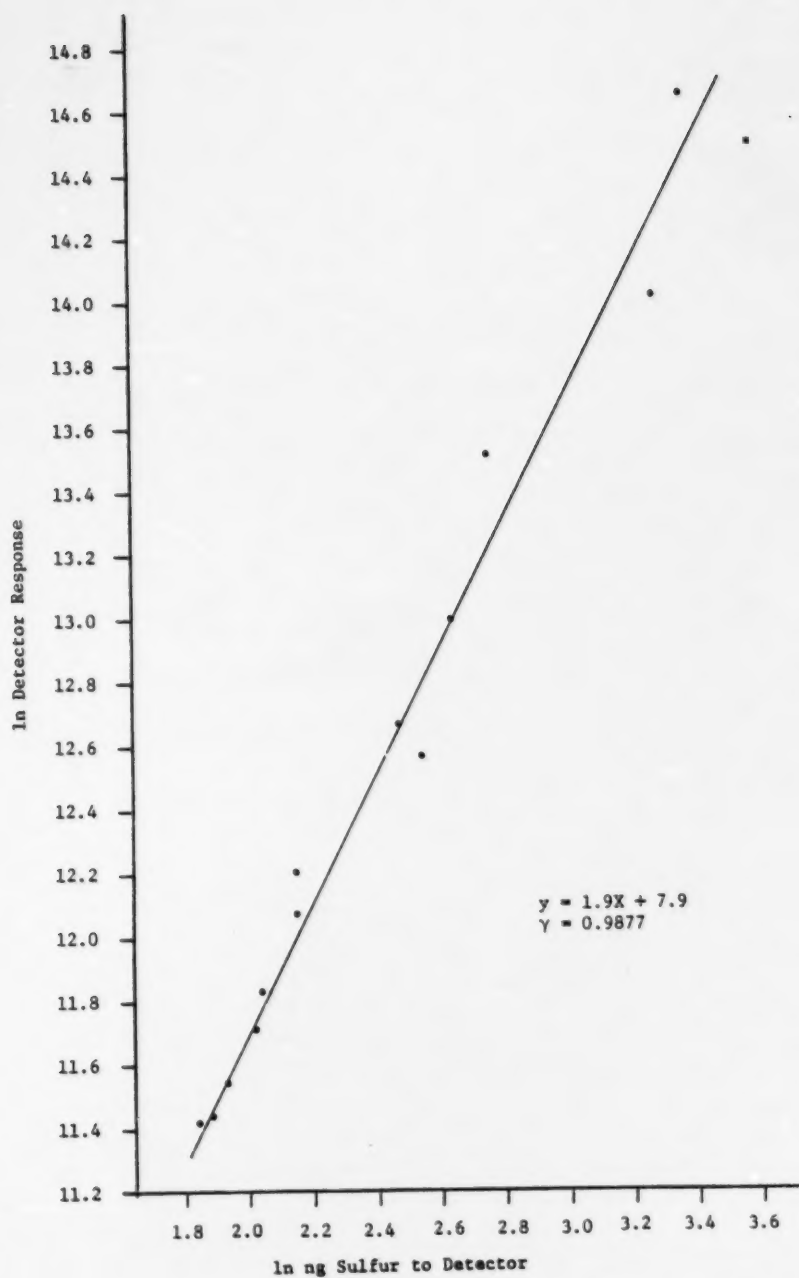


Figure D-16. Carbon disulfide (CS₂) calibration curve.

<u>Species</u>	<u>STD #1</u>	<u>STD #2</u>	<u>STD #3</u>
CH ₄	0.62%	4.04%	2.10%
CO	24.7%	10.0%	40.08%

Figures D-17 through D-20 show the calibration curves that were obtained.

D.5.3 Light Hydrocarbons

C₁-C₆ hydrocarbons were analyzed on site using the following instrument parameters:

Instrument:	Tracor Model 560 FID-GC
Column:	6 ft. x 1/8 inch OD stainless steel, 80/100 mesh Porapak Q
Carrier:	Nitrogen at 30 mL/min.
Detector Temperature:	300°C
Oven Program:	50°C for 2 minutes; 10°C/min. to 200°C; 200°C until baseline reestablished

Calibration curves were obtained using certified gas mixtures of C₁-C₆ paraffins and C₂-C₆ olefins. Response factors for methane, ethane, and ethylene were determined individually. Average response factors for the paraffins and corresponding olefins for C₃-C₆ hydrocarbons was used. Calibration curves are presented in Figure D-21.

D.6.0 ORGANIC ANALYSES

Preliminary analyses of the samples were done using established methods (Ref. D-8) with modifications. These modifications are included in the following descriptions of the analytical methods. Detailed analyses were done using a computer assisted gas chromatography - mass spectrometry system

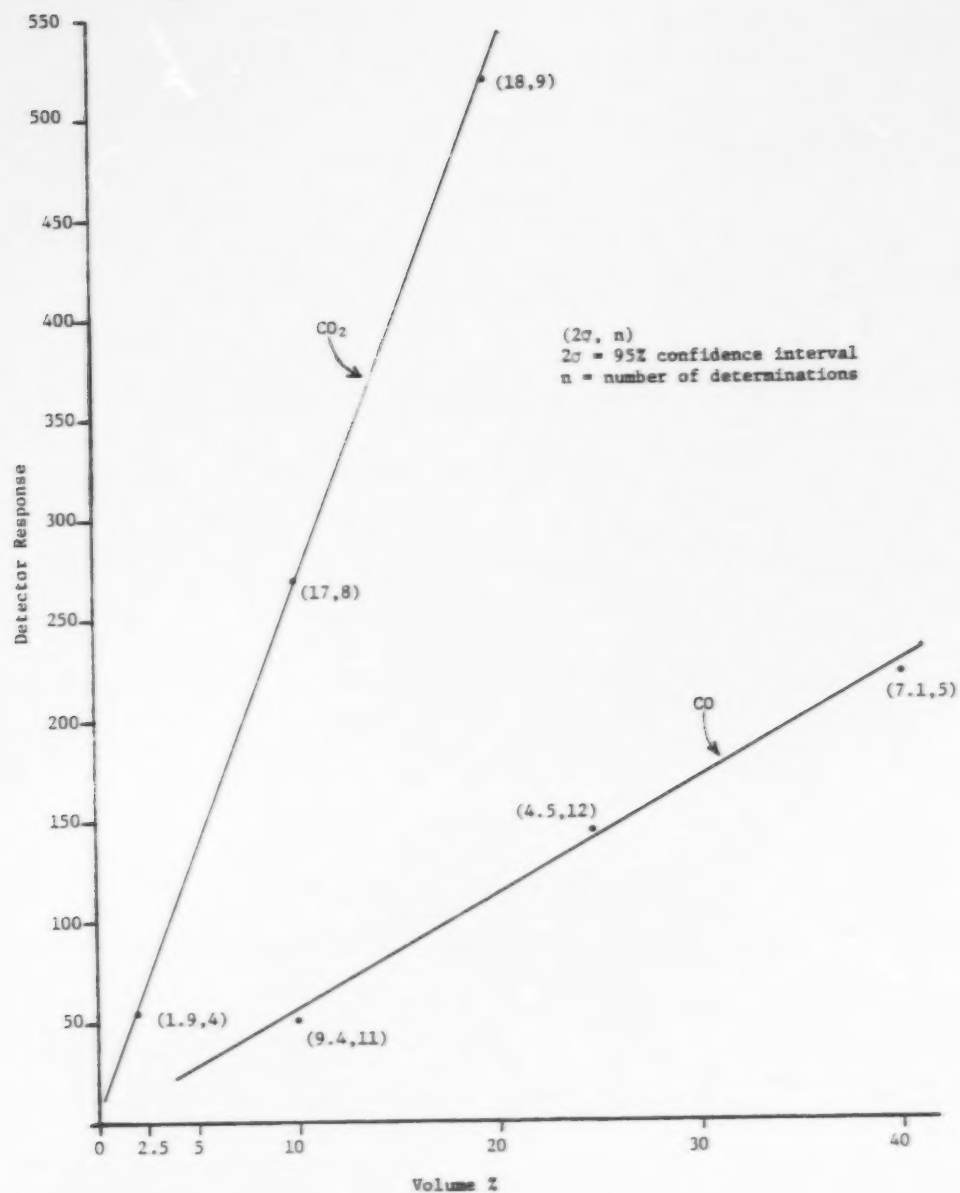


Figure D-17. Calibration curves for carbon dioxide (CO₂) and carbon monoxide (CO).

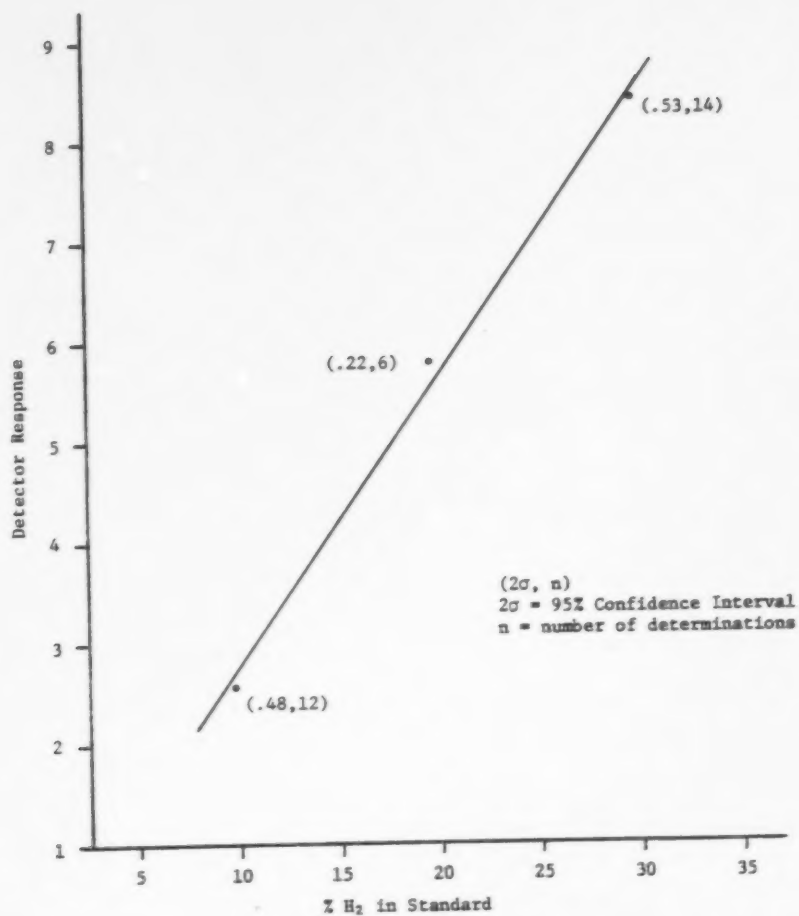


Figure D-18. Hydrogen calibration curve.

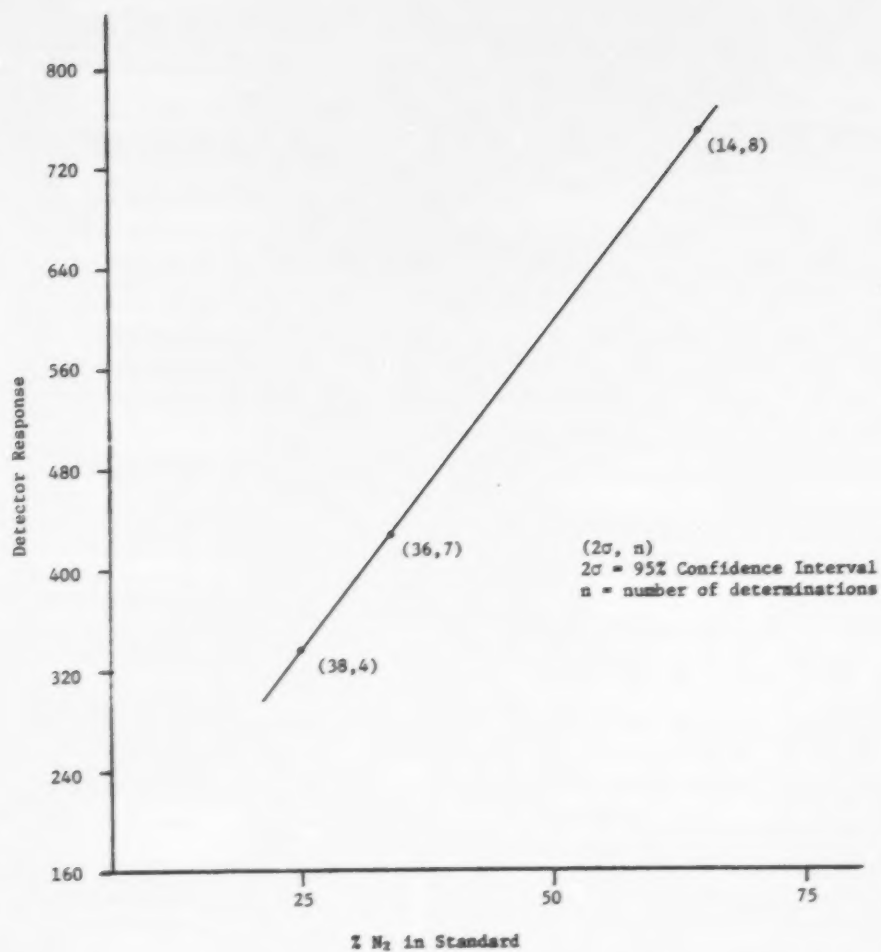


Figure D-19. Nitrogen calibration curve.

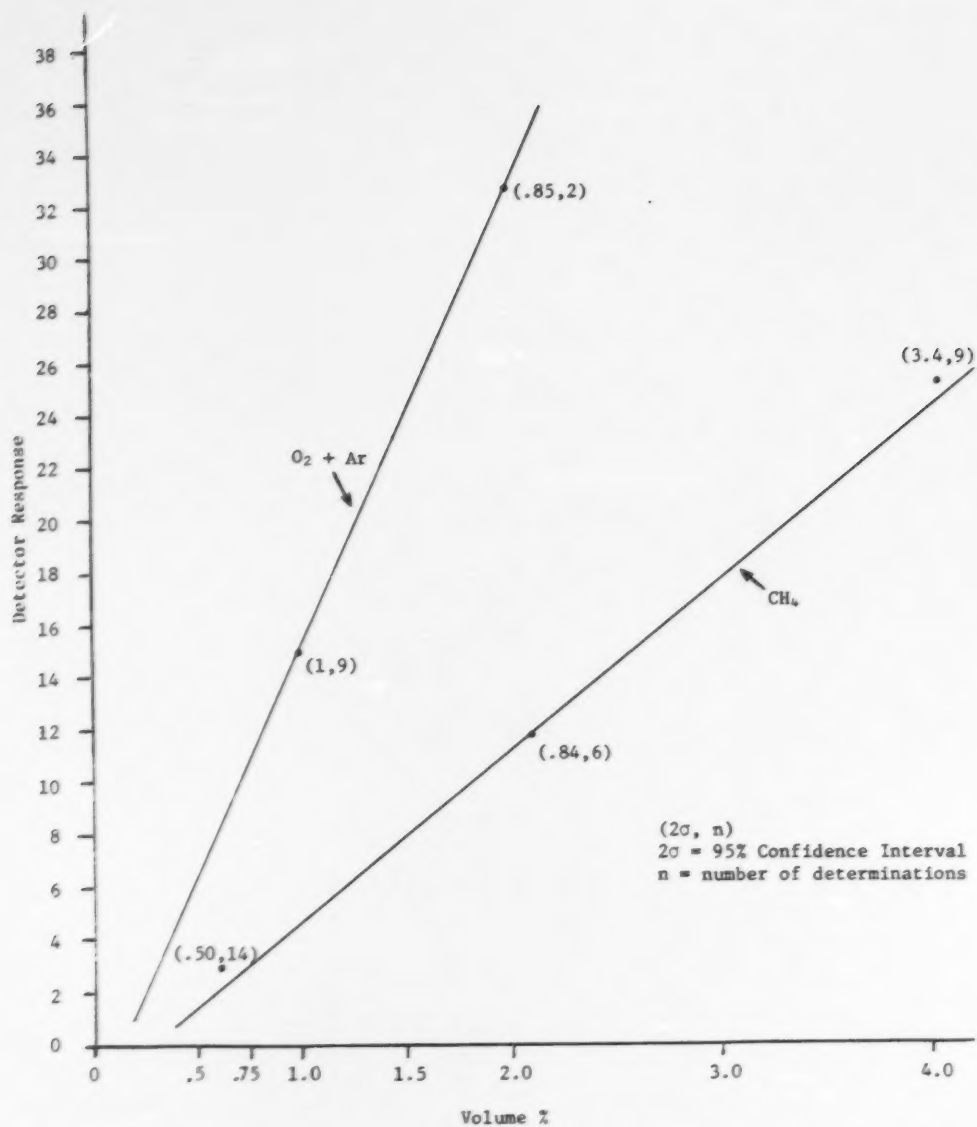


Figure D-20. Calibration curves for methane (CH_4) and oxygen ($O_2 + Ar$).

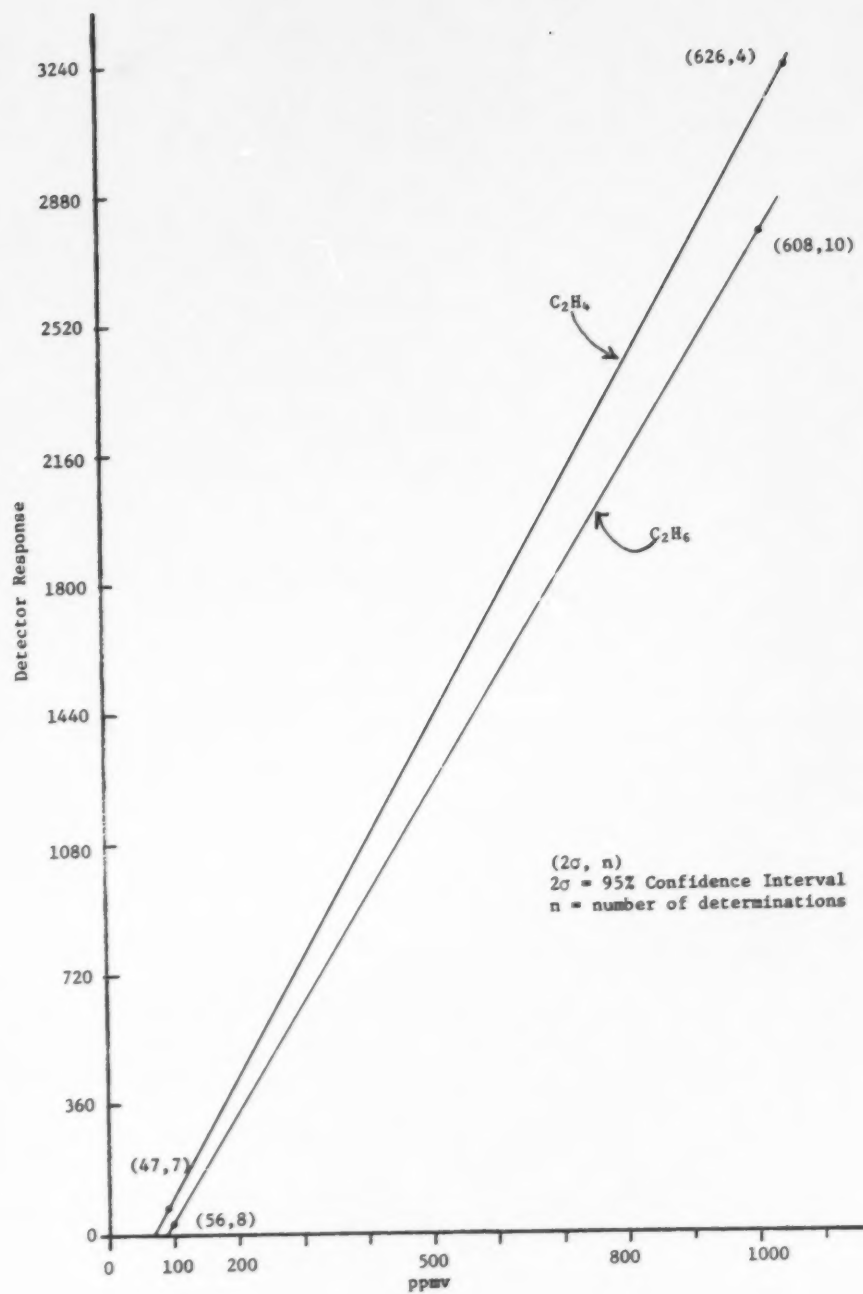


Figure D-21. Ethane (C_2H_6) and ethylene (C_2H_4) calibration curves.

(GC-MS). GC-MS analysis was done after sample screening by gas chromatograph (GC) using a selective detector in nitrogen-, sulfur-, and halogen-specific modes and after screening for PCB's. These GC analyses yielded data pertaining to organic compounds and compound classes which contain specific heteroatoms at very low additional cost to the comprehensive analysis.

D.6.1 Extractions

Aliquots of the aqueous samples were extracted at both pH=2 and pH=12. Serial extractions using methylene chloride followed by diethyl ether were done at each pH to obtain a wider range of extractable organics than extraction with a single solvent would have yielded. After extraction, the methylene chloride and ether fractions were combined to yield one extract for each sample. The extracts were concentrated using Kuderna-Danish concentrators equipped with 3-ball Snyder columns.

Aliquots of solid samples were extracted in Soxhlet extractors for 24 hours with methylene chloride followed by 24 hours with ether. After extraction, the 2 fractions were combined to yield one extract for each sample.

The use of diethyl ether to extract the aqueous and solid samples was an addition to the cited methods (Ref. D-8).

D.6.2 Total Chromatographable Organics Analysis (TCO)

TCO analysis is a means of quantitating organics with boiling points in the range of 100°C to 300°C using gas chromatography. Quantitation was performed by comparing sample chromatograms to an n-C₇ - C₁₇ boiling point standard and an n-C₈, C₁₂, C₁₆ quantitative standard. The quantitative standard was analyzed at concentrations approximating sample concentrations. The following conditions were used:

Instrument:	Tracor Model 560 FID-GC
Column:	3 m x 2 mm ID glass, 10% OV-101 on 60/80 mesh Supelcoport
Carrier:	Nitrogen at 40 mL/min
Injector:	265°C
Detector:	300°C
Oven Program:	50°C for 5 min; 20°C/min to 250°C; 250°C hold until baseline reestablished

Chromatograms were recorded and integrated with a Hewlett-Packard 3390 integrator. Solvent blanks were analyzed for all samples and subtracted from sample values.

D.6.3 Gravimetric Analysis (GRAV)

GRAV analysis is a means of quantitating the non-volatile portion of a sample extract. Theoretically, this includes all compounds with a boiling point greater than 300°C. GRAV analyses were done on initial extracts and on liquid chromatographic (LC) fractions. A Mettler H-80 4-place analytical balance was used for all weighings.

GRAV analyses on the unfractionated extracts and LC fractions were done by taking volumetric aliquots of the concentrated extracts or LC fractions. The GRAV aliquots were placed in tared beakers and allowed to evaporate to dryness at room temperature in a desiccator. If less than 10 mg of material was present, additional aliquots were added until approximately 10 mg of GRAV material was present. In some cases for the LC fractions, this was not done due to low total masses in the fractions. Weighings were done at minimum 24-hour intervals until a maximum variation of ± 0.1 mg was observed. Solvent and LC fraction blanks were analyzed along with the samples and subtracted from sample values.

D.6.4 Gas Chromatographic Analysis - Benzene, Toluene, and Xylenes

Benzene, toluene, and xylene samples were taken using 2-section glass tubes packed with activated charcoal. After sampling, the tubes were refrigerated until they could be extracted. Extraction was carried out by separate elution of the front and back sections of charcoal with carbon disulfide for 30 minutes with shaking. Analyses of the front and back sections of charcoal extracts were done separately using the following instrument parameters:

Instrument:	Tracor Model 560 FID-GC
Column:	1.8 m x 2 mm ID glass, 5% SP-1200/1.75% Bentone 34 on 100/120 mesh Supelcoport
Carrier:	Nitrogen at 25 mL/min
Injector Temperature:	275°C
Detector Temperature:	300°C
Oven Program:	60°C for 3 min; 10°C/min to 120°C

Calibration was performed by analyzing known quantities of benzene, toluene, and xylenes and establishing response factors for the compounds.

D.6.5 Gas Chromatographic Analysis - Hall Detector

Gas chromatography was performed using a Hall 700-A electrolytic conductivity detector (HECD). The HECD may be used in either the halogen, sulfur, or nitrogen specific mode. Three analyses were performed on each extract, one in each of the the three detector modes. This provided measurements analogous to TCO values for classes of organic compounds containing specific heteroatoms. The analytical parameters were:

Instrument:	Tracor Model 560 HECD-GC
Columns:	3 m x 2 mm ID glass, 10% OV-101 on 60/80 mmesh Supelcoport (halogen and sulfur modes); 10% Apiezon L/2% KOH on 80/100 mesh Chromosorb W AW (nitrogen mode).
Carrier:	Helium at 40 mL/min
Oven Program:	50°C for 5 min; 20°C/min to 250°C; 250°C hold until baseline reestablished.
Injector Temperature:	265°C
Detector Temperature:	265°C
Standards:	Halogen mode - hexachlorobenzene Sulfur mode - ethyl parathion Nitrogen mode - pyridine

Samples were solvent exchanged with cyclopentane prior to analysis. Chromatograms were plotted and integrated with a Hewlett-Packard Model 3388 data processor. Standards were analyzed to provide boiling point references and concentration correlations. Solvent blanks were analyzed with the samples and subtracted from sample values.

D.6.6 Gas Chromatographic Analysis - Electron Capture Detector

Extracts were analyzed for PCBs by electron capture detector using the following parameters:

Instrument:	Tracor Model 560 GC equipped with Ni ⁶³ Electron Capture Detector
Column:	6 ft x 1/8 in. ID glass, 1.5% SP-2401 on 100/120 Supelcoport
Carrier:	Argon/Methane at 30 mL/min
Make-up:	Argon/Methane at 30 mL/min

Injector Temperature:	225°C
Detector Temperature:	350°C
Oven Program:	180°C for 5 min; 8°C/min. to 210°C; 210°C for 25 min.

Samples were solvent exchanged with cyclopentane prior to analysis. Chromatograms were plotted and integrated with a Hewlett-Packard Model 3388 data processor. Standards were analyzed to provide boiling point references and concentration correlations. Solvent blanks were analyzed with the samples and subtracted from sample values.

D.6.7 Mass Spectrometric Analysis

Composited extracts for each of seven samples were analyzed using gas chromatography - mass spectrometry (GC-MS). The samples were analyzed for the base-neutral and acid extractable EPA Priority Pollutants given in Table D-1 using a computer assisted search. This technique searches for specific compounds based on the appearance of three characteristic fragment ions within a known relative retention time window. Quantitation was done by response factor comparison to a known amount of d₁₀-anthracene.

Samples were analyzed by GC-MS using the following parameters:

Instrument:	Hewlett-Packard 5985 GC-MS system
Column:	30 meter SE-54 fused silica capillary
Carrier:	Hydrogen at 35 cm/sec
Injector Temperature:	250°C
Oven Program:	25°C for 3 min; 16°C/min to 60°C and 6°C/min to 300°C
Ionizing Voltage:	70 eV

TABLE D-1. BASE/NEUTRAL AND ACID EXTRACTABLE PRIORITY POLLUTANTS
ANALYZED BY GC-MS

Base/Neutral Extractable		Acid Extractable	
Compound	m/e	Compound	m/e
1. 1,3-Dichlorobenzene	146	1. 2-Chlorophenol	128
2. 1,4-Dichlorobenzene	146	2. 2-Nitrophenol	139
3. Hexachloroethane	117	3. Phenol	94
4. 1,2-Dichlorobenzene	146	4. 2,4-Dimethylphenol	12
5. Bis(2-Chloroethyl) ether	93	4. 2,4-Dichlorophneol	162
6. Bis(2-Chloroisopropyl) ether	121	6. 2,4,6-Trichlorophenol	196
7. Nitrobenzene	77	7. p-Chloro-m-cresol	142
8. N-Nitrosodimethylamine	74	8. 2,4-Dinitrophenol	184
9. N-Nitrosodi-n-propylamine	130	9. 4,6-Dinitro-o-cresol	198
10. Hexachlorobutadiene	225	10. Pentachlorophenol	266
11. Isophorone	82	11. 4-Nitrophenol	139
12. 1,2,4-Trichlorobenzene	180	12. 2,2-Difluorobiphenyl	190
13. Bis(2-Chloroethoxy) methane	93		
14. Naphthalene	128		
15. Hexachlorocyclopentadiene	237		
16. 2-Chloronaphthalene	162		
17. Acenaphthylene	152		
18. Acenaphthene	154		
19. Dimethyl phthalate	163		
20. 2,6-Dinitrotoluene	63		
21. 4-Chlorophenyl phenyl ether	204		
22. Fluorene	166		
23. 2,4-Dinitrotoluene	63		
24. Diethyl phthalate	149		
25. N-Nitrosodiphenylamine	169		
26. 1,2-Diphenyl Hydrazine	77		
27. Hexachlorobenzene	284		
28. 4-Bromophenyl phenyl ether	248		
29. Phenanthrene	178		
30. Anthracene	178		
31. Di-n-butyl phthalate	149		
32. Fluoranthene	202		
33. Pyrene	202		
34. Benzidine	184		
35. Butyl benzyl phthalate	149		
36. Bis(2-Ethylhexyl) phthalate	149		
37. ICDD	322		
38. Chrysene	228		
39. Benz(a)anthracene	228		

(Continued)

TABLE D-1. (Continued)

Base/Neutral Extractable		Acid Extractable	
Compound	m/e	Compound	m/e
40. 3,3-Dichlorobenzidine	252		
41. Di-n-octyl phthalate	149		
42. Benzo(b)fluoranthene	252		
43. Benzo(k)fluoranthene	252		
44. Benzo(a)pyrene	252		
45. Indeno (1,2,3-c,d)pyrene	276		
46. Benzo(a,h)anthracene	278		
47. Benzo(g,h,i)perylene	276		

¹ratio of mass to electron charge

D.6.8 Liquid Chromatographic Separation (LC)

The LC separation gives a fractionation of the sample extract into compound classes based on polarity. This is not a high resolution technique, and compound classes may overlap in individual fractions. However, with adequate care, reproducible, predictable separations can be achieved. Table D-2 presents the organic compound classes as separated into the seven LC fractions.

Samples were prepared for fractionation using recommended procedures (Ref. D-8). All extracts were solvent exchanged from methylene chloride/diethyl ether to cyclopentane over 200 mg silica gel. Using the recommended solvent exchange results in a solvent mixture containing less than 5% methylene chloride/ether. This is sufficiently low not to alter the polarity of the elution series. The exchanged extract and 200 mg silica gel were transferred to the LC column and eluted with a series of 7 solvent mixtures on the basis of increasing polarity. The LC column parameters were:

Column:	200 mm x 10.5 mm ID glass with Teflon stopcock, glass solvent reservoir; water-jacketed, outlet water temperature 20°C.
Absorbent:	Silica gel, 60-200 mesh, Davidson, Grade 950. Soxhlet extracted serially with methanol, methylene chloride, and pentane, activated at 110°C for 2 hours. Slurry packed in n-pentane.
Drying Agent:	Sodium sulfate, reagent grade, fired at 500°C for 4 hours. 3 g added to top of column.
Elution Series:	Pentane; methylene chloride/pentane 1:4; methylene chloride/pentane 1:1; methylene chloride; methanol/methylene chloride 1:19; methanol/methylene chloride 1:4; methanol/methylene chloride 1:1. Pentane fraction = 25 mL, all others = 10 mL.

TABLE D-2. COMPOUND CLASSES SEPARATION USING LIQUID CHROMATOGRAPHY TECHNIQUE†

Category (Subcategory)	Most Probable		Category (Subcategory)	Most Probable	
	Fraction*	LC		Fraction*	LC
Aliphatic hydrocarbons	1		Phenols		6
(Alkanes)	1		(Alkyl, etc.)		6
(Alkenes)	1		(Halogenated phenols)		6
(Alkynes)	1		(Nitrophenols)		6
Halogenated aliphatics	1,2		Esters		6
(Saturated)	1,2		(Phthalates)		6
(Unsaturated)	1,2		Ketones		6
Aromatic hydrocarbons	2,3		Amines		6
(Benzenes)	2,3		(Primary, secondary, tertiary)		6
Halogenated aromatic hydrocarbons	2,3		(Hydrazines, azo compounds)		6
Nitro aromatic hydrocarbons	4,5		(Nitrosoamines)		6
Fused alternate, nonalternate hydrocarbons	2,3		Heterocyclic nitrogen compounds		4
MW < 216 (methyl pyrene)	2,3		(Indoles, carbazoles)		6
MW > 216	2,3		(Quinolines, soridines)		6
Ethers	4		Alkyl sulfur compounds		6
(Halogenated ethers)	4		(Mercaptans)		6
Epoxides	4		(Sulfides, disulfides)		6
Aldehydes	4		Heterocyclic sulfur compounds		4
Heterocyclic oxygen compounds	4		(Benzothiophenes)		7
Nitriles	4		Sulfonic acids, sulfoxides		6
(Aliphatic)	4		Amides		6,7
(Aromatic)	4		Carboxylic acids		2,3,4
Alcohols	6		Silicones		5,6,7
(Primary, secondary, tertiary)	6		Phosphates		
(Glycols)	6				

*Possible assignments. Fractions 4-5, 5-6, 6-7 generally overlap to a considerable extent. Also, additional components of a particular molecule may cause it to elute in an LC fraction other than that expected. For example, a short-chain ester would probably elute in LC fraction 5 or 6, whereas a long-chain ester would elute in Fractions 3 or 4.

†Reference D-8.

REFERENCES

- D-1 American Public Health Association, American Water Works Association, and Water Pollution Control Federation, Standard Methods for the Examination of Water and Wastewater, 14th edition. Washington, D.C., American Public Health Association, 1975.
- D-2 Luthy, R.G., Manual of Methods: Preservation and Analysis of Coal Gasification Wastewaters, Environmental Studies Institute, Carnegie-Mellon University, Pittsburgh, Pennsylvania, 1977.
- D-3 Baker, R.L., "Determination of Fluoride in Vegetation Using the Specific Ion Electrode," Anal. Chem. 44(7), 1326, (1972).
- D-4 Baumann, Elizabeth W., "Trace Fluoride Determination with Specific Ion Electrode," Anal. Chem. Acta 42, 127-132, (1968).
- D-5 Thomas, Josephus, Jr. and Harold J. Gluskoter, "Determination of Fluoride in Coal with the Fluoride Ion-Selective Electrode," Anal. Chem. 46(9), 1321, (1974).
- D-6 American Society for Testing and Materials, 1974 Annual Book of ASTM Standards Part 31: Water, Philadelphia, Pennsylvania, 1977.
- D-7 Code of Federal Regulations, 40; Protection of the Environmental Parts 53 to 80, Revised 1 July 1980.
- D-8 Lentzen, D.E., D.E. Waggoner, E.D. Estes, and W.F. Gutknecht, IERL-RTP Procedures Manual: Level 1 Environmental Assessment (Second Edition), NTIS No. Pb-293 795, Research Triangle Institute, Research Triangle Park, North Carolina, October 1978. (EPA-600/7-78-201)
- D-9 U.S. EPA, Methods for Chemical Analysis of Water and Wastewater, EPA 600/4-79 020, March 1979.
- D-10 American Society for Testing and Materials, 1977 Annual Book of ASTM Standards Part 26, Gaseous Fuels: Coal and Coke: Atmospheric Analysis, Philadelphia, Pennsylvania, 1977.
- D-11 Resource Conservation and Recovery Act, Public Law 95-580.

APPENDIX E
PROCESS DATA

Appendix E presents process data for the Foster Wheeler/Stoic gasifier at UMD. The gasifier hourly log from February 23, 1981 through March 3, 1981 is presented in Table E-1. Table E-2 presents the hourly log for the boilers. The flow rate, molecular weight, and moisture content of the product and combustion emission streams are presented in Table E-3.

Figures E-1 through E-8 present the results of continuous SO_x and NO_x monitoring by UMD personnel for the period 2/24/81 through 3/3/81.

NOTES ON DULUTH GASIFIER HOURLY LOG

1. CALCULATION OF FLOW RATES

Steam/FT21A, Steam/FT21B read the same flow; readings are high and low range meters, respectively, and are a measure of the jacket steam flow to the blast air.

For FT21A, divide the reading by 100, determine the square root, and multiply by 4,000 to obtain indicated steam flow in lbs/hr.

For FT21B, divide the reading by 100, determine the square root, multiply by 971 to obtain indicated steam flow in lbs/hr.

AIR TO VEN. (air to vent). Multiply reading by 970 (blast air flow) to obtain flow rate in lbs/hr.

Water TO Jack. (water to jacket). This value is the feed water flow rate to the gasifier steam jacket in gal/min.

2. TAR

Tank Level value is percent of tank level where 0-100% = 0-32", 0" = 1 ft. in tank or 0% = 1 ft. in tank.

3. WATER

Jacket FEED. This value is the cumulative quantity of feed water to the gasifier jacket. Subtract consecutive values to obtain gallons added during a specific time period.

4. COAL

COAL. This is the cumulative number of dumps. Subtract consecutive values, multiply by the coal scale calibration factor, and on all sheets (except 2-24-81 TEST RUN 1 sheet) multiply by 0.94 to obtain lbs of coal.

5. ELECTRIC

ELECTRIC. Multiply difference by 120 to obtain KWH (cumulative).

6. TAR

TAR. Cumulative amount of tar pumped in gallons.

7. WATER

WATER. This is the total (cumulative) water to gasifier building.
Multiply the difference by 100 to obtain gallons.

8. AIR FLOW

AIR FLOW. Multiply the difference by 10 to obtain cumulative lbs air.

9. BOILER FEEDWATER TO JACKET

BFW TO JACK. (This flow totalizer reading did not change, instrument was not working.)

10. HEAT TRACE

HEAT TRACE. Multiply difference by 100 to obtain cumulative lbs of steam.

11. ESP VOLTAGE

ESP LOAD VOLTS. Multiply by 1000 to obtain ESP voltage.

12. ESP CURRENT

ESP LOAD CUR. Value is in milliamps.

BOILER HOURLY LOG

1. GAS PRESS. Gas Pressure is in inches of H₂O.
2. STEAM. Readings times 35 equals lbs of steam/hr.
3. COND. Difference in readings times 1000 equals lbs of boiler feedwater.
4. GAS. Difference in readings times 12.4 equals lbs of gas (fuel).
5. CONDENSATE. Same as COND. above.
6. (OIL) TAR. Difference in readings equals gallons.

WET ASH TOTALS

0800 hrs, 2-24-81 to 1900 hrs, 2-24-81	4144 lbs
1900 hrs, 2-24-81 to 1200 hrs, 2-25-81	6720 lbs
1200 hrs, 2-25-81 to 1050 hrs, 2-27-81	7460 lbs

CYCLONE DUST TOTALS

0800 hrs, 2-24-81 to 1900 hrs, 2-24-81	9 lbs
1900 hrs, 2-24-81 to 1200 hrs, 2-25-81	14 lbs
1200 hrs, 2-26-81 to 1500 hrs, 2-27-81	54 lbs

TABLE E-1. UMD GASIFIER HOURLY LOG

Date
2/23

DULUTH GASIFIER HOURLY LOG

TEMPERATURE °F												PRESSURE " H ₂ O												FLOW RATE				TAR		Water		COAL		COMMENTS																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																									
1	2	3	4	5	6	7	8	9	TOP BOILER (Base to Top of First Floor)	Bottom of Ventrals	Top of Ventrals	Flow to Top of Ventrals	Flow to Bottom of Ventrals	Flow to Water	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level		Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to 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Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to 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Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level	Flow to Level

(Continued)

TABLE E-1. (Continued)

TEMPERATURE °F										PRESSURE " H ₂ O		FLOW RATE		TAR		COAL		COMMENTS
1	2	3	4	5	6	7	8	9	TOP TO BOT EAS	ABOVE VEN	AIR TO VEN	TO JACK	TO LEVEL	WATER JACK	FEED			
91	196	378	112	854	259	218	117	117	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
900	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
830	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
900	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
930	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1000	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1030	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1060	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1090	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1120	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1150	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1180	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1210	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1240	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1270	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1300	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1330	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1360	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1390	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1420	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1450	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1480	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1510	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1540	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1570	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1600	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1630	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1660	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1690	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1720	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1750	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1780	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1810	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1840	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1870	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1900	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1930	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1960	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
1990	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2020	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2050	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2080	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2110	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2140	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2170	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2200	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2230	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2260	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2290	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2320	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2350	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2380	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2410	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2440	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2470	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2500	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2530	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2560	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2590	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2620	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2650	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2680	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2710	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2740	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2770	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2800	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2830	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2860	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2890	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2920	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2950	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
2980	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3010	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3040	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3070	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3100	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3130	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3160	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3190	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3220	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3250	188	367	112	852	350	216	118	125	24	30	33	24	26	22	3214.0	2151	Start Test at 2	
3280	188	367	112	852	350	216	118	125	24	30	33	24	26	22</				

(Continued)

TABLE E-1. (Continued)

[illegible]

(Continued)

TABLE E-1. (Continued)

DULUTH GASIFIER HOURLY LOG												
Date 7/25	TEMPERATURE °F					PRESSURE " H ₂ O					FLOW RATE	
	1	2	3	4	5	6	7	8	9	TOP GAS PRESSURE	DOWN TO VEN. GAS	DOWN TO FEED
81	198	386	97	976	392	214	117	101	136	24	24	36
100	198	386	97	976	392	214	117	101	136	24	24	36
200	201	386	96	915	393	226	116	104	125	24	24	36
300	201	387	95	914	395	218	115	105	125	24	24	36
400	193	364	94	914	383	201	115	107	139	24	24	36
500	195	346	92	910	346	226	115	108	134	24	24	36
600	197	387	91	912	381	207	115	108	133	24	24	36
700	196	398	91	919	412	219	114	107	133	24	24	36
800	193	443	90	911	428	201	113	107	134	24	24	36
900	193	418	89	914	432	217	113	107	134	24	24	36
1000	192	421	88	915	433	208	113	107	133	24	24	36
1100	191	422	87	918	433	196	114	106	133	24	24	36
1200	189	424	86	914	435	202	117	106	134	24	24	36
1300	186	377	86	910	357	210	117	106	134	24	24	36
1400	188	412	85	914	435	197	117	105	133	24	24	36
1500	184	385	85	910	406	219	117	105	133	24	24	36
1600	183	402	84	916	414	207	116	105	133	24	24	36
1700	185	383	84	920	365	205	114	105	133	24	24	36
1800	192	470	83	930	445	196	112	104	134	24	24	36
1900	192	460	83	922	370	233	112	104	133	24	24	36
2000	187	412	82	921	425	187	110	104	132	24	24	36
2100	183	415	82	924	418	191	109	103	132	24	24	36
2200	183	415	82	924	418	191	109	103	132	24	24	36
2300	183	415	82	924	418	191	109	103	132	24	24	36
2400	183	415	82	924	418	191	109	103	132	24	24	36
2500	183	415	82	924	418	191	109	103	132	24	24	36
2600	183	415	82	924	418	191	109	103	132	24	24	36
2700	183	415	82	924	418	191	109	103	132	24	24	36
2800	183	415	82	924	418	191	109	103	132	24	24	36
2900	183	415	82	924	418	191	109	103	132	24	24	36
3000	183	415	82	924	418	191	109	103	132	24	24	36
3100	183	415	82	924	418	191	109	103	132	24	24	36
3200	183	415	82	924	418	191	109	103	132	24	24	36
3300	183	415	82	924	418	191	109	103	132	24	24	36
3400	183	415	82	924	418	191	109	103	132	24	24	36
3500	183	415	82	924	418	191	109	103	132	24	24	36
3600	183	415	82	924	418	191	109	103	132	24	24	36
3700	183	415	82	924	418	191	109	103	132	24	24	36
3800	183	415	82	924	418	191	109	103	132	24	24	36
3900	183	415	82	924	418	191	109	103	132	24	24	36
4000	183	415	82	924	418	191	109	103	132	24	24	36
4100	183	415	82	924	418	191	109	103	132	24	24	36
4200	183	415	82	924	418	191	109	103	132	24	24	36
4300	183	415	82	924	418	191	109	103	132	24	24	36
4400	183	415	82	924	418	191	109	103	132	24	24	36
4500	183	415	82	924	418	191	109	103	132	24	24	36
4600	183	415	82	924	418	191	109	103	132	24	24	36
4700	183	415	82	924	418	191	109	103	132	24	24	36
4800	183	415	82	924	418	191	109	103	132	24	24	36
4900	183	415	82	924	418	191	109	103	132	24	24	36
5000	183	415	82	924	418	191	109	103	132	24	24	36
5100	183	415	82	924	418	191	109	103	132	24	24	36
5200	183	415	82	924	418	191	109	103	132	24	24	36
5300	183	415	82	924	418	191	109	103	132	24	24	36
5400	183	415	82	924	418	191	109	103	132	24	24	36
5500	183	415	82	924	418	191	109	103	132	24	24	36
5600	183	415	82	924	418	191	109	103	132	24	24	36
5700	183	415	82	924	418	191	109	103	132	24	24	36
5800	183	415	82	924	418	191	109	103	132	24	24	36
5900	183	415	82	924	418	191	109	103	132	24	24	36
6000	183	415	82	924	418	191	109	103	132	24	24	36
6100	183	415	82	924	418	191	109	103	132	24	24	36
6200	183	415	82	924	418	191	109	103	132	24	24	36
6300	183	415	82	924	418	191	109	103	132	24	24	36
6400	183	415	82	924	418	191	109	103	132	24	24	36
6500	183	415	82	924	418	191	109	103	132	24	24	36
6600	183	415	82	924	418	191	109	103	132	24	24	36
6700	183	415	82	924	418	191	109	103	132	24	24	36
6800	183	415	82	924	418	191	109	103	132	24	24	36
6900	183	415	82	924	418	191	109	103	132	24	24	36
7000	183	415	82	924	418	191	109	103	132	24	24	36
7100	183	415	82	924	418	191	109	103	132	24	24	36
7200	183	415	82	924	418	191	109	103	132	24	24	36
7300	183	415	82	924	418	191	109	103	132	24	24	36
7400	183	415	82	924	418	191	109	103	132	24	24	36
7500	183	415	82	924	418	191	109	103	132	24	24	36
7600	183	415	82	924	418	191	109	103	132	24	24	36
7700	183	415	82	924	418	191	109	103	132	24	24	36
7800	183	415	82	924	418	191	109	103	132	24	24	36
7900	183	415	82	924	418	191	109	103	132	24	24	36
8000	183	415	82	924	418	191	109	103	132	24	24	36
8100	183	415	82	924	418	191	109	103	132	24	24	36
8200	183	415	82	924	418	191	109	103	132	24	24	36
8300	183	415	82	924	418	191	109	103	132	24	24	36
8400	183	415	82	924	418	191	109	103	132	24	24	36
8500	183	415	82	924	418	191	109	103	132	24	24	36
8600	183	415	82	924	418	191	109	103	132	24	24	36
8700	183	415	82	924	418	191	109	103	132	24	24	36
8800	183	415	82	924	418	191	109	103	132	24	24	36
8900	183	415	82	924	418	191	109	103	132	24	24	36
9000	183	415	82	924	418	191	109	103	132	24	24	36
9100	183	415	82	924	418	191	109	103	132	24	24	36
9200	183	415	82	924	418	191	109	103	132	24	24	36
9300	183	415	82	924	418	191	109	103	132	24	24	36
9400	183	415	82	924	418	191	109	103	132	24	24	36
9500	183	415	82	924	418	191	109	103	132	24	24	36
9600	183	415	82	924	418	191	109	103	132	24	24	36
9700	183	415	82	924	418	191	109	103	132	24	24	36
9800	183	415	82	924	418	191	109	103	132	24	24	36
9900	183	415	82	924	418	191	109	103	132	24	24	36
10000	183	415	82	924	418	191	109	103	132	24	24	36
10100	183	415	82	924	418	191	109	103	132	24	24	36
10200	183	415	82	924	418	191	109	103	132	24	24	36
10300	183	415	82	924	418	191	109	103	132	24	24	36
10400	183	415	82	924	418	191	109	103	132	24	24	36
10500	183	415	82	924	418	191	109	103	132	24	24	36
10600	183	415	82	924	418	191	109	103	132	24	24	36
10700	183	415	82	924	418	191	109	103	132	24	24	36
10800	183	415	82	924	418	191	109	103	132	24	24	36
10900	183	415	82	924	418	191	109	103	132	24	24	36
11000	183	415	82	924	418	191	109	103	132	24	24	36
11100	183	415	82									

TABLE E-1. (Continued)

DULUTH GASIFIER HOURLY LOG											
TEMPERATURE °F						PRESSURE " H ₂ O			FLOW RATE		
1	2	3	4	5	6	7	8	9	TOP GAS	MID GAS	BOTTOM GAS
100	110	120	130	140	150	160	170	180	TO JACH	TO JACH	TO JACH
190	200	210	220	230	240	250	260	270	TO JACH	TO JACH	TO JACH
280	290	300	310	320	330	340	350	360	TO JACH	TO JACH	TO JACH
370	380	390	400	410	420	430	440	450	TO JACH	TO JACH	TO JACH
460	470	480	490	500	510	520	530	540	TO JACH	TO JACH	TO JACH
550	560	570	580	590	600	610	620	630	TO JACH	TO JACH	TO JACH
640	650	660	670	680	690	700	710	720	TO JACH	TO JACH	TO JACH
730	740	750	760	770	780	790	800	810	TO JACH	TO JACH	TO JACH
820	830	840	850	860	870	880	890	900	TO JACH	TO JACH	TO JACH
910	920	930	940	950	960	970	980	990	TO JACH	TO JACH	TO JACH
1000	1010	1020	1030	1040	1050	1060	1070	1080	TO JACH	TO JACH	TO JACH
1090	1100	1110	1120	1130	1140	1150	1160	1170	TO JACH	TO JACH	TO JACH
1180	1190	1200	1210	1220	1230	1240	1250	1260	TO JACH	TO JACH	TO JACH
1270	1280	1290	1300	1310	1320	1330	1340	1350	TO JACH	TO JACH	TO JACH
1360	1370	1380	1390	1400	1410	1420	1430	1440	TO JACH	TO JACH	TO JACH
1450	1460	1470	1480	1490	1500	1510	1520	1530	TO JACH	TO JACH	TO JACH
1540	1550	1560	1570	1580	1590	1600	1610	1620	TO JACH	TO JACH	TO JACH
1630	1640	1650	1660	1670	1680	1690	1700	1710	TO JACH	TO JACH	TO JACH
1720	1730	1740	1750	1760	1770	1780	1790	1800	TO JACH	TO JACH	TO JACH
1810	1820	1830	1840	1850	1860	1870	1880	1890	TO JACH	TO JACH	TO JACH
1900	1910	1920	1930	1940	1950	1960	1970	1980	TO JACH	TO JACH	TO JACH
1990	2000	2010	2020	2030	2040	2050	2060	2070	TO JACH	TO JACH	TO JACH
2080	2090	2100	2110	2120	2130	2140	2150	2160	TO JACH	TO JACH	TO JACH
2170	2180	2190	2200	2210	2220	2230	2240	2250	TO JACH	TO JACH	TO JACH
2260	2270	2280	2290	2300	2310	2320	2330	2340	TO JACH	TO JACH	TO JACH
2350	2360	2370	2380	2390	2400	2410	2420	2430	TO JACH	TO JACH	TO JACH
2440	2450	2460	2470	2480	2490	2500	2510	2520	TO JACH	TO JACH	TO JACH
2530	2540	2550	2560	2570	2580	2590	2600	2610	TO JACH	TO JACH	TO JACH
2620	2630	2640	2650	2660	2670	2680	2690	2700	TO JACH	TO JACH	TO JACH
2710	2720	2730	2740	2750	2760	2770	2780	2790	TO JACH	TO JACH	TO JACH
2800	2810	2820	2830	2840	2850	2860	2870	2880	TO JACH	TO JACH	TO JACH
2890	2900	2910	2920	2930	2940	2950	2960	2970	TO JACH	TO JACH	TO JACH
2980	2990	3000	3010	3020	3030	3040	3050	3060	TO JACH	TO JACH	TO JACH
3070	3080	3090	3100	3110	3120	3130	3140	3150	TO JACH	TO JACH	TO JACH
3160	3170	3180	3190	3200	3210	3220	3230	3240	TO JACH	TO JACH	TO JACH
3250	3260	3270	3280	3290	3300	3310	3320	3330	TO JACH	TO JACH	TO JACH
3340	3350	3360	3370	3380	3390	3400	3410	3420	TO JACH	TO JACH	TO JACH
3430	3440	3450	3460	3470	3480	3490	3500	3510	TO JACH	TO JACH	TO JACH
3520	3530	3540	3550	3560	3570	3580	3590	3600	TO JACH	TO JACH	TO JACH
3610	3620	3630	3640	3650	3660	3670	3680	3690	TO JACH	TO JACH	TO JACH
3700	3710	3720	3730	3740	3750	3760	3770	3780	TO JACH	TO JACH	TO JACH
3790	3800	3810	3820	3830	3840	3850	3860	3870	TO JACH	TO JACH	TO JACH
3880	3890	3900	3910	3920	3930	3940	3950	3960	TO JACH	TO JACH	TO JACH
3970	3980	3990	4000	4010	4020	4030	4040	4050	TO JACH	TO JACH	TO JACH
4060	4070	4080	4090	4100	4110	4120	4130	4140	TO JACH	TO JACH	TO JACH
4150	4160	4170	4180	4190	4200	4210	4220	4230	TO JACH	TO JACH	TO JACH
4240	4250	4260	4270	4280	4290	4300	4310	4320	TO JACH	TO JACH	TO JACH
4330	4340	4350	4360	4370	4380	4390	4400	4410	TO JACH	TO JACH	TO JACH
4420	4430	4440	4450	4460	4470	4480	4490	4500	TO JACH	TO JACH	TO JACH
4510	4520	4530	4540	4550	4560	4570	4580	4590	TO JACH	TO JACH	TO JACH
4600	4610	4620	4630	4640	4650	4660	4670	4680	TO JACH	TO JACH	TO JACH
4690	4700	4710	4720	4730	4740	4750	4760	4770	TO JACH	TO JACH	TO JACH
4780	4790	4800	4810	4820	4830	4840	4850	4860	TO JACH	TO JACH	TO JACH
4870	4880	4890	4900	4910	4920	4930	4940	4950	TO JACH	TO JACH	TO JACH
4960	4970	4980	4990	5000	5010	5020	5030	5040	TO JACH	TO JACH	TO JACH
5050	5060	5070	5080	5090	5100	5110	5120	5130	TO JACH	TO JACH	TO JACH
5140	5150	5160	5170	5180	5190	5200	5210	5220	TO JACH	TO JACH	TO JACH
5230	5240	5250	5260	5270	5280	5290	5300	5310	TO JACH	TO JACH	TO JACH
5320	5330	5340	5350	5360	5370	5380	5390	5400	TO JACH	TO JACH	TO JACH
5410	5420	5430	5440	5450	5460	5470	5480	5490	TO JACH	TO JACH	TO JACH
5500	5510	5520	5530	5540	5550	5560	5570	5580	TO JACH	TO JACH	TO JACH
5590	5600	5610	5620	5630	5640	5650	5660	5670	TO JACH	TO JACH	TO JACH
5680	5690	5700	5710	5720	5730	5740	5750	5760	TO JACH	TO JACH	TO JACH
5770	5780	5790	5800	5810	5820	5830	5840	5850	TO JACH	TO JACH	TO JACH
5860	5870	5880	5890	5900	5910	5920	5930	5940	TO JACH	TO JACH	TO JACH
5950	5960	5970	5980	5990	6000	6010	6020	6030	TO JACH	TO JACH	TO JACH
6040	6050	6060	6070	6080	6090	6100	6110	6120	TO JACH	TO JACH	TO JACH
6130	6140	6150	6160	6170	6180	6190	6200	6210	TO JACH	TO JACH	TO JACH
6220	6230	6240	6250	6260	6270	6280	6290	6300	TO JACH	TO JACH	TO JACH
6310	6320	6330	6340	6350	6360	6370	6380	6390	TO JACH	TO JACH	TO JACH
6400	6410	6420	6430	6440	6450	6460	6470	6480	TO JACH	TO JACH	TO JACH
6490	6500	6510	6520	6530	6540	6550	6560	6570	TO JACH	TO JACH	TO JACH
6580	6590	6600	6610	6620	6630	6640	6650	6660	TO JACH	TO JACH	TO JACH
6670	6680	6690	6700	6710	6720	6730	6740	6750	TO JACH	TO JACH	TO JACH
6760	6770	6780	6790	6800	6810	6820	6830	6840	TO JACH	TO JACH	TO JACH
6850	6860	6870	6880	6890	6900	6910	6920	6930	TO JACH	TO JACH	TO JACH
6940	6950	6960	6970	6980	6990	7000	7010	7020	TO JACH	TO JACH	TO JACH
7030	7040	7050	7060	7070	7080	7090	7100	7110	TO JACH	TO JACH	TO JACH
7120	7130	7140	7150	7160	7170	7180	7190	7200	TO JACH	TO JACH	TO JACH
7210	7220	7230	7240	7250	7260	7270	7280	7290	TO JACH	TO JACH	TO JACH
7300	7310	7320	7330	7340	7350	7360	7370	7380	TO JACH	TO JACH	TO JACH
7390	7400	7410	7420	7430	7440	7450	7460	7470	TO JACH	TO JACH	TO JACH
7480	7490	7500	7510	7520	7530	7540	7550	7560	TO JACH	TO JACH	TO JACH
7570	7580	7590	7600	7610	7620	7630	7640	7650	TO JACH	TO JACH	TO JACH
7660	7670	7680	7690	7700	7710	7720	7730	7740	TO JACH	TO JACH	TO JACH
7750	7760	7770	7780	7790	7800	7810	7820	7830	TO JACH	TO JACH	TO JACH
7840	7850	7860	7870	7880	7890	7900	7910	7920	TO JACH	TO JACH	TO JACH
7930	7940	7950	7960	7970	7980	7990	8000	8010	TO JACH	TO JACH	TO JACH
8020	8030	8040	8050	8060	8070	8080	8090	8100	TO JACH	TO JACH	TO JACH
8110	8120	8130	8140	8150	8160	8170	8180	8190	TO JACH	TO JACH	TO JACH
8200	8210	8220	8230	8240	8250	8260	8270	8280	TO JACH	TO JACH	TO JACH
8290	8300	8310	8320	8330	8340	8350	8360	8370	TO JACH	TO JACH	TO JACH
8380	8390	8400	8410	8420	8430	8440	8450	8460	TO JACH	TO JACH	TO JACH
8470	8480	8490	8500	8510	8520	8530	8540	8550	TO JACH	TO JACH	TO JACH
8560	8570	8580	8590	8600	8610	8620	8630	8640	TO JACH	TO JACH	TO JACH
8650	8660	8670	8680	8690	8700	8710	8720	8730	TO JACH	TO JACH	TO JACH
8740	8750	8760	8770	8780	8790	8800	8810	8820	TO JACH	TO JACH	TO JACH
8830	8840	8850	8860	8870	8880	8890	8900	8910	TO JACH	TO JACH	TO JACH
8920	8930	8940	8950	8960	8970	8980	8990	9000	TO JACH	TO JACH	TO JACH
9010	9020	9030	9040	9050	9060	9070	9080	9090	TO JACH	TO JACH	TO JACH
9100	9110	9120	9130	9140	9150	9160	9170	9180	TO JACH	TO JACH	TO JACH
9190	9200	9210	9220	9230	9240	9250	9260	9270	TO JACH	TO JACH	TO JACH
9280	9290	9300	9310	9320	9330	9340	9350	9360	TO JACH	TO JACH	TO JACH
9370	9380	9390	9400	9410	9420	9430	9440	9450	TO JACH	TO JACH	TO JACH
9460	9470	9480	9490	9500	9510	9520	9530	9540	TO JACH	TO JACH	TO JACH
9550	9560	9570	9580	9590	9600	9610	9620	9630	TO JACH	TO JACH	TO JACH
9640	9650	9660	9670	9680	9690	9700	9710	9720	TO JACH	TO JACH	TO JACH
9730	9740	9750	9760	9770	9780	9790	9800	9810	TO JACH	TO JACH	TO JACH
9820	9830	9840	9850	9860	9870	9880	9890				

TABLE E-1. (Continued)

DULUTH GASIFIER HOURLY LOG												COMMENTS													
TEMPERATURE °F												PRESSURE " H ₂ O		FLOW RATE		TAR		COAL		COMMENTS					
1 2 3 4 5 6 7 8 9												TOP BOT GAS AIR TO VENT GAS		Inlet Steam To Feed		Inlet Steam To Feed		Inlet Steam To Feed		Inlet Steam To Feed					
81	100	118	510	75	999	524	112	91	136	24	30	32	36	25	36	1.8	34	5.3	2.6	10	OFF	14247	308422	Gst 136 16	
200	191	498	75	1002	532	112	91	136	24	30	32	34	34	22	30	4.3	34	5.3	2.6	10	OFF	14247	308460		
300	191	481	75	1001	522	112	91	136	24	30	32	34	34	22	30	4.3	34	5.3	2.6	10	OFF	14247	308465	Grate A 6.5 3.00	
400	191	471	75	993	510	112	100	136	24	27	32	32	34	22	30	4.3	34	5.3	2.6	10	OFF	14247	308465		
500	191	483	75	991	499	111	100	136	24	27	32	32	34	22	30	4.3	34	5.3	2.6	10	OFF	14247	308465	1	
600	190	480	75	985	482	110	99	135	24	28	34	35	34	22	30	4.3	34	5.3	2.6	10	OFF	14247	308465		
700	188	526	74	1017	538	115	110	90	125	24	24	29	30	24	33	2.4	40	5.3	2.2	6	OFF	14247	308465	Grate A 6.5 8.40	
800	192	440	74	972	398	118	118	93	136	24	26	31	34	36	24	38	1.5	30	5.2	2.8	9	OFF	14247		308465
900	191	447	75	992	521	117	114	93	136	24	26	31	34	36	24	38	1.5	30	5.2	2.8	9	OFF	14247	308465	Grate A 6.2 10.00
1000	193	422	75	988	449	112	112	94	125	24	28	32	34	34	38	2.4	41	5.3	2.4	17	OFF	14247	308465		
1100	196	447	75	986	462	113	113	97	136	24	28	32	34	34	38	2.4	41	5.3	2.4	17	OFF	14247	308465	Grate A 6.5 12.15 Carl scale 4074	
1200	194	453	75	986	466	112	112	98	136	24	28	32	34	34	38	2.4	41	5.3	2.4	17	OFF	14247	308465		
1300	192	454	75	984	464	111	111	99	136	24	28	32	34	34	38	2.4	41	5.3	2.4	17	OFF	14247	308465	Grate A 6.2 13.45	
1400	190	428	76	965	401	113	123	101	134	24	32	36	38	24	40	2.3	41	5.2	2.2	8	OFF	14247	308465		
1500	191	428	76	965	401	113	123	101	134	24	32	36	38	24	40	2.3	41	5.2	2.2	8	OFF	14247	308465	Grate A 6.5 2.10	
1600	193	446	76	996	534	116	116	103	136	24	26	30	32	34	36	1.5	34	5.1	1.8	10	OFF	14247	308465		
1700	192	489	76	990	516	115	115	104	136	24	26	32	32	34	34	2.1	39	5.1	1.8	12	OFF	14247	308465		
1800	195	484	77	996	502	115	115	105	136	24	26	32	32	34	34	1.7	39	5.1	1.8	15	OFF	14247	308465		
1900	190	447	78	979	485	115	116	106	135	24	30	34	36	24	38	2.1	44	5.1	2.0	16	OFF	14247	308465		
2000	199	457	78	975	485	117	117	107	136	24	26	32	32	34	36	2.1	44	5.1	2.0	16	OFF	14247	308465		
2100	195	456	79	991	512	117	117	107	135	24	26	32	32	34	38	2.3	45	5.0	2.0	0	OFF	14247	308465		
2200	193	434	79	986	454	118	118	108	135	24	30	32	34	34	36	2.1	33	5.2	2.2	0	OFF	14247	308465		
2300	196	447	80	988	458	114	118	108	136	24	28	32	34	34	36	2.1	33	5.2	2.2	0	OFF	14247	308465	A 434 Bucket Trouble 35 min.	
2400	196	448	80	986	457	118	118	108	136	24	28	32	34	34	36	2.1	33	5.2	2.2	0	OFF	14247	308465		
GAS ANALYSIS												1 PERCENT				1 PERCENT				1 PERCENT					
ELECTRIC												1200 HRS				2400 HRS				1200 HRS					
TAR												ESP LOAD VARS				ESP LOAD CUR				ESP SPARK RATE					
WATER												1 TBI TOP GAS AFTER ESP				2 MIXED GAS, BOILER 1				3 MIXED GAS, BOILER 2					
AIR FLOW												4 T64 BOTTOM GAS				5 T66 MIXED GAS				6 T68 TOP GAS BEFORE ESP					
BFW TO JACK												7 T68 BLOWER AIR TEMP				8 T68 ASH PAN				9 T70 BST					
HEAT TRACE												1 TBI TOP GAS AFTER ESP				2 MIXED GAS, BOILER 1				3 MIXED GAS, BOILER 2					
COAL SCALE CALIBRATION												156 lbs/dump = 308723				156 lbs/dump = 308723				156 lbs/dump = 308723					
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TABLE E-1. (Continued)

Date 2/28		TEMPERATURE °F										PRESSURE " H ₂ O		FLOW R/H		TAR		COAL		COMMENTS									
		1	2	3	4	5	6	7	8	9	TOP BOT GAS	AIR W/ TO VEN	Flow Rate	Flow Rate	Wt %	Grain	Grain	Grain											
81	100	182	448	81	965	451	182	119	110	137	24	28	21	34	25	36	3.6	41.1	5.3	2.4	0	OFF	14118	30197					
200	183	447	81	963	457	189	118	110	137	24	28	32	34	24	36	2.3	40.9	5.2	1.2	1	OFF	14312	30198						
300	183	448	81	964	458	197	122	110	135	24	28	24	34	24	36	1.8	35.6	5.2	1.3	4	OFF	14381	30198						
400	183	449	82	966	459	197	126	111	130	24	28	24	34	24	36	1.4	31.6	5.5	1.4	6	OFF	14397	30199						
500	182	448	82	957	458	188	122	112	132	24	28	32	34	24	36	1.3	31.2	5.4	4.0	10	OFF	14388	30198						
600	182	448	83	959	458	203	120	113	132	24	28	32	34	24	36	1.4	31.6	5.5	2.0	12	OFF	14367	30196						
700	182	448	84	965	458	197	120	113	132	24	28	32	34	24	36	1.7	32.1	5.3	1.7	15	OFF	14197	30193						
800	182	448	84	970	458	184	119	113	136	24	28	32	34	24	36	1.6	31.0	5.2	1.8	18	OFF	14187	30191						
900	185	469	85	983	540	186	118	102	140	24	25	29	30	24	34	2.4	46.3	5.0	1.6	21	OFF	14198	30137						
1000	170	447	85	980	465	168	120	108	140	24	28	32	32	24	36	2.1	37.1	5.0	1.8	25	OFF	14182	30134						
1100	167	450	85	983	441	174	122	109	137	24	28	32	33	24	36	2.0	37.1	5.2	0	0	OFF	14187	30191						
1200	167	450	86	970	420	181	121	111	142	24	30	34	36	24	38	2.2	37.9	5.0	0	0	OFF	14187	30191						
1300	168	525	86	995	575	163	121	113	135	24	25	30	30	24	34	2.4	46.3	5.3	5.2	0	OFF	14213	30124						
1400	164	550	86	995	575	163	121	113	135	24	25	30	30	24	34	2.4	46.3	5.3	5.2	0	OFF	14213	30124						
1500	164	550	87	995	575	163	121	113	135	24	25	30	30	24	34	2.4	46.3	5.3	5.2	0	OFF	14213	30124						
1600	164	550	87	995	575	163	121	113	135	24	25	30	30	24	34	2.4	46.3	5.3	5.2	0	OFF	14213	30124						
1700	171	484	87	967	502	175	120	113	141	24	26	30	32	24	34	3.0	53.0	4.7	9.0	10	OFF	14161	30123						
1800	168	500	87	969	509	175	120	114	136	24	26	30	32	24	34	2.1	48.8	5.0	0	12	OFF	14520	30139						
1900	166	502	88	964	506	168	120	114	136	24	26	30	32	24	34	2.0	36.1	5.0	1.0	14	OFF	14516	30134						
2000	164	502	87	960	505	176	120	115	136	24	26	30	32	24	34	2.0	36.1	5.2	1.4	16	OFF	14176	30119						
2100	168	428	88	981	406	194	119	115	132	24	28	32	36	24	36	2.5	46.3	5.2	1.6	20	OFF	14553	30153						
2200	175	446	88	965	468	194	116	115	135	24	26	30	34	24	36	2.3	41.0	5.2	2.8	0	OFF	14597	30152						
2300	178	447	88	926	398	206	113	115	135	24	29	33	34	24	36	2.3	41.0	5.2	2.8	0	OFF	14597	30152						
2400	181	473	88	963	592	188	113	114	135	24	24	30	30	24	33	2.3	43.9	5.2	3.0	0	OFF	14592	30162						
TEMPERATURES																				GAS ANALYSIS									
1200 HRS																				1 TBI TOP GAS AFTER ESP									
2400 HRS																				2 MIXED GAS, CHILER									
1400 HRS																				3 MIXED GAS, CHILER									
1600 HRS																				4 T64 BOTTOM GAS									
1800 HRS																				5 T66 MIXED GAS									
2000 HRS																				6 T65 TOP GAS BEFORE ESP									
2200 HRS																				7 T66 BLOWER AIR TEMP									
2400 HRS																				8 T69 ASH PAN									
2600 HRS																				9 T70 BST									
ELECTRIC																				ESP LOAD VARS									
TAR																				ESP LOAD Cw									
WATER																				ESP SPARK RATE									
AIR FLOW																													
BFW TO JACK																													
HEAT TRACE																													
COAL SCALE CALIBRATION																													
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TABLE E-1. (Continued)

DULUTH GASIFIER HOURLY LOG												COMMENTS			
TEMPERATURE °F						PRESSURE " H ₂ O		FLOW R.U.E		TAR % PUMP ON/OFF	WATER % PUMP ON/OFF		COAL		
1	2	3	4	5	6	7	8	9	10					11	12
81	179	444	88	931	418	206	112	132	112	132	4	OFF	11697	309781	A-6.5 A-6.0 A-6.8 at 10:50 A-6.0 at 11:00
100	178	480	88	958	505	172	112	131	111	131	6	OFF	141085	309799	
200	183	420	88	943	442	265	112	108	132	112	8	OFF	141219	309748	
300	182	439	88	942	453	187	112	108	132	112	12	OFF	141447	309735	
400	181	442	87	943	453	185	112	107	132	112	14	OFF	141586	309752	
500	180	443	87	944	453	201	111	105	132	111	12	ON	141739	309768	
600	180	443	87	944	453	201	111	105	132	111	12	ON	141739	309768	
700	183	443	87	950	465	207	110	96	131	110	3	OFF	141918	309777	
800	184	445	87	955	456	204	115	97	131	115	6	OFF	141965	309761	
900	184	445	87	955	456	204	115	97	131	115	6	OFF	142285	309761	
1000	180	448	87	948	580	165	121	102	133	121	8	OFF	142326	309735	
1100	177	526	86	965	579	157	122	103	132	122	12	OFF	142555	309744	
1200	171	523	87	970	555	168	124	104	133	124	15	OFF	142849	309760	
1300	167	533	87	943	533	161	124	104	133	124	15	OFF	143231	309752	
1400	164	533	87	975	530	161	124	104	133	124	16	OFF	143195	309756	
1500	161	533	87	970	531	163	128	100	133	128	16	OFF	143186	309744	
1600	164	537	88	947	422	160	126	102	133	126	10	ON	143311	309765	
1700	168	429	89	946	434	183	126	102	133	126	0	OFF	143422	309748	
1800	170	427	89	946	434	183	122	103	133	126	0	OFF	143647	309745	
1900	172	426	90	948	435	193	123	103	133	126	0	OFF	143744	309767	
2000	174	426	90	947	435	193	120	103	134	126	0	OFF	143723	309777	
2100	177	427	90	960	437	200	117	104	133	126	0	OFF	150492	309748	
2200	175	425	90	973	517	176	117	104	136	126	1	OFF	150338	309767	
2300	170	479	91	938	506	153	115	104	133	126	2	OFF	150423	309777	
2400	165	478	91	923	502	153	115	104	132	126	3	OFF	150566	309726	
GAS ANALYSIS															
1 TBI TOP GAS AFTER ESP												H ₂			
2 MIXED GAS, BOILER												H ₂			
3 MIXED GAS, BOILER												H ₂			
4 T64 BOTTOM GAS												H ₂			
5 T66 MIXED GAS												H ₂			
6 T65 TOP GAS BEFORE ESP												H ₂			
7 T68 BLOWER AIR TEMP												CO			
8 T69 ASH PILE												CH ₄			
9 T70 BST												CO ₂			
												H ₂ O			
COAL SCALE CALIBRATION 163 164449 309772 309773															
ELECTRIC 1446 1501															
TAR 302.654 302.65															
WATER 60.156 60.263															
AIR FLOW 614487 619444															
BFW TO JACK 999291															
HEAT TRACE 2651 2659															

(Continued)

TABLE E-1. (Continued)

DULUTH GASIFIER HOUR Y LOG													COMMENTS									
TEMPERATURE °F													PRESSURE " H ₂ O		FLOW RATE		TAR		WATER		COAL	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
1200 HRS.													1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.		1200 HRS.	
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TABLE E-1. (Continued)

[illegible]

(Continued)

TABLE E-1. (Continued)

[illegible]

(Continued)

TABLE E-1. (Continued)

DIALUTH GASIFIER HOURLY LOG										COMMENTS									
TEMPERATURE °F										COAL									
PRESSURE " H ₂ O										TAR									
FLOW RATE										WATER									
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TABLE E-2. UMD BOILER HOURLY LOG

DATE
7 February
23, 1981

DULUTH BOILER HOURLY LOG

BOILER NO. 1 DATA											BOILER NO. 2 DATA					BOILER NO. 3 DATA (CONT.)					COMMENTS (ON LINE TIME, OFF LINE TIME, ETC.)
GAS PRESS.	STEAM	COND.	GAS	OIL	GAS PRESS.	STEAM	COND.	GAS	OIL	STEAM	COND.	GAS	OIL	STEAM	COND.	GAS	OIL	STEAM	COND.	GAS	
100	20	307288	92437	329814																	
200	20	307180	92444	33044																	
300	19	30833	92460	331055																	
400	19	308778	92477	331625																	
500	20	309204	92494	332272																	
600	20	309699	92508	332802																	
700	19	310205	92525	333461																	
800	19	310638	92541	334048																	
900	20	310759	92553	334598																	
1000	20	311183	92558	334950																	
1100	20	311615	92571	335097																	
1200	20	311936	92582	335322																	
1300	20	312402	92596	336136																	
1400	19	312503	92606	336458																	
1500	20	312382	92615	336940																	
1600	20	312822	92640	337891																	
1700	20	313042	92651	338691																	
1800	20	313746	92672	339221																	
1900	20	314203	92685	339541																	
2000	20	314740	92705	340176																	
2100	20	315027	92722	340621																	
2200	20	315653	92731	341146																	
2300	19	316171	92744	342333																	
2400	19	317640	92770	343450																	

4/3/79

(Continued)

TABLE E-2. (Continued)

DULUTH BOILER HOURLY LOG													
TIME	BOILER NO. 1 DATA				BOILER NO. 2 DATA				BOILER NO. 3 DATA (ON LINE)				COMMENTS (ON LINE TIME, OFF LINE TIME, ETC.)
	GAS PRESS.	STEAM	COND.	GAS	OIL	GAS PRESS.	STEAM	COND.	GAS	OIL	STEAM	CONDENSATE	OIL TARI
1400	19	31811	32765	34356									
1500	19	31856	32802	34417									
1600	19	31909	32819	34480									
1700	19	31953	32835	34543									
1800	19	32006	32852	34603									
1900	19	32053	32868	34663									
2000	19	32100	32885	34723									
2100	19	32147	32901	34783									
2200	19	32194	32917	34843									
2300	19	32240	32935	34903									
2400	19	32287	32954	34963									
2500	19	32334	32973	35024									
2600	19	32381	32991	35084									
2700	19	32428	33010	35144									
2800	19	32475	33029	35204									
2900	19	32522	33048	35264									
3000	19	32569	33067	35324									
3100	19	32616	33086	35384									
3200	19	32663	33105	35444									
3300	19	32710	33124	35504									
3400	19	32757	33143	35564									
3500	19	32804	33162	35624									
3600	19	32851	33181	35684									
3700	19	32898	33200	35744									
3800	19	32945	33219	35804									
3900	19	32992	33238	35864									
4000	19	33039	33257	35924									
4100	19	33086	33276	35984									
4200	19	33133	33295	36044									
4300	19	33180	33314	36104									
4400	19	33227	33333	36164									
4500	19	33274	33352	36224									
4600	19	33321	33371	36284									
4700	19	33368	33390	36344									
4800	19	33415	33409	36404									
4900	19	33462	33428	36464									
5000	19	33509	33447	36524									
5100	19	33556	33466	36584									
5200	19	33603	33485	36644									
5300	19	33650	33504	36704									
5400	19	33697	33523	36764									
5500	19	33744	33542	36824									
5600	19	33791	33561	36884									
5700	19	33838	33580	36944									
5800	19	33885	33599	37004									
5900	19	33932	33618	37064									
6000	19	33979	33637	37124									
6100	19	34026	33656	37184									
6200	19	34073	33675	37244									
6300	19	34120	33694	37304									
6400	19	34167	33713	37364									
6500	19	34214	33732	37424									
6600	19	34261	33751	37484									
6700	19	34308	33770	37544									
6800	19	34355	33789	37604									
6900	19	34402	33808	37664									
7000	19	34449	33827	37724									
7100	19	34496	33846	37784									
7200	19	34543	33865	37844									
7300	19	34590	33884	37904									
7400	19	34637	33903	37964									
7500	19	34684	33922	38024									
7600	19	34731	33941	38084									
7700	19	34778	33960	38144									
7800	19	34825	33979	38204									
7900	19	34872	33998	38264									
8000	19	34919	34017	38324									
8100	19	34966	34036	38384									
8200	19	35013	34055	38444									
8300	19	35060	34074	38504									
8400	19	35107	34093	38564									
8500	19	35154	34112	38624									
8600	19	35201	34131	38684									
8700	19	35248	34150	38744									
8800	19	35295	34169	38804									
8900	19	35342	34188	38864									
9000	19	35389	34207	38924									
9100	19	35436	34226	38984									
9200	19	35483	34245	39044									
9300	19	35530	34264	39104									
9400	19	35577	34283	39164									
9500	19	35624	34302	39224									
9600	19	35671	34321	39284									
9700	19	35718	34340	39344									
9800	19	35765	34359	39404									
9900	19	35812	34378	39464									
10000	19	35859	34397	39524									

4/15/78

(Continued)

TABLE E-2. (Continued)

[illegible]

(Continued)

TABLE E-2. (Continued)

DULUTH BOILER HOURLY LOG

DULUTH BOILER HOURLY LOG												
BOILER NO. 1 DATA					BOILER NO. 2 DATA					BOILER NO. 3 DATA (ON ...)		
GAS PRESS.	STEAM	COND.	GAS	OIL	GAS PRESS.	STEAM	COND.	GAS	OIL	STEAM	CONDENSER	(OIL) TAR
19	341034	93559	37360									
19	341517	93576	37430									
19	342049	93593	37497									
19	342562	93611	37566									
19	343071	93628	37635									
19	343582	93645	37697									
19	344092	93663	37764									
19	344602	93680	37830									
19	345112	93697	37896									
19	345622	93715	37962									
19	346132	93732	38032									
19	346642	93749	38100									
19	347152	93766	38166									
19	347662	93783	38232									
19	348172	93800	38298									
19	348682	93817	38364									
19	349192	93834	38430									
19	349702	93851	38496									
19	350212	93868	38562									
19	350722	93885	38628									
19	351232	93902	38694									
19	351742	93919	38760									
19	352252	93936	38826									
19	352762	93953	38892									

COMMENTS

(ON LINE TIME, OFF LINE TIME, ETC.)

26, 1921

4/3/78

(Continued)

TABLE E-2. (Continued)

DULVTH BOILER HOURLY LOG

COMMENTS
(ON LINE TIME, OFF LINE TIME, ETC.)

BOILER NO. 3 DATA (UN. UNIT)

CONDENSATE

STEAM

OIL

GAS

COND.

STEAM

GAS PRESS.

OIL

GAS

COND.

STEAM

GAS PRESS.

TIME

100 19 35259 3157 31975

200 19 35319 31575 319657

300 19 35395 3189 31211

400 19 35407 4005 31190

500 19 35451 4020 31232

600 19 35505 4058 31323

700 19 35527 4007 31405

800 19 35587 4019 31466

900 19 35610 4004 31575

1000 19 35627 4112 31623

1100 19 35630 4117 31676

1200 19 35605 4132 31743

1300 19 35676 4148 31801

1400 19 35715 4148 31820

1500 19 35723 4183 31940

1600 19 35886 4204 32022

1700 19 36053 4218 32091

1800 19 36074 4238 32157

1900 19 36131 4257 32241

2000 19 36174 4264 32281

2100 19 36253 4283 32354

2200 18 36270 4301 32430

2300 18 36354 4321 32503

2400 18 36377 4356 32564

(Continued)

TABLE E-2. (Continued)

DULUTH BOILER HOURLY LOG													
TIME	BOILER NO. 1 DATA				BOILER NO. 2 DATA				BOILER NO. 3 DATA (ON LINE)				COMMENTS (ONLINE TIME, OFFLINE TIME, ETC.)
	GAS PRESS	STEAM	COND.	GAS	OIL	GAS PRESS	STEAM	COND.	GAS	OIL	STEAM	CONDENSATE	(OIL) TAR
100	18	346248	99553	96276									
200	18	346722	99388	96096									
300													
400	18	346553	99403	96055									
500	18	346235	99417	96007									
600	18	346763	99432	96073									
700	18	346248	99452	96054									
800	18	346763	99468	96073									
900	18	346248	99481	96054									
1000	18	346763	99493	96073									
1100	18	346248	99501	96054									
1200	18	346763	99514	96073									
1300	18	346248	99527	96054									
1400	18	346763	99540	96073									
1500	18	346248	99553	96054									
1600	18	346763	99566	96073									
1700	18	346248	99579	96054									
1800	18	346763	99592	96073									
1900	18	346248	99605	96054									
2000	18	346763	99618	96073									
2100	18	346248	99631	96054									
2200	18	346763	99644	96073									
2300	18	346248	99657	96054									
2400	18	346763	99670	96073									

(Continued)

TABLE E-2. (Continued)

DULUTH BOILER HOURLY LOG

DULUTH BOILER HOURLY LOG									
TIME	BOILER NO. 1 DATA				BOILER NO. 2 DATA				BOILER NO. 3 DATA (CON. LINE TIME, OFF LINE TIME, ETC.)
	GAS PRESS.	STEAM	COND.	GAS	OIL	GAS PRESS.	STEAM	COND.	
1911									
100	18	57668	9727	422782					
200	18	57607	9750	423351					
300	18	57652	9766	424003					
400	18	57706	9782	424740					
500	18	57750	9798	425406					
600	18	57795	9815	426084					
700	18	57840	9831	426786					
800	18	57885	9847	427455					
900	18	57930	9863	428120					
1000	20	57975	9879	428792					
1100	20	58020	9895	429462					
1200	19	58065	9911	430130					
1300	19	58110	9927	430806					
1400	19	58155	9943	431481					
1500	19	58200	9959	432156					
1600	19	58245	9975	432831					
1700	19	58290	9991	433506					
1800	19	58335	10007	434181					
1900	19	58380	10023	434856					
2000	31	58425	10039	435531					
2100	19	58470	10055	436206					
2200	31	58515	10071	436881					
2300	24	58560	10087	437556					
2400	24	58605	10103	438231					

(Continued)

TABLE F-2. (Continued)

TIME	DULUTH BOILER HOURLY LOG										COMMENTS	
	BOILER NO. 1 DATA					BOILER NO. 2 DATA					(ON LINE TIME, OFF LINE TIME, ETC.)	
	GAS PRESS	STEAM	COND.	GAS	OIL	GAS PRESS	STEAM	COND.	GAS	OIL		
100	24	385.782	95070	43495								
200	24	386.047	95079	43735								
300	22	386.546	95087	43768								
400	22	386.656	95096	43804								
500	22	386.971	95106	43848								
600	22	387.279	95115	43886								
700	22	387.587	95124	43920								
800	22	387.891	95133	43954								
900	22	388.195	95142	43988								
1000	22	388.499	95150	44022								
1100	22	388.803	95159	44056								
1200	22	389.107	95168	44090								
1300	22	389.411	95177	44124								
1400	22	389.715	95186	44158								
1500	22	389.930	95192	44200								
1600	22	390.185	95199	44254								
1700	22	390.439	95206	44307								
1800	22	390.724	95215	44333								
1900	22	390.990	95223	44372								
2000	22	391.152	95232	44409								
2100	22	391.440	95236	44451								
2200	22	391.627	95241	44483								
2300	22	391.978	95248	44525								
2400	22	392.070	95254	44577								

(Continued)

TABLE E-2. (Continued)

DULUTH BOILER HOURLY LOG															COMMENTS (ON LINE TIME, OFF LINE TIME, ETC.)	
BOILER NO. 1 DATA					BOILER NO. 2 DATA					BOILER NO. 3 DATA (ON TANK)						
GAS PRESS.	STEAM	COND.	GAS	OIL	GAS PRESS.	STEAM	COND.	GAS	OIL	STEAM	CONDENSATE	(OIL) TANK				
19 81																
100	22	952511	95261	949595						555107	43548	1486680				
200	22	972553	95268	946208						555490	43571	1486910				
300	22	972591	95276	946235						555866	43594	1487120				
400	22	973109	95284	947237						556232	43616	1487340				
500	22	973380	95292	947204						556606	43639	1487560				
600	22	973655	95300	948156						557009	43664	1487800				
700		172021	95310	948156						557237	43682	1487990				
800										557710	43710	1488230				
900										558193	43732	1488510				
1000										558697	43767	1488810				
1100										559056	43788	1489020				
1200										559402	43804	1489220				
1300										560103	43836	1489480				
1400										560255	43861	1489740				
1500										560677	43887	1490000				
1600										561101	43910	1490250				
1700										561505	43933	1490500				
1800										561920	43951	1490690				
1900										562225	43971	1490840				
2000										562552	43994	1491110				
2100																
2200																
2300																
2400																

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TABLE E-3. GAS STREAMS FLOW RATES

	grams/ second	Nm ³ /min. (25°C, latm)	SCF/min. (60°F, latm)	Molecular Weight (dry)	Moisture Content (%)
Combined Product Gas	930	57	2000	23.8	9.6
Cleaned Top Product Gas	560	33	1200	24.4	19.2
Raw Top Product Gas	530	33	1100	23.9	19.3
Top Product Gas	540	33	1100	24.2	19.2
Cleaned Bottom Product Gas	460	26	890	25.8	4.2
Raw Bottom Product Gas	670	38	1300	26.2	4.5
Bottom Product Gas	560	32	1100	26.0	4.4
Product Gas Boilers Combustion Emissions	5600	280	9400	29.8	9.2
Boiler No. 3 Firing ESP Oils Combustion Emissions	5200	250	8700	30.1	7.0
Boiler No. 3 Firing Natural Gas Combustion Emissions	3700	180	6200	29.8	15.0

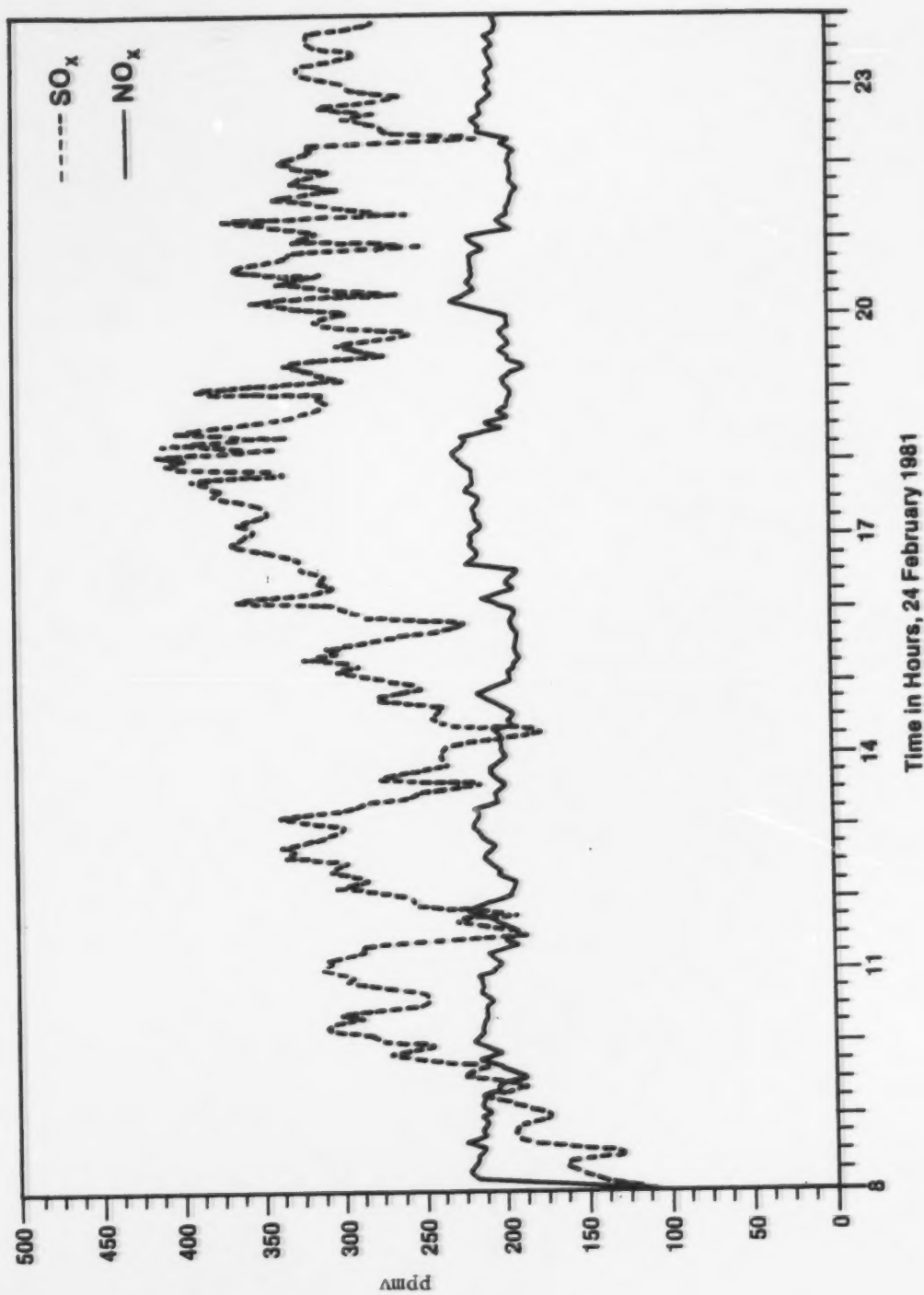


Figure E-1. NO_x and SO_x emissions from product gas boilers.

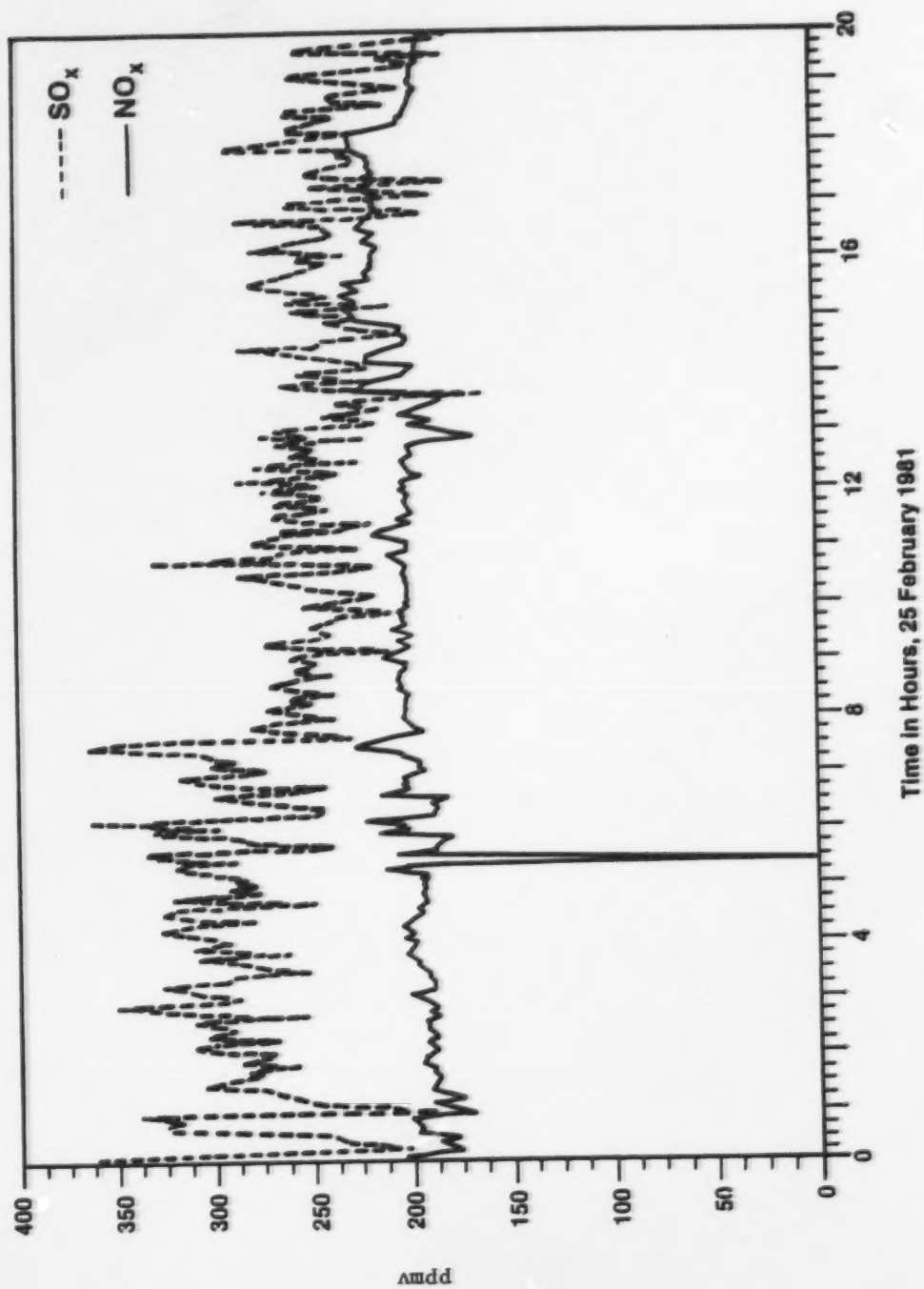
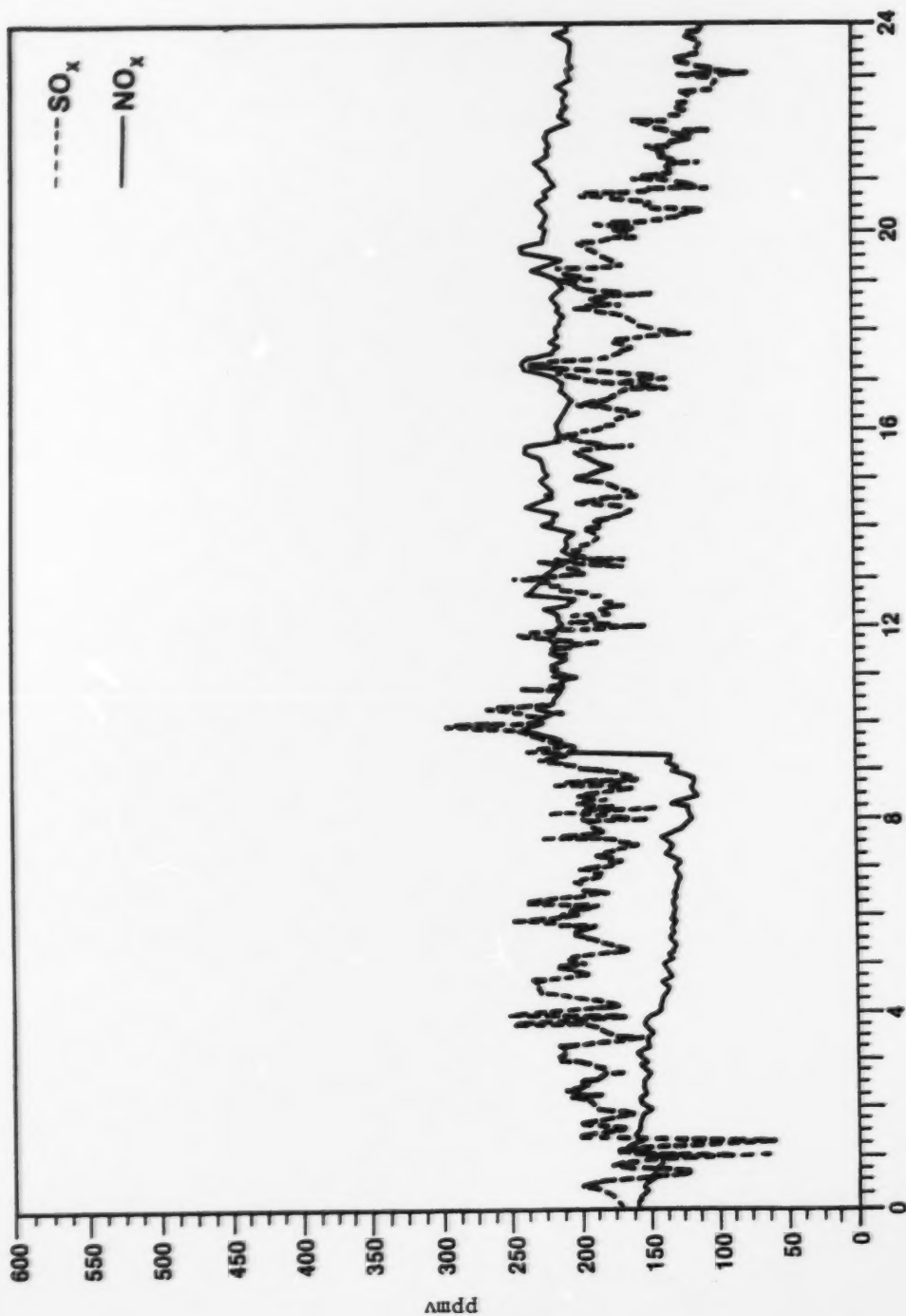


Figure E-2. NO_x and SO_x emissions from product gas boilers.



Time in Hours, 26 February 1981
Figure E-3. NO_x and SO_x emissions from product gas boilers.

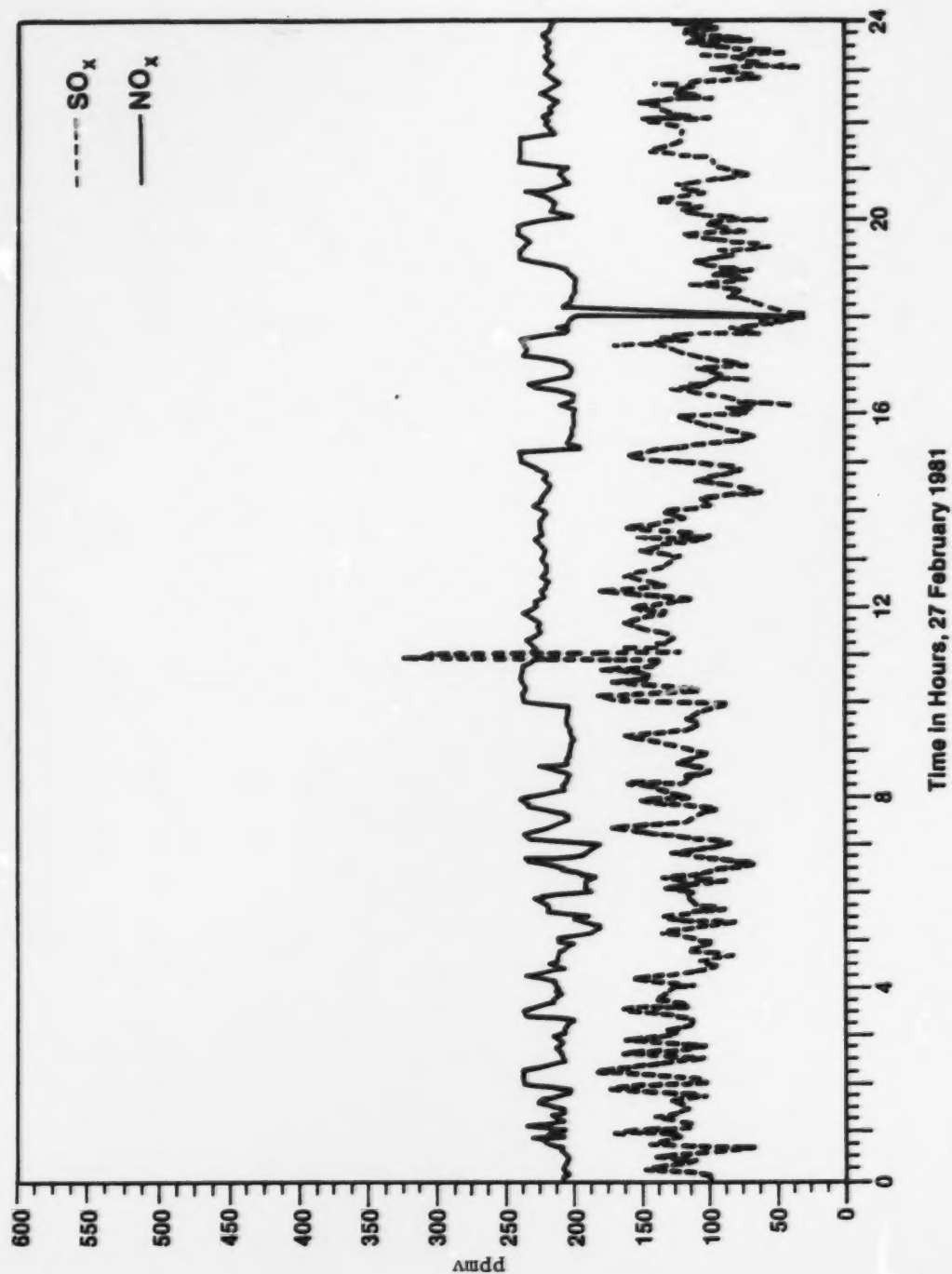


Figure E-4. NO_x and SO_x emissions from product gas boilers.

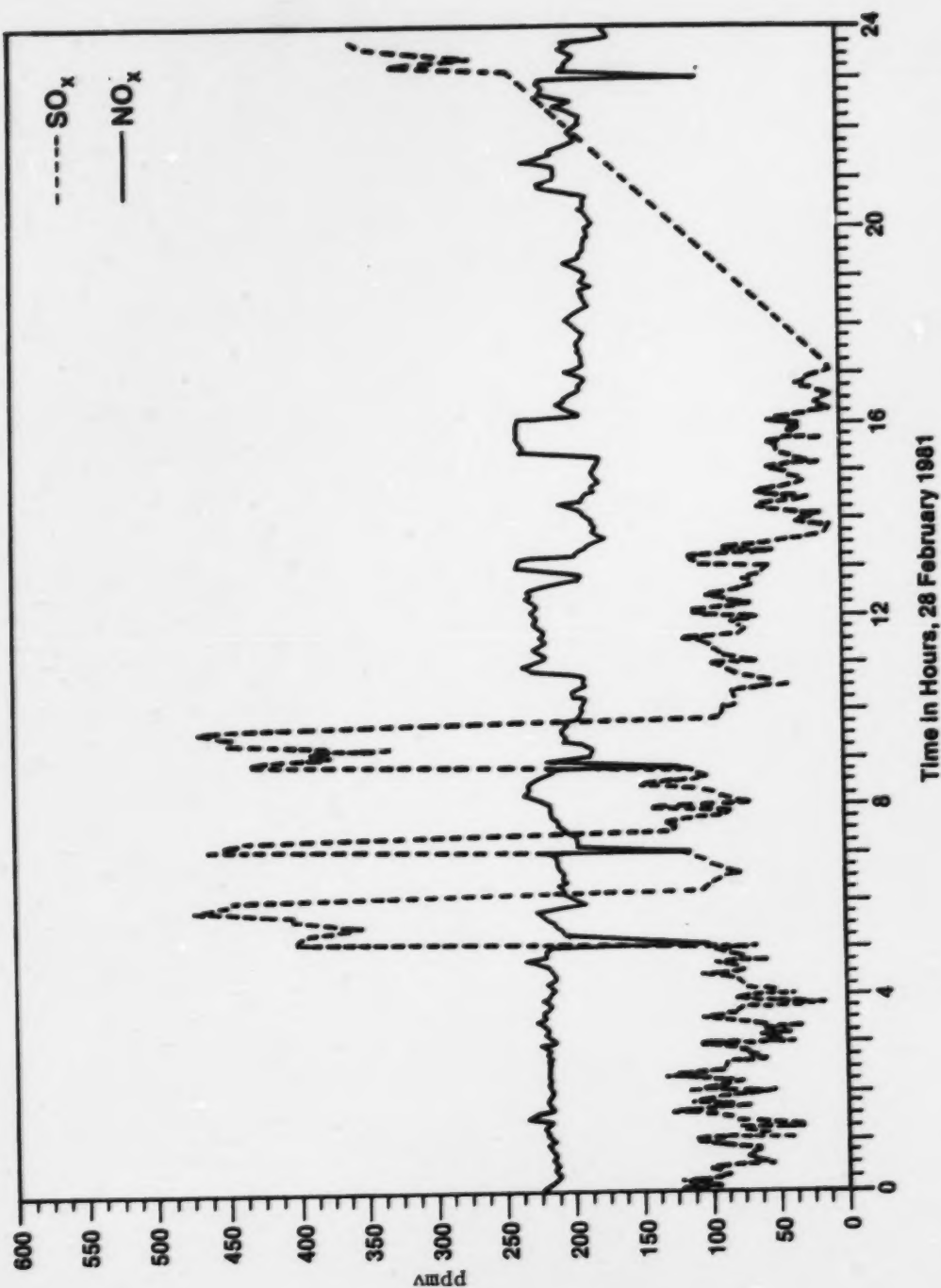


Figure E-5. NO_x and SO_x emissions from product gas boilers.

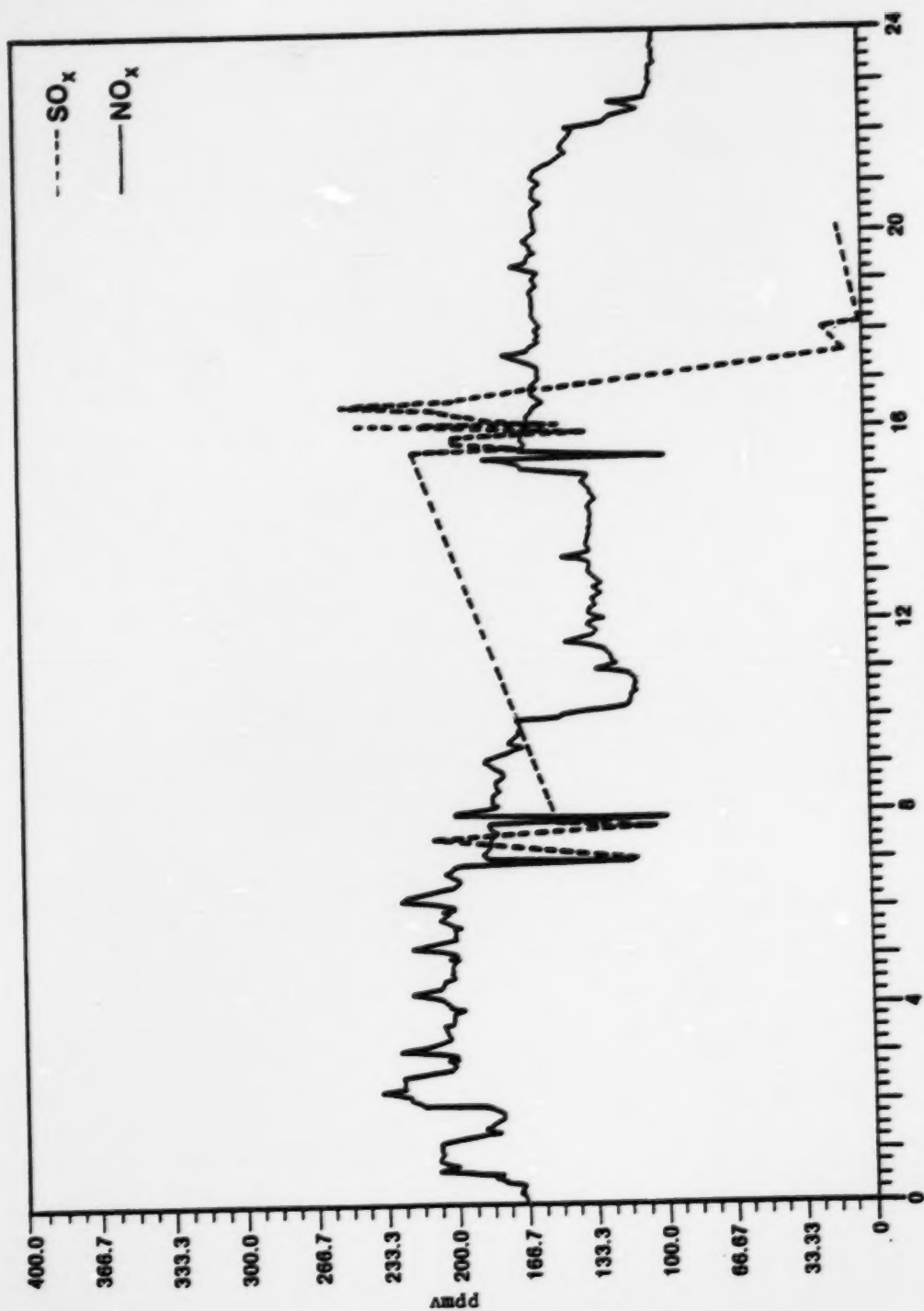
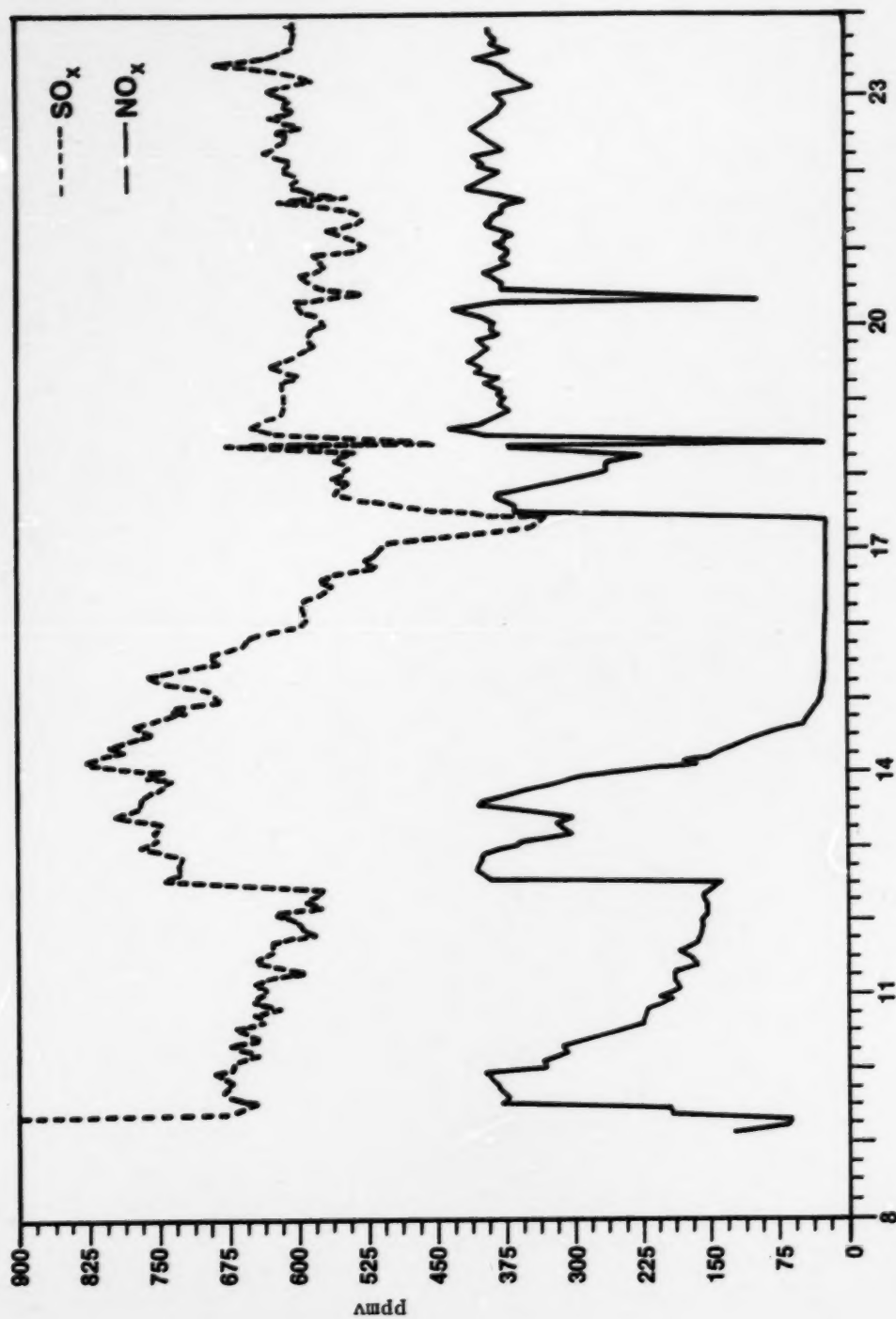


Figure E-6. NO_x and SO_x emissions from product gas boilers.



Time in Hours, 2 March 1981

Figure E-7. NO_x and SO_x emissions from product gas boilers.

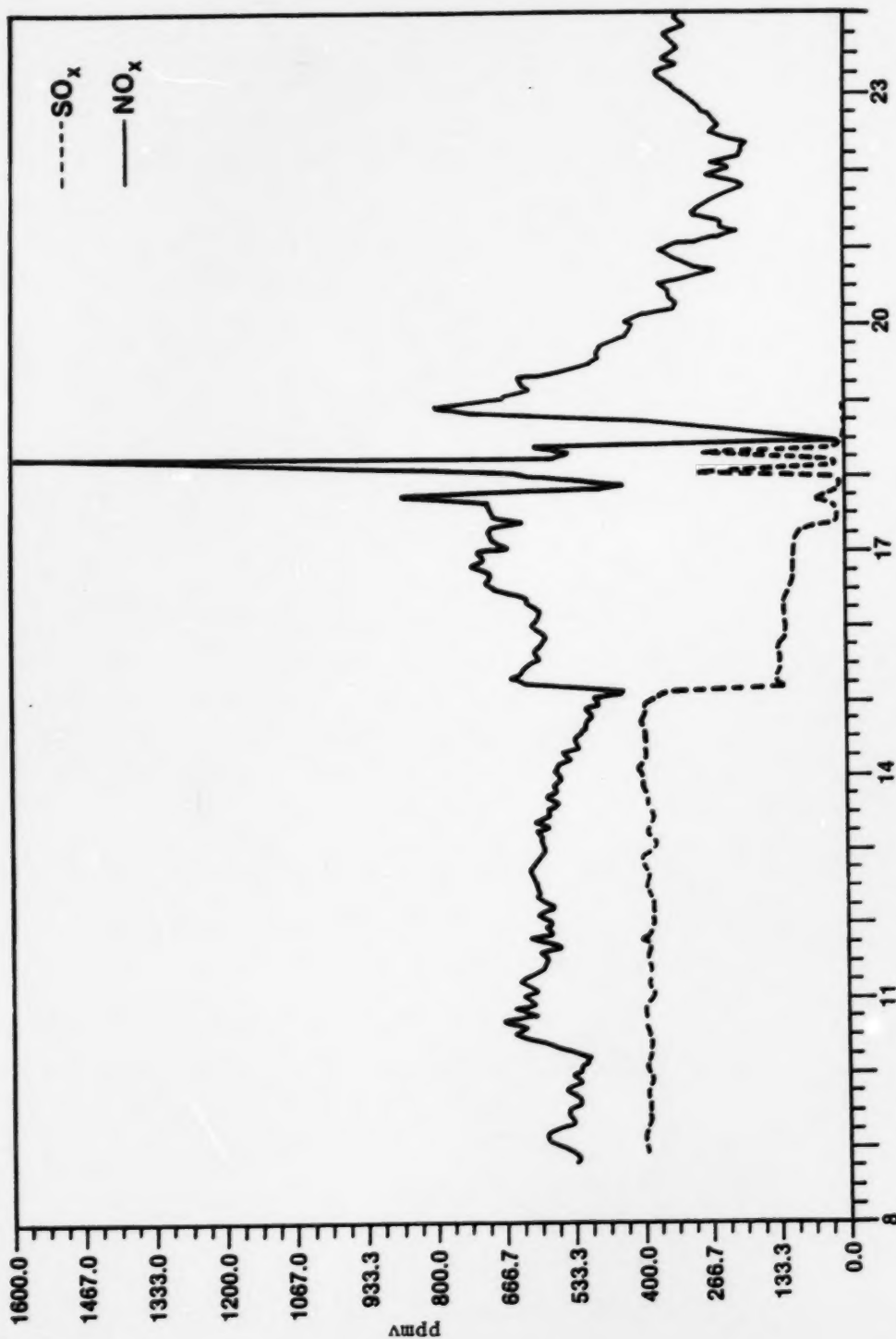


Figure E-8. NO_x and SO_x emissions from product gas boilers.